FURTHER FLAVONOL GLYCOSIDES OF EMBELIA SCHIMPERI LEAVES

Lawrence O. Arot Manguro^{1*}, Ivar Ugi² and Peter Lemen²

¹Nairobi University, Chemistry Department, P.O. Box 30197, Nairobi, Kenya ²Technische Universität München, Institut für Organische Chemie und Biochemie, Lerhstul 1, Lichtenbergstrasse 4, 85747-Graching, Germany

(Received April 13, 2003; revised October 28, 2003)

ABSTRACT. Fractionation of the methanolic extract of *Embelia schimperi* leaves has led to the isolation of two novel flavonol glycosides. The compounds were characterized as isorhamnetin 3-O- β -galactosysl (1 \rightarrow 4)- β -galactoside and quercetin 3-O-[α -rhamnosyl (1 \rightarrow 2)] [α -rhamnosyl (1 \rightarrow 4)]- α -rhamnoside. Also reported from the same extracts were known compounds quercetin, myricetin, quercetin 3-O- α -rhamnoside, quercetin 3-O- β -glucoside, quercetin 3-O-rutinoside, myricetin 3-O- β -xyloside, isorhamentin 3-O- β -glucoside and myricetin 3-O- β -glucoside. Their structural elucidation was accomplished using spectral measurements and chemical methods.

KEY WORDS: *Embelia schimperi*, Flavonol glycosides, Isorhamnetin 3-O-β-galactoysyl (1 \rightarrow 4)-β-galactoside, Quercetin 3-O-[α-rhamnosyl (1 \rightarrow 2)] [α-rhamnosyl (1 \rightarrow 4)]-α-rhamnoside

INTRODUCTION

Embelia schimperi Vatke is one of the five Myrsinaceae species endemic to Kenya and is widely used in traditional medicine as an anthelmintic and anti-microbial [1]. Phytochemically, the plant is typified by the presence of benzoquinone derivatives; 2,5-dihydroxy-3-undecyl-1,4-benzoquinone (trivial name: embelin) and 2,5-dihydroxy-3-tridecyl-1,4-benzoquinone (trivial name: rapanone) [2, 3]. In a recent study [4], a novel flavonol glycoside along with four known ones has been isolated from this source. In continuing investigation of a methanolic extract of the plant leaves, we now report the isolation and characterization of two new flavonol glycosides (1 and 2). Also being reported are eight known ones identified as isorhamnetin 3-O-β-glucoside (3), quercetin 3-O-β-glucoside (4), quercetin (5), myricetin (6), quercetin 3-O-β-glucoside (7), myricetin 3-O-β-glucoside (8), myricetin 3-O-β-xyloside (9) and quercetin 3-rutinoside (10) [5-7].

1 R = galactosyl (1 \rightarrow 4)-galactoside, R₁ = Me; 2 R = [rhamnosyl (1 \rightarrow 2)][rhamnosyl (1 \rightarrow 4)]-rhamnoside, R₁ = H; 3 R = glucose, R₁ = Me; 4 R = rhamnose R₁ = H.

^{*}Corresponding author. E-mail: kamanguro@yahoo.com

RESULTS AND DISCUSSION

Compound 1 afforded UV data, which suggested free hydroxyls at C-5, C-7, and C-4 [8-10]. The assignment of the 1H and ^{13}C NMR data supported by NOESY and HMBC correlations confirmed the aglycone as isorhamnetin (3'-O-methylquercetin) substituted at C-3. Acid hydrolysis gave galactose as the sugar residue. The FAB mass spectrum results indicated galactosylgalactoside bios as evidenced by the cleavage behaviour (see experimental section). In the ^{13}C NMR spectrum a signal due to C-4" of residual galactose shifted downfield at δ 82.40 in comparison to isorhamnetin 3-O- β -glucoside (3), suggesting interglycosidic linkage at this position. This was corroborated by HMBC correlation between the anomeric proton (δ 4.65) of terminal galactose and a carbon (δ 82.40) assigned to C-4" of the residual galactose and further confirmed by NOESY cross peaks between the terminal galactose anomeric proton (δ 4.65) and the C-4" proton (δ 3.68). Therefore compound 1 was concluded as isorhamnetin 3-O- β -galactosyl ($1\rightarrow$ 4)- β -galactoside.

Figure 1. Pertinent HMBC and NOESY correlations observed with compound 2.

