Bull. Chem. Soc. Ethiop. **2017**, 31(3), 519-534. © 2017 Chemical Society of Ethiopia and The Authors DOI: <u>http://dx.doi.org/10.4314/bcse.v31i3.16</u> ISSN 1011-3924 Printed in Ethiopia

USES OF ACETOACETANILIDE FOR THE SYNTHESIS OF THIOPHENE DERIVATIVES WITH CYTOTOXIC ACTIVITIES

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(Received October 21, 2017; Revised December 13, 2017; Accepted December 16, 2017)

ABSTRACT. The reaction of acetoacetanilide 1 with elemental sulfur and malononitrile 2 gave the thiophene derivative 3. The latter was the key starting material for the synthesis of different thiophene and fused derivatives through its reaction with aryldiazonium chloride, benzaldehyde and acetic anhydride. Moreover, the reaction with phenylisothiocyanate produced fused derivative. The reaction of compound 3 with elemental sulfur produced dithiophene derivative In addition, the reaction of compound 3 with ethylcyanoacetate produced N-cyanoacetamide derivative that was capable for further heterocyclizations through its reaction with different reagents. The cytotoxicity of the newly synthesized products was evaluated against the three cancer cell lines namely MCF-7 (breast adenocarcinoma), NCI-H460 (non-small cell lung cancer), and SF-268 (CNS cancer) where some compounds 5a, 5b, 8, 22, and 23 exhibited optimal cytotoxic effect against cancer cell lines.

KEY WORDS: Thiophene, Thieno[2,3-*d*]pyrimidine, Coumarin, Cytotoxicity

INTRODUCTION

Thiophene derivatives represent a class of important and well-studied heterocycles [1]. The interest in this kind of heterocycles has spread from early dye chemistry [2] to modern drug design [3], biodiagnostics [4], electronic and optoelectronic devices [5], conductivity-based sensors [6], and self-assembled superstructures [7]. The general synthetic approaches to such kind of compounds either involve the functionalization at the positions a and b to the sulfur atom of the pre-constructed thiophene nucleus [8], or the construction of thiophene ring from appropriately substituted open chain precursors [9]. The latter becomes much attractive for its general applicability to achieve more complicated substitution patterns [10]. Gewald and coworkers developed the synthesis of 2-aminothiophenes from the multicomponent condensation of ketones or aldehydes, cyanoacetate and elemental sulfur [11]. Later on, there are several reviews and papers reported on the variations and improvements on the originally published Gewald's synthesis of polysubstituted thiophenes [12]. Recently, we were involved through comprehensive program involving the synthesis of thiophene [13, 14] derivatives together with their further reactions with chemical reagents to give heterocyclic and fused heterocyclic derivatives with antitumor activities. Moreover, we synthesized different thiophene derivatives that have been screened for antitumor activity against breast adenocarcinoma (MCF-7), nonsmall cell lung cancer (NCI-H460) and CNS cancer (SF-268); this is besides studying their cytotoxicity against the normal human cell line normal fibroblast cells WI 38 [15]. In this work we make extension for this program through the synthesis of thiophene derivatives using acetoacetanilide and maononitrile in the presence of elemental sulfur to produce potentially antitumor thiophene derivatives [16-23].

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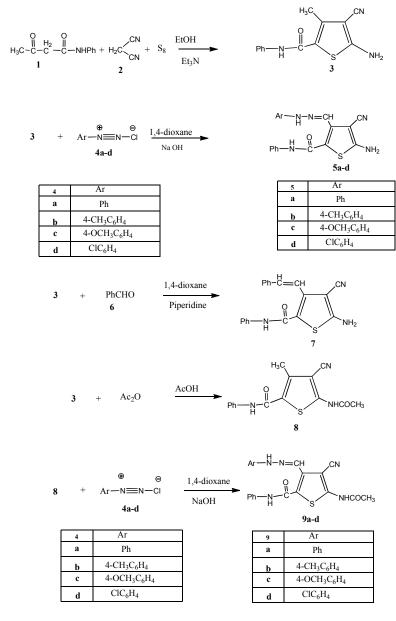
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RESULTS AND DISCUSSION

In the present work, acetaacetanilide **1** reacted with elemental sulfur and malononitrile **2** in the presence of triethylamine to give the thiophene derivative **3**. The structure of compound **3** was based on analytical and spectral data. Thus, the ¹H NMR spectrum showed one singlet at δ 2.5 ppm corresponding to the presence of CH₃ group, one singlet at δ 3.33 ppm corresponding to the NH₂ group, two multiplets at δ 7.00-7.71 ppm corresponding to C₆H₅ group and a singlet at δ 9.54 ppm for the NH group. Moreover, the ¹³C NMR data revealed the presence of signals at 16.9 (CH₃), 116.3 (CN), 119.0, 120.2, 123.8, 125.5, 134.2, 139.6, 143.2, 144.1 (C₆H₅, thiophene C), 163.9 (CO).

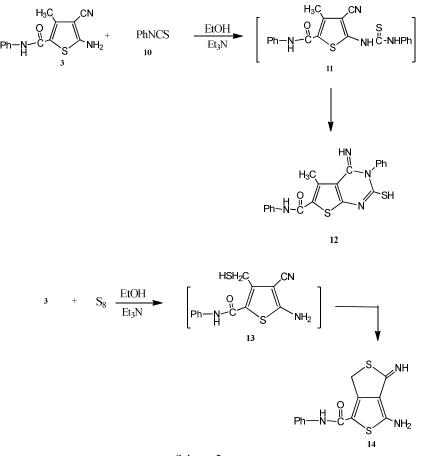
The 4-methyl group present in compound **3** showed interesting reactivity due to the effect of the electron withdrawal effect of the *o*-cyano group. Thus, compound **3** coupled with the aryldiazonium salts **4a-d** to give the aryl hydrazone derivatives **5a-d**, respectively. The analytical and spectral data of **5a-d** were the basis of their structure elucidation. On the other hand compound **3** reacted with benzaldehyde **6** to give the benzal derivative **7**. Compound **3** reacted with acetic acid/acetic anhydride (3:1) to give the 2-N-acetyl derivative **8**. The analytical and spectral data of the latter product were in agreement with its structure. Thus, ¹H NMR spectrum of **8** showed two singlet at δ 2.50 and 2.51 ppm corresponding to the presence of two CH₃ groups, one multiplets at δ 7.05-7.66 ppm corresponding to the C₆H₅ group and two singlets at δ 10.00, 11.96 ppm corresponding to the presence of two NH group. Moreover, ¹³C NMR showed the presence of signals at 19.6, 20.3 (2CH₃), (C=C), 116.3 (CN), 123.2, 124.8, 125.3, 126.9, 128.9, 130.2, 132.7, 134.6, 142.2, 143.4 (C₆H₅, thiophene C), 164.3, 165.8 (2CO).

