

Geochemical Characterization and Mineralization Potential of Kaffo Albite Arfvedsonite Granites in the Riruwai Younger Granite Complex, Nigeria

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Abstract

This study presents geochemical analysis of the Kaffo albite arfvedsonite granites, a subset of the Riruwai Younger Granite Complex, Nigeria, to assess their mineralization potential and tectonic significance. Major and trace element compositions were analyzed using energy dispersive x-ray fluorescence (EDXRF), and Pearson correlation coefficients were calculated to establish elemental associations and identify mineralization patterns. The results highlight significant geochemical factors representing magmatic differentiation, crustal assimilation, and late-stage fluid processes. Strong negative correlations between SiO₂ and major oxides (CaO, MgO, Fe₂O₃) suggest fractional crystallization and crustal contamination, while positive correlations among Fe₂O₃, TiO₂, and trace metals (Cu, Zn, Pb) indicate possible ore mineralization associated with Fe-Ti oxides. The rare earth element (REE) distribution further suggests magmatic differentiation and hydrothermal influence, reinforcing the high potential for rare-metal mineralization, including Nb, Ta, Zr, and REEs. These findings enhance the understanding of the metallogenic evolution of the region and have significant implications for mineral exploration and mining.

Keywords: Petrogenesis, Geochemistry, Granite, Mineralization and Rare-earth-elements

INTRODUCTION

The study of granitic and alkaline rocks provides critical insights into magmatic evolution, mineralization potential, and tectonic settings. Albite-arfvedsonite granites are a distinct class of peralkaline igneous rocks commonly associated with rare metal mineralization, including elements such as Nb, Ta, Zr, Sn, and REEs (Ogunleye et al., 2006, Olasehinde & Ashano 2021). The Kaffo Albite Arfvedsonite suite represents a significant geological formation within the Nigerian Younger Granite Province, a region known for its diverse magmatic history and

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metallogenic significance (Kinnaird, 1979, 1985). The Pearson correlation coefficient is a statistical tool used to measure the strength and direction of linear relationships between two variables. In geochemistry, it helps in understanding elemental associations and their implications for rock formation, alteration processes, and mineralization potential.

This study examines the Pearson correlation coefficients of major oxides and trace elements of the Kaffo Albite Arfvedsonite in the Riruwai Younger Granite Complex. By analyzing these correlations, we can infer magmatic differentiation trends, source characteristics, and potential ore-bearing mineral phases. Strong positive correlations suggest a common source, similar geochemical behavior, or co-precipitation during magmatic or hydrothermal processes. Negative correlations may indicate fractionation, substitution, or different petrogenetic pathways. Understanding these relationships is crucial for mineral exploration, as they highlight possible ore-related enrichment zones and the geotectonic environment of rock formation. This analysis provides insights into the geochemical controls on mineralization and the tectonic significance of the studied rocks. By comparing our results with other peralkaline granite occurrences worldwide, such as those in the East African Rift, the Massif Central (France), and the Ilímaussaq Complex (Greenland), we provide a broader geodynamic and metallogenic context for the Kaffo suite. This will help refine existing models of rare-metal enrichment in peralkaline granites and contribute to exploration strategies in similar geological settings. The implications of element intercorrelations and their alignment with mineralization trends will be discussed in relation to other documented studies on rift-related granitic systems.

The Geological Setting

The Riruwai Younger Granite Complex is one of the several Jurassic anorogenic intrusions in north-central Nigeria (Figure 1), forming part of the larger Younger Granites Province. This geological province is composed of high-level granitic intrusions that were emplaced within the Pan-African basement complex during the Mesozoic Era. The Riruwai Complex, like other Younger Granites, provides critical insights into the geodynamic processes that shaped the region during the breakup of Gondwana (Jacobson & MacLeod, 1977, Bowden et al., 1987). The Younger Granites Province, including the Riruwai Complex, is closely linked to extensional tectonics associated with lithospheric thinning and mantle upwelling (Coulon et al., 1996). The Riruwai Complex (Figure 2) consists of multiple ring intrusions that exhibit a variety of rock types, reflecting a complex magmatic history (Bowden & Turner, 1974). These lithologies suggest a progression from mafic to felsic compositions, indicative of fractional crystallization and crustal assimilation processes that contributed to the final geochemical signatures of the granites. Structurally, the Riruwai Complex exhibits circular and elliptical ring structures, characteristic of forceful magmatic emplacement and subsequent caldera collapse. These features are consistent with the extensional tectonic environment that facilitated magma ascent and emplacement (Kinnaird, 1985).

The Riruwai Complex shares several geochemical and structural similarities with other Younger Granite complexes, such as Jos, Afu, and Bauchi (Bowden & Turner, 1974). However, it is distinguished by its unique peralkaline granites and significant rare-metal mineralization. The presence of high concentrations of elements such as tin (Sn), niobium (Nb), and tantalum (Ta) makes Riruwai economically significant compared to some other complexes that lack such mineralization and rare-earth elements (REEs) (Bowden et al., 1976; Jacobson & MacLeod, 1977, Kinnaird, 1985).

