Immobilization of the *Candida rugosa* lipase onto a *Scirpus grossus* L.f. fiber as biocatalyst for biodiesel synthesis via hydrolysis-esterification

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This study aimed to immobilize the *Candida rugosa* lipase (*C. rugosa* lipase) on ground *Scirpus grossus* L.f. fibers by glutaraldehyde-crosslinking to form a hydrolysis-esterification catalyst for biodiesel synthesis. The effects of different glutaraldehyde concentrations and solvent for 3-aminopropyltriethoxysilane (3-APTES) activation of the fibers on the resultant immobilized lipase activity, protein loading, degree of immobilization and catalytic efficiency were investigated. The optimum condition found was to first activate the *S. grossus* L.f. fibers using 2% (w/v) of 3-APTES in distilled water and subsequently crosslink with 0.2% (w/v) glutaraldehyde prior to *C. rugosa* lipase immobilization at pH 7. The immobilized *C. rugosa* lipase was then evaluated as a biocatalyst for biodiesel synthesis via the hydrolysis-esterification of palm oil and bioethanol through monitoring the production of free fatty acids (FFAs) and fatty acid ethyl ester (FAEE, biodiesel). The reusability of immobilized lipase was also determined. The immobilized *C. rugosa* lipase yielded a higher hydrolysis-esterification efficiency (that is, FFA and FAEE formation) than that of the free lipase with the immobilized form looking promising for FAEE biodiesel production. The *C. rugosa* lipase immobilized with 0.2% (w/v) glutaraldehyde exhibited the highest reuse stability, retaining some hydrolysis and esterification activity for up to six uses, whereas crosslinking with higher [0.5% or 0.8% (w/v)] glutaraldehyde levels resulted in a loss of both activities within four uses.

Key words: Immobilization, lipase, *Scirpus grossus* L.f. fiber, 3-aminopropyltriethoxysilane, glutaraldehyde, biodiesel.

INTRODUCTION

Currently, the limited petroleum-based oil reserves, the increase of oil price due to the decrease of crude oil production because of the political problem of the middle-east countries, and the environmental concerns from CO₂ emission have urged many researchers to develop alternative renewable energy sources. Biodiesel is viewed as one of the new potential alternative renewable, non-toxic, biodegradable and clean energy sources for...
petroleum-based fuel replacement (Gerpen, 2005). It is claimed as a nonpolluting fuel because the burning of biodiesel does not contribute to global warming due to its large reduction of CO₂ emission compared with petroleum-based oil. Although, biodiesel can be successfully produced by chemical catalysis, this is a high energy consumption process and the recovery of the biodiesel from the glycerol byproduct is difficult. Moreover, this process also produces large amounts (ca. 10% of the volume of biodiesel produced) of non-environmentally friendly wastewater that is difficult to treat and also expensive. Recently, the use of lipase as biocatalyst instead of acid or alkali catalysts has attracted considerable attention because byproduct glycerol can be recovered easily and the purification of the biodiesel (fatty acid methyl esters, FAME) is simple to accomplish. In addition, this process is non-toxic, has a lower energy requirement, and also has biodegradable characteristics (Shimada et al., 1999; Nelson et al., 1996). However, the high cost of lipase makes enzymatic processes commercially unattractive. Therefore, immobilization of lipase has advantages over free lipase in terms of its reusability, its ease of removal from the reaction mixture and its adaptability to various engineering designs (Xie and Ma, 2009). Nevertheless, the high cost of comercial immobilized lipase due in part to the requirement for expensive support materials, makes production of biodiesel by this approach more expensive than that produced by chemical catalysis and so it is not economically viable. Thus, cheaper supports such as clays, silica gels and glasses have been evaluated as alternative adsorbents to the expensive polymeric resins (Hung et al., 2003; Minovska et al., 2005; Lee et al., 2006; Sugunan et al., 2007).