Compound 8 coupled with the aromatic diazonium salts 4a-d to give the arylhydrazone derivatives 9a-d (Scheme 1). Encouraged by the excellent results, we next investigated the reactivity of compound 3 towards phenylisothiocyanate 10, where the reaction afforded the theino[d] pyrimidine derivative 12. The reaction took place through the intermediate formation of the N-phenylthiourea derivative 11. The structure of compound 12 was confirmed on the basis of analytical and spectral data. Thus, the ¹H NMR spectrum showed a singlet at δ 2.51 ppm for the CH₃ group, a multiplet at δ 7.30-7.66 ppm corresponding to the two C₆H₅ groups, a singlet at δ 7.71 ppm for SH group and two singlets at δ 9.54, 11.08 ppm corresponding to the presence of two NH group. Compound 3 reacted with elemental sulfur in the presence of triethylamine to give the thieno [c] thiophene derivative 14. The reaction took place through the first reaction of the CH₃ group with elemental sulfur to give the thiol derivative **13** followed by the addition of the SH group to the CN group to give 14 (Scheme 2). On the other hand, the reaction of compound 3 with malononitrile (15) in 1,4-dioxane containing a catalytic amount of triethylamine gave the thieno [b] pyridine derivative 17. Formation of the latter product took place through the intermediate formation of the acyclic intermediate 16. The 2-amino group present in **3** showed interesting reactivity towards diazodization and coupling. Thus, compound **3** reacted with a cold solution of sodium nitrite in the presence of acetic acid/hydrochloric acid (3:1) to give the intermediate diazonium salt 18. The latter reacted with acetyl acetone 19 to give the hydrazone derivative 20 (Scheme 3). The analytical and spectral data of the latter product were the basis of its structure elucidation. Thus the ¹H NMR spectrum showed three singlets at δ 2.32, 2.45, 2.62 ppm corresponding to the presence of three CH₃ groups, a multiplet at δ 7.27-7.62 ppm corresponding to the C₆H₅ group and two singlets at δ 7.72, 9.54 ppm for the two NH groups. In addition, the ¹³C NMR spectrum showed the presence of signals at 19.8, 20.6, 21.3 (3 CH₃), 117.0 (CN), 120.4, 121.7, 122.9, 124.7, 125.9, 128.2, 130.1, 138.0 (C₆H₅ thiophene C), 163.2, 164.0, 165.2 (3CO), 173.2 (C=N).



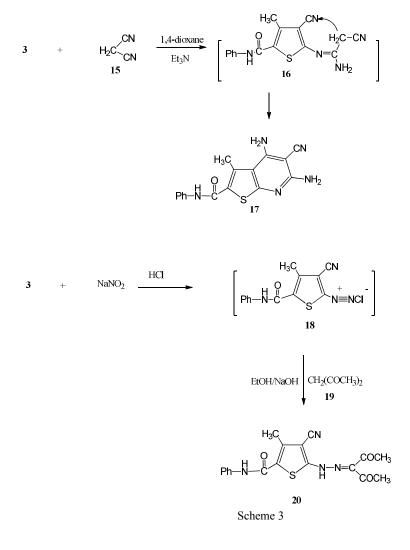
Scheme 1

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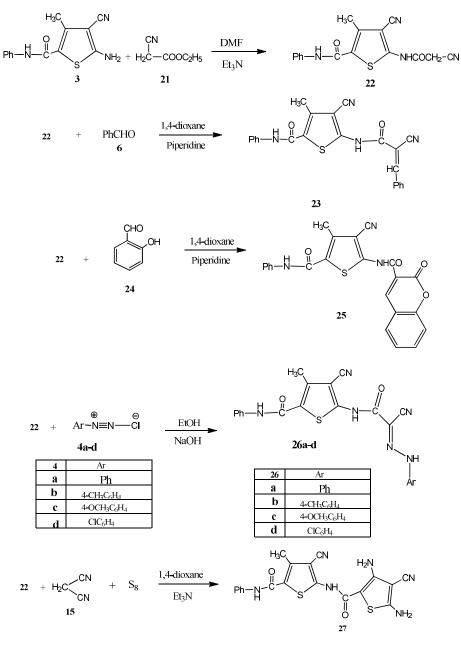


Scheme 2

Thus, compound **3** reacted with ethyl cyanoacetate **21** in refluxing dimethylformamide to give the amide derivative **22**. The analytical and spectral data of compound **22** are in agreement with its structure. The reaction of **22** with benzaldehyde **6** gave the benzylidene derivative **23**. Moreover, the reaction of **22** with salicyladehyde **24** gave the coumarin derivative **25**. The excellent yield of compound **22** encouraged us to study its reactivity towards aromatic diazonium salts in the aim of synthesizing new aryl hydrazine derivatives with different cytotoxic activities. Thus, the reaction of **22** with aromatic diazonium salts **4a-d** gave the analytical and spectral data of the latter products were the basis of their structure elucidation. Finally the reaction of compound **22** with elemental sulfur and malononitrile **15** gave the thiophene derivative **27** (Scheme 4).



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Scheme 4

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Structure activity relationship

From Table 1 it is clear that the benzimidazole moiety was found to be crucial for the cytotoxic effect of the cyclic compounds 3 to 27. Compounds 5a, 5b, 8, 22, and 23 exhibited optimal cytotoxic effect against cancer cell lines. On the other hand compound 9c, 26d and 27 showed very low activities toward the three cancer cell lines. Compounds 3, 5c, 5d, 7, 9d, 14, 20, 25 and 26c showed moderate activities. The thiophene derivative 3 showed low potency, however its reaction with difference aryl diazonium salts gave the arylhydrazone derivatives 5a-c. It is obvious that compounds 5a (R = H) and $5b (R = CH_3)$ showed the highest cytotoxicity among the three compounds. The reaction of compound 3 with benzaldehyde produced compound 7 with low potency. The thiophene derivative 7 showed low potency although its acetylation gave compound 8 with high potency against the three cancer cell lines. For the thiophenes derivatives **9a-d**, it is clear that compound **9b** with 6- the 4-methylphenyl group showed the highest activity among the four compounds which is considered as a moderate activity. Compound 12 which was a thieno[d]pyrimidine derivative showed highest activity among the tested compounds. Similarly, compound 17 which is a thieno[b]pyridine derivative showed high potency towards the three cancer cell lines. Compounds 22, 23, 26a and 26b showed relatively high potency and such activity is attributed to the presence of the N-acyl moieties. From the activity of the compounds towards the three cancer cell lines it is obvious that the presence of N-rich and Scontaining heterocyclic ring was the main effect through the high potency of the compound.

Table 1. Effect of the newly synthesized compounds on the growth of three human tumor cell lines $GI_{50} \pmod{L^{-1}}$.

Compound No.	MCF-7	NCI-H460	SF-268
3	22.6 ± 1.4	24.9 ± 2.8	13.8 ± 3.8
<u> </u>	0.4 ± 0.2	0.1 ± 0.02	0.9 ± 0.08
5b	0.01 ± 0.002	0.01 ± 0.004	0.04 ± 0.01
5c	30.6 ± 10.2	32.6 ± 8.6	24.4 ± 12.8
5d	18.1 ± 1.6	6.2 ± 2.8	8.6±2.6
7	28.6 ± 12.2	12.6 ± 8.6	52.4 ± 14.6
8	0.7 ± 0.50	0.2 ± 0.08	1.0 ± 0.02
9a	70.7 ± 18.5	40.2 ± 12.8	52.4 ± 8.6
9b	10.8 ± 2.6	4.5 ± 0.8	4.8 ± 1.8
9c	60.6 ± 16.9	38.9 ± 10.8	50.8 ± 8.6
9d	26.4 ± 2.2	34.1 ± 0.8	18.8 ± 4.8
12	0.01 ± 0.006	0.03 ± 0.001	0.02 ± 0.004
14	18.1 ± 0.6	16.3 ± 1.4	12.3 ± 1.5
17	0.06 ± 0.004	0.03 ± 0.003	0.1 ± 0.02
20	32.8 ± 0.6	36.5 ± 0.8	30.7 ± 1.6
22	0.3 ± 0.1	0.2 ± 0.08	0.5 ± 0.01
23	0.9 ± 0.2	0.1 ± 0.02	0.3 ± 0.05
25	22.6 ± 2.6	24.3 ± 0.8	30.9 ± 3.8
26a	0.01 ± 0.003	0.02 ± 0.001	0.01 ± 0.001
26b	0.02 ± 0.008	0.03 ± 0.006	0.05 ± 0.00
26c	22.4 ± 2.2	32.6 ± 1.4	26.8 ± 6.4
26d	55.1 ± 2.7	23.2 ± 4.8	14.4 ± 2.6
27	36.0 ± 1.8	43.0 ± 0.8	30.5 ± 1.1
Doxorubicin	0.04 ± 0.008	0.09 ± 0.008	0.09 ± 0.007

Results are given in concentrations that were able to cause 50% of cell growth inhibition (GI_{50}) after a continuous exposure of 48 h and show means ± SEM of three-independent experiments performed in duplicate.