Geochemical Characterization and Mineralization Potential of Kaffo Albite Arfvedsonite Granites in the Riruwai Younger Granite Complex, Nigeria

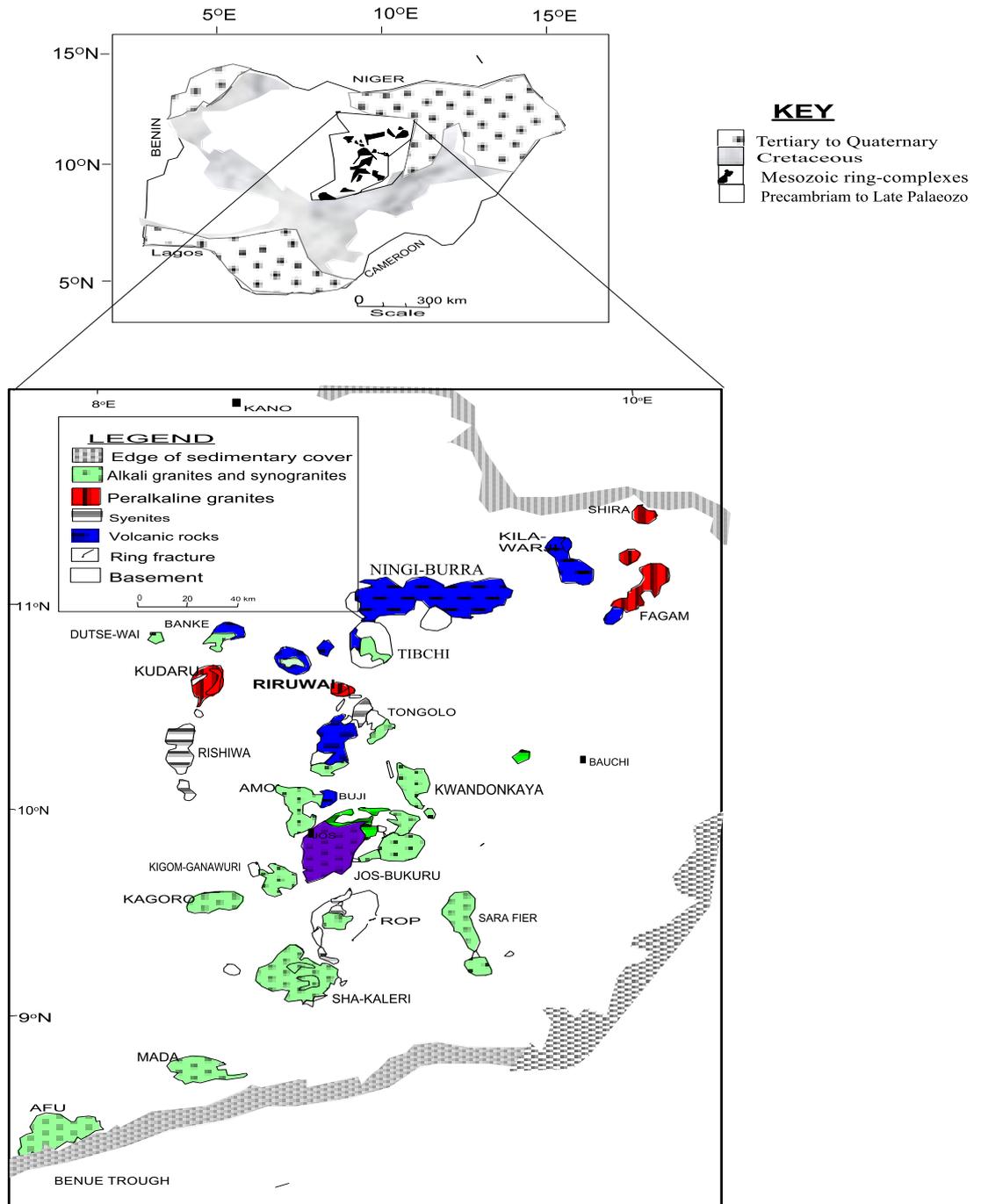


Figure 1 The Younger Granite Complexes of Nigeria Map. Modified after (Kinnaird, 1985; Ogunleye, et al. 2006)