Compound **2** showed a molecular ion peak at m/z 741 [M+H]⁺, analyzing for C_{33} H₄₀O₁₉. Acid hydrolysis afforded rhamnose as the sugar residue. Its ¹H NMR data showed a 2H AX and a 3H ABX system characteristic of quercetin moiety [11] while the UV data recorded in MeOH and with addition of common shift reagents [8, 9] suggested that the residual sugar was attached at C-3 position of the aglycone, a fact further supported by HMBC correlations as outlined in Figure 1. From the mass spectrum data, fragments at m/z 595 [M-146 + H]⁺ (loss of rhamnose), 449 [M-2 x 146 + H]⁺ (loss of two rhamnoses), 303 [M-3 x 146 + H]⁺ (loss of three rhamnoses), 339 [3 x rhamnose]⁺, 293 [2 x rhamnose]⁺ (Figure 2) together with three anomeric protons of rhamnose units at δ 5.10 (d, J = 1.1 Hz), 4.75 (d, J = 1.2 Hz) and 4.50 (d, J = 1.0 Hz) indicated the presence of three sugar residues in the molecule. Furthermore, a comparative analysis of the ¹³C NMR chemical shift of the sugar unit with those of quercetin 3-O- α -rhamnoside (4) showed glycosylation shift for C-2" (11.80 ppm) and C-4" (10.30 ppm) in the residual ramnose unit, thus suggesting the presence of 2",4"-substituted rhamnose unit bearing two other terminal ones. The signal at δ 82.30 attributed to C-2" of the residual rhamnose suggested rhamnosyl (1 \rightarrow 2)-rhamnoside moiety [12] while the signal at δ 81.60 attributed to C-4" of the same rhamnose

Bull. Chem. Soc. Ethiop. 2004, 18(1)

indicated a rhamnosyl (1 \rightarrow 4)-rhamnoside moiety [13], a fact further supported by NOESY and HMBC spectra correlations (Figure 1). On this basis compound **2** was established as quercetin 3-O-(α -rhamnosyl (1 \rightarrow 2)] [α -rhamnosyl (1 \rightarrow 4)]- α -rhamnoside.

Figure 2. Fragmentation of compound 2 during FAB.

Bull. Chem. Soc. Ethiop. 2004, 18(1)

Table 1. ¹³C NMR data of compounds 1-4.

Carbon No.	1	2	3	4
1				
2	156.80	157.30	158.70	158.30
3	135.75	134.20	135.40	135.60
4	177.40	177.70	179.40	178.65
5	161.50	161.30	163.04	162.30
6	99.01	98.70	98.86	98.80
7	164.45	164.20	165.91	163.70
8	94.00	93.80	94.73	94.02
9	156.10	156.80	158.20	157.40
10	104.10	104.00	105.80	104.50
1'	120.75	120.70	123.10	122.35
2'	113.90	115.60	116.00	116.60
3'	148.35	147.80	148.40	149.10
4'	147.16	145.20	150.80	148.20
5'	115.65	115.40	114.40	116.00
6'	121.90	121.10	123.80	123.50
1"	103.80	101.20	104.71	102.10
2"	74.30	82.30	75.90	70.50
3"	76.50	78.01	77.08	71.90
4''	82.40	81.60	71.50	71.30
5"	73.00	76.80	76.50	69.70
6''	61.90	18.80	62.60	17.80
1'"	104.80	99.80		
2'"	72.45	70.40		
3'"	74.85	72.90		
4'"	68.70	71.00		
5'"	75.30	70.10		
6'''	60.50	18.30		
1''''		101.10		
2''''		71.40		
3''''		73.50		
4''''		72.00		
5''''		69.80		
6''''		17.70		
3'-OMe	57.60		58.20	

EXPERIMENTAL

General. UV spectra were recorded on a 8452 A Hewlett Packard Array spectrophotometer. ¹H and ¹³C NMR spectra were recorded in CDCl₃-DMSO-d₆ mixture and DMSO-d₆ on a Bucker WM instrument operating at 360 and 90 MHz, respectively. EIMS were measured using 70 eV MAT 8200 A Varian Bremen instrument. Polish Academy of Sciences provided FAB-MS spectra. Silica gel for column and TLC plates were impregnated with 2% oxalic acid solution. Semi-preparative high performance liquid chromatography (HPLC) was performed on a Bischoff instrument connected to a 785 programmable absorbance detector and a programmable monitor 8252 dual pen recorder.

Plant material. Embelia schimperi leaves were collected from Ngong hills in June 1997 and voucher specimens were identified after comparison with authentic samples at Botany Department Herbarium, Nairobi University.

Extraction and isolation. The CH₂Cl₂ defatted powdered leaves (approx. 1 kg) was further extracted with MeOH (2.5 L x 3) for one week. The extracts were combined and freed from the solvent under reduced pressure to give a dark green residue (105 g). A portion of the extract (100 g) was subjected to column chromatography with CH₂Cl₂-MeOH gradient to pure MeOH affording 160 fractions of 100 mL each. The composition of the fractions were determined by TLC using solvent system CH₂Cl₂-MeOH (9:1, 3:2) and those showing similar profiles were combined to give three major pools. The first pool (fractions 20-80, 25 g) was subsequently subjected to repeated flash chromatography with CH₂Cl₂-MeOH (95:5) followed by the same solvent system in the ratio 9:1 to give 5 (29 mg), 6 (55 mg) and 4 (24 mg). Fractions 81-105 (Pool II) was evaporated to give 17 g which upon repeated flash chromatography with CH₂Cl₂-MeOH (4:1) and collecting 10 mL each, afforded 7 (66 mg), 8 (34 mg), 3 (65 mg) and 9 (35 mg). Pool III (fractions 106-160, 35 g) mainly from methanol elution was further purified by column chromatography over sephadex LH-20 using different solvent systems. Final purification of the eluates by preparative HPLC on reverse phase (RP-18) using MeOH-H₂O (7:3) gave 10, 1 and 2 in 75 mg, 48 mg, and 55 mg, respectively.

Isorhamnetin 3-O-β-galactosyl ($I \rightarrow A$)-β-galactoside (I). Amorphous yellow powder. UV λ_{max} (MeOH) nm: 257, 268, 358; (+AlCl₃) 268, 301, 360, 398; (+AlCl₃/HCl) 262, 302, 402, 424; (+NaOMe) 272, 324, 420; (+NaOAc) 270, 322, 408; (NaOAc/H₃BO₃) 256, 266, 362. IR ν_{max} (KBr) cm⁻¹: 3510, 3450, 1680, 1650, 1580, 1460, 1410, 1390, 1365, 1250, 1200, 1180, 1050, 900. ¹H NMR (CDCl₃+one drop DMSO-d₆) δ: 12.64 (s, OH-5, D₂O exchang.), 10.20 (br s, OH-7, D₂O exchang.), 8.30 (br s, OH-4', D₂O exchang.), 7.90 (d, J = 2.3 Hz, H-2'), 7.80 (dd, J = 8.2, 2.3 Hz, H-6'), 6.98 (d, J = 8.2 Hz, H-5'), 6.50 (d, J = 1.8 Hz, H-8), 6.35 (d, J = 1.8 Hz, H-6), 3.85 (s, OMe); residual galactose: 5.30 (d, J = 7.7 Hz, H-1"), 3.78 (d, J = 12.1, 3.2 Hz, H-6"_B), 3.68 (m, H-4"), 3.60 (dd, J = 12.1, 5.4 Hz, H-6"_A), 3.40 (m, H-3"), 3.25 (m, H-5"), 3.16 (m, H-2"); terminal galactose: 4.65 (d, J = 7.60 Hz, H-1"'), 3.97 (m, H-6"_B), 3.73 (m, H-6"_A), 3.55 (m, H-4"'), 3.45 (m, H-3"'), 3.30 (m, H-5"'), 3.27 (m, H-2"'). ¹³C NMR data: see Table 1. EIMS (70 eV): m/z (%) 302 (100), 153 (27), 137 (18), 136 (9). FAB-MS (positive ion mode): 641 [M+H]⁺, 479 [M-162+H]⁺, 317 [M-2 x 162+H]⁺, 153, 137.

Quercetin 3-O- $[\alpha$ -rhamnosyl $(1\rightarrow 2)]$ $[\alpha$ -rhamnosyl $[1\rightarrow 4]$ - α -rhamnoside (2). A yellow powder. UV λ_{max} (MeOH) nm: 258, 301, 358; (+AlCl₃) 274, 300, 430; (+AlCl₃/HCl) 270, 300, 402; (+NaOMe) 272, 328, 414; (+NaOAc) 273, 324, 380, 418; (NaOAc/ H_3BO_3) 260, 379. IR v_{max} (KBr) cm⁻¹: 3500, 1665, 1570, 1510, 1430, 1250, 950. ¹H NMR (99% DMSO-d₆) δ: 12.80 (s, OH-5, D₂O exchang.), 10.30 (br s, OH-7, D₂O exchang.), 9.20-8.50 (br s, OH-3' and 4', D₂O exchang.), 7.80 (d, J = 2 Hz, H-2'), 7.64 (dd, J = 8.4, 2 Hz, H-6'), 6.80 (d, J = 8.3 Hz, H-5'), 6.40 (d, J = 2.2 Hz, H-8), 6.24 (d, J = 2.2 Hz, H-6); residual rhamnose: 5.10 (d, J = 1.1 Hz, H-1), 3.70 (d, J = 9.3, 7.6 Hz, H-2"), 3.47 (t, J = 9.2 Hz, H-3"), 3.36 (t, J = 9.1 Hz, H-4"), 3.20 (ddd, J = 9.1 Hz, H-4")= 9.1, 5.3, 2.3Hz, H-5"), 1.20 (d, J = 6.5 Hz, Me-6"); 2"-O-rhamnose: 4.50 (d, J = 1.0 Hz, H-1""), 3.45 (dd, J = 9.3, 7.5 Hz, H-2'''), 3.39 (t, J = 9 Hz, H-3'''), 3.34 (t, J = 9.1 Hz, H-4'''), 3.24 (dd, J = 9.1 Hz, H-2'''), 3.24 (dd, J = 9.1 Hz, H-2''''), 3.= 9.1, 5.2, 2.5 Hz, H-5"), 1.0 (d, J = 6.6 Hz, Me-6"); 4"-O-rhamnose: 4.75 (d, J = 1.2 Hz, H-1""), 3.85 (d, J = 3.3 Hz, H-2""), 3.60 (dd, J = 8.5, 3.3 Hz, H-3""), 3.30 (d, J = 11, 6.2 Hz, H-5""), 3.20 (dd, J = 9.5, 7.8 Hz, H-4""), 0.80 (d, J = 6.7 Hz, Me-6""). ¹³C NMR data: see Table 1. EIMS (70 eV): m/z (%) 302 (100), 162 (4), 153 (20), 147 (7), 137 (9), FAB-MS (positive ion mode): 741 [M+H]⁺, 595 [M-146+H]⁺, 449 [M-2 x 146+H]⁺, 303 [M-3 x 146+H]⁺, 339, 293, 180, 151, 146, 137.

Isorhamnetin 3-O-β-glucoside (3). A pale yellow powder. UV λ_{max} (MeOH) nm: 260, 304, 360; (+AlCl₃) 268, 302, 402; (+AlCl₃/HCl) 272, 302, 398; (+NaOMe) 272, 396; (+NaOAc) 270, 320, 360, 406; (NaOAc/H₃BO₃) 260, 370. IR ν_{max} (KBr) cm⁻¹: 3450, 1650, 1600, 1580, 1440, 1360, 1290, 1140, 1050, 990. ¹H NMR (CDCl₃+one drop DMSO-d₆) δ: 12.50 (s, OH-5, D₂O exchang.), 10.25 (br s, OH-7, D₂O exchang.), 8.60 (br s, OH-4', D₂O exchang.), 7.82 (d, J = 1.9 Hz, H-2'), 7.46 (dd, J = 8.4, 2.0 Hz, H-6'), 6.85 (d, J = 8.4, 2.0 Hz, H-5'), 6.31 (d, J = 2.1 Hz, H-8), 6.16 (d, J = 2.1 Hz, H-6), 3.85 (s, OMe); residual glucose: 5.10 (d, J = 7.6 Hz, H-1"), 3.80 (dd, J = 11.6, 3.6 Hz, H-6"_B), 3.66 (dd, J = 11.6, 5.8 Hz, H-6"_A), 3.50 (t, J = 8.4 Hz, H-4"), 3.45 (m, H-3"), 3.37 (dd, J = 10.2,7.7 Hz, H-2"), 3.28 (m, H-4"). ¹³C NMR data: see Table 1. EIMS (70 eV): m/z (%) 316 (100), 245 (10), 150 (5), 128 (10), 83 (57). FAB-MS (positive ion mode): 449 [M+H]⁺, 317 [M-162+H]⁺, 151, 137.

Quercetin 3-O-α-rhamnoside (4). A yellow amorphous powder. UV λ_{max} (MeOH) nm: 260, 300, 357; (+AlCl₃) 272, 301, 428; (+AlCl₃/HCl) 272, 301, 401; (+NaOMe) 271, 330, 414; (+NaOAc) 274, 320, 384; (NaOAc/H₃BO₃) 260, 382. IR ν_{max} (KBr) cm⁻¹: 3500, 1665, 1580, 1520, 1480, 1450, 1120, and 1020. ¹H NMR (CDCl₃+ DMSO-d₆) δ: 12.45 (s, OH-5, D₂O exchang.), 10.50 (br s, OH-7, D₂O exchang.), 9.20-8.50 (br s, OH-3' and 4', D₂O exchang.), 7.86 (d, J = 2.2 Hz, H-2'), 7.70 (dd, J = 8.4, 2.3 Hz, H-6'), 6.90 (d, J = 8.4 Hz, H-5'), 6.40 (d, J = 2 Hz, H-8), 6.20 (d, J = 2 Hz, H-6); rhamnose: 5.15 (d, J = 0.9 Hz, H-1"), 3.65 (br d, J = 3.5 Hz, H-2"), 3.55 (dd, J = 9.1, 3.1Hz, H-3"), 3.44 (m, H-4"), 3.35 (m, H-5"), 1.00 (d, J = 6.5 Hz, Me-6"). ¹³C NMR data: see Table 1. EIMS (70 eV): m/z (%) 302 (100), 153 (21), 137 (30).

Acid hydrolysis. Compounds (1-4), 15 mg each in a mixture of 8% HCl (2 mL) and MeOH (20 mL) were separately refluxed for 2 h. The reaction mixtures were reduced under pressure to dryness, dissolved in H_2O (3 mL) and neutralized with NaOH. The neutralized products were subjected to TLC analysis (eluent, EtOAc-MeOH- H_2O -HOAc, 6:2:1:1). The chromatograms were sprayed with aniline hydrogen phthalate followed by heating at 100 ^{0}C . The sugars were identified after comparison with authentic samples.

ACKNOWLEDGEMENTS

The authors are thankful to Alexander von Humboldt Foundation for awarding Lawrence Manguro a fellowship at Technische Universitaet Muenchen. Polish Academy of Sciences is acknowledged for providing FAB-MS data. Mr S.G. Mathenge of Botany Department, University of Nairobi is appreciated for identification and collection of plant material.

REFERENCES

- Kokwaro, J.O. Medicinal Plants of East Africa, East African Literature Bureau: Nairobi; 1976; p 49.
- 2. Midiwo, J.O.; Manguro Arot, L.O.; Mbakaya, C.L. Bull. Chem. Soc. Ethiop. 1988, 2, 83.
- 3. Midiwo, J.O.; Manguro Arot, L.O.; Odingo, J.O. Bull. Chem. Soc. Ethiop. 1990, 4, 71.
- 4. Manguro Arot, L.O.; Williams, L.A.D. Phytochemistry 1997, 44, 1398.
- 5. Manguro Arot, L.O.; Williams, L.A.D. Planta Med. 1996, 62, 178.
- 6. Manguro Arot, L.O.; Midiwo, J.O.; Kraus, W. Phytochemistry 1996, 43, 1107.
- 7. Manguro Arot, L.O.; Midiwo, J.O.; Kraus, W. Nat. Prod. Lett. 1996, 9, 121.

- 8. Markham, K.R.; Chari, V.M.; Mabry, T.J. in *The Flavonoid Advances in Research*, Harbone, J.B.; Mabry, T.J. (Eds.); Chapman and Hall: London; **1982**; pp 28-52.
- 9. Mabry, T.J.; Markham, K.R.; Thomas, M.B. *The Systematic Identification of Flavonoids*, Springer-Verlag: Berlin; **1970**; Chapter 1.
- 10. Markham, K.R. *Techniques in Flavonoids Identification*, Academic Press: London; **1982**; Chapters 1 and 2.
- 11. Manguro Arot, L.O.; Mukonyi, K.W.; Githiomi, J.K. Nat. Prod. Lett. 1995, 7, 163.
- 12. Markham, K.R.; Geiger, H.; Jaggy, H. Phytochemistry 1992, 31, 1009.
- 13. Shao, Y.; Poobrasert, O.; Kenelly, E.J.; Chin, C.; Chi-Tang, H.; Huang, M.; Garrison, S.A.; Cordell, G.A. *Planta Med.* **1997**, 63, 258.