Other attractive supports are plant fibers that are composed of cellulose because of their accessibility, cheapness, hydrophilic nature and the great number of hydroxyl groups on the surface that are capable of chemical reaction (Przybyt and Sugier, 1988). In addition, most of them possess a large number of pores, allowing the lipase to be immobilized onto the plant fibers through simple adsorption. Scirpus grossus L.f. or Cyperus papyrus L., is an abundant biomass cellulose fiber in Thailand. Due to its very low cost, lack of toxicity and its chemical reactivity that allows it to react with the enzyme easily, S. grossus L.f. fibers are viewed as a non-toxic, economic and environmental friendly support for lipase immobilization. Lipase immobilization onto S. grossus L.f. fibers by physical adsorption is simple but appears to be that they cannot maintain their active configuration and so lose catalytic activity with time for biodiesel production via transesterification. Lee et al. (2006) developed a suitable method for the immobilization of Rhizopus oryzae lipase on silica gel using glutaraldehyde as a crosslinking agent, reporting on the optimizing of each step of the immobilization procedures, such as the pretreatment of silica gel, silanization, crosslinking and glutaraldehyde modification.

Silanization on the silica gel surface using 3-aminopropyltriethoxysilane (3-APTES) exhibited the highest resultant catalytic activity of the immobilized lipase. Furthermore, after 20 consecutive uses in a 30 min reaction with repeated washing between each reaction, some 80% of the enzyme activity of the immobilized lipase was still retained. Wang et al. (2008) reported that the immobilization of Candida rugosa lipase (C. rugosa lipase) onto the dense surface of a polysulfone ultrafiltration membrane by filtration and then crosslinking with glutaraldehyde solution greatly improved the lifetime of the enzyme. The apparent reaction rate and stability of the immobilized lipase were greatly improved after crosslinking. Xie and Ma (2009) reported that immobilized lipase on Fe₃O₄ nanoparticles treated with 3-APTES was using as biocatalyst for biodiesel production via transesterification. The conversion of soybean oil to biodiesel fuels reached over 90% by the three-step addition of methanol when 60% immobilized lipase was employed. In addition, it could be used four times without significant decrease of activity. Thus, to reduce the hydrophilicity of S. grossus L.f. and enhance the stability and retain a high catalytic activity of the immobilized lipase, it is necessary to modify or activate the surface properties of cellulose fibers enriched with hydroxyl groups using a silane coupling agent and subsequent crosslinking of the lipase to the support material using glutaraldehyde.

In this study, glutaraldehyde-crosslinking based immobilization of the C. rugosa lipase onto 3-APTES activated ground S. grossus L.f. fiber was investigated for its suitability for biodiesel synthesis. Protein loading, lipase activity and degree of immobilization of immobilized the C. Rugosa lipase were subsequently determined. The catalytic activities of the immobilized C. rugosa lipase in a hydrolysis-esterification reaction using palm oil and bioethanol for FAME biodiesel synthesis was also examined to ascertain its biocatalyst potential including the reusability of the immobilized C. rugosa lipase.

**MATERIALS AND METHODS**

C. rugosa lipase powder (EC 3.1.1.3) was purchased from Sigma-Aldrich (USA). Plant fibers of S. grossus L.f. from Nakhon Ratchasima Province (Thailand) were used as the support. They were dried and ground into fine powder using a Pulverizer model T15 before use. 3-APTES used for the activation of hydroxyl group of S. grossus L.f. and glutaraldehyde (50% in water) used as a crosslinking agent, were purchased from Fluka (Switzerland). Palm oil was obtained from Morakot Industries (Thailand) and used as the substrate triglyceride. Bioethanol was kindly provided by the Petroleum Authority of Thailand.