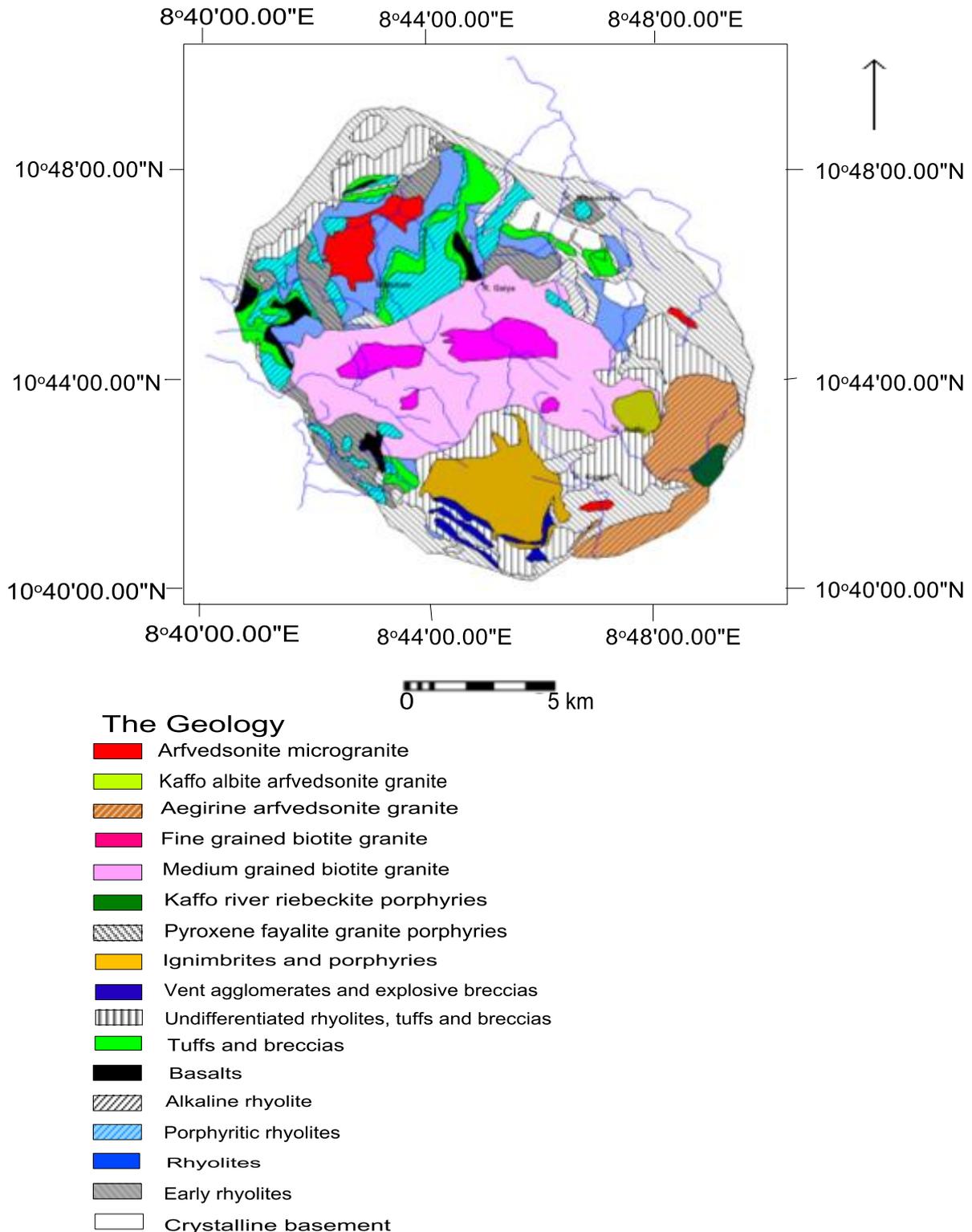


Figure 2 Geology Map of the Riruwai Complex. Modified after Ogunleye, et al. (2006)

METHODOLOGY

The methodology for analyzing the Pearson correlation coefficient and its implications for mineralization involves several key steps, including sample collection, geochemical analysis, data processing, and statistical evaluation. Rock samples were collected from the Riruwai Younger Granite Complex. Sampling focused on fresh, unweathered rock exposures to ensure

the accuracy of geochemical data. Sampling and Analysis follows guide lines provided by Ashano (2009). Thirty-nine (39) samples were analyzed for major and trace elements which include SiO₂, CaO, Fe₂O₃, K₂O, Na₂O, TiO₂, MnO, P₂O₅, Al₂O₃, V, Cr, Cu, Sr, Zr, Ba, Zn, Ce, Pb, Sn, Ga, Y, Ni, Rb, Nb, La, Eu, Mo, Co, Ta, W, Hf, Pr, Lu, Gd, U, Th, and N of which three (3) albite arfvedsonite granites formed part of this study. The samples weighed from ~ 4 to ~ 5 kg. Samples were crushed, powdered, and homogenized before geochemical analysis. The major oxides and trace element compositions of the rock samples were determined using advanced analytical techniques. Energy Dispersive x-ray fluorescence (EDXRF) spectrometer of model "Minipal 4" was used for the analysis at the National Geoscience Research Laboratory (NGRL) Kaduna to quantify bulk chemical composition. The analysis had a precision that was better than 1% for the oxides over 10 wt% and 10% for the oxides below 10 wt%. The trace elements measurement also showed an error below 5%. The geochemical data were compiled, and descriptive statistics were generated to assess data distribution. Pearson correlation coefficients were calculated for all element pairs to determine relationships. The correlation values range from -1 to +1, where: +1 indicates a perfect positive correlation, -1 indicates a perfect negative correlation, 0 indicates no correlation and strong correlations between elements were interpreted in terms of magmatic differentiation, fractionation, and mineralization processes.

RESULT AND DISCUSSION

The result of the analysis of Kaffo Albite Arfvedsonite (Table 1 and 2) highlights key element groupings that provide insights into its petrogenesis, tectonic setting, and mineralization. By comparing these findings with other peralkaline granitic systems worldwide, we can better understand the processes controlling rare metal enrichment.

Table 1 Major and trace element concentration of the kaffo albite arfvedsonite granite

Rock Type	Y41 Kaffo albite Arfvedsonite Granite 10°44'42.60"N 8°43'59.09"E	Y42 Kaffo Albite Arfvedsonite Granite 10°43'31.89"N 8°42'38.61"E	Y44 Kaffo Albite Arfvedsonite Granite 10°43'13.18"N 8°46'38.61"E
SiO ₂	72.2	73.54	74.5
CaO	0.25	0.19	0.18
MgO	0.07	0.08	0.03
K ₂ O	4.34	3.75	3.74
Na ₂ O	5.66	5.84	5.06
TiO ₂	0.11	0.12	0.08
MnO	0.06	0.05	0.04
P ₂ O ₅	0.01	0.01	0.01
Fe ₂ O ₃	2.41	2.36	2.33
Al ₂ O ₃	12.24	12.31	11.86
LOI	1.68	1.34	1.84
Total	99.03	99.59	99.67
A/CNK	1.19	1.26	1.32
V	0.03	0.02	0.02
Cr	0.02	0.04	0.04
Cu	390	220	200
Sr	11	9.45	6.21
Zr	4300	5100	4620
Ba	20	40	26
Zn	1400	1250	1470
Ce	380	291	212
Pb	366	310	370
Sn	175	186	143
Ga	86	73	78
Y	410	419	394.4

Geochemical Characterization and Mineralization Potential of Kaffo Albite Arfvedsonite Granites in the Riruwai Younger Granite Complex, Nigeria

