



Ambient Air Ammonia (NH₃) Concentration in Two Solid Waste Dump Sites in Abakaliki, Ebonyi State, Nigeria

^{1,2}Ngele, S. O., ²Uduma, A. U. and ²Alisi, I. O.

¹Department of Industrial Chemistry, Ebonyi State University, Abakaliki

²Department of Applied Chemistry, Federal University, Dutsin- Ma, Katsina State

Email: udumas96@gmail.com

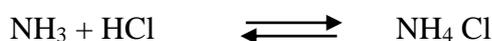
ABSTRACT

This work monitored the level of ammonia in the ambient air of two major solid waste dump sites in Abakaliki urban, Ebonyi State, in the morning for a period of a month on the onset of wet season, using portable monitor based on electrochemical sensor. The result showed that site 1 had a relatively higher mean ammonia level of 0.152 ± 0.003 ppm as against 0.09 ± 0.002 ppm for site 2 while the corresponding minimum levels were 0.027 ± 0.001 ppm and 0.03 ± 0.001 ppm for site 1 and 2 respectively. The ammonia levels peak on weekends (Friday to Sunday) and the beginning of the week day (Monday). The one-way Analysis of variance of mean difference in NH₃ concentration in the two sites monitored revealed a no statistical significant difference in the mean ($p < 0.05$). There is no national/ international threshold limit for ammonia but the concentrations of ammonia obtained in this study is within the values obtained in some studies in Asian and European cities reported in literature.

Keywords: Abakaliki, Ambient Air, Ammonia, Dump Sites, Portable Monitor

INTRODUCTION

Ammonia is the most abundant form of reduced nitrogen in the gas-phase within the atmosphere (Behera *et al.*, 2013). Ammonia contributes to both formation of particulate and deposition of reactive nitrogen in the environment (Reis *et al.*, 2009, Aneja *et al.*, 2012). The deposition of nitrogen in the form of ammonia can result in eutrophication of sensitive ecosystem and to acidification of the soil (Bouwman *et al.*, 1997). An enhanced load of nitrogen in terrestrial ecosystem has been found to correlate with loss of biodiversity and can increase ecosystem vulnerability to extreme weather and insect attacks (Behera *et al.*, 2013). NH₃ plays a decisive role in particulate formation chemistry by determining the amount of ammonium sulphate and nitrate as particulate constituents (Sharma. *et al.*, 2007, Baek and Aneja, 2004). Agriculture is a major source of NH₃ as it is released from animal urine and dung and from volatilization of applied fertilizers



It has been estimated that ammonia emissions from agriculture give a substantial contribution (13%) to the particulate concentration in Europe (Aneja, *et al.*, 2008, Pinder *et al.*, 2007) and thereby adds significantly to the external costs

(Miroslav and Vladimir, 1998). Although NH₃ can cause injury to plants at very high concentrations (20ppm), typical levels encountered in the atmosphere (5–25ppbv) are not considered to be harmful (Miroslav and Vladimir, 1998)

If present in sufficient amounts, ammonia can neutralize atmospheric acid vapours (sulphuric, nitric and hydrochloric) by forming salt aerosols. In air, NH₃ is the precursor gas of NH₄⁺ in particles (as shown in the reactions below). The gas-to-particle conversion processes in ambient atmosphere may produce inorganic ammonium salts of ammonium bisulfate (NH₄HSO₄), ammonium sulfate ((NH₄)₂SO₄), ammonium nitrate (NH₄NO₃), and ammonium chloride (NH₄Cl) (Tsai *et al.*, 2014, Gong *et al.*, 2013, Behera *et al.*, 2013).

The latter two reactions are reversible and the acid gases can be released under specific atmospheric conditions such as warm weather (Miroslav and Vladimir, 1998).

related to air pollution in Europe (Erismann and Schaap, 2004, Battye *et al.*, 2003, Werner *et al.*, 2015). According to Huang *et al.*, (2012) and Zhou *et al.*, (2015), Meng, *et al.*, (2011) more than 60% of total ammonia emission in Beijing comes from

livestock and farm-land. Other sources, including human excrement, waste disposal, biomass burning, chemical industry and traffic, totally contributed 14.9–35.5% to the total budget with vehicular source accounting for only about 5%.

Ammonia is a critical nitrogen compound that alone has a major effect on global biogeochemical nitrogen cycle, atmospheric reactions leading to particulate formation, climate change, health effects and more lasting cascading effects in the ecosystem. Hence in the recent years, the sources, transport and fate of atmospheric ammonia has been widely studied (Myhre *et al.*, 2009), due to its role in global climate change. For instance, SO_4^{2-} and NO_3^- aerosols have important effects on global radiation budgets because of their ability to scatter the incoming solar radiation, act as cloud condensation nuclei and indirectly increase cloud life time (Myhre *et al.*, 2009).

In some cities of the world, measurements of ambient NH_3 have been reported, for example, in Rome (Perrino *et al.*, 2002), New York city (Li *et al.*, 2006), Manchester (Whitehead *et al.*, 2007) and Barcelona (Pandolfi *et al.*, 2012)

There are currently no regulations or incentive programmes in most counties of the world including Nigeria for reduction in NH_3 emission. This is in contrast to other primary gaseous pollutants such as SO_2 , NO_x and VOCs, where extensive control measures and guidelines

exist for the reduction in their emissions. Extensive measures have not been taken to mitigate NH_3 emission despite the fact that all these pollutants make similar contributions to PM mass loading, visibility degradation and / or acidification / eutrophication.

MATERIALS AND METHOD

Study Area

Abakaliki is the capital of Ebonyi state. It is predominantly urban, covering a total area of 5533 km². According to National Population Commission (NPC) 2006 census figure, Ebonyi State had a population of 2, 176, 947, out of which the Abakaliki capital territory (ACT) consists of 271, 833. The capital territory is located between longitude 6°25’N and latitude 8°08’E (Fig.1). Urban activities in the ACT include; commercial, education and industrial development as well as rapidly expanding residential areas. Improved living standards of people in ACT due to its socio-economic development have led to the generation of enormous quantity of solid waste. Abakaliki Capital Territory is facing a crisis in solid waste management with overflowing waste in designated dumping sites that stays for days and sometimes weeks before they are cleared with attendant emitted foul odour for overstayed decaying/decomposing waste.

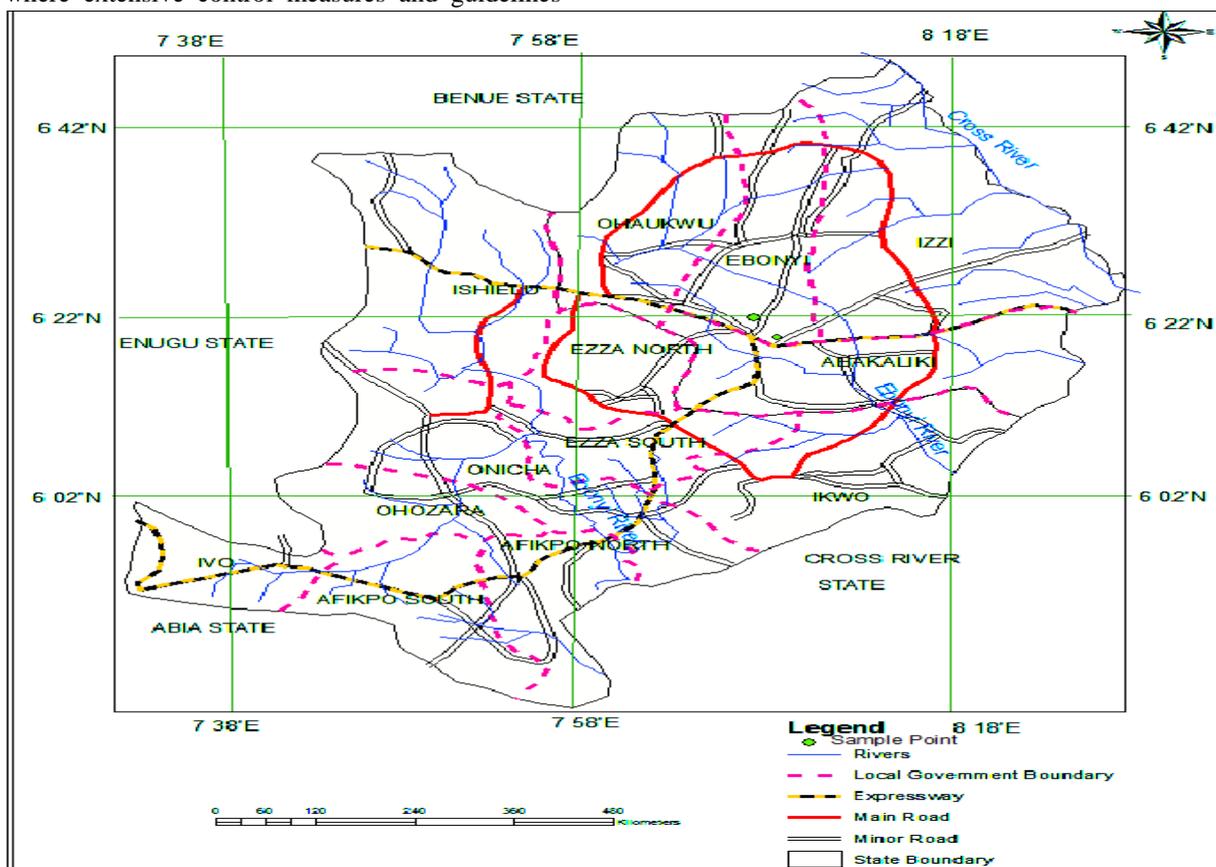


Fig. 1. Map of Ebonyi State Showing The Sampling Points

Site Selection and Monitoring Protocol

Two biggest solid waste dump sites in Abakaliki urban namely, Juju Hill by water works road dump site and Kpirikpiri market by Ogbuga road dump sites were selected for this study, because of the high volume of solid waste dumped on these sites.

Monitoring of Ammonia

The levels of ammonia at the two sites were monitored using the Crowcon Gasman ammonia portable monitor.

The instrument is equipped with NH₃ detecting electrochemical sensor. The range of detection is between 0- 50ppm and the detection limit is 0.01ppm.

The monitor was switched on to the gas position and hand held to a height of two meters in the direction of the prevailing wind. The instrument reading was recorded at stability. The reading was taken in triplicate at each instance and averaged in the morning hours (8-9 am each day) in each of the sites. Huang *et al.*, (2012) had earlier reported morning and night peaking in ammonia concentration in a study in Beijing China. This study spanned for 26 days in the wet season (June, 2015).

Data Analysis

The data collected from the 26 days monitoring campaign was computed for descriptive statistics, one-way analysis of variance of mean values of NH₃ levels in the two sites using version-15 Minitab statistical software.

RESULTS AND DISCUSSION

The mean levels of ammonia in sites 1 and 2 were presented in Table 1. The maximum and

minimum ammonia concentration during the 26 days monitoring in site 1 were 0.152 ± 0.003 and 0.027 ± 0.001 ppm respectively while the corresponding values for site 2 were 0.09 ± 0.002 and 0.03 ± 0.001 ppm respectively. The ammonia levels peak on weekend (Friday to Saturday) and the beginning of the week day (Monday). The maximum ammonia concentration of 0.152ppm in site 1 was recorded on Saturday (13/6/2015) while maximum in site 2 (0.09 ppm) was recorded on Monday (8/6/2015) and Thursday (13/6/2017) (Table 1 and Fig.2)

The mean concentration of ammonia in the study period gave 0.070538 ± 0.029972 and 0.049231 ± 0.018414 in sites 1 and 2 respectively (Table 2). The mean levels of ammonia obtained in this study were higher than the mean of 0.2ppb and range of 0.2-1.5 ppb reported by Mukhtar *et al.*, (2008) in a study using Ogawa passive sampler. The level of NH₃ in this study may be attributed to the multiple impacted nature of the sites including but not limited to agricultural (site 1 is situated close to a rice farm where ammonium fertilizer is being used and Kpirikpiri market), also automobile impacted, as the two sites are situated along major roads and are further impacted by ammonia emissions from the decaying and sometimes burnt organic matter in the dump sites. This is in line with Huang *et al.*, (2012) and Zhou *et al.*, (2015), who reported that more than 60% of total ammonia emission in Beijing comes from livestock and farm-land. Other sources, including human excrement, waste disposal, biomass burning, chemical industry and traffic, totally contributed 14.9–35.5% to the total budget with vehicular source accounting for about 5%.

Table 1: Mean Levels of Ammonia from Sites 1 and 2

Day	Date	Kpirikpiri/Ogbuaga market waste dump mean NH ₃ Conc. (ppm) (site1)	Juju hill/water works waste dump mean NH ₃ Conc.(ppm) (site2)
Mon	1/6/2015	0.09 ±0.002	0.08 ± 0.002
Tue	2/6/2015	0.03 ±0.001	0.04 ± 0.001
Wed	3/6/2015	0.066 ± 0.002	0.05 ± 0.001
Thur.	4/6/2015	0.146 ± 0.003	0.03 ± 0.001
Fri	5/6/2015	0.095 ± 0.003	0.04 ± 0.001
Sat	6/6/2015	0.027 ± 0.002	0.05 ± 0.001
Mon	8/6/2015	0.079 ± 0.002	0.09 ± 0.002
Tue	9/6/2015	0.027 ±0.001	0.05 ± 0.001
Wed	10/6/2015	0.07 ±0.002	0.03 ± 0.001
Thur.	11/6/2015	0.05 ± 0.001	0.08 ± 0.002
Fri	12/6/2015	0.079 ±0.002	0.04 ± 0.001
Sat	13/6/2015	0.152 ±0.003	0.05 ± 0.001
Mon	15/6/2015	0.076 ± 0.002	0.03 ±0.001
Tue	16/6/2015	0.097 ±0.001	0.04 ±0.001
Wed	17/6/2015	0.05 ±0.001	0.05 ± 0.001
Thur.	18/6/2015	0.04 ± 0.001	0.09 ±0.002
Fri	19/6/2015	0.08 ±0.002	0.05 ±0.001
Sat	20/6/2015	0.06 ±0.001	0.03 ± 0.001
Mon	22/6/2015	0.07 ±0.001	0.04 ±0.001
Tue	23/6/2015	0.06 ± 0.001	0.05 ± 0.001
Wed	24/6/2015	0.06 ± 0.001	0.05 ±0.001
Thur.	25/6/2015	0.05 ± 0.001	0.07 ± 0.002
Fri	26/6/2015	0.07 ± 0.001	0.03 ±0.001
Sat	27/6/2015	0.06 ± 0.001	0.05 ± 0.001
Mon	29/6/2015	0.08 ± 0.002	0.04 ± 0.001
Tue	30/6/2015	0.07 ±0.001	0.03 ± 0.001

(NB. Monitoring was not done on Sundays).

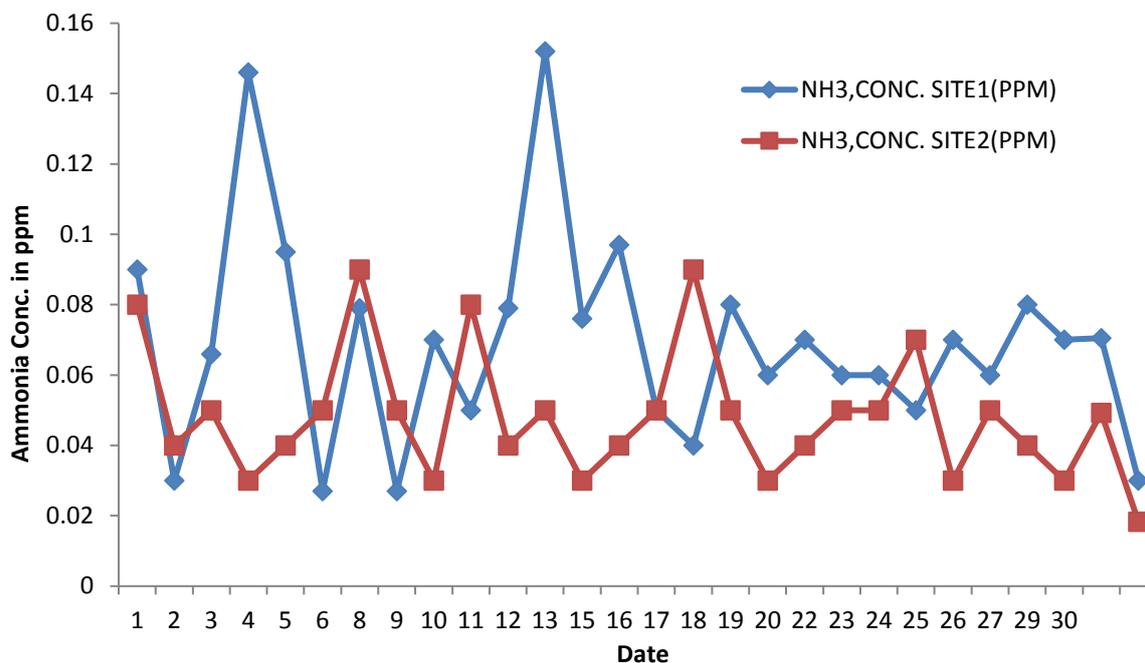


Fig. 2 Date (Day) Ammonia Levels in the two Sites within the Study Period

Table 2 Descriptive Statistics of NH₃ Concentrations (ppm) in Sites 1 and 2

Sites	N	Mean	StDev
NH3_Conc	26	0.070538	0.029972
Site_1			
NH3_Conc	26	0.049231	0.018314
Site_2			

Table 3 ONE-WAY ANOVA for NH₃ Concentration (ppm) at Sites 1 and 2

Source	DF	SS	MS	F	P
Factor	1	0.006789	0.006789	8.15	0.007
Error	40	0.033306	0.000833		
Total	41	0.040096			

Site-wise comparison showed that site 1 had the higher mean concentrations of NH₃ relative to site 2. However, one-way analysis of variance (ANOVA) for the differences in the mean levels of NH₃ in the two sites (Table 3) revealed that although there is variation in the mean levels of NH₃ in the sites, such variations were not statistically significant($p < 0.05$).

CONCLUSION

This study revealed that NH₃ concentrations in sites 1 and 2 within the period of the study ranged between 0.027-0.152 and 0.03-0.09 ppm with mean of 0.152 ± 0.003 and 0.09 ± 0.002 ppm for sites 1 and 2 respectively. Furthermore, the observed difference in the mean levels of NH₃ in the two sites may have occurred

by marginal difference in the emission sources in the two sites. The level of NH₃ in this work is within the values reported in literature in some Asian and European cities.

REFERENCES

- Aneja, V. P., Schlesinger, W. H. and Erisman, J. W. (2008). Farming Pollution. *Nat. Geosci* 1:409-411
- Aneja, V. P., Schlesinger, W. H., Erisman, J. W., Behera, S. N., Shama, M. and Battye, W. (2012). Reactive Nitrogen Emissions from Crop and Livestock Farming in India. *Atmos. Environ.* 47 : 92-103
- Baek, B. H. and Aneja, V. P. (2004). Measurement and Analysis of the Relation between Ammonia, Acid Gases and fine Particles in Eastern North Carolina. *J. Air Waste Management Assoc.* 54:623-633.
- Battye, W., Aneja, V. P. and Roelle, P. A. (2003). Evaluation and Improvement of Ammonia Emission Inventories. *Atmos. Environ.* 37:3873-3883
- Behera, S. N., Sharma, M., Aneja, V. P., Balasubramanian, R. (2013). Ammonia in the Atmosphere: A Review on Emission Sources, Atmospheric Chemistry and Deposition on Terrestrial Bodies. *Environ. Sci. Pollut. Res.* 20: 8092– 8131.
- Bouwman, A. F., Lee, D. S., Asman, W. A., Dentener, F. J., Van der Hoeck, K. W. and Olivier, J. G. (1997). A Global High-Resolution Emission Inventory for Ammonia. *Global Biogeochem. Cycles* 11:561-587
- Erisman, J. W. and Schaap, M. (2004). The need for Ammonia Abatement with Respect to Secondary Particulate Reduction in Europe. *Environ. Pollution.* 129:159-163
- Gong, L., Lewicki, R., Griffin, R.J., Tittel, F.K., Lonsdale, C.R., Stevens, R.G., Pierce, J.R., Malloy, Q.G.J., Travis, S.A., Bobmanuel, L.M., Lefer, B.L. and Flynn, J.H. (2013). Role of Atmospheric Ammonia in Particulate Matter Formation in Houston during Summer time. *Atmos. Environ.* 77: 893–900.
- Huang, X., Song, Y., Li, M., Li, J., Huo, Q., Cai, X., Zhu, T., Hu, M. and Zhang, H. (2012). A High-Resolution Ammonia Emission Inventory in China. *Global Biogeochem. Cycles* 26: GB1030.
- Li, Y., Schwab, J.J. and Demerjian, K. L. (2006). Measurements of Ambient Ammonia Using a Tunable Diode Laser Absorption Spectrometer Characteristics of Ambient Ammonia Emissions in an Urban Area of New York City. *J. Geophys. Res.* 111:D110502
- Meng, Z., Lin, W., Jiang, X., Yan, P., Wang, Y., Zhang, Y., Jia, X. and Yu, X. (2011). Characteristics of Atmospheric Ammonia over Beijing, China. *Atmos. Chem. Phys.* 11: 6139–6151.
- Miroslav, R. and Vladimir, B. (1998). *Practical Environmental Analysis*, Royal Society of Chemistry U. K. pp. 74-137.
- Mukhtar, C. S., Mutlu, A., Capareda, S. C. and Parnell, C. B. (2008). Seasonal and Spatial Variations of Ammonia Emissions from an Open-Lot Dairy Operation. *J. Air Waste Management Association* 58: 369- 376 .
- Myhre, G., Berglen, T.F., Johnsrud, M., Hoyle, C.R., Bernsten, T.K., Christopher, S.A., Fahey, D.W., Isaksen, I.S., Jones, T.A., Kahn, R.A., Loco, N. Quinn, P., Remer, L., Schwarz, J.P. and Yttri, K.E. (2009). Modelled Radiative Forcing of the Direct Aerosol Effect with Multi-Observation Evaluation. *Atmos. Chem. and Phys.* 9:1365-1392
- Ohia, G.N. Amasiatu, G.I. Ajagbe, J.O. Ojokuku, G.O. and Mohammed, U. (2005). *Comprehensive Certificate Chemistry* 2nd ed. University Press PLC Ibadan, pp. 112, 278-280, 286-295, 298-308.
- Pandolfi, M., Amato, F., Reche, C., Alastuey, A., Otjes, R. P., Blom, M. J. and Querol, X. (2012). Summer Ammonia Measurements in a Densely Populated Mediterranean City (Barcelona). *Atmoschem .phys.* 12:7557-7575
- Pinder, R. W., Adams, P. J. and Pandis, S.N. (2007). Ammonia Emission Controls as a Cost Effective Strategy for Reducing Atmospheric Particulate Matter in Eastern United States. *Environ. Sci. Technol.* 41:380-386
- Perrino, C., Catrambone, M., Menno Di-Bucchianico, A and Allegrini, I. (2002). Gaseous Ammonia in the Urban Area of Rome, Italy and its Relationship with Traffic Emissions. *Atmos. Environ.* 36:5385-5394
- Reis, S., Pinder, R. W., Zhang, M., Lijie, G. and Sutton, M. A. (2009). Reactive Nitrogen in Atmospheric Emission Inventories. *Atmos. Chem. and Phys.* 9: 7657-7677
- Sharma, M., Kisshore, S., Tripathi, S.N. and Behera, S.N. (2007). Role of Atmospheric Ammonia in the Formation of Inorganic Secondary Particulate Matter: A Study of Kanpur, India. *J. Atmos. Chem.* 58:1-17.
- Tsai, J., Chang, L. and Chiang, H. (2014). Airborne Pollutant Characteristics in an Urban, Industrial and Agricultural Complex Metroplex with High Emission Loading and Ammonia Concentration. *Sci. Total Environ.* 494–495: 74–83.
- Weatherburn, M. W. (1967). Phenol–Hypochlorite Reaction for the Determination of Ammonia. *Anal. Chem.* 39:971-974.

- Werner, M., Kryza, M., Geels, C., Ellermann, T. and Ambelas-Skjoth, C. (2015). Spatial, Temporal and Vertical Distribution of Ammonia Concentrations over Europe: Comparing a Static and Dynamic Approach with WRF- Chem. Atmos. Chemistry and Physics Discussions 15: 22935-22973
- Whitehead, J. D., Longley, I. D. and Gallagher, M. W. (2007). Seasonal and Diurnal Variation in Atmospheric Ammonia in an Urban Environment Measured using a Quantum Cascade Laser Absorption Spectrometer. Water, Air and Soil Pollution 183:317-329
- Zhou, Y., Cheng, S., Lang, J., Chen, D., Zhao, B., Liu, C., Xu, R. and Li, T. (2015). A Comprehensive Ammonia Emission Inventory with High-Resolution and its Evaluation in the Beijing-Tianjin-Hebei (BTH) Region, China. Atmos. Environ. 106: 305–317.