EXPERIMENTAL

Antitumor and normal cell line

Reagents. Fetal bovine serum (FBS) and L-glutamine were from Gibco Invitrogen Co. (Scotland, UK). RPMI-1640 medium was from Cambrex (New Jersey, USA). Dimethyl sulfoxide (DMSO), doxorubicin, penicillin, streptomycin and sulforhodamine B (SRB) were from Sigma Chemical Co. (Saint Louis, USA).

Cell cultures. Three human tumor cell lines, MCF-7 (breast adenocarcinoma), NCI-H460 (nonsmall cell lung cancer), and SF-268 (CNS cancer) were used. MCF-7 was obtained from the European Collection of Cell Cultures (ECACC, Salisbury, UK) and NCI-H460 and SF-268 were kindly provided by the National Cancer Institute (NCI, Cairo, Egypt). They grow as monolayer and routinely maintained in RPMI-1640 medium supplemented with 5% heat inactivated FBS, 2 mM glutamine and antibiotics (penicillin 100 U/mL, streptomycin 100 mg/mL), at 37 °C in a humidified atmosphere containing 5% CO₂. Exponentially growing cells were obtained by plating 1.5 x 10⁵ cells/mL for MCF-7 and SF-268 and 0.75 x 10⁴ cells/mL for NCI-H460, followed by 24 h of incubation. The effect of the vehicle solvent (DMSO) on the growth of these cell lines was evaluated in all the experiments by exposing untreated control cells to the maximum concentration (0.5%) of DMSO used in each assay.

Tumor cell growth assay. The effects of **3-27** on the in vitro growth of human tumor cell lines were evaluated according to the procedure adopted by the National Cancer Institute (NCI, USA) in the 'In vitro Anticancer Drug Discovery Screen' that uses the protein-binding dye sulforhodamine B to assess cell growth. Briefly, exponentially, cells growing in 96-well plates were then exposed for 48 h to five serial concentrations of each compound, starting from a maximum concentration of 150 μ M. Following this exposure period adherent cells were fixed, washed, and stained. The bound stain was solubilized and the absorbance was measured at 492 nm in a plate reader (Bio-Tek Instruments Inc., Powerwave XS, Wincoski, USA). For each test compound and cell line, a dose–response curve was obtained and the growth inhibition of 50% (GI₅₀), corresponding to the concentration of the compounds that inhibited 50% of the net cell growth was calculated as described elsewhere. Doxorubicin was used as a positive control and tested in the same manner.

Instruments

All melting points are uncorrected. IR spectra were recorded for KBr discs on a PyeUnicam SP-1000 spectrophotometer. ¹H NMR spectra were measured on a Varian EM-390–200 MHz in DMSO as solvent using TMS as internal standard and chemical shifts are expressed as δ . Analytical data were obtained from the Microanalytical Data Unit at Cairo University, Giza, Egypt.

General procedure for the synthesis of 5-amino-4-cyano-3-methyl-N-phenylthiophene-2-carboxamide 3

To a solution of 3-oxo-N-phenylbutanamide (17.7 g, 0.1 mol) in absolute ethanol (50 mL) containing triethylamine (2.0 mL), malononitrile (6.6 g, 0.1 mol) and sulfur (3.2 g, 0.1 mol) was added. The reaction mixture was heated under reflux for 3 h then left to cool. The formed solid produce was collected by filtration and dried. The obtained product was crystallized from ethanol to give pale yellow crystals. Yield 20.5 g (80%); m.p. 225 °C; anal. calcd for $C_{13}H_{11}N_3OS$ (257.06): C, 60.68; H, 4.31; N, 16.33; S, 12.46%; found: C, 60.76; H, 4.47; N,

16.60; S, 12.51%. IR (KBr) v/cm⁻¹ 3466-3320 (NH₂, NH), 3055 (CH aromatic), 2220 (CN), 1688 (CO), 1630 (C=C). ¹H NMR (DMSO-d₆) δ 2.50 (s, 3H, CH₃), 3.33 (s, 2H, NH₂), 7.00–7.71 (m, 5H, C₆H₅), 9.54 (s, 1H, NH). ¹³C NMR (DMSO-d₆) δ 16.9 (CH₃), 116.3 (CN), 119.0, 120.2, 123.8, 125.5, 134.2, 139.6, 143.2, 144.1(C₆H₅, thiophene C), 163.9 (CO).

General procedure for the synthesis of the thiophenederivatives 5a-d

To a cold solution of **3** (2.57 g, 0.01 mol) in 1,4-dioxane (40 mL) containing sodium hydroxide (2.5 g) a cold solution of the respective diazonium salt [prepared by the addition of sodium nitrite solution (0.70 g, 0.01 mol) to a cold solution of either aniline (0.94 g, 0.01 mol), *p*-toluidine (1.15 g, 0.01 mol), *p*-methoxy aniline (1.3 g, 0.01 mol) or *p*-chloroaniline (1.29 g, 0.01 mol) in concentrated hydrochloric acid (12 mL) with continuous stirring] was added while stirring. The formed solid product, in each case, upon stirring at room temperature for 1 h was collected by filtration and dried. The obtained product was crystallized from ethanol to give brown crystals.

5-Amino-4-cyano-N-phenyl-3-((2-phenylhydrazono) methyl) thiophene-2-carboxamide (*5a*). Yield 2.7 g (75%); m.p. 135-140 °C; anal. calcd. for $C_{19}H_{15}N_5OS$ (361.42): C, 63.14; H, 4.18; N, 19.38; S, 8.87%; found: C, 63.06; H, 4.27; N, 19.44; S, 8.91%. IR (KBr) v/cm⁻¹ 3472-3328 (2NH, NH₂), 3055 (CH aromatic), 2220 (CN), 1687 (CO), 1656 (C=N), 1628 (C=C).¹H NMR (DMSO-d₆) δ 3.33 (s, 2H, NH₂), 6.39 (s, 1H, CH),6.40 (s,1H, NH), 7.05–7.71 (m,10H, 2C₆H₅), 9.54 (s, 1H, NH).¹³C NMR (DMSO-d₆) δ 116.6 (CN), 119.8, 121.6, 124.3, 125.8, 128.0, 129.6, 130.2, 134.5, 139.8, 143.8, 144.7 (two C₆H₅, thiophene C), 164.2 (CO), 172.1 (C=N).

5-Amino-4-cyano-N-phenyl-3-((2-(p-tolyl) hydrazono) methyl) thiophene-2-carboxamide (**5b**). Yield 2.62 g (70%); m.p. 140-150 °C; anal. calcd. for $C_{20}H_{17}N_5OS$ (375.54): C, 63.98; H, 4.56; N, 18.65; S, 8.54%; found: C, 63.79; H, 4.31; N, 18.72; S, 8.76%. IR (KBr) v/cm⁻¹ 3477-3318 (NH₂, 2NH), 3062 (CH aromatic), 2223 (CN), 1688 (CO), 1660 (C=N), 1628 (C=C).¹H NMR (DMSO-d₆) δ 2.29 (s, 3H, CH₃), 3.30 (s, 2H, NH₂),7.05 (s,1H, CH), 7.12 (s,1H, NH), 7.32–7.71 (m, 9H, C₆H₄, C₆H₅), 9.54 (s, 1H, NH).¹³C NMR (DMSO-d₆) δ 19.8 (CH₃), 116.6 (CN), 119.3, 122.5, 124.6, 125.2, 128.3, 129.8, 131.4, 132.9, 138.4, 142.7, 143.9 (C₆H₅, C₆H₄, thiophene C).