Ni	245	184	194
Rb	1300	1181	1054
Nb	1446	1292	1560
La	143	134	126
Eu	2.7	2.2	2.1
Mo	11.5	12.3	9.1
Co	1.2	0.9	1.1
Ta	170	145.67	17.78
W	1	1	1
Hf	115.16	102.28	114.8
Pr	54.7	37.2	66.8
Lu	5.65	6.68	7.01
Gd	56.17	65.34	43.12
U	121	105	111
Th	201	199	232
Nd	112	91	222
REE	754.22	627.42	679.03
LREE/HREE,Y	1.80	1.46	1.67

Table 2 Pearson correlation of major and trace element concentration of the kaffo albite arfvedsonite granite

	SiO ₂	CaO	MgO	K ₂ O	Na ₂ O	TiO ₂	MnO	P ₂ O ₅	Fe ₂ O ₃	Al ₂ O ₃	V	Cr	Cu	Sr	Zr	Ba	Zn	Ce	Pb	Sa	Ga	Y	Ni	Rb	Nb	La	Eu	Mo	Co	Ta	W	Hf	Pr	Lu	Gd	U	Th	Nd					
SiO ₂	1.00																																										
CaO	-0.96	1.00																																									
MgO	-0.69	0.45	1.00																																								
K ₂ O	-0.92	0.99	0.34	1.00																																							
Na ₂ O	-0.67	0.42	1.00	0.31	1.00																																						
TiO ₂	-0.65	0.40	1.00	0.29	1.00	1.00																																					
MnO	-1.00	0.92	0.76	0.87	0.74	0.72	1.00																																				
P ₂ O ₅	0.10	-0.38	0.66	-0.49	0.68	0.69	0.00	1.00																																			
Fe ₂ O ₃	-1.00	0.97	0.66	0.93	0.63	0.61	0.99	-0.14	1.00																																		
Al ₂ O ₃	-0.72	0.49	1.00	0.38	1.00	1.00	0.79	0.62	0.69	1.00																																	
V	-0.91	0.99	0.33	1.00	0.30	0.28	0.87	-0.50	0.93	0.37	1.00																																
Cr	0.91	-0.99	-0.33	-1.00	-0.30	-0.28	-0.87	0.50	-0.93	-0.37	-1.00	1.00																															
Cu	-0.95	1.00	0.42	1.00	0.39	0.37	0.91	-0.42	0.96	0.46	1.00	-1.00	1.00																														
Sr	-0.96	0.83	0.87	0.76	0.86	0.84	0.98	0.20	0.94	0.89	0.75	-0.75	0.81	1.00																													
Zr	0.48	-0.72	0.30	-0.79	0.33	0.35	-0.40	0.92	-0.52	0.26	-0.80	0.80	-0.74	-0.21	1.00																												
Ba	0.38	-0.64	0.41	-0.72	0.43	0.45	-0.29	0.96	-0.43	0.36	-0.73	0.73	-0.66	-0.10	0.99	1.00																											
Zn	0.22	0.07	-0.86	0.19	-0.87	-0.88	-0.31	-0.95	-0.17	-0.83	0.21	-0.21	0.11	-0.50	-0.75	-0.82	1.00																										
Ce	-1.00	0.94	0.73	0.89	0.71	0.70	1.00	-0.03	0.99	0.76	0.88	-0.88	0.92	0.97	-0.43	-0.33	-0.28	1.00																									
Pb	-0.04	0.33	-0.70	0.43	-0.72	-0.74	-0.06	-1.00	0.08	-0.67	0.45	-0.45	0.36	-0.26	-0.89	-0.84	0.97	-0.03	1.00																								
Sa	-0.65	0.40	1.00	0.29	1.00	1.00	0.72	0.70	0.61	1.00	0.27	-0.27	0.36	0.84	0.36	0.46	-0.89	0.69	-0.74	1.00																							
Ga	-0.68	0.87	-0.06	0.92	-0.09	-0.11	0.61	-0.79	0.72	-0.01	0.92	-0.92	0.88	0.44	-0.97	-0.94	0.56	0.64	0.76	-0.12	1.00																						
Y	-0.55	0.28	0.98	0.17	0.89	0.89	0.63	0.78	0.51	0.98	0.15	-0.15	0.25	0.77	0.47	0.56	-0.94	0.60	-0.82	0.99	-0.24	1.00																					
Ni	-0.84	0.96	0.18	0.99	0.15	0.13	0.78	-0.63	0.86	0.22	0.99	-0.99	0.97	0.64	-0.89	-0.83	0.35	0.80	0.58	0.12	0.97	0.00	1.00																				
Rb	-0.99	0.92	0.77	0.86	0.75	0.73	1.00	0.02	0.99	0.80	0.86	-0.86	0.90	0.98	-0.38	-0.27	-0.33	1.00	-0.08	0.73	0.60	0.64	0.77	1.00																			
Nb	0.34	-0.05	-0.91	0.07	0.93	0.93	-0.42	-0.91	-0.29	-0.89	0.09	-0.89	-0.01	-0.60	-0.66	-0.74	0.99	-0.39	0.93	-0.94	0.46	-0.97	0.24	-0.44	1.00																		
La	-1.00	0.94	0.73	0.89	0.71	0.70	1.00	-0.03	0.99	0.76	0.88	-0.88	0.92	0.97	-0.43	-0.33	-0.28	1.00	-0.03	0.69	0.64	0.60	0.80	1.00	-0.39	1.00																	
Eu	-0.96	1.00	0.47	0.99	0.44	0.42	0.93	-0.36	0.98	0.51	0.99	-0.99	1.00	0.84	-0.70	-0.62	0.85	0.95	0.30	0.42	0.85	0.31	0.95	0.93	-0.07	0.95	1.00																
Mo	-0.65	0.40	1.00	0.29	1.00	1.00	0.72	0.69	0.61	1.00	0.28	-0.28	0.37	0.84	0.35	0.45	-0.88	0.70	-0.74	1.00	-0.11	0.99	0.13	0.73	-0.93	0.70	0.42	1.00															
Co	-0.42	0.66	-0.37	0.75	-0.40	-0.42	0.33	-0.95	0.46	-0.33	0.76	-0.76	0.69	0.13	-1.00	-1.00	0.80	0.36	0.92	-0.43	0.95	-0.53	0.85	0.31	0.72	0.36	0.65	-0.42	1.00														
Ta	-0.89	0.72	0.94	0.64	0.93	0.92	0.93	0.37	0.87	0.96	0.62	-0.62	0.70	0.99	-0.83	0.88	-0.64	0.92	-0.42	0.92	0.28	0.87	0.50	0.94	-0.73	0.92	0.74	0.92	-0.04	1.00													
W	0.12	-0.13	-0.95	-0.02	-0.96	-0.96	-0.87	-0.37	-0.93	0.00	0.00	-0.10	-0.66	-0.60	-0.68	0.98	-0.47	0.89	-0.96	0.38	-0.99	0.15	-0.52	1.00	-0.47	-0.16	-0.96	0.66	-0.78	1.00													
Hf	-0.12	0.40	-0.64	0.51	-0.66	-0.68	0.03	-1.00	0.17	-0.60	0.52	-0.52	0.44	-0.38	-0.93	-0.96	0.94	0.06	1.00	-0.68	0.81	-0.76	0.65	0.01	0.90	0.06	0.38	-0.68	0.95	-0.34	0.85	1.00											
Pr	0.32	-0.03	-0.91	0.09	-0.92	-0.93	-0.41	-0.91	-0.27	-0.89	0.11	-0.11	0.01	-0.58	-0.68	-0.76	1.00	-0.38	0.94	-0.93	0.48	-0.97	0.26	-0.42	1.00	-0.38	-0.05	-0.93	0.73	-0.71	0.99	0.90	1.00										
Lu	0.98	-1.00	-0.54	-0.98	-0.51	-0.49	-0.96	0.29	-0.99	-0.58	-0.97	0.97	-0.99	-0.88	0.64	0.55	0.03	-0.97	-0.23	-0.49	-0.81	-0.38	-0.93	-0.95	0.15	-0.97	-1.00	-0.49	-0.50	-0.79	0.23	-0.31	0.13	1.00									
Gd	-0.51	0.23	0.97	0.12	0.98	0.98	0.58	0.81	0.46	0.96	0.10	-0.10	0.20	0.74	0.51	0.61	-0.95	0.56	-0.85	0.99	-0.29	1.00	-0.05	0.60	-0.98	0.50	0.25	0.98	-0.50	0.84	-1.00	-0.80	-0.98	-0.33	1.00								
U	-0.69	0.87	-0.05	0.92	-0.08	-0.10	0.62																																				

