## SHORT COMMUNICATION

# SYNTHESIS AND CRYSTAL STRUCTURE OF 1-[2-(3-ETHYL-2,2-DIMETHYLCYCLOBUTYL)ACETYL]-3-PHENYLTHIOUREA 

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#### Abstract

The title compound 1-[2-(3-ethyl-2,2-dimethylcyclobutyl)acetyl]-3-phenylthiourea has been synthesized and its crystal structure was studied. The crystal belongs to triclinic system, space group $P-1, a=$ $10.200(2) \AA, b=12.395(3) \AA, c=15.679(3) \AA, \alpha=92.99(2)^{\circ}, \beta=106.00(3)^{\circ}, \gamma=111.95(3)^{\circ}, V=1740.4(6) \AA^{3}$, $Z=2, \mu=0.187 \mathrm{~mm}^{-1}, D_{\mathrm{c}}=1.162 \mathrm{~g} / \mathrm{cm}^{3}, F(000)=656, R=0.0784, w R 2=0.1505$, formula unit $\mathrm{C}_{17} \mathrm{H}_{24} \mathrm{~N}_{2} \mathrm{OS}$. The title compound has a fragment of 2,2-dimethylcyclobutane and its conformation represents semi-chair. The intermolecular and intramolecular hydrogen bonds are revealed.


KEY WORDS: Synthesis, Single crystal X-ray diffraction, Crystal structure, 2,2-Dimethylcyclobutane

## INTRODUCTION

Many thiourea compounds exhibit strong structure stabilization and interesting biological activities. So in recent years the synthesis and their biological activities have been studied. Many cyclobutyl analogues also have favorable bioactivities [1-3]. So the thiourea compounds containing cyclobutane ring may have interesting biological activities. In this paper we reported the synthesis and crystal structure of 1-[2-(3-ethyl-2,2-dimethylcyclobutyl)acetyl]-3-phenylthiourea. In the present contribution the synthesis of the title compound has been performed according to the scheme shown in Figure 1.


Figure 1. The synthesis scheme of the title compound
In order to determine the structure and configuration of the title compound X-ray diffraction study has been carried out.

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## EXPERIMENTAL

Synthesis. 2,2-Dimethyl-3-ethylcyclobutaneacetic acid was prepared in our laboratory according to literature [4]. The title compound was synthesized according to following method: 2,2-dimethyl-3-ethylcyclo-butaneacetic acid ( $5.0 \mathrm{~g}, 0.029 \mathrm{~mol}$ ) was dissolved in the mixture of dichloromethane ( 40 mL ) and $\mathrm{SOCl}_{2}(4.5 \mathrm{~mL}, 0.062 \mathrm{~mol})$, and the mixture was stirred and refluxed for 4 h . Then the excessive solution was removed by decompression distillation. The residue was added dropwise to the mixture of $\mathrm{KSCN}(3.0 \mathrm{~g}, 0.030 \mathrm{~mol})$ and acetonitrile ( 30 $\mathrm{mL})$. The resulting mixture was stirred for 4 h at room temperature. Aniline ( $2.70 \mathrm{~g}, 0.029 \mathrm{~mol}$ ) was added into the mixture. The reaction mixture was refluxed for 7 h . Then the excessive solution was removed by decompression distillation and the residue was poured into water (70 $\mathrm{mL})$. Finally pumping filtration left the crude product as yellow powder. The solid was purified by recrystallization with ethanol in $83.9 \%$ yield. Melting point: 99.0-100.0 ${ }^{\circ} \mathrm{C}$. $\mathrm{IR}(\mathrm{KBr}) v: \mathrm{cm}^{-1}$ $3191(\mathrm{~N}-\mathrm{H}), 3036(\mathrm{C}=\mathrm{H}), 1695(\mathrm{C}=\mathrm{O}), 1252(\mathrm{C}-\mathrm{N}), 1164(\mathrm{C}=\mathrm{S}) .{ }^{1} \mathrm{H}$ NMR ( $\left.\delta, \mathrm{ppm}, \mathrm{CDCl}_{3}\right)$ : $12.38(\mathrm{~s}, 1 \mathrm{H}, \mathrm{NH}), 9.05(\mathrm{~s}, 1 \mathrm{H}, \mathrm{NH}), 7.65\left(\mathrm{~d}, 2 \mathrm{H}, \mathrm{C}_{6} \mathrm{H}_{5}\right), 7.38\left(\mathrm{~m}, 2 \mathrm{H}, \mathrm{C}_{6} \mathrm{H}_{5}\right), 7.25(\mathrm{~m}, 1 \mathrm{H}$, $\mathrm{C}_{6} \mathrm{H}_{5}$ ), 2.51, $2.30\left(\mathrm{~m}, 2 \mathrm{H}, \mathrm{CH}_{2}\right), 2.40,2.18\left(\mathrm{~m}, 2 \mathrm{H}, \mathrm{CH}_{2}\right), 1.77\left(\mathrm{~m}, 2 \mathrm{H}, \mathrm{CH}_{2}\right), 1.35(\mathrm{~m}, 1 \mathrm{H}, \mathrm{CH})$, $1.20(\mathrm{~m}, 1 \mathrm{H}, \mathrm{CH}), 1.08-0.89\left(\mathrm{~m}, 6 \mathrm{H}, \mathrm{CH}_{3}\right), 0.80\left(\mathrm{q}, 3 \mathrm{H}, \mathrm{CH}_{3}\right)$.

X-ray diffraction study. Experimental X-ray diffraction data were obtained on a Enraf-Nonius CAD4 diffractometer (graphite-monochromated $\operatorname{Mo} K_{\alpha}$ radiation, $\lambda=0.71073 \AA$ ) by $\omega / 2 \theta$ scanning. Data were collected in the $\theta$ range of $1.4-25.3^{\circ}$ (range of indices: $0 \leq h \leq 12,-14 \leq k \leq 13$, $-18 \leq l \leq 18$ ). The structure was solved and refined with SHELX-97 software. All H atoms were placed geometrically, with the $\mathrm{C}-\mathrm{H}$ distances in the range $0.93-0.98 \AA$ and $\mathrm{N}-\mathrm{H}$ distances 0.86 $\AA$, and included in the refinement in riding motion approximation with $U_{\text {iso }}(\mathrm{H})=1.2$ or $1.5 U_{\text {eq }}(\mathrm{H})$ of the carrier atom. Selected crystallographic data and experiment parameters are listed in Table 1. Full crystallographic data for the compound of this study have been deposited with Cambridge Crystallographic Data Centre (www.ccdc.cam.ac.uk), CCDC deposition number 729335 .

Table 1. Selected crystallographic data and experiment parameters.

| Gross formula | $\mathrm{C}_{17} \mathrm{H}_{24} \mathrm{~N}_{2} \mathrm{OS}$ |
| :--- | :--- |
| Molecular weight | 304.44 |
| Temperature, K | $293(2)$ |
| Symmetry and space group | Triclinic, $P-1$ |
| $\mathrm{a}, \mathrm{b}, \mathrm{c}, \AA$ | $10.200(2), 12.395(3), 15.679(3)$ |
| $\alpha, \beta, \gamma, \mathrm{deg}$ | $92.99(2), 106.00(3), 111.95(3)$ |
| $\mathrm{V}, \AA^{3}$ | $1740.4(6)$ |
| Z | 2 |
| $\mathrm{P}_{\text {calc }}, \mathrm{g} / \mathrm{cm}^{3}$ | 1.162 |
| Crystal dimensions, $\mathrm{mm}^{\text {Absorption coefficient, } \mathrm{mm}^{-1}} \mathrm{0.30} \mathrm{\times 0.20} \mathrm{\times 0.10mm}$ |  |
| Measured/observed reflections | 0.187 |
| $R_{\text {int }}$ | $6718 / 6332$ |
| Refined parameters | 0.0310 |
| $R$ factor $[I>2 \sigma(I)]$ | 349 |
| $R$ factor $($ all data $)$ | $R 1=0.0784, w R 2=0.1505$ |
| Goof value | $R 1=0.1498, w R 2=0.1774$ |

## RESULTS AND DISCUSSION

The IR and ${ }^{1} \mathrm{H}$ NMR for the product are in good agreement with the title compound. In order to determine the structure and configuration of title compound X-ray diffraction study has been carried out. Figure 2 illustrates the structure of the title compound. As a whole, the molecule is substantially non-planar. The 2,2-dimethylcyclobutane fragment is not flat and the conformation represents semi-chair. The cyclobutane ring is flexed as though folded from the dimethylsubstituted C atom to the unsubstituted C atom, with a dihedral angle of $17.5^{\circ}$. This is a little different from other compounds containing cyclobutane rings. In ( $\pm$ )-2-[(1S,3S)-3-acetyl-2,2-dimethylcyclobutyl]- N -( $p$-tolyl)acetamide the dihedral angle is $23.7^{\circ}$ [5]; in ( $\pm$ )-cis-pinonic acid [6] and ( $1 S, 3 S$ )-(+)-cis-3-acetyl-2,2-dimethylcyclobutaneacetic acid [7] the dihedral angle is $29.8^{\circ}$; in $(+)$-trans-pinonic acid [8] the angle is $19.1^{\circ}$; in methyl $( \pm)-2-((1 R, 3 R)-3-\{2-[(3 S)-1-$ ethyl-3-hydroxy-2-oxo-2,3-dihydro-1 H -3-indolyl]acetyl $\}$-2,2-dimethylcyclobutyl) acetate [9] the angle is $18.6^{\circ}$; in (-)-cis-3-acetyl-2,2-dimethylcyclobutanecarboxylic acid [10] the angle is $25.5^{\circ}$. The selected bond lengths and bond angles are listed in Table 2 and 3.


Figure 2. View of the structure of the title compound. Ellipsoids are drawn at the $50 \%$ probability level.

Table 2. Selected bond lengths ( $\AA$ ).

| Bond | Length | Bond | Length | Bond | Length | Bond | Length |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| S1-C11 | $1.642(7)$ | O1-C10 | $1.222(8)$ | C7-H7C | 0.9600 | C8-H8A | 0.9600 |
| N1-C10 | $1.377(8)$ | N1-C11 | $1.399(8)$ | C8-H8B | 0.9600 | C8-H8C | 0.9600 |
| N1-H1A | 0.8600 | N2-C11 | $1.312(8)$ | C9-C10 | $1.512(9)$ | C9-H9A | 0.9700 |
| N2-C12 | $1.430(8)$ | N2-H2A | 0.8600 | C9-H9B | 0.9700 | C12-C17 | $1.370(10)$ |
| C1-C2 | $1.473(8)$ | C1-H1B | 0.9600 | C12-C13 | $1.373(9)$ | C13-C14 | $1.399(10)$ |
| C1-H1C | 0.9600 | C1-H1D | 0.9600 | C13-C18 | $1.551(10)$ | C14-C15 | $1.359(11)$ |
| C2-C3 | $1.450(7)$ | C2-H2B | 0.9700 | C14-H14A | 0.9300 | C15-C16 | $1.350(11)$ |
| C2-H2C | 0.9700 | C3-C6 | $1.461(10)$ | C6-C7 | $1.466(7)$ | C16-C17 | $1.392(10)$ |
| C3-C4 | $1.560(10)$ | C3-H3A | 0.9800 | N4-H4C | 0.8600 | C6-C8 | $1.533(11)$ |
| C4-C5 | $1.634(10)$ | C4-H4A | 0.9700 | C7-H7A | 0.9600 | C7-H7B | 0.9600 |
| C4-H4B | 0.9700 | C5-C6 | $1.469(10)$ | C5-C9 | $1.502(10)$ | C5-H5A | 0.9800 |

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Table 3. Selected bond angles and torsion angles.

| Angles | $\left({ }^{\circ}\right)$ | Angles | $\left({ }^{\circ}\right)$ | Angles | $\left({ }^{\circ}\right)$ |
| :---: | :---: | :---: | :---: | :---: | :---: |
| C10-N1-C11 | $129.5(6)$ | C10-N1-H1A | 115.2 | C3-C2-C1 | $116.6(8)$ |
| C11-N1-H1A | 115.2 | C11-N2-C12 | $128.1(6)$ | C1-C2-H2B | 108.1 |
| C11-N2-H2A | 115.9 | C12-N2-H2A | 115.9 | C3-C2-H2B | 108.1 |
| C2-C1-H1B | 109.5 | C2-C1-H1C | 109.5 | C3-C2-H2C | 108.1 |
| H1B-C1-H1C | 109.5 | C2-C1-H1D | 109.5 | C1-C2-H2C | 108.1 |
| H1B-C1-H1D | 109.5 | H1C-C1-H1D | 109.5 | H2B-C2-H2C | 107.3 |
| C1-C2-C3-C5 | $5.6(16)$ | C1-C2-C3-C6 | $-174.4(10)$ | C5-C4-C9-C10 | $-72.3(10)$ |
| C6-C4-C9-C10 | $172.1(6)$ | C6-C4-C5-C3 | $-11.8(7)$ | C6-C3-C5-C4 | $12.2(7)$ |
| H4A-C4-C5-H5B | -22 | H4A-C4-C6-C3 | -95 | H4A-C4-C6-C7 | 150 |
| H4A-C4-C6-C8, | 25 | C2-C3-C5-C4 | $-167.8(10)$ | C6-C3-C5-C4 | $12.2(7)$ |
| C2-C3-C6-C4 | $168.3(12)$ | C2-C3-C6-C7 | $-77.4(13)$ | C2-C3-C6-C8 | $49.6(16)$ |
| C5-C3-C6-C4 | $-11.7(6)$ | C5-C3-C6-C7 | $102.6(8)$ | C5-C3-C6-C8 | $-130.4(8)$ |

The C12 atom containing in the benzene ring, $\mathrm{S} 1, \mathrm{~N} 1, \mathrm{~N} 2$ and C 11 are in the same plane. The dihedral angel between the plane and the benzene ring is $49.9(2)^{\circ}$. There are two intramolecular hydrogen bonds and two intermolecular hydrogen bonds in the title compound which contributes to the stabilization of the crystal structure. The hydrogen bonds are shown in the Figure 3 and Table 4.


Figure 3. The hydrogen bonds of the title compound.
$\mathrm{N} 4, \mathrm{H} 4 \mathrm{~B}$ and O 2 form a intramolecular hydrogen bond and on account of the hydrogen bond $\mathrm{N} 4, \mathrm{C} 28, \mathrm{~N} 3, \mathrm{C} 27$ and O 2 form a 6 -membered ring. The $\mathrm{C}-\mathrm{N}$ bond length on the sides of $\mathrm{C} 28=\mathrm{S} 2$ bond are varied greatly. The bond length of C28-N4 (1.326(7) $\AA$ ) is shorter than that of
$\mathrm{C} 28-\mathrm{N} 3(1.383(7) \AA)$. The bond length of $\mathrm{C} 28=\mathrm{S} 2$ is 1.66 (6) $\AA$ longer than the normal bond length of $\mathrm{C}=\mathrm{SH}(1 \mathrm{~A})$ due to the intermolecular hydrogen bond of $\mathrm{N}(1)-\mathrm{H}(1 \mathrm{~A}) \ldots \mathrm{S}(2)$.

Table 4. Hydrogen-bond geometry $\left(\AA{ }^{\circ}{ }^{\circ}\right)$ of the title compound.

| D—H $\ldots \mathrm{A}$ | D—H | H $\ldots \mathrm{A}$ | D $\ldots \mathrm{A}$ | D—H...A |
| :---: | :---: | :---: | :---: | :---: |
| $\mathrm{N}(1)-\mathrm{H}(1 \mathrm{~A}) \ldots \mathrm{S}(2)^{\mathrm{a}, \mathrm{b}}$ | 0.86 | 2.62 | $3.460(6)$ | 165 |
| $\mathrm{~N}(2)-\mathrm{H}(2 \mathrm{~A}) \ldots \mathrm{O}(1)^{\mathrm{c}}$ | 0.86 | 1.96 | $2.666(5)$ | 138 |
| $\mathrm{~N}(3)-\mathrm{H}(3 \mathrm{~A}) \ldots \mathrm{S}(1)^{\mathrm{a}, \mathrm{b}}$ | 0.86 | 2.64 | $3.466(5)$ | 163 |
| $\mathrm{~N}(4)-\mathrm{H}(4 \mathrm{~B}) \ldots \mathrm{O}(2)^{\mathrm{c}}$ | 0.86 | 1.92 | $2.637(7)$ | 140 |
| $\mathrm{C}(26)-\mathrm{H}(26 \mathrm{~B}) \ldots \mathrm{S}(1)^{\mathrm{a}, \mathrm{b}}$ | 0.97 | 2.78 | $3.701(8)$ | 159 |
| $\mathrm{C}(34)-\mathrm{H}(34 \mathrm{~A}) \ldots \mathrm{S}(2)^{\mathrm{c}}$ | 0.93 | 2.81 | $3.267(6)$ | 111 |

Note. Elements of symmetry transformation: ${ }^{\text {a }} 1-\mathrm{x},-\mathrm{y}, 1-\mathrm{z} ;{ }^{\mathrm{b}}$ intermolecular hydrogen bond; ${ }^{\mathrm{c}}$ intramolecular hydrogen bond.

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