**C. rugosa lipase immobilization**

The procedure for C. rugosa lipase immobilization was composed
of two main steps. The first step was to activate the ground fiber using 3-APTES solution in either distilled water (H treatment) or in acetone (A treatment). The second step was to immobilize the *C. rugosa* lipase before crosslinking with glutaraldehyde (A1 and H1 treatments) or to simultaneously immobilize and crosslink the lipase onto the fibers with glutaraldehyde (A2 and H2 treatments), as detailed below. The flow diagram of the *C. rugosa* lipase immobilization is shown in Figure 1.

**Activation of *C. rugosa* L.f. fibers with 3-APTES in distilled water (H treatments) or in acetone (A treatments)**

Five grams of ground *S. grossus* L.f. fibers was mixed with 100 ml of 2% (w/v) of 3-APTES solution in either distilled water or in acetone and stirred at 360 rpm for 30 min. The activated ground fiber was then washed with distilled water three times and left to dry at 80°C for 6 h.

**Lipase immobilization using glutaraldehyde as a crosslinking agent**

**Procedure 1** (immobilization after crosslinking): One gram of the H or A activated ground *S. grossus* L.f. fibers crosslinked with 20 ml of 0.5% (w/v) glutaraldehyde in 0.05 M phosphate buffer (pH 7). The reaction mixture was stirred at 360 rpm for 1 h at room temperature and then washed with distilled water three times, vacuum filtered, left to dry and kept in a desiccator. Thereafter, 0.5 g of the crosslinked-ground fiber was mixed with 10 ml of a 0.01 g/ml *C. rugosa* lipase solution in 1 M phosphate buffer (pH 7) and stirred at 360 rpm for 12 h at room temperature. The immobilized lipase was then washed six times, each with 10 ml of 0.05 M phosphate buffer solution (pH 7), vacuum filtered and left to dry in a desiccator for overnight prior to storage at 4°C.

**Procedure 2** (simultaneous immobilization and crosslinking): To 0.5 g of A or H treated ground, *S. grossus* L.f. fibers was mixed with 10 ml of 0.01 g/ml *C. rugosa* lipase solution in 1 M phosphate buffer (pH 7), and 100 µl of glutaraldehyde [at 0.2, 0.5 or 0.8% (w/v)] was then added and stirred at 360 rpm for 12 h at room temperature. The immobilized lipase was then washed six times, each with 10 ml of 0.05 M phosphate buffer (pH 7), vacuum filtered, dried and kept in a desiccator overnight and stored at 4°C.

**Morphology of immobilized lipase**

Scanning Electron Microscopy (SEM), using a Jeol JSM-5800 LV electron microscope (Japan) under an accelerated voltage of 15 KV, was performed at a magnification of 750x to examine the morphology of the ground *S. grossus* L.f. fibers and the adsorption of the *C. rugosa* lipase to the fiber support after immobilization.

**Determination of *C. rugosa* lipase activity**

The activity of the free and immobilized lipase was determined using 0.5% (w/v) *p*-nitrophenyl palmitate (*p*-NPP) in ethanol as the substrate. The increase in absorbance at 410 nm caused by the
release of p-nitrophenol, was measured with a UV spectrophotometer (ANTHOS Zenyth 200 microplate spectrophotometer). One unit (U) of lipase activity was defined as the amount of enzyme required to hydrolyze 1 nmol/min of p-NPP under the experimental conditions.

**Determination of protein loading**

Protein concentration was spectrophotometrically determined with a UV spectrophotometer (ANTHOS Zenyth 200 Microplate Spectrophotometer) according to the Bradford method using bovine serum albumin (BSA) as the standard (Bradford, 1976). The protein loading level is shown as the amount of lipase incorporated in the fibers as a percentage of the total amount added.

**Determination of immobilization efficiency**

The efficiency of each immobilization technique, \( \eta \), was estimated by calculating the activity of the lipase solution before (\( E_0 \)) and after (\( E_i \)) immobilization, using Equation 1:

\[
\eta = \frac{E_0 - E_i}{E_0}
\]
\[
\eta = \frac{E_0V_0 - EV_f}{E_0V_0} \times 100
\]

Where, \(V_0\) is the initial volume of lipase solution (ml) and \(V_f\) is the total volume of lipase solution after filtration (ml). The activities are given in U/ml.

**Biodiesel synthesis**

In this study, the catalytic activities of the immobilized \(C.\ rugosa\) lipase for the hydrolysis-esterification reaction of palm oil and bioethanol as substrates for FAEE biodiesel synthesis were studied through the monitoring of the free fatty acid (FFA) and FAEE production levels as follows:

**Hydrolysis of palm oil**

Palm oil was first hydrolyzed by mixing 3 g of palm oil and 10\% (of the weight of palm oil) of the immobilized \(C.\ rugosa\) lipase suspension or the free \(C.\ rugosa\) lipase solution. Then, 300 \(\mu\)l of distilled water was added to the mixture and continuously stirred at 600 rpm for 12 h at 50°C. From this, 100 \(\mu\)l of the hydrolitic product was removed and centrifuged at 13,000 rpm for 30 min. The upper phase was used to quantify the FFA composition by high performance liquid chromatography (HPLC) with a Shimadzu LC-20A series (Japan).

**Esterification**

The stepwise ethanolysis of palm oil was conducted using a 1:3 molar ratio of palm oil/bioethanol as this ratio has been reported to be optimal (Shimada et al., 1999). However, the bioethanol was added to the hydrolyzed palm oil in three stages feeding of bioethanol at each of 0, 8 and 16 h into palm oil and stirring at 600 rpm for 24 h at 50°C. At completion, 100 \(\mu\)l of the product was removed and centrifuged at 13,000 rpm for 30 min. The upper phase was used to quantify the FFA and FAEE composition by HPLC as described earlier.

**Evaluation of the reusability of immobilized \(C.\ rugosa\) lipase**

After being used as a catalyst for the hydrolysis-esterification, the immobilized \(C.\ rugosa\) lipase was filtered, washed with 20 ml heptane and store in a desiccator for 12 h prior to reuse for the next batch reaction. The conversion of palm oil / bioethanol to FFA and FAEE at the end of the reaction was evaluated by HPLC as described earlier. From these results, the immobilized \(C.\ rugosa\) lipase prepared by first activating the \(S.\ grossus\) L.f. fibers with 3-APTES (in distilled water) and then immobilization of the lipase after crosslinking with glutaraldehyde (treatment H1; Table 1). The highest lipase activity, protein loading and degree of immobilization of the immobilized lipase was achieved by first activating the \(S.\ grossus\) L.f. fibers with 3-APTES (in distilled water) and then immobilization of the lipase after crosslinking with glutaraldehyde (treatment H1; Table 1). However, with respect to the catalytic efficiency, defined as the ratio between the lipase activity and the protein loading level, this method gave a marginally lower catalytic efficiency than when 3-APTES activated in acetone (A1 in Table 1). The higher the catalytic efficiency, the lower the amount of lipase that has been inactivated by the immobilization (Salis et al., 2008).

**RESULTS AND DISCUSSION**

**Morphology of the immobilized \(C.\ rugosa\) lipase on \(S.\ grossus\) L.f. fibers**

Representative SEM images of the free lipase, ground \(S.\ grossus\) L.f. fibers and the 0.5\% (w/v) glutaraldehyde crosslinked and immobilized \(C.\ rugosa\) lipase on \(S.\ grossus\) L.f. fibers prepared by different procedures are shown in Figure 2. The presence of lipase particles was seen on the surface and into the pores of the fiber support prepared by all procedures. The activation of the ground \(S.\ grossus\) L.f. fibers with 3-APTES in distilled water and subsequently immobilizing the \(C.\ rugosa\) lipase after crosslinking with glutaraldehyde (treatment
The effects of varying the glutaraldehyde concentration on the resulting immobilized lipase activity, protein loading, degree of immobilization and catalytic efficiency of the immobilized *C. rugosa* lipase. The immobilized *C. rugosa* lipase was prepared by activation of *S. grossus* L.f. fibers with 3-APTES in distilled water and immobilized after glutaraldehyde crosslinking (treatment H1).

<table>
<thead>
<tr>
<th>Glutaraldehyde concentrations [% (w/v)]</th>
<th>Lipase activity (U/g-fiber)</th>
<th>Protein loading (%)</th>
<th>Degree of immobilization (%)</th>
<th>Catalytic efficiencya</th>
</tr>
</thead>
<tbody>
<tr>
<td>0.2</td>
<td>3.01</td>
<td>80.4</td>
<td>76.6</td>
<td>0.037</td>
</tr>
<tr>
<td>0.5</td>
<td>3.31</td>
<td>79.6</td>
<td>84.2</td>
<td>0.042</td>
</tr>
<tr>
<td>0.8</td>
<td>2.93</td>
<td>77.7</td>
<td>74.5</td>
<td>0.038</td>
</tr>
</tbody>
</table>

*aCatalytic efficiency is the ratio of the lipase activity to the protein loading level.*

Effects of pH on the degree of *C. rugosa* lipase immobilization and hydrolysis-esterification catalytic efficiency

The effects of pH on the resultant lipase activity, protein loading, degree of immobilization and catalytic efficiency of the immobilized *C. rugosa* lipase were shown in Table 3. Immobilized *C. rugosa* lipase was prepared by activation of *S. grossus* L.f. fibers with 3-APTES in distilled water crosslinking with 0.2% (w/v) glutaraldehyde and then the lipase was immobilized (treatment H1) at pH 5, 7 and 9. The highest lipase activity, degree of immobilization were achieved at pH 7, whereas, the catalytic efficiency at pH 5 and 7 were similar. Thus, the pH optimum for the *C. rugosa* lipase immobilization was pH 7 or mild pH 5 rather than pH 9. This may be the bond between lipase and the support is unstable in basic reaction media (pH > 7) and stable in acid and neutral media (pH 5 and 7).

Production of FFAs and FAEE biodiesel from palm oil and bioethanol

The % conversion of palm oil to FFA (hydrolysis activity) and subsequently with the bioethanol to FAEE (esterification activity) is summarized in Table 4. The three immobilized *C. rugosa* lipase preparations all yielded a higher FFA and especially and importantly,
FAEE production level than that of the free *C. rugosa* lipase. The immobilized *C. rugosa* lipase could catalyze the hydrolysis and esterification reaction relatively efficiently. Minovska et al. (2005) reported previously that the free enzyme must be used at three-times higher quantities than the immobilized lipase, whilst the free lipase may lose its enzymatic activity under severe environments and with a lower surface area to catalyze the reaction. Of the immobilized *C. rugosa* lipase preparations tested here, that produced by crosslinking with 0.2% (w/v) glutaraldehyde provided the highest levels of FFA and FAEE, and so the highest hydrolysis-esterification catalytic activity. Too high concentrations of glutaraldehyde [0.8% (w/v)] gave lower yields, presumably due to the strong covalent binding of the lipase and glutaraldehyde multiple sites causing the enzyme denaturation and loss of some active sites of the lipase.

**Table 3.** Effects of pH on lipase activity, protein loading, degree of immobilization and catalytic efficiency of the immobilized *C. rugosa* lipase. The immobilized *C. rugosa* lipase was prepared by activation of *S. grossus* L.f. fibers with 3-APTES in distilled water and immobilized after crosslinking with 0.2% (w/v) glutaraldehyde (treatment H1).

<table>
<thead>
<tr>
<th>pH</th>
<th>Lipase activity (U/g-fiber)</th>
<th>Protein loading (%)</th>
<th>Degree of immobilization (%)</th>
<th>Catalytic efficiencya</th>
</tr>
</thead>
<tbody>
<tr>
<td>5</td>
<td>2.77</td>
<td>72.7</td>
<td>76.6</td>
<td>0.038</td>
</tr>
<tr>
<td>7</td>
<td>3.01</td>
<td>80.4</td>
<td>84.2</td>
<td>0.037</td>
</tr>
<tr>
<td>9</td>
<td>2.16</td>
<td>63.2</td>
<td>74.5</td>
<td>0.034</td>
</tr>
</tbody>
</table>

aCatalytic efficiency is the ratio of the lipase activity to the protein loading level.

**Table 4.** Conversion of free fatty acid (FFA) and fatty acid ethyl ester (FAEE; biodiesel) produced by the immobilized *C. rugosa* lipase from a 3:1 molar ratio of palm oil: bioethanol. The immobilized *C. rugosa* lipase was prepared by activation of *S. grossus* L.f. fibers with 3-APTES in distilled water and immobilized after glutaraldehyde crosslinking at various concentrations (treatment H1).

<table>
<thead>
<tr>
<th>Immobilized <em>C. rugosa</em> lipase [% (w/v) glutaraldehyde]</th>
<th>Free fatty acid (FFA) (%)</th>
<th>Fatty acid ethyl ester (FAEE) (%)</th>
</tr>
</thead>
<tbody>
<tr>
<td>Free lipase</td>
<td>41.6</td>
<td>14.7</td>
</tr>
<tr>
<td>0.2</td>
<td>52.6</td>
<td>49.6</td>
</tr>
<tr>
<td>0.5</td>
<td>43.9</td>
<td>32.2</td>
</tr>
<tr>
<td>0.8</td>
<td>43.5</td>
<td>23.7</td>
</tr>
</tbody>
</table>

**Reusability of immobilized enzyme for biodiesel synthesis**

The stability of the immobilized lipase is important for any subsequent biodiesel synthesis. Therefore, the long-term stability of the prepared immobilized lipase after biodiesel synthesis was evaluated. The immobilized lipase activity declined after each use for both the hydrolysis (FFA production; Figure 3A) and esterification (FAEE production; Figure 3B) catalytic activities. However, note that the esterification activity is dependent upon the availability of FFA and so a reduction in the hydrolysis efficiency may be manifest as an apparent but false reduction in the esterification catalytic efficiency when FFA levels are low.

The lipase immobilized with 0.2% (w/v) glutaraldehyde exhibited the highest reuse stability with some 50% residual FFA and FAEE levels after three or one reuse, respectively, and with some hydrolysis-esterification activity remaining at up to six reuses (Figure 3B). In contrast, the lipase immobilized with 0.8% (w/v) glutaraldehyde exhibited a loss of lipase activity with four uses.

**Conclusions**

Optimal immobilization of the *C. rugosa* lipase on the ground biomass of *S. grossus* L.f. fibers was achieved by prior activation of the fibers with 2% (w/v) 3-APTES in distilled water followed by crosslinking with 0.2% (w/v) glutaraldehyde prior to immobilization of the *C. rugosa* lipase. A higher concentrations of glutaraldehyde is less efficient presumably because the unreacted aldehyde groups of glutaraldehyde remain after crosslinking with an amino group of the lipase and so can attach to other amino groups of the lipase, thereby reducing the lipase's activity.

In terms of FAEE biodiesel synthesis, the immobilized *C. rugosa* lipase-catalyzed hydrolysis-esterification of palm oil and bioethanol yielded a higher release of FFAs and conversion to FAEE than those of the free *C. rugosa*. With the 0.2% (w/v) glutaraldehyde immobilized *C. rugosa* lipase also exhibited moderate reuse ability for up to six reuses.
Figure 3. The % production of (A) free fatty acids (FFAs) and (B) fatty acid ethyl esters (FAEE; biodiesel) after repeated use of the immobilized \textit{C. rugosa} lipase on \textit{S. grossus} L.f. fibers crosslinked with various glutaraldehyde (GA) concentrations [as % (w/v)].

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