The high correlation of Na₂O with Al₂O₃ and K₂O in the Kaffo suite suggests feldspar fractionation, with albite playing a key role in the late-stage crystallization of the magma. This is consistent with the observations of Bowden and Turner (1974) in the Younger Granites of Nigeria, where differentiation trends are controlled by feldspar and amphibole fractionation. The Fe₂O₃ and MnO contents in the Kaffo suite are highly correlated, which is characteristic of arfvedsonite-bearing granites. The negative correlation of Fe₂O₃ and MnO with SiO₂ and positive correlation with CaO and K₂O suggests their crystallization from a Fe-rich melt before the final silica saturation. This is consistent with the petrogenetic model proposed for the Ilímaussaq Complex, where Fe-rich amphiboles control iron distribution.

The element associations in the Kaffo suite suggest a petrogenetic evolution characterized by differentiation through fractional crystallization, feldspar fractionation, and crystallization of Fe-rich amphiboles. These trends are consistent with those observed in similar peralkaline granitic complexes worldwide, such as the Ilímaussaq Complex and the Younger Granites of Nigeria. The study of the Kaffo suite provides valuable insights into the petrogenetic evolution of peralkaline granites and highlights the importance of major element associations in understanding the geological history of these complexes.

Trace Element Clustering and Rare-Metal Mineralization

Positive correlation with elements like Zr (0.48) and Hf (-0.12), which are typically enriched in highly fractionated granites. The geochemical data from the Kaffo suite reveals positive correlations between elements such as Zr, Hf, Nb, Ta, and Y, which are typically enriched in highly fractionated granites (René, 2014; Wu, et al, 2017). These correlations suggest that silica enrichment is linked to extensive magmatic differentiation, with depletion of major elements due to feldspar fractionation. This trend is consistent with the Kaffo suite, where differentiation through fractional crystallization leads to the formation of peralkaline granites. The clustering of Zr, Nb, Ta, and Y in the study indicates their geochemical coherence, reflecting their association with peralkaline magmatism (Xie, et al., 2024). This trend is similar to those observed in other peralkaline granite complexes worldwide, such as the Kymi Granite of Finland (Rämö & Haapala, 2005) and the Eastern Desert Granites of Egypt (Abdel-Rahman, 1987; El-Bialy & Omar, 2015; .Shahin, 2016). In these complexes, HFSE enrichment is linked to magma-fluid interactions in a rift-related setting, suggesting a similar petrogenetic evolution for the Kaffo suite. The strong positive correlation between Sn, Nb, Ta, and W suggests potential for rare-metal mineralization, similar to the Abu Khruq Complex (Egypt), where Nb-Ta mineralization occurs in association with arfvedsonite granites (Helba et al., 1997; Moghazi, et al, 1999). This correlation highlights the importance of considering the geochemical coherence of these elements in the context of rare-metal mineralization.

Peralkaline Magmatism and Late-Stage Magmatic Processes

The REE geochemistry of the Kaffo suite provides valuable insights into the petrogenetic evolution of the complex and the late-stage magmatic processes that controlled the distribution of these elements. The strong positive correlation between Ce, La, Nd, and Pr indicates REE enrichment, a common feature of peralkaline granites (Mohamed et al., 2015; Anenburg, 2020). This mirrors studies on other peralkaline complexes worldwide, such as the Thor Lake Peralkaline Complex (Canada) (Pinckston & Dorian, 2011) and the Amis Complex (Namibia) (Kogarko, 1990). The negative correlation of Eu with other REEs suggests Eu depletion due to feldspar fractionation, a signature of highly evolved peralkaline systems. This trend is consistent with the petrogenetic model proposed for the Kaffo suite, where differentiation through fractional crystallization leads to the formation of peralkaline granites.

The REEs also correlate negatively with elements like Th, U, and Hf, suggesting different modes of enrichment and mobility in late-stage magmatic processes. Nb and Ta show similar trends, indicating their common association with pegmatitic and hydrothermal processes. This suggests that the REEs and other incompatible elements were mobilized and concentrated in late-stage magmatic fluids, which is consistent with the formation of REE-bearing phases such as eudialyte and xenotime in peralkaline granitic environments. The high positive correlation between Ce, La, Nd, and Pr indicates that these elements were enriched together in the late-stage magmatic fluids, which is consistent with the formation of REE-rich minerals such as bastnäsite and monazite (Mohamed et al., 2015). The negative correlation of Eu with other REEs suggests that Eu was depleted due to feldspar fractionation, which is consistent with the formation of Eu-depleted minerals such as quartz and feldspar.

The Alkali and High Field Strength Elements Implications for Late-Stage Fluid Processes and Mineralization

Na₂O shows a strong positive correlation with Y (0.99), Sn (1.00), and Mo (1.00), implying an association with late-stage fluid processes, which often mobilize HFSEs in peralkaline systems. Rb and K₂O are highly correlated (0.86), indicating their control by potassic feldspar and mica. The strong Rb-Ta (-0.73) and Rb-Nb (-0.44) anti-correlation suggests that Ta-Nb mineralization occurs independently of K-feldspar fractionation and is likely related to late-stage fluid mobilization. The geochemical data from the Kaffo suite reveals several significant correlations and anti-correlations between alkali and high field strength elements (HFSEs), which provide valuable insights into the late-stage fluid processes and mineralization in the complex.

The strong positive correlation between Na₂O and Y, Sn, and Mo implies an association with late-stage fluid processes, which often mobilize HFSEs in peralkaline systems. This suggests that the Na₂O-rich fluids played a crucial role in the mobilization and concentration of these elements in the Kaffo suite. Similar associations have been observed in other peralkaline complexes worldwide, such as the Ilímaussaq Complex of Greenland (Hunt et al., 2014, Lindhuber, et al., 2015) and the Amis Complex of Namibia (Schmitt, et al., 2002). The high correlation between Rb and K₂O indicates their control by potassic feldspar and mica, which are common minerals in peralkaline granites. This suggests that the Rb and K₂O contents in the Kaffo suite were influenced by the crystallization of these minerals. The strong Rb-Ta and Rb-Nb anti-correlation suggests that Ta-Nb mineralization occurs independently of K-feldspar fractionation and is likely related to late-stage fluid mobilization. This implies that the Ta-Nb mineralization in the Kaffo suite was controlled by a separate fluid phase that was enriched in these elements. Similar anti-correlations have been observed in other peralkaline complexes worldwide, such as the Thor Lake Peralkaline Complex of Canada (Pinckston & Dorian, 2011) and the Eastern Desert Granites (Egypt) (Mohamed & El-Sayed, 2008; Zoheir, 2019).

Implications for Mineral Exploration

The correlation between Nb and Ta with Na₂O and Sn suggests a pegmatitic to hydrothermal enrichment mechanism, which is consistent with the formation of rare-metal deposits in peralkaline granites (Larsen & Sørensen 1987; Kogarko & Nielsen 2020). The positive correlation between Zr and Hf indicates potential for zircon and Hf-bearing accessory minerals, which are common in peralkaline granites. The negative correlation between U and Th with REEs implies that REE mineralization is largely decoupled from uranium-bearing phases, which is consistent with the formation of REE deposits in peralkaline granites. The

strong correlation between Sn and W suggests Sn-W mineralization linked to late-stage pegmatitic activity, which is a common feature of peralkaline granites.

The study suggests that the Kaffo suite shares geochemical affinities with other rift-related peralkaline granites known for rare-metal mineralization, such as the Mount Weld of Australia (Lottermoser, 1990; Zhukova, et al., 2021; Cook, et al., 2023) and Strange Lake of Canada deposits (McConnell & Batterson, 1987; Salvi & Williams-Jones, 2006; McClenaghan, et al., 2019). The strong Nb-Ta-Zr-Y association highlights the potential for economic deposits similar to those in Mount Weld and Strange Lake. This geochemical characterization provides a framework for exploration, suggesting that detailed mineralogical and fluid inclusion studies should be conducted to confirm ore potential. The Kaffo suite's geochemical characteristics are consistent with the formation of rare-metal deposits in peralkaline granites, including:

- (a) Pegmatitic to hydrothermal enrichment mechanism, as indicated by the correlation between Nb and Ta with Na₂O and Sn.
- (b) Potential for zircon and Hf-bearing accessory minerals, as indicated by the positive correlation between Zr and Hf.
- (c) Decoupling of REE mineralization from uranium-bearing phases, as indicated by the negative correlation between U and Th with REEs.
- (d) Sn-W mineralization linked to late-stage pegmatitic activity, as indicated by the strong correlation between Sn and W.

CONCLUSION

The Kaffo Albite Arfvedsonite suite reveals strong geochemical similarities with well-known peralkaline granite-related rare-metal deposits worldwide. The clustering of HFSEs, REEs, and Sn-W mineralization indicators suggests a high potential for economic deposits. The Pearson's correlation analysis of Albite-Arfvedsonite granite reveals a complex magmatic evolution with significant mineralization potential for Nb-Ta, Sn, Zr, and REEs. The data align with global studies on peralkaline granites and suggest that hydrothermal fluids played a key role in concentrating economic minerals. Further exploration and detailed mineralogical and fluid inclusion studies are recommended to confirm ore potential.

REFERENCES

- Abdel-Rahman, A.M. (1987). Crystallization of amphiboles in the plutonic complexes of northeastern Egypt: implications for magma evolution. *Neues Jahrb Mineral* 157:319-335
- Anenburg, M. (2020). Rare earth mineral diversity controlled by REE pattern shapes. *Mineral. Mag.* 84, 629-639. <https://doi.org/10.1180/mgm.2020.70>.
- Ashano, E. C. (2009). Measurement of Parameters for construction of Geochemical Maps. In: D.O Lambert-Aikhionbare and A.I Olayinka (Eds.) *proceedings of field mapping standardization workshop* (pp 165-179). Ibadan University press
- Bailey, J.C., Sørensen, H., Andersen, T., Kogarko, L.N., Rose-Hansen, J. (2006). On the origin of microrhythmic layering in arfvedsonite lujavrite from the Ilímaussaq alkaline complex, South Greenland. *Lithos* 91, 301-318.
- Bowden, P., & Turner, D. C. (1974). Peralkaline and Associated Ring Complexes in the Nigeria-Niger Province, West Africa. In: H. Sorensen, Ed., *Alkaline Rocks*, John Wiley & Sons New York, 330-351.
- Bowden, P., & Turner, D.C. (1974). Peralkaline and associated ring-complex granites of Nigeria. *Journal of the Geological Society of London*, 130, 105-122.

- Bowden, P., Bennett, J.N., Whitley, J.E., & Moyes, A.B. (1979). Rare earths in Nigerian Mesozoic granites and related rocks. In: Ahrens LH (ed) Origin and distribution of the elements, 2nd edn, Pergamon, Oxford, pp 479–491
- Bowden, P, Black, R, Martin, R. F, Ike, E. C, Kinnaird, J.A, & Batchelor, R.A. (1987) Niger-Nigerian alkaline ring complexes: a classic example of African Phanerozoic anorogenic mid-plate magmatism. In: Fitton JG, Upton BGJ (eds) Alkaline igneous rocks. *Geological Society of London Special Publication*, 30:357–379
- Cook, C.N., Ciobanu, C.L., Wade, B.P., Gilbert, S.E., & Alford, R. (2023). Mineralogy and Distribution of REE in Oxidised Ores of the Mount Weld Laterite Deposit, Western Australia. *Minerals* 13, (656) pp1-34. <https://doi.org/10.3390/min13050656>
- Coulon, C., Vidal, P., Dupuy, C., Baudin, P., Popoff, M., Maluski, H., & Hermitte, D. (1996). The Mesozoic to early Cenozoic magmatism of the Benue Trough (Nigeria); geochemical evidence for the involvement of the St. Helena plume. *Journal of Petrology* 37, 1341–135
- El-Bialy M.Z & Omar M.M. (2015). Spatial association of Neoproterozoic continental arc I-type and post-collision A-type granites in the Arabian-Nubian Shield: The Wadi Al-Baroud Older and Younger Granites, North Eastern Desert, Egypt. *Journal of African Earth Sciences* 103: 1-29.
- Helba, H., Trumbull, R. B., & Morteani, G. (1997). Rare-metal mineralization in arfvedsonite granites, Abu Khruq Complex, Egypt. *Mineralium Deposita*, 32(6), 607-623.
- Hunt, E.J., Finch, A.A., Donaldson, C.H., (2014). Magma mixing in layered kakortokites – Ilímaussaq Complex. S. Greenland, AGU Fall Meeting, San Francisco
- Jacobson, R.R.E., & MacLeod, W.N. (1977). Geology of the Jos Plateau, Nigeria. *Bulletin of the Geological Survey of Nigeria*, 32, 1-91.
- Kinnaird, J. A. (1979). Mineralisation associated with the Nigerian mesozoic ring complexes studies in geology. *Salamanca* 14, 189-220.
- Kinnaird, J.A. (1985). Hydrothermal alteration and mineralization of the Younger Granite Complexes of Nigeria. *Journal of African Earth Sciences*, 3(1), 185-222
- Kogarko, L. N. (1990). Ore-forming potential of alkaline magmas. *Lithos*, 26, (1-2), 167-175 [https://doi.org/10.1016/0024-4937\(90\)90046-4](https://doi.org/10.1016/0024-4937(90)90046-4).
- Kogarko, L & Nielsen, T.F.D. (2020). Chemical Composition and Petrogenetic Implications of Eudialyte-Group Mineral in the Peralkaline Lovozero Complex, Kola Peninsula, Russia. *Minerals*, 10, 1036; doi: 10.3390/min10111036
- Larsen, L.M., & Sørensen, H. (1987). The Ilímaussaq intrusion – progressive crystallisation and formation of layering in an agpaitic magma. *Geological Society of London, Special Publication* 30, 473–488.
- Lindhuber, M.J., Marks, M.A.W., Bons, P.D., Wenzel, T., & Markl, G. (2015). Crystal mat-formation as an igneous layering-forming process: textural and geochemical evidence from the ‘lower layered’ nepheline syenite sequence of the Ilímaussaq complex, South Greenland. *Lithos* 224, 295–309.
- Lottermoser, B. G. (1990). Rare-earth element mineralisation within the Mt. Weld carbonatite laterite, Western Australia, *Lithos*, 24 (2), 151-167. [https://doi.org/10.1016/0024-4937\(90\)90022-S](https://doi.org/10.1016/0024-4937(90)90022-S).
- McClenaghan, M.B., Paulen, R.C., & Kjarsgaard, I.M. (2019). Rare metal indicator minerals in bedrock and till at the Strange Lake peralkaline complex, Quebec and Labrador, Canada. *Canadian Journal of Earth Science*, 56: 857–869 [dx.doi.org/10.1139/cjes-2018-0299](https://doi.org/10.1139/cjes-2018-0299)
- McConnell, J. W. & Batterson, M.J. (1987). The strange lake Zr-Y-Nb-Be-REE deposit, labrador: A geochemical profile in till, lake and stream sediment, and water. *Journal of*

- Geochemical Exploration*, 29,(1-3) 105-127, [https://doi.org/10.1016/0375-6742\(87\)90073-2](https://doi.org/10.1016/0375-6742(87)90073-2).
- Mohamed F.H & El-Sayed M.M. (2008). Post-orogenic and anorogenic A-type fluorite-bearing granitoids, Eastern Desert, Egypt: petrogenetic and geotectonic implications. *Chemie der Erde* 68: 431-450.
- Mohamed, E. I., Baher A. E., Gehan M.A., Amera, M. E., & Koichiro, W. (2015). Altered granitic rocks, Nusab El Balmum Area, Southwestern Desert, Egypt: Mineralogical and geochemical aspects of REEs, *Ore Geology Reviews*,70, 252-261
- Mokhtar H., Surour, A.A., Azer, M.K, Ren, M., & Said, A. (2024) Geochemistry and mineral chemistry of granitic rocks from west Wadi El Gemal area, southern Eastern Desert of Egypt: indicators for highly fractionated syn- to post-collisional Neoproterozoic felsic magmatism. *Acta Geochim.* <https://doi.org/10.1007/s11631-024-00714-1>
- Ogunleye, P. O., Garba. I., & Ike, E. C. (2006). Factors contributing to enrichment and crystallization of niobium in pyrochlore in the Kaffo albite arfvedsonite granite, Riruwai Complex, Younger Granites province of Nigeria. *Journal of African Earth Sciences*, 44(3), 372-382.
- Olasehinde, A & Ashano E.C. (2021): Data Driven Predictive Modelling of Mineral Prospectivity Using Principal Component Analysis: A Case Study of Riruwai Complex. *Advances in Applied Science Research*. Vol 12 no 7:33 pp1-7
- Pinckston, D. & Dorian, S. (2011). Mineralogy of the Lake zone, Thor Lake rare-metals deposit, N.W.T., Canada. *Canadian Journal of Earth Sciences*. 32. 516-532. 10.1139/e95-044.
- Rämö, O. T., & Haapala, I. (2005). Petrogenesis of the Kymi granite, Finland: A study of the whole-rock geochemistry and mineral chemistry. *Lithos*, 80(1-4), 1-22.
- René M. (2014). Composition of coexisting zircon and xenotime in rare-metal granites from the Krušné Hory/Erzgebirge Mts. (Saxothuringian Zone, Bohemian Massif). *Mineral Petrol*, 108: 551-569
- Salvi, S & Williams-Jones, A.E. (2006). Alteration, HFSE mineralisation and hydrocarbon formation in peralkaline igneous systems: Insights from the Strange Lake Pluton, Canada, *Lithos*, 91, Issues 1-4, 19-34
- Schmitt, A. K., Trumbull, R.B., Dulski, P., & Emmermann R. (2002). Zr-Nb-REE Mineralization in Peralkaline Granites from the Amis Complex, Brandberg (Namibia): Evidence for Magmatic Pre-form Melt Inclusions. *Economic Geology* 97 (2): 399-413. doi: <https://doi.org/10.2113/gsecongeo.97.2.399>
- Moghazi, A.M., Mohamed, F.H., & Kanisawa, S. (1999): Geochemistry and petrogenesis of late Proterozoic plutonic rock suites in the Homrit Waggat and El Yatima areas, Eastern Desert Egypt. *J. African Earth Sci.* 29: 535- 549.
- Sørensen H., & Larsen L.M (2001). The hyper-agpaitic stage in the evolution of the Ilímaussaq alkaline complex, South Greenland. *Geol Greenl Surv Bull* 190:83-94
- Wu, F., Liu, X., Ji, W., Wang, J., & Yang, L. (2017). Highly fractionated granites: Recognition and research; *Science China Earth Sciences* 60, 1201-1219. doi: 10.1007/s11430-016-5139-1
- Xie, M., Fan, H., Sakyi, P.A., Yang, K., Li, X., She, H., Liang, G., & Han, C (2024). Mineralizations of Nb-Ta-Rb-Zr and rare-earth elements in Boziguoer, South Tianshan, NW China: Geochronology and geochemistry of monazite and bastnäsite. *Ore Geology Reviews*, (168). <https://doi.org/10.1016/j.oregeorev.2024.106034>
- Zhukova, I.A., Stepanov, A.S., Jiang, S., Murphy, D., Mavrogenes, J., Allen, C., Chen, W., & Bottrill, R. (2021). Complex REE systematics of carbonatites and weathering products from uniquely rich Mount Weld REE deposit, Western Australia, *Ore Geology Reviews*, 139B, <https://doi.org/10.1016/j.oregeorev.2021.104539>

Zoheir, B., Goldfarb, R., Holzheid, A., Helmy, H., & El Sheikh, A. (2019). Geochemical and geochronological characteristics of the Um Rus granite intrusion and associated gold deposit, Eastern Desert, Egypt. *Geosci Front.* <https://doi.org/10.1016/j.gsf.2019.04.012>