5-Amino-3-((2-(4-methoxyphenyl))hydrazono)methyl)-4-cyano-N-phenylthiophene-2-carboxamide (5c). Yield 3.24 g (83%); m.p. 120-125 °C; Anal. Calcd. for $C_{20}H_{17}N_5O_2S$ (391.45): C, 61.37; H, 4.38; N, 17.89;S, 8.19%; found: C, 61.43; H, 4.23; N, 18.02; S, 8.22%. IR (KBr) v/cm⁻¹ 3484-3322 (NH₂, 2NH), 3062 (CH aromatic), 2221 (CN), 1689 (CO), 1662 (C=N), 1625 (C=C).¹H NMR (DMSO-d₆) δ 3.74 (s, 2H, NH₂), 3.75 (s, 3H,CH₃),6.93 (s,1H,CH), 7.01 (s,1H, NH), 7.30–7.67 (m,9H, C₆H₅, C₆H₄), 9.50 (s, 1H, NH).¹³C NMR (DMSO-d₆) δ 19.3 (CH₃), 116.3 (CN), 119.9, 123.2, 124.9, 125.1, 128.6, 129.1, 132.0, 132.9, 138.4, 142.7, 143.4 (C₆H₅, C₆H₄, thiophene C), 164.8 (CO), 171.2 (C=N).

5-*Amino-3-((2-(4-chlorophenyl)hydrazono)methyl)-4-cyano-N-phenylthiophene-2-carboxamide* (*5d*). Yield 3.36 g (85 %); m.p. 120 °C; anal. calcd. for $C_{19}H_{14}CIN_5OS$ (395.87): C, 57.65; H, 3.56; N, 17.69; S, 8.10%; found: C, 57.90; H, 3.47; N, 17.88; S, 8.28%. IR (KBr) v/cm⁻¹ 3489-3328 (NH₂, 2NH), 3051 (CH aromatic), 2220 (CN), 1687 (CO), 1665 (C=N), 1621 (C=C).¹H NMR (DMSO-d₆) δ 3.81 (s, 2H, NH₂), 6.90 (s, 1H, CH),7.02 (s,1H, NH), 7.30–7.67 (m, 9H, C₆H₄, C₆H₅), 9.52 (s, 1H, NH).¹³C NMR (DMSO-d₆) δ 116.4 (CN), 119.9, 123.8, 124.9, 125.4, 128.1, 128.4, 132.3, 132.6, 134.6, 142.9, 143.5 (C₆H₅, C₆H₄, thiophene C), 164.1 (CO), 171.1 (C=N).

5-Amino-4-cyano-N-phenyl-3-styrylthiophene-2-carboxamide (7)

To a solution of **3** (2.57 g, 0.01 mol) in 1,4-dioxane (40 mL) containing piperidine (1.0 mL), benzaldehyde (1.0 g, 0.01 mol) was added. The reaction mixture, in each case, was heated under reflux for 8 h then left to cool. The solid product formed upon evaporating the solution under vacuum followed by triturating the remaining product with ethanol, was collected by filtration and dried. The obtained product was crystallized from ethanol to give pale brown crystals. Yield 2.76 g (80%); m.p. 180 °C; anal. calcd. for $C_{20}H_{15}N_3OS$ (345.42): C, 69.54; H, 4.38; N, 12.17; S, 9.28%; found: C, 69.69; H, 4.29; N, 12.28; S, 9.41%; IR (KBr) v/cm⁻¹ 3426–3365 (NH₂, NH), 3054 (CH, aromatic), 2219 (CN),1801 (C=O),1660 (C=N); ¹H NMR (DMSO-d₆) δ 3.31 (s, 2H, NH₂), 7.05–7.27 (2d, 2H, CH=CH), 7.30–7.59 (m,10H, 2C₆H₅), 9.54 (s, 1H, NH), ¹³C NMR (DMSO-d₆) δ 90.6, 98.3 (C=C), 116.5 (CN), 120.4, 122.6, 124.6, 126.9, 128.3, 128.9, 132.1, 132.8, 134.6, 142.9, 143.9 (two C₆H₅, thiophene C), 164.13 (CO).

5-Acetamido-4-cyano-3-methyl-N-phenylthiophene-2-carboxamide (8)

To a solution of **3** (2.57 g, 0.01 mol) in acetic acid (30 mL), acetic anhydride (1.02 g, 0.01 mol) was added. The reaction mixture was heated under reflux for 3 h then left to cool. The solid product formed upon pouring onto ice/water containing a few drops of hydrochloric acid was collected by filtration and dried. The obtained product was crystallized from ethanol to give pale yellow crystals. Yield 2.33 g (78%), m.p. 260 °C. anal. calcd. for $C_{15}H_{13}N_3O_2S$ (299.35): C, 60.18; H, 4.38; N, 14.04; S, 10.71%; found: C, 62.42; H, 4.36; N, 14.11; S, 10.84%. IR (KBr) v/cm⁻¹ 3446–3268 (NH₂, 2NH), 3051 (CH, aromatic), 2221 (CN), 1703 (C=O), 1643 (C=C). ¹H NMR (DMSO-d₆) δ 2.50,2.51 (2s, 6H, 2CH₃), 7.05–7.66 (m, 5H, C₆H₅), 10.00,11.96 (2s, 2H, 2NH). ¹³C NMR (DMSO-d₆) δ 19.6, 20.3 (2CH₃), (C=C), 116.3 (CN), 123.2, 124.8, 125.3, 126.9, 128.9, 130.2, 132.7, 134.6, 142.2, 143.4 (C₆H₅, thiophene C), 164.3, 165.8 (2CO).

General procedure for the synthesis of thiophene-2-carboxamidederivatives 9a-d

To a cold solution of **8** (2.99 g, 0.01 mol) in ethanol (40 mL) containing sodium acetate (2.5 g) a cold solution of the respective diazonium salt [prepared by the addition of sodium nitrite solution (0.70 g, 0.01 mol) to a cold solution of either aniline (0.94 g, 0.01 mol), *p*-toluidine (1.15 g, 0.01 mol), *p*-methoxy aniline (1.3 g, 0.01 mol) or *p*-chloroaniline (1.29 g, 0.01 mol) in concentrated hydrochloric acid (12 mL) with continuous stirring] was added while stirring. The formed solid product, in each case, upon stirring at room temperature for 1 h was collected by filtration and dried. The obtained product was crystallized from ethanol to give yellowish brown crystals for **9a** and **9b** and reddish yellow crystals for **9c** and **9d**.

5-Acetamido-4-cyano-N-phenyl-3-((2-phenylhydrazono)methyl)thiophene-2-carboxamide (9a). Yield 2.82 g (70%); m.p. over 300 °C; anal. calcd. for $C_{21}H_{17}N_5O_2S$ (403.46): C, 62.52; H, 4.25; N, 17.36; S, 7.95%; found: C, 62.44; H, 4.11; N, 17.47; S, 8.05%. IR (KBr) v/cm⁻¹: 3488-3328 (2NH), 3050 (CH aromatic), 2220 (CN), 1685, 1680 (2CO), 1660 (C=N), 1630 (C=C). ¹H NMR (DMSO-d₆) δ 2.26 (s, 3H, CH₃), 7.07 (CH=N), 7.08 (s, 1H, NH), 7.12–7.66 (m, 10H, 2C₆H₅), 10.00, 11.96 (2s, 2H, 2NH).¹³C NMR (DMSO-d₆) δ 19.8 (CH₃), 116.8 (CN), 119.4, 121.7, 122.0, 123.8, 124.9, 130.3, 133.8, 134.2, 138.5, 140.4, 144.2 (two C₆H₅, thiophene C), 163.8, 164.2 (2CO), 172.3 (C=N).

