



## Synthesis, growth and characterization of magnesium chloride doped L-alanine cadmium chloride single crystal: For nonlinear optical application

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### ABSTRACT

#### KEYWORDS:

Solution method;  
Doping;  
Crystal Growth;  
Harmonic generation;  
Nonlinear optics

The aim of this research was to synthesize and characterize pure and magnesium chloride ( $MgCl_2$ ) doped L-alanine cadmium chloride (LACC) single crystals. Pure and  $MgCl_2$  doped LACC single crystals were synthesized by solution method with slow evaporation solution growth technique at room temperature. The single crystal X-ray diffraction studies of pure, 1 and 2 mol%  $MgCl_2$  doped LACC single crystals revealed monoclinic crystal structure with  $C_2$  space group. The optical properties of pure and  $MgCl_2$  doped LACC single crystals investigated by UV-VIS/NIR spectrometer confirmed that the crystals were transparent in the wavelength range of 230-1100 nm. The optical band gap energy of pure and doped LACC single crystals were found to have the same value of 5.4 eV. The energy dispersive X-ray analysis indicated the incorporation of magnesium and chlorine atoms in LACC single crystal. The second harmonic generation efficiency of 1 and 2 mol%  $MgCl_2$  doped LACC crystals were analyzed by Kurtz-Perry powder technique and found to be 1.75 and 2 times greater than that of the standard potassium dihydrogen phosphate crystal, respectively.

### INTRODUCTION

The core attention of nonlinear optical (NLO) studies is to modify the phase, frequency or amplitude of intense electromagnetic input field using NLO materials for photonic applications (Boyd, 2019). Nonlinear optical materials have been extensively studied in the recent years, due to their potential applications on optical computing, laser technology, harmonic generation, optical communication, optical data

storage technology, signal processing and manipulation. Therefore, currently there is a need to produce high efficiency single crystals of NLO materials (Marudhu *et al.*, 2013; Raguram *et al.*, 2016). Much attention has been paid to organic NLO materials due to their promising properties, such as fast optical response time and high nonlinearity, compared to the inorganic materials. The aggregate of materials which have large nonlinear optical properties with resistance to physical and

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chemical attack has led to the investigation of semi-organic materials (Kaliammal *et al.*, 2020). Amino acids are favorable organic materials displaying specific features of interest, such as molecular chirality which secures non-centrosymmetric crystallographic structure, absence of strongly conjugated bonds which leads to wide transparency ranges in the visible and ultra-violet spectral regions and zwitter ionic nature of the molecule which favors crystal hardness for applications in devices (Natarajan *et al.*, 2006; Raguram *et al.*, 2016). L-alanine is an amino acid group that forms a series of complexes upon reaction with di erent acids.

Reports are available on L-alanine based NLO crystals such as: L-alanine cadmium chloride (Dhanuskodi *et al.*, 2007), L-alanine acetate (Kumar *et al.*, 2005), L-alanine Hydrogen chloride (Rose *et al.*, 2010), L-alanine sodium nitrate (Fleck and Petrosyan, 2009), L-alanine potassium chloride (Prabha and Palaniswamy, 2010) and L-alanine maleate (Karunanithi *et al.*, 2012). L-alanine cadmium chloride is an amino acid derivative NLO material with high second harmonic generation (SHG) e ciency and it crystallizes in the monoclinic crystal system with non-centrosymmetric space group (Jothimani and Selvarajan, 2017; Radhika *et al.*, 2013).

Few reports are available on the synthesis of L-alanine cadmium chloride single crystals by solution growth method. As far as the authors are aware, there is no report on magnesium chloride doped L-Alanine cadmium chloride single crystals. Since doping can influence many of the useful properties like, optical transparency, second harmonic generation (SHG), crystalline perfection which may in turn

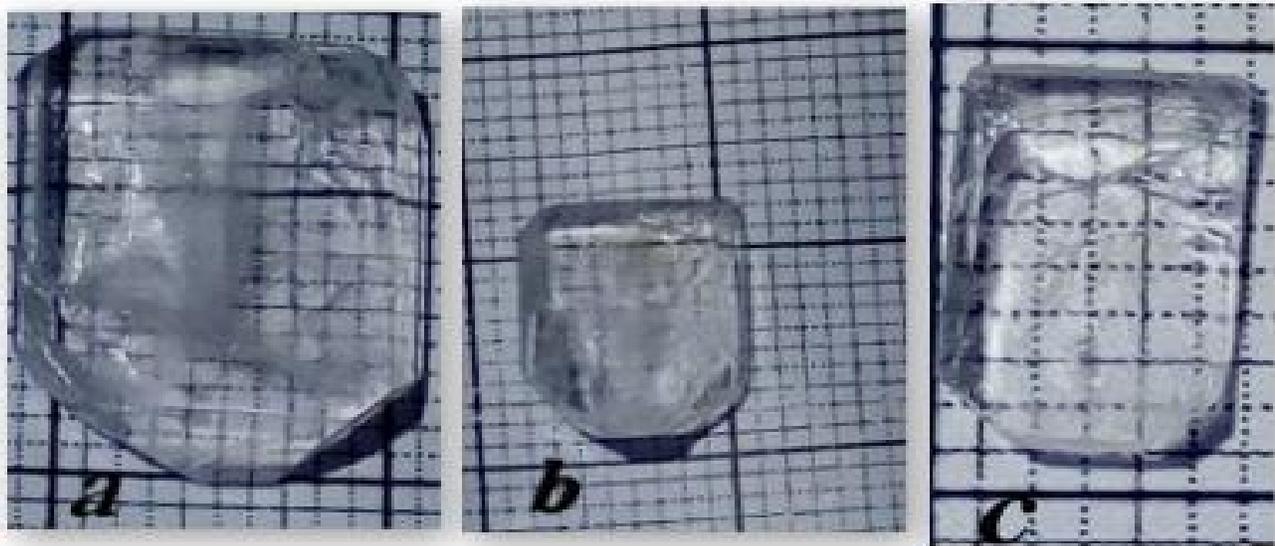
influence the physical properties depending on the degree of doping and the accommodating capability of the parent crystal (Shkir *et al.*, 2014). In the current work magnesium chloride doped L-alanine cadmium chloride single crystals were grown by slow evaporation solution growth method and the results of structural, optical, SHG and composition studies of the grown crystals were reported.

## MATERIALS AND METHODS

In this work analytical grade L-Alanine (Sisco Research laboratories, 99%), cadmium chloride (Uni-CHEM, 99%) and magnesium chloride (Uni-CHEM, 98%) were used without further process. The parent compound was synthesized by taking an equimolar ratio of L-alanine and cadmium chloride that is obtained by dissolving 8.0901 g L-Alanine and 22.8353 g cadmium chloride. The calculated amount of L-alanine has been dissolved by distilled water in a beaker and placed on a magnetic hot plate regulated at 30°C then cadmium chloride was added. It has been stirred continuously for four hours to obtain homogenous supersaturated solution. Afterwards the solution has filtered by using Whatman filter paper into a 500 ml beaker. The filtered solution has been kept free from dust and other contamination by covering it with porous cover so the rate of evaporation could be minimized. 1 mol% and 2 mol% MgCl<sub>2</sub> doped L-Alanine cadmium chloride single crystals were synthesized by adding 2.0333 g and 4.0666 g MgCl<sub>2</sub> into the parent compound solution respectively. A similar stirring process has been followed for 1 and 2 mol% of MgCl<sub>2</sub> doped L-alanine cadmium chloride. After a period of 30 days, optically clear transparent crystals were harvested from supersaturated

solution. Thus, the grown pure, 1 and 2 mol%  $\text{MgCl}_2$  doped LACC crystals were shown in Fig.

1.



**Fig. 1. Photograph of (a) undoped, (b) 1 mol%  $\text{MgCl}_2$  doped and (c) 2 mol%  $\text{MgCl}_2$  doped LACC single crystals respