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Bioassay Guided Chromatographic Isolation Of Lactation Inducing Agent From *Hippocratea Obtusifolia*

¹Datti, Y. and ²Idris, M. M.

¹Department of Chemistry, Northwest University, Kano. ²Department of Pure and Applied Chemistry, Bayero University, Kano. Email: whydee@yahoo.co.uk

ABSTRACT

The powdered leaves of *Hippocratea obtusifolia* (commonly known as *gwadayi* in Hausa) was percolated with ethanol and then sequentially extracted with petroleum ether, chloroform, ethyl acetate and methanol. The extracts were respectively labeled HO1, HO2, HO3, and HO4, with the ethanolic extract labeled HO. The extracts were subjected to bioactivity against lactation inducement, and the chloroform extract labeled HO2 was found to be the most active. The HO2 extract was then subjected to column and thin layer chromatography with similar fractions pooled together. The pooled fractions were then subjected to bioactivity testing against lactation inducement, and the fraction labeled HO2-13 was found to be the most active. The fraction labeled HO2-13 was found to be the most active to bioactivity testing against lactation inducement, and the fraction labeled HO2-13 was found to be the most active. The fraction HO2-13-111 was subjected to gether. The pooled fraction HO2-13-111 was subjected to be the most active. The fraction HO2-13-111 was subjected spectroscopic analysis, using Infrared, Mass spectroscopy, and both ¹H and ¹³C NMR data obtained. The structure of the compound present in fraction HO2-13-111 was then elucidated based on the obtained data.

Keywords: Force-Feeding, Lactation, Progesterone, Thin-Layer Column Chromatography.

INTRODUCTION

One outstanding feature that distinguishes the class mammalia from the rest of the members of the kingdom animalia, is that their young ones are fed from the milk secreted by the mothers mammary glands (Neville and Peaker 1984). The provision of a specialized maternal body fluid for neonatal nutrition allows birth to occur at an early stage of development and provide time of intense maternal infant interaction during early behavioral development. In addition, the nutritional reserves of the mothers can sustain the infant through a period of famine, as it contains all the nutrients required for the development of the body (Collier *et al*, 1977). These nutrients include carbohydrates, proteins, lipids and minerals (Hurley, 1998).

Lactogenesis is the term meaning the initiation of lactation. Lactation is the process of functional differentiation which mammary tissue undergoes when changing from a non-lactating to a lactating stage. This process is normally associated with the end of pregnancy and around the time of parturition (Nagasawa *et al.*, 1969). Because lactogenesis is particularly dependent upon specific set of hormones (called lactogenic complex of hormones). Mammary tissue from most state of non-lactating mammary gland also can be made to undergo some degree of lactogenesis by administration of high amount of such hormones even in non-pregnant animals (Davis *et al.*, 1983). The production of breast milk is controlled by an interplay of various hormones, with prolactin being the predominant hormone involved. The maturation of breast tissue, resulting in milk production, is controlled by many other factors besides prolactin, including estrogen, progesterone, insulin, growth hormone, cortisol, thyroxine, and human placental lactogen (Michael, 2008).

Throughout human history, there have been infants who have been nursed by surrogate mothers. Non-maternal lactation may have occurred as a results of maternal death or illness or because the birth mother gave over, or shared the care of her baby with another mother (Akers *et al.*, 1981). Sometimes the surrogate mother was already breastfeeding another baby and milk supply simply increased due to additional demand, to meet the growth needs for two or more babies. In the event that no already-nursing mother was available, reports from several continents describe effort by non-lactating woman to induce lactation by putting the baby to breast (Turner *et al.*, 1956).

The inducement of lactation without the needs of a proceeding pregnancy could offer several practical and economical advantages (Bauman *et al.*, 1977). Lactation has been induced in ruminants using steroids (progesterone and estrogen) as a short-term treatment followed by a

wide variety of drugs aimed in increasing prolactin secretion. Those treatment protocols have resulted in lactation with milk production ranging between 25% and 82% of physiological post-partum lactation (Palmer *et al.*, 2002).

Inducement of lactation has also been achieved both locally and traditionally, especially Hausa communities. the Factors among necessitating the need to have the lactation induced, include among others, the insufficiency of the milk from the birth mother, and the need to induce lactation on a birth mother, (who could not produce the milk). The method employed is mainly dependent on the necessitating factor: As for the insufficiency of the milk from the birth mother, the use of a local porridge called kunun kanwa (literally potash gruel) is employed. By immediately taking this porridge, hot and steaming, it is widely believed and accepted that the milk content will be increased to nearly maximum. Even though, there has not been any scientific proof to this, this tradition alongside the role played by the famous wankan jego (the hot water birth) in solving the problem of milk insufficiency, is widely accepted within the Hausa culture and tradition.

However, when it comes to the situation where there is no milk at all, in many cases use is made of some plants that are believed to be useful for that purpose. In India lactation inducing herbal remedies are commonly used by nursing mothers to meet the nutritional requirement of their children (Nudrat et al., 2007). Notable plants used by the Warlis tribe in India to induce lactation in nursing mothers include Carica papava Linn, Asparagus racemosa Wt, Trigonella foenum graecum Linn and Madhuca indica Gmellin (Nudrat et al., 2007). Other notable plants believed to be useful for the purpose of inducement of lactation, are Kigelia aethiopica, known as hantsar giwa in Hausa, as well as Hippocratea obtusofolia or gwadayi in Hausa.

Hippocratea obtusofolia is a species of flowering plant in the family *Hippocrateacea* which occur in some part of West African countries like Nigeria, Niger, Senegal and Cameroon. *Hippocratea obtusifolia*, a shrub with a green leaves and stalk growing up to 1m tall, is an evergreen where rainfall occurs throughout the year, but deciduous where there is long dry season, it produces sticky white fluid of detached (Huxley, 1992).

AIMS AND OBJECTIVES

This research is aimed at evaluating, isolating and structure elucidation of the bioactive lactating inducing agent from *Hippocratea obtusifolia*.

MATERIALS AND METHOD The Plant Materials

The leaves and stalk of *Hippocratea* obtusofolia was collected at Dawaki, in Dawakin Tofa Local Government Area of Kano state, and were identified and authenticated at the Department of Biological Science, Bayero University, Kano.

Extraction Of The Plant

Five hundred grams of the dried and ground form of the plant Hippocratea obtusofolia was put in 2.5dm³ brown capacity brown bottle, and 1.5dm³ of ethanol added. The set up was left for two weeks with constant shaking, after which the extract was filtered and then concentrated using rotavapor (Vishnoi, 1979) to get the crude extract labeled HO. Part of the crude residue HO was then dissolved using aqueous methanol, and then partitioned with petroleum ether using separating funnel to get the petroleum ether fraction. The petroleum ether fraction was concentrated, dried, weighed and labeled HO1. The aqueous methanol portion was further partitioned with chloroform to get the chloroform fraction. The chloroform fraction was concentrated, dried, weighed and labeled as HO2. The aqueous methanol portion was then partitioned with ethyl acetate to get the ethyl acetate fraction. The ethyl acetate fraction was concentrated, dried, weighed and labeled as HO3. The final aqueous methanol portion was then concentrated, dried, weighed and labeled as HO4. Each of the five extracts HO, HO1, HO2, HO3 and HO4 was tested for lactation inducing activity.

Activity Testing

All the five fractions HO, HO1, HO2, HO3 and HO4 were subjected to bio-activity test. Twenty four female guinea pigs of average weight of 200g were randomly divided into 6 groups of four each. The first five groups A, B, C, D and E were each force-fed with the labeled portion HO, HO1, HO2, HO3 and HO4 of Hippocratea obtusofolia respectively for a period of two weeks. Each guinea pig received 2g/Kg body weight of the extract dissolved in $2cm^3$ of distilled water. The force-feeding was carried out using a 2cm³ syringe and endogastric tube. The last group F served as the control and was given distilled water free from the sample extract (Salim, et al., 2005). After 2 weeks each group was quantitatively examined for the presence or any sign of lactation (Palmer et al., 2002). At the end the chloroform extract HO2 was found to be the most active.

Purification Of The Chloroform Fraction (HO2)

Having found that the chloroform fraction HO2 is the most active, it was then subjected to column chromatography. The column was eluted using the solvents in the following order: petroleum ether (1dm³), petroleum ether:dichloromethane (7:3, 2dm³), petroleum ether:dichloromethane (3:2,

 $1.5 dm^3$), petroleum ether:dichloromethane (1:1, 1dm³), petroleum ether:dichloromethane (2:3, 1.2dm^3), petroleum ether:dichloromethane (3:7, $1 dm^{3}$). dichloromethane $(1.5 dm^3).$ dichloromethane:ethylacetate (4:1. $1 dm^{3}$), dichloromethane:ethylacetate (1:1, $1 dm^{3}$), dichloromethane:ethylacetate (1:4, $1 dm^{3}$), ethylacetate (1dm³). The eluents were collected in 100cm³ volumes. All the fractions were spotted on a TLC plate coated with silica gel, and fractions that are similar were combined and labeled as pooled fractions HO2-13, HO2-40, HO2-60, HO2-79, HO2-90 and HO2-110.

Bioactive Evaluation Of The Pooled Fractions

Three of the five pooled fractions, HO2-13, HO2-60 and HO2-90 that were the most abundant, were subjected to bio-activity test. Eight female guinea pigs of average weight of 200g were randomly divided into 4 groups of 2 guinea pigs each (Salim, et al., 2005). The first three groups G, H and I were each force-fed with the three pooled fractions HO2-13, HO2-60 and HO2-90, for a period of two weeks. Each guinea pig received 1g/kg body weight of the extract dissolved in $2cm^3$ of distilled water and force-feeding was carried out using a 2cm³ syringe and endogastric tube. The last group J served as the control and was given distilled water free from the sample extract (Salim, et al., 2005). After 2 weeks each group was quantitatively examined for the presence or any sign of lactation (Palmer et al., 2002). At the end the chloroform extract HO2-13 was found to be the most active.

Purification Of Fraction HO2-13 (Active Fraction)

Having found that fraction HO2-13 showed appreciable activity, it was then subjected to micro-column chromatography. The order of elution was petroleum ether (500mls), petroleum ether:dichloromethane (9:1, 500mls), petroleum ether:dichloromethane (4:1, 500mls), petroleum ether:dichloromethane (7:3, 500mls), petroleum ether:dichloromethane (3:2, 500mls), petroleum

ether:dichloromethane (1:1, 500mls), petroleum ether:dichloromethane (2:3, 500mls), petroleum ether:dichloromethane (3:7, 500mls), petroleum ether:dichloromethane (1:4, 500mls), petroleum ether:dichloromethane (1:9, 500mls), dichloromethane (500mls). The eluents were collected in 20 cm^3 volumes. All the fractions were spotted on a TLC plate coated with silica gel, and fractions that are similar were combined and labeled as pooled fractions HO2-13-15, HO2-13-111 and HO2-13-135.

Bioactivity Evaluation Of The Pooled Fractions (II)

The pooled fractions, HO2-13-111 and HO2-13-149 were the subjected to bio-activity test. Six female guinea pigs of average weight of 200g were randomly divided into 3, with the first two groups K and L each force-fed with the two pooled fractions HO2-13-111 and HO2-13-149, for a period of two weeks. Each guinea pig received 0.5g/kg body weight of the extract dissolved in $2cm^3$ of distilled water and force-feeding was carried out using a $2cm^3$ syringe and endogastric tube. The last group M served as the control and was given distilled water free from the sample extract (Salim et al., 2005). After 2 weeks each group was quantitatively examined for the presence or any sign of lactation (Palmer et al., 2002). At the end the chloroform extract HO2-13-111 was found to be pure and the most active. The purity was confirmed by subjecting the isolated HO2-13-111 to TLC using varying solvent ratios, and in each case a single spot was obtained.

Spectroscopic Analysis

The active fraction HO2-13-111 was then subjected to spectroscopic analysis, and the parameters obtained are Infrared, mass spectroscopy and both ¹H and ¹³C-nmr.

RESULTS AND DISCUSSION

After subjecting all the five extracts to bioassay, the result obtained is as tabulated in Table 1.

Table 1: R	Result of E	Bioactivitv	Test 1
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EXTRACT	HO	HO1	HO2	HO3	HO4	CONTROL
ACTIVITY	+	-	+	-	-	-
Key: $+ = Extr$	act is active					
- = Extra	ct is not activ	e				

After subjecting the HO2 fraction to chromatography and TLC, as well as pooling the

fractions that were found to be similar, the following weights were obtained (Table 2).

FRACTION	R _F -VALUES	POOLED INTO	FRACTION	WEIGHT (g)		
			CODE			
6-34	0.72	13	HO2-13	2.23		
39-41	0.36	40	HO2-40	1.05		
44-70	0.29	60	HO2-60	1.32		
73-80	0.55	79	HO2-79	0.98		
85-99	0.42	90	HO2-90	1.97		
103-116	0.98	111	HO2-110	0.67		
Table 3: Result of Bioactivity Test 2						
EXTRACT F	HO2-13 HO2-4	40 HO2-60	НО2-79 Н	O2-90 HO2-110		

Table 2: Weight and Code of the Pooled Fractions 1

EXTRACT	HO2-13	HO2-40	HO2-60	HO2-79	HO2-90	HO2-110
ACTIVITY	+	-	-	-	-	-
Key: $+ = E$	xtract is active					
-=Ex	stract is not active	e				

After subjecting the HO2-13 fraction to chromatography and TLC, as well as pooling the

fractions that were found to be similar, the following weights were obtained (Table 4).

Table 4: Weight a	nd Code of the	Pooled Fracti	ons 2
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	FRACTION	R _F -VALUES	POOLED INTO	FRACTION	WEIGHT (mg)
				CODE	
	17-50	0.87	15	HO2-13-15	57.23
	60-123	0.56	111	HO2-13-111	98.32
	129-146	0.31	135	HO2-13-135	47.70

After subjecting the pure isolate HO2-13-111 to spectroscopic analysis using Infrared, mass spectroscopy, ¹H-nmr and ¹³C-nmr, and the results obtained are as shown in Figures 1- 6 on the attached spectral data sheets.

The IR (Fig.1) of HO2-13-111 shows a peak at 1727.4 cm⁻¹ representing a carbonyl functional group, a peak at 2936.7cm⁻¹ representing a C-H stretching, while the band at 1661.2cm⁻¹ can be attributed to =C-H bending of alkenes, and a peak at 1246.1cm⁻¹ representing a carbon-oxygen single bond stretching, while the peak seen at 1450.6cm⁻¹ may be attributed to to C-H bending of alkanes (Williams and Fleming 2004). The absence of an OH signal (3300-2500cm⁻¹) in the IR indicates that the carbonyl groups are not from carboxylic acids, and this left us with the possibility of having either a ketonic carbonyl group, or an ester, or both. However, the fact that we do not have an IR signal at 1750-1735 cm⁻¹, which is characteristic of esters, rather we have a peak 1727.4cm⁻¹, which is characteristic of ketone, means that the carbonyl group is from a ketone.

The ¹Hnmr (Fig. 2) reveals a cluster of peaks from 0-2ppm, which according to many literature consulted (Gonzalez *et al.*, 1981; Molnar *et al.*, 2006; Badgujar and Jain, 2009; Anand *et al.*, 2011) can be attributed to triterpenes. Other peaks

from the Hmnr include a sharp singlet at 2.00ppm, a singlet at 4.70ppm, a doublet at 2.25ppm, a quartet at 4.10ppm, a doublet of triplet at 4.38ppm, a triplet at 4.60ppm, a doublet of quartet at 2.31ppm, and a singlet at 3.7ppm.

The ¹³Cnmr data (Fig. 3) of HO2-13-111 showed the presence of resonances signifying the presence of 32 carbons. The results obtained from the dept spectrum (Fig. 4) of this compound reveal that 25 out of the 32 carbons are protonated, while the remaining 7 are unprotonated. Deduction from the dept (Table 5) shows that there are 9 methyl carbons, 11 methylene carbons, 5 methine carbons, and consequently 7 quaternary carbons. Notable signals from the ¹³Cnmr (Fig. 3) include a peak at $\partial 14.7$ ppm which may be associated with carbon 23, a peak at $\partial 171.8$ ppm which can be attributed to carbon 3, a peak at ∂142.5ppm which can be attributed to carbon 24, a peak at $\partial 129.8$ ppm which can be attributed to carbon 25, a peak at ∂ 81.5ppm which can be due to carbon 15, a peak at ∂ 55.6ppm which can be due to carbon 2, and a peak at $\partial 43.4$ which can be due to carbon 13. The fact that the proton on carbon 15 appeared as a doublet of triplet instead of a triplet, is an indication that there is a shift of a methyl group from carbon 14 to carbon 8 (Venkata and Chaturvedula 2013).

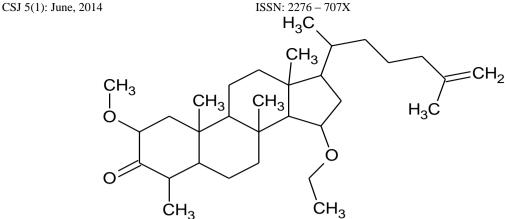
Joone

Table 5: (13C - Adept of HO2 – 13 – 111)

ADEPT SPECTRUM ANALYSIS

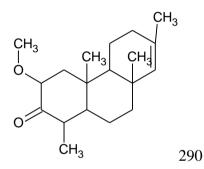
AUDEFI				
index 1 D 2 D 3 D 4 D 5 T 6 D 7 T 8 T 9 T 10 T 11 Q 12 Q 13 Q 14 T 15 T 16 Q 17 T 18 T 19 Q 20 T 21 T 22 Q 23 Q 24 Q 25 Q		frequency 6530.1 4096.3 2800.0 2576.1 1945.9 1935.6 1900.5 1883.7 1740.7 1680.4 1579.7 1469.8 1407.2 1388.6 1321.8 1273.0 1194.8 1188.3 1073.8 1068.1 917.4 847.2 833.9 812.5 736.2	ppm 129.859 81.459 55.682 51.229 38.696 38.491 37.794 37.460 34.615 33.416 31.413 29.229 27.985 27.613 26.285 25.314 23.759 23.630 21.354 21.240 18.244 16.848 16.582 16.158 14.640	intensity 35.080 50.885 37.741 35.589 33.347 35.447 34.954 33.459 33.459 33.842 34.166 44.341 44.597 54.820 32.783 34.450 46.078 45.803 25.486 79.335 30.582 35.972 60.810 55.942 49.973 48.337
Number	of	protonated	carbons: 25	
CH : CH2 : CH3 :	5 11 9			

Combining the information above, we can come up with a structure having the framework of a triterpene with one ketonic group at carbon 3, a double bond at carbon 24, and two C-O-C bonds at carbons 2 and 15. Having a clear quatet Hnmr at 4.10ppm is an indication of the presence of a highly shielded ethyl group, with the methylene group splitted into a quatet by the methyl group. This means that one of the two C-O-C bonds is ethoxy, while the remaining three may be methoxy, and preferably the C-O-C on carbon 15 is chosen to be the one having the ethoxy group. The protons on the double bond split into a singlet at 4.7ppm, while the protons on the methyl group attached to the double split into a sharp singlet peak at 2.00ppm, the proton on carbon 3 will split into a doublet of quatet at 2.31ppm, and the methoxy proton will split into a singlet at 3.7ppm. Consequently, the structure of the isolated compound HO2-13-111 is proposed as;



The proposed structure whose formula weight is 486, with molecular formula $C_{32}H_{54}O_2$ has the following fragmentation pattern, as evidently shown by the mass spectroscopy (Fig. 5).

Heterolytic cleavage of structure 486 by shifting the bond between C14 and 15 to the bond



When the proposed structure heterolytic cleaves by losing a propylene group, a structure with formula weight 445 results, which can itself heterolytically loose a methoxy group to yield another structure with formula weight 414. Structure 414 can heterolytically cleave to generate another structure with formula weight 318, which itself can lose a methyl group to yield a structure with formula weight 304.

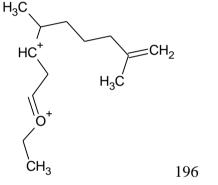
CONCLUSION

The chloroform fraction HO2 of Hippocratea obtusofolia was found to be the most active as far as inducement of lactation is concerned. Activity guided fractionation and chromatography of HO2 led to the isolation of a fraction HO2-13 which retained the observed activity. This was further chromatographed to obtain fractions HO2-13-111 and HO2-13-149, with the latter being the most active.

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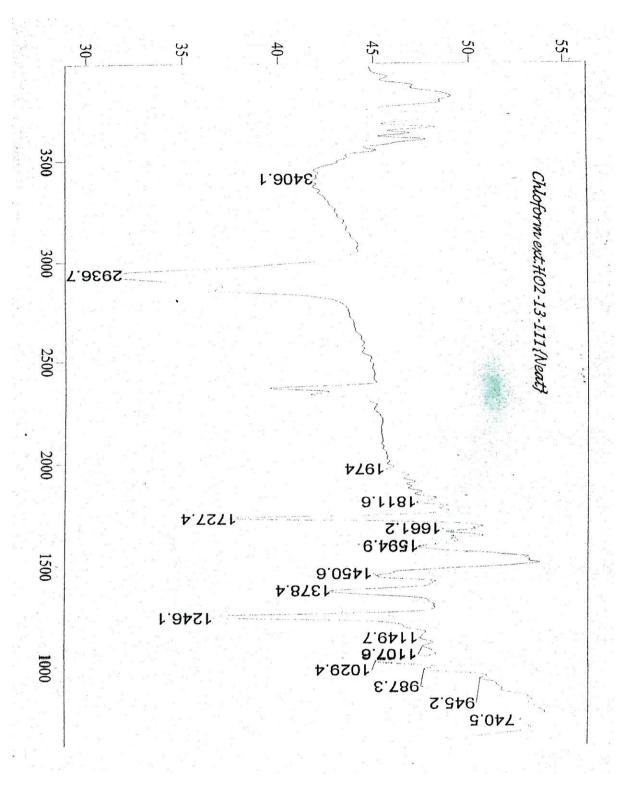
between C15 and O (creating a double bond) will create a positive charge on C14, which will cause the bond between C13 and 17 to shift to the bond between C13 and 14 (creating a double bond). This splits the structure into two structures with formula weights 290 and 196 (base peak).

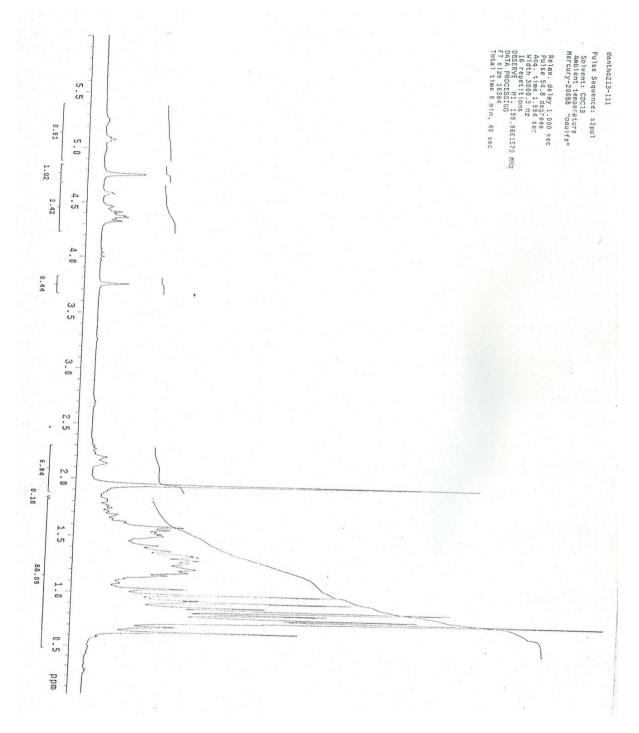


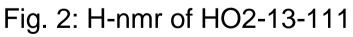
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pp

.17

.234

HEIGHT 28. 30.

dantho213-111 Pulse Sequence: s2pul Solvent: CDCl3 Ambient temperature Mercury-2008B "oauife"

Pulse 97.5 degrees Acq. time 1.498 sec

3, 50.2815182 1, 199.9671274 tions HZ

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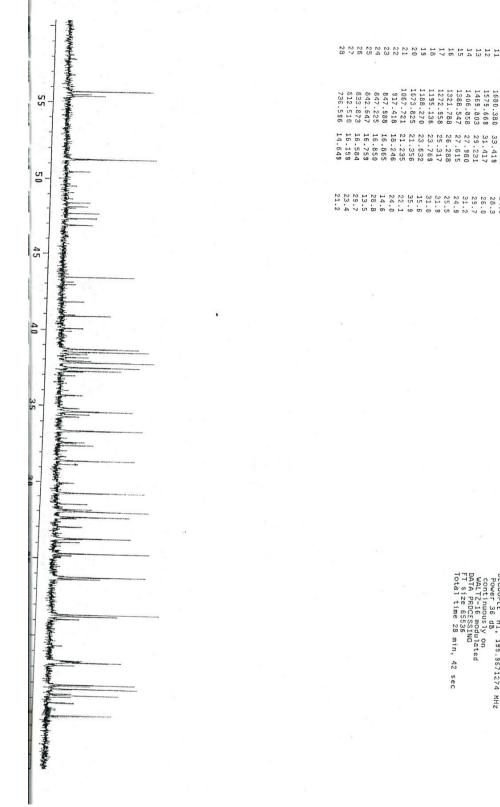


Fig. 3: C-nmr of HO2-13-111

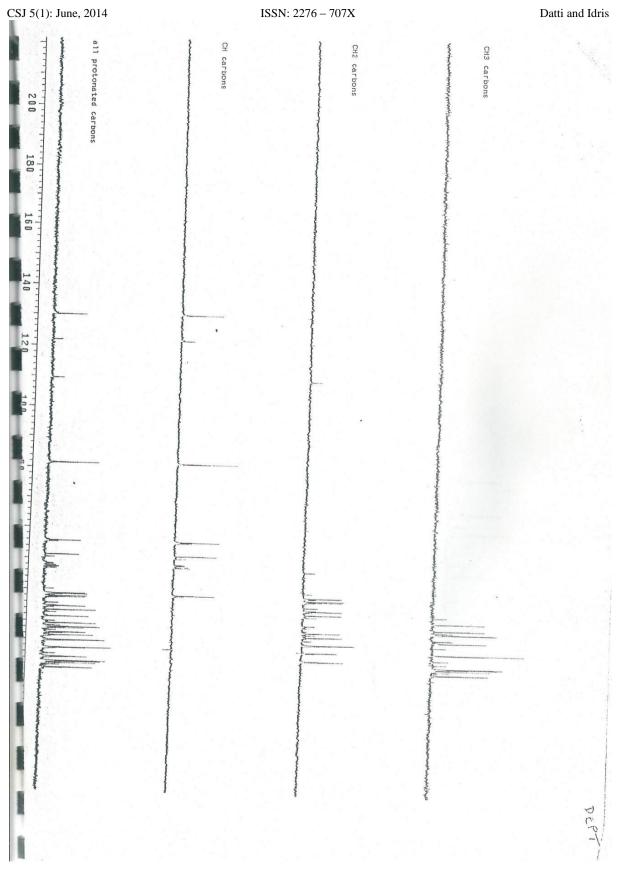


Fig. 4 (C-Dept of HO2-13-111)

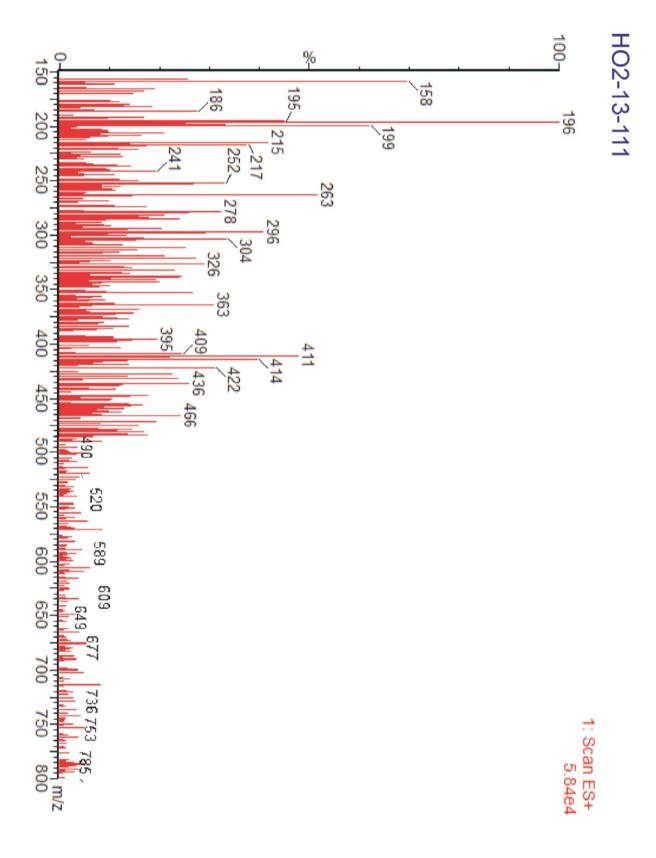


Fig. 5: Mass Spect. of HO2-13-111