5-Acetamido-4-cyano-N-phenyl-3-((2-(p-tolyl)hydrazono) methyl) thiophene-2-carboxamide (**9b**). Yield 3.33 g (80%); m.p. 290 °C; anal. calcd. for $C_{22}H_{19}N_5O_2S$ (417.48): C, 63.29; H, 4.59; N, 16.78; S, 7.68%; found: C, 62.88; H, 4.33; N, 16.92; S, 7.82%. IR (KBr) v/cm⁻¹ 3447–3267 (3NH), 3052 (CH, aromatic), 2970 (CH₃), 2223 (CN), 1704 (C=O),1643 (C=N),1529 (C=C); ¹H

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NMR (DMSO-d₆) δ 2.25,2.50 (2s, 6H, 2CH₃), 7.07 (s, 1H, NH), 7.09 (s,1H,CH=N), 7.30–7.60 (m, 9H, C₆H₅, C₆H₄), 10.01,11.97 (2s, 2H, 2NH).¹³C NMR (DMSO-d₆) δ 19.8, 21.6 (2CH₃), 116.4 (CN), 120.4, 121.8, 122.0, 124.1, 124.9, 130.6, 133.9, 134.3, 138.8, 140.1, 144.9 (C₆H₅, C₆H₄, thiophene C), 164.1, 164.8 (2CO), 172.1 (C=N).

5-Acetamido-4-cyano-3-((2-(4-methoxyphenyl) hydrazono) methyl)-N-phenylthiophene-2carboxamide (9c). Yield 3.76 g (87%); m.p. 280°C; anal. calcd. for $C_{22}H_{19}N_5O_3S$ (433.48): C, 60.96; H, 4.42; N, 16.16; S, 7.40%; found: C, 60.37; H, 4.36; N, 16.09; S, 7.66%. ¹H NMR (DMSO-d₆) δ 2.50,3.30 (2s, 6H, 2CH₃), 6.93 (s, 1H, CH), 7.32–7.71 (m, 9H, C₆H₅, C₆H₄), 9.93, 9.94, 11.95 (3H, 3H, 3NH). ¹³C NMR (DMSO-d₆) δ 19.8, 22.4 (2CH₃), 116.8 (CN), 120.5, 121.1, 122.3, 124.4, 124.9, 130.6, 132.3, 133.6, 138.9, 140.3, 144.5 (C₆H₅, C₆H₄, thiophene C), 164.3, 164.9 (2CO), 172.3 (C=N).

5-Acetamido-3-((2-(4-chlorophenyl)hydrazono)methyl)-4-cyano-N-phenylthiophene-2-carboxamide (9d). Yield 3.63 g (83%); m.p. 295°C; anal. calcd for $C_{21}H_{16}ClN_5O_2S$ (437.9): C, 57.60; H, 3.68; N, 15.99; S, 7.32%; found: C, 57.76; H, 3.45; N, 16.09; S, 7.46%. ¹H NMR (DMSO-d₆) δ 2.25 (s, 3H, CH₃), 7.07 (s, 1H, NH), 7.09 (s, 1H, CH=N), 7.30–7.60 (m, 9H, C₆H₅, C₆H₄), 10.01, 11.97 (2s, 2H, 2NH). ¹³C NMR (DMSO-d₆) δ 21.9 (CH₃), 116.6 (CN), 120.6, 121.4, 122.8, 124.2, 124.5, 130.8, 132.1, 133.6, 138.9, 140.2, 144.9 (C₆H₅, C₆H₄, thiophene C), 164.6, 164.8 (2CO), 172.2 (C=N).

3,4-Dihydro-4-imino-2-mercapto-5-methyl-N,3-diphenylthieno[2,3-d]pyrimidine-6-carboxamide (12)

To a solution of **3** (2.57 g, 0.01 mol) in absolute ethanol (40 mL) containing triethylamine (1.0 mL), phenyl isothiocyanate (1.35 g, 0.01 mol) was added. The reaction mixture was heated underreflux for 2 h then left to cool. The solid product formed upon evaporating the solution under vacuum followed by triturating the remaining product with ethanol, was collected by filtration and dried. The obtained product was crystallized from ethanol to give pale yellow crystals. Yield 2.70 g (75%); m.p. 120-135 °C; anal. calcd. for $C_{20}H_{16}N_4OS_2$ (392.5): C, 61.20; H, 4.11; N, 14.27; S, 16.34%; found: C, 61.31; H, 4.21; N, 14.38; S, 16.44%. IR (KBr) v/cm⁻¹ 3480-3320 (SH, 2NH), 3055 (CH aromatic), 1688 (CO) 1660 (exocyclic C=N), 1628 (C=C); ¹H NMR (DMSO-d₆) δ 2.51 (s, 3H, CH₃), 7.30–7.66 (m, 10H, 2C₆H₅), 7.71 (s, 1H, SH), 9.54, 11.08 (2s, 2H, 2NH). ¹³C NMR (DMSO-d₆) δ 19.6 (CH₃), 119.8, 120.3, 122.4, 123.7, 126.0, 128.9, 130.2, 131.2, 133.8, 138.9, 140.2, 142.6, 144.6, 146.8 (two C₆H₅, thiophene C), 164.2 (CO), 174.8 (C=N).

3-Amino-4,6-dihydro-4-imino-N-phenylthieno[3,4-c]thiophene-1-carboxamide (14)

To a solution of **3** (2.57 g, 0.01 mol) in absolute ethanol (50 mL) containing triethylamine (2.0 mL), element sulfur (3.2 g, 0. 1 mol) was added. The reaction mixture was heated under reflux for 3 h then left to cool. The formed solid product was collected by filtration and dried. The obtained product was crystallized from ethanol to give pale yellow crystals. Yield 2.02 g (70%); m.p. 190 °C; anal. calcd. for $C_{13}H_{11}N_3OS_2$ (289.38): C, 53.96; H, 3.83; N, 14.52; S, 22.16%; found: C, 52.77; H, 3.67; N, 14.69; S, 22.09%. IR (KBr) v/cm-¹ 3364–3316 (NH₂, NH), 3057 (CH, aromatic), 1718 (C=O), 1624 (C=N); ¹H NMR (DMSO-d₆) δ 3.56 (s, 2H, NH₂), 7.05(s, 1H, thiophene CH₂), 7.27–7.60 (m, 5H, C₆H₅), 7.71, 9.54 (2H, 2H, 2NH). ¹³C NMR (DMSO-d₆) δ 64.6 (CH₂),119.4, 120.8, 122.6, 124.2, 126.5, 128.4, 131.2, 133.8, 138.9, 140.2, 142.6, 144.6 (C₆H₅, thiophene C), 164.8 (CO), 175.9 (C=N).

4,6-Diamino-5-cyano-3-methyl-N-phenylthieno[2,3-b]pyridine-2-carboxamide (17)

To a solution of **3** (2.57 g, 0.01 mol) in absolute ethanol (50 mL) containing triethylamine (1.5 mL), malononitrile (0.66 g, 0.01 mol) was added. The reaction mixture was heated under reflux for 3 h. The solid product formed upon evaporating the solution under vacuum followed by triturating the remaining product with ethanol, was collected by filtration and dried. The obtained product was crystallized from ethanol to give pale brown crystals. Yield 2.45 g (76%); m.p. 185 °C; anal. calcd. for $C_{16}H_{13}N_5OS$ (323.37): C, 59.43; H, 4.05; N, 21.66; S, 9.92%; found: C, 59.21; H, 3.89; N, 21.88; S, 10.05%. IR (KBr) v/cm⁻¹ 3424–3282 (2 NH₂, NH), 3056 (CH, aromatic), 2220 (CN), 1680 (C=O), 1629 (C=C). ¹H NMR (DMSO-d₆) & 2.51 (s, 3H, CH₃), 2.44, 3.56 (2s, 4H, NH₂), 7.05–7.64 (m, 5H, C_6H_5), 9.54 (s, 1H, NH), ¹³C NMR (DMSO-d₆) & 22.3 (CH₃), 116.9 (CN), 119.4, 121.3, 122.6, 124.2, 126.8, 129.1, 130.5, 133.8, 138.3, 140.1, 143.8, 144.1 (C_6H_5 , thiophene, pyridine C), 164.2 (CO), 172.2 (C=N).

4-Cyano-5-(2-(2,4-dioxopentan-3-ylidene)hydrazinyl)-3-methyl-N-phenylthiophene-2-carboxamide (20)

To a cold solution of acetyl acetone (1.00 g, 0.01 mol) in ethanol (40 mL) containing sodium hydroxide (2.5 g) a cold solution of the respective diazonium salt [prepared by the addition of sodium nitrite solution (0.70 g, 0.01 mol) to a cold solution of **3** (2.57 g, 0.01 mol) in concentrated hydrochloric acid (12 mL) with continuous stirring] was added while stirring. The formed solid product, upon stirring at room temperature for 1 h was collected by filtration and dried. The obtained product was crystallized from ethanol to give black crystals. Yield 2.87 g (78%); m.p. 160-170 °C; anal. calcd. for $C_{18}H_{16}N_4O_3S$ (368.41): C, 58.68; H, 4.38; N, 15.21; S, 8.70%; found: C, 58.89; H, 4.25; N, 15.53; S, 8.88%. IR (KBr) v/cm⁻¹ 3422–3280 (2NH), 3053 (CH, aromatic), 2222 (CN), 1684 (C=O), 1622 (C=C); ¹H NMR (DMSO-d₆) & 2.32, 2.45, 2.62 (3s, 9H, 3CH₃), 7.27–7.62 (m, 5H, C_6H_5), 7.72, 9.54 (2s, 2H, 2NH), ¹³C NMR (DMSO-d₆) δ 19.8, 20.6, 21.3 (3 CH₃), 117.0 (CN), 120.4, 121.7, 122.9, 124.7, 125.9, 128.2, 130.1, 138.0 (C_6H_5 , thiophene C), 163.2, 164.0, 165.2 (3CO), 173.2 (C=N).

4-Cyano-5-(2-cyanoacetamido)-3-methyl-N-phenylthiophene-2-carboxamide (22)

To a solution of **3** (2.57 g, 0.01 mol) in absolute ethanol (40 mL) containing triethylamine (1.5 mL), ethyl cyanoacetate (1.13 g, 0.01 mol) was added. The reaction mixture was heated under reflux for 2 h. The solid product formed upon pouring onto ice/water containing a few drops of hydrochloric acid was collected by filtration and dried. The obtained product was crystallized from ethanol to give brown crystals. Yield 2.59 g (80%); m.p. 195 °C; anal. calcd. for $C_{16}H_{12}N_4O_2S$ (324.36): C, 59.25; H, 3.73; N, 17.27; S, 9.89%; found: C, 59.86; H, 3.65; N, 17.06; S, 9.93. IR (KBr) v/cm⁻¹ 3477-3318 (2NH), 3045 (CH aromatic); ¹H NMR (DMSO-d₆) δ 2.54 (s, 3H, CH₃), 4.42 (s, 2H, CH₂). 7.09–7.60 (m, 5H, C₆H₅), 7.71, 9.54 (2s, 2H, 2NH). ¹³C NMR (DMSO-d₆) δ 20.6 (CH₃), 62.1 (CH₂), 116.8, 117.3 (2CN), 119.9, 120.6, 121.9, 123.1, 124.7, 125.5, 126.8, 130.2, 138.3, 140.2 (C₆H₅, thiophene C), 164.4, 164.5 (2CO).

5-(2-Cyano-3-phenylacrylamido)-4-cyano-3-methyl-N-phenylthiophene-2-carboxamide (23)

To a solution of **22** (3.24 g, 0.01 mol) in 1,4-dioxane (40 mL) containing piperidine (1.0 mL), benzaldehyde (1.0 g, 0.01 mol) was added. The reaction mixture was heated under reflux for 3 h then left to cool. The solid product formed upon evaporating the solution under vacuum followed by triturating the remaining product with ethanol, was collected by filtration and dried. The obtained product was crystallized from 1,4-dioxane to give yellowish green crystals. Yield 3.37 g (82%); m.p. 175 °C; anal. calcd. for $C_{23}H_{16}N_4O_2S$ (412.46): C, 66.97; H, 3.91; N, 13.58;

S, 7.77%; found: C, 66.87; H, 3.69; N, 13.73; S, 8.01%. IR (KBr) v/cm⁻¹ 3424–3362 (2NH), 3059 (CH, aromatic), 2924 (CH₃), 1687, 1685 (2C=O), 1657 (C=C); ¹H NMR (DMSO-d₆) δ 2.53 (s, 3H, CH₃), 7.11 (s, 1H, CH=C), 7.39–7.68 (m, 10H, 2C₆H₅), 8.88, 10.27 (2s, 2H, 2NH). ¹³C NMR (DMSO-d₆) δ 19.6 (CH₃), 87.3, 92.5 (CH=C), 116.4, 116.9 (2CN), 119.3, 120.8, 122.4, 123.6, 124.8, 125.9, 126.8, 130.1, 132.1, 136.2, 140.9, 142.6 (two C₆H₅, thiophene C), 164.6, 166.2 (2CO).

N-(3-Cyano-4-methyl-5-(phenylcarbamoyl)thiophen-2-yl)-2-oxo-2H-chromene-3-carboxamide (25)

To a solution of **22** (3.24 g, 0.01 mol) in 1,4-dioxane (40 mL) containing piperidine (1.0 mL), salicyaldehyde (1.22 g, 0.01 mol) was added. The reaction mixture was heated under reflux for 2 h then left to cool. The solid product formed upon evaporating the solution under vacuum followed by triturating the remaining product with ethanol, was collected by filtration and dried. The obtained product was crystallized from ethanol to give orange yellow crystals. Yield 3.00 g (70%); m.p. 240-250 °C; anal. calcd. for $C_{23}H_{15}N_3O_4S$ (429.45): C, 64.33; H, 3.52; N, 9.78; S, 7.47%; found: C, 64.27; H, 3.34; N, 9.92; S, 7.64%. IR (KBr) v/cm⁻¹ 3479–3321 (2NH), 3053 (CH, aromatic), 2921 (CH₃), 2220 (CN), 1689-1685 (3C=O), 1653 (C=C). ¹H NMR (DMSO-d₆) δ 2.53 (s, 3H, CH₃), 7.00 (s, H, coumarin H-4), 7.34–7.90 (m, 9H, C₆H₅, C₆H₄), 9.07, 10.24 (2s, 2H, 2NH). ¹³C NMR (DMSO-d₆) δ 18.3 (CH₃), 116.3 (CN), 96.8 (coumarin C-4), 120.5, 121.2, 121.9, 123.9, 125.2, 125.6, 129.1, 130.4, 132.7, 134.9, 139.2, 140.9, 141.8, 143.6, 152.0 (C₆H₅, coumarin, thiophene C), 164.1, 165.9, 166.3 (3CO).

General procedure for the synthesis of 2-cyanoacetamido)-4-cyano-3-methyl-N-phenylthiophene-2-carboxamide derivatives **26a-d**

To a cold solution of **22** (3.24 g, 0.01 mol) in 1,4-dioxane (40 mL) containing sodium hydroxide (2.5 g) a cold solution of the respective diazonium salt [prepared by the addition of sodium nitrite solution (0.70 g, 0.01 mol) to a cold solution of either aniline (0.94 g, 0.01 mol), *p*-toluidine (1.15 g, 0.01 mol), *p*-methoxy aniline (1.3 g, 0.01 mol) or *p*-chloroaniline (1.29 g, 0.01 mol) in concentrated hydrochloric acid (12 mL) with continuous stirring]was added while stirring. The formed solid product, in each case, upon stirring at room temperature for 1 h was collected by filtration and dried. The obtained product was crystallized from ethanol to give brown crystals for **26a** and **26c** and black crystals for **26b** and **26d**.

2-(2-Phenylhydrazono)-2-cyanoacetamido)-4-cyano-3-methyl-N-phenylthiophene-2-carboxamide (26a). Yield 3.5 g (82%); m.p. 70-80 °C; anal. calcd. for $C_{22}H_{16}N_6O_2S$ (428.47): C, 61.67; H, 3.76; N, 19.61; S, 7.48. found: C, 61.21; H, 3.58; N, 19.88; S, 7.58%. IR (KBr) v/cm⁻¹ 3492– 3329 (3NH), 3056 (CH, aromatic), 2923 (CH3), 2222, 2220 (2CN), 1689, 1687 (2C=O), 1660 (C=N), 1633 (C=C). ¹H NMR (DMSO-d₆) δ 2.52 (s, 3H, CH₃), 7.30 (s, 1H, NH), 7.40–7.66 (m, 10H, 2C₆H₅), 8.40, 9.48 (2s, 2H, 2NH). ¹³C NMR (DMSO-d₆) δ 18.8 (CH3), 116.7, 117.2 (2CN), 121.0, 121.8, 122.3, 123.5, 124.2, 125.9, 129.1, 132.7, 132.9, 140.9, 141.8, 143.6 (two C6H5, thiophene C), 164.4, 165.2 (2CO), 172.0 (C=N).

2-(2-p-Tolylhydrazono)-2-cyanoacetamido)-4-cyano-3-methyl-N-phenylthiophene-2-carboxamide (**26b**). Yield 3.53 g (80%); m.p. 110-120 °C; anal. calcd. for C₂₃H₁₈N₆O₂S (442.49): C, 62.43; H, 4.10; N, 18.99; S, 7.25%; found: C, 62.38; H, 3.91; N, 19.06; S, 7.41%. IR (KBr) v/cm⁻¹ 3488–3339 (3NH), 3053 (CH, aromatic), 2923 (CH₃), 2222, 2220 (2CN), 1693, 1689 (2C=O), 1661 (C=N), 1630 (C=C). ¹H NMR (DMSO-d₆) δ 2.39, 2.40 (2s, 6H, 2CH₃), 7.09 (s, 1H, NH), 7.37–7.71 (m, 9H, C₆H₅, C₆H₄), 8.80, 10.13 (2s, 2H, 2NH).¹³C NMR (DMSO-d₆) δ

18.3, 19.6 (2CH₃), 116.2, 117.0 (2CN), 119.8, 122.0, 122.3, 123.8, 124.6, 126.4, 129.8, 132.2, 133.6, 140.5, 141.8, 143.8 (two C_6H_5 , thiophene C), 164.2, 165.7 (2CO), 172.1 (C=N).

2-(2-(4-Methoxyphenyl)hydrazono)-2-cyanoacetamido)-4-cyano-3-methyl-N-phenylthiophene-2carboxamide (26c). Yield 3.89 g (85%); m.p. 155°C; anal. calcd. for $C_{23}H_{18}N_6O_3S$ (458.49): C, 60.25; H, 3.96; N, 18.33; S, 6.99%; found: C, 59.89; H, 3.78; N, 18.09; S, 7.11%. IR (KBr) v/cm⁻¹ 3447–3267 (3NH), 3052 (CH, aromatic), 2970 (CH₃), 2223 (CN), 1704 (C=O), 1643 (C=N), 1529 (C=C). ¹H NMR (DMSO-d₆) δ 2.49, 2.68 (2s, 6H, 2CH₃), 7.09 (s, 1H, NH), 7.39– 7.79 (m, 9H, C₆H₅, C₆H₄), 10.55, 11.11 (2s, 2H, 2NH). ¹³C NMR (DMSO-d₆) δ 18.6, 21.3 (2CH₃), 116.4, 117.3 (2CN), 120.9, 122.9, 122.3, 123.8, 124.8, 126.4, 129.3, 132.4, 133.8, 140.9, 141.3, 143.1 (two C₆H₅, thiophene C), 164.3, 165.9 (2CO), 172.0 (C=N).

5-(2-(2-(4-Chlorophenyl)hydrazono)-2-cyanoacetamido)-4-cyano-3-methyl-N-phenylthiophene-2-carboxamide (26d). Yield 3.65 g (79%); m.p. 135°C; anal. calcd. for C₂₂H₁₅ClN₆O₂S (462.91):C, 57.08; H, 3.27; N, 18.15; S, 6.93%; found: C, 56.82; H, 3.12; N, 17.99; S, 7.14%. IR (KBr)v/cm⁻¹ 3476–3320 (3NH), 3055 (CH, aromatic), 2924 (CH₃), 2224, 2220 (2CN), 1690, 1689 $(2C=O), 1661 (C=N), 1624 (C=C). ¹H NMR (DMSO-d₆) <math>\delta$ 2.62 (s, 3H, CH₃), 7.06 (s, 1H, NH), 7.30–7.79 (m, 9H, C₆H₅, C₆H₄), 10.27, 11.10 (2s, 2H, 2NH). ¹³C NMR (DMSO-d₆) δ 18.9 (CH₃), 116.2, 117.1 (2CN), 120.8, 122.5, 122.8, 123.1, 124.8, 126.2, 129.3, 132.6, 133.8, 140.3, 142.8, 143.7 (two C₆H₅, thiophene C), 164.6, 165.4 (2CO), 172.1 (C=N).

3,5-Diamino-4-cyano-N-(3-cyano-4-methyl-5-(phenylcarbamoyl)thiophen-2-yl)thiophene-2carboxamide (27)

To a solution of **22** (3.24 g, 0.01 mol) in 1,4-dioxane (50 mL) containing triethylamine (1.5 mL), malononitrile (0.66 g, 0.01 mol) was added followed by elemental sulfur (0.32 g, 0.01 mol). The reaction mixture was heated under reflux for 3 h. The solid product formed upon evaporating the solution under vacuum followed by triturating the remaining product with ethanol, was collected by filtration and dried. The obtained product was crystallized from ethanol to give pale brown crystals. Yield 3.16 g (75%); m.p. 185 °C; anal. calcd. for $C_{19}H_{14}N_6O_2S_2$ (422.48): C, 54.01; H, 3.34; N, 19.89; S, 15.18%; found: C, 53.72; H, 3.28; N, 20.11; S, 15.23. IR (KBr) v/cm⁻¹ 3491–3316 (3NH), 3056 (CH, aromatic), 2921 (CH₃), 2227, 2220 (2CN), 1692, 1686 (2C=O), 1660 (C=N), 1623 (C=C). ¹H NMR (DMSO-d₆) δ 2.50 (s, 3H, CH₃), 3.45–3.48 (2s, 4H, 2NH₂). 7.03–7.60 (m, 5H, C₆H₅), 7.70, 9.54 (2s, 2H, 2NH). ¹³C NMR (DMSO-d₆) δ 18.6 (CH₃), 116.1, 116.9 (2CN), 120.9, 122.8, 123.0, 124.3, 124.8, 126.8, 129.3, 132.9, 136.2, 140.7, 142.3, 143.1 (C₆H₅, two thiophene C), 164.3, 165.7 (2CO).

CONCLUSION

A series of thiophene derivatives were synthesized and evaluated against three cancer cell lines. The above results allow the conclusion that administration of the tested compounds to the cancer cell line showed promising anticancer activity. Compounds **5a**, **5b**, **8**, **22**, and **23** exhibited optimal cytotoxic effect against cancer cell lines.

REFERANCES

- 1. Mohanakrishnan, A.K.; Amaladass, P.; Clement, J.A. Synthesis of end-blocked thienyl oligomers incorporating benzo[c]thiophene. *Tetrahedron Lett.* **2007**, 48, 779-784.
- 2. Ferreira, I.C.; Queiroz M.J., Kirsch, G. Tandem palladium-catalyzed borylation and suzuki coupling (BSC) to thienocarbazole precursors. *Tetrahedron Lett.* **2003**, 44, 4327-4343.

- Queiroz, M.J.; Ferreir, I.C.; Gaetano, Y.D.; Kirsch, G.; Calhelha, R.C.; Estevinho, L.M. Synthesis and antimicrobial activity studies of ortho-chlorodiarylamines and heteroaromatic tetracyclic systems in the benzo[b]thiophene series. J. Bioorg. Med. Chem. 2006, 14, 6827-6848.
- Sall, D.J.; Briggs, S.L.; Chirgadze, N.Y.; Clawson, D.K.; Gifford-Moore, D.S.; Klimkowski, V.J., McCowan J.R., Smith G.F., Wikel J.H. Dibasic benzo[b]thiophene derivatives as a novel class of active site directed thrombin inhibitors. II: Exploring interactions at the proximal (S2) binding site. *Bioorg. Med. Chem. Lett.* **1998**, 8, 2527-2538.
- Lee, S.; Lee, H.; Yi, K.Y.; Lee, B.H.; Cho, N.S. 4-Substituted(benzo[b]thiophene-2carbonyl)guanidines as novel Na⁺/H⁺ exchanger isoform-1 (NHE-1) inhibitors. *Bioorg. Med. Chem. Lett.* 2005, 15, 2998-3001.
- Bastian, J.A.; Chrigadze; N.; Denny, M.L.; Gifford-Moore, D.S.; Sall, D.J.; Smith, G.F.; Wikel, J.H. Diamino benzo[b]thiophene derivatives as a novel class of active site directed thrombin inhibitors. III: Enhancing activity by imposing conformational restriction in the C-4" side chain. *Bioorg. Med. Chem. Lett.* **1999**, 9, 363-368.
- Wang, Y.; Burton, D.J. Site-specific preparation of 4-substituted-6-fluoro(carboalkoxyl) benzo[b]furans and benzo[b]thiophenes via base-catalyzed cyclization of enyne derivatives. *J. Fluorine Chem.* 2007, 128, 1052-1057.
- Pinto, E.; Queiroz, M.J.; Vale-Silva, L.A.; Oliveira, J.F.; Begouin, A.; Begouin, J.M.; Kirsch, G. Antifungal activity of synthetic di(hetero)arylamines based on the benzo[b]thiophene moiety. J. Bioorg. Med. Chem. 2008, 16, 8172-8177.
- Eicher, T.; Hauptman, S.; Speicher, A. The Chemistry of Heterocycles: Chapter 5, Five Membered Heterocycles, Section 5.6 Thiophene, Wiley-VCH: New York; 2003.
- King, W.J.; Nord, F.F. Studies in the thiophene series. V: Wolff-Kishner reductions. J. Org. Chem. 1949, 14, 638-642.
- Wu, C.; Decker, E.R.; Blok, N.; Bui, H., You, T.J.; Wang, J.; Bourgoyne, A.R.; Knowles, V.; Berens, K.L.; Holland, G.W.; Brock, T.A.; Dixon, R.A.F. Discovery, modeling, and human pharmacokinetics of N-(2-acetyl-4,6-dimethylphenyl)-3-(3,4-dimethylisoxazol-5ylsulfamoyl)thiophene-2-carboxamide (TBC3711), a second generation, ET A selective, and orally bioavailable endothelin antagonist 1. J. Med. Chem. 2004, 47, 1969-1986.
- Dure, K.; Dubus, S.; Ho, H.A.; Levesque, I.; Brunette, M.; Corbeil, G.; Boissinot, M.; Boivin, G.; Bergeron, M.G.; Boudreau, D.; Leclerc, M. Fluorescent polymeric transducerfor the rapid, simple, and specific detection of nucleic acids at the zeptomole level. J. Am. Chem. Soc. 2004, 126, 4240-4244.
- Mohareb, R.M.; Wardakhan, W.W.; Ibrahim, R.A. Synthesis of pyridine, pyran and thiazole containing thiophene derivatives and their anti-tumor evaluations. *Med. Chem. Res.* 2016, 25, 2187-2204.
- Mohareb, R.M.; Ibrahim, R.A. Design, cytotoxicity and toxicity of new thiophene and thieno [2,3-b] pyridine derivatives. *Med. Chem. Res.* 2017, 26, 587-602.
- Mohareb, R.M.; Abbas N.S., Ibrahim, R.A. New approaches for the synthesis of thiophene derivatives with anti-tumor activities. *Acta Chim. Slov.* 2013, 60, 583-594.
- Rost, C.; Karg, S.; Riess, W.; Loi, M.A.; Murgia, M.; Muccini, M. Ambipolar light emitting organic field-effect transistor. *Appl. Phys. Lett.* 2004, 85, 1613-1615.
- Vriezema, D.M.; Hoogboom, J.; Veloonia, K.; Takazawa, K.; Christianen, P.C.M.; Maan, J.C.; Rowan, A.E.; Nolte R.J.M. Vesicles and polymerized vesicles from thiophenecontaining rod-coil block copolymers. *Angew. Chem. Int. Ed.* 2003, 42, 772-776.
- Yu, H.; Pullen A.E.; Beuschel, M.G.; Swager, T.M. Charge-specific interactions in segmented conducting polymers: An approach to selective ionoresistive responses. *Angew. Chem. Int. Ed.* 2004, 43, 3700-3703.
- 19. Sally, E.; Wenzel, M.D. New approaches to anti-inflammatory therapy for asthma. *Am. J. Med.* **1998**, 104, 287-300.

- Angerer, E.; Erber, S. 3-Alkyl-2-phenylbenzo[b]thiophenes: Nonsteroidal estrogen antagonists with mammary tumor inhibiting activity. J. Steroid Biochem. Mol. Biol. 1992, 41, 557-562.
- Malaisa, O.; Neuprez, A.; Reginster, J.Y. Raitements non hormonaux de l'ostéoporose postménopausique. *Gynécologie Obstétrique & Fertilité* 2008, 36, 815-822.
- Luo, Z.; Liu, Z.; Yang, Z. The synthesis and photoactivated cytotoxicity of novel 5-phenyl 3-(2,2':5',2"-terthien-5-yl)-4,5-dihydro-1H-pyrazolines. *Chin. Chem. Lett.* 2014, 25, 333-336.
- Shchekotikhin, A.E.; Glazunova, V.A.; Dezhenkova, L.G; Luzikov, Y.N.; Sinkevich, Y.B.; Kovalenko, L.V.; Buyanov, V.N.; Balzarini, J.; Huang, F.; Lin, J.; Huang, H.; Shtil, A.A.; Preobrazhenskaya M.N. Synthesis and cytotoxic properties of 4,11-bis[(aminoethyl)amino]anthra[2,3-b]thiophene-5,10-diones, novel analogues of antitumor anthracene-9,10. *Bioorg. Med. Chem.* 2009, 17, 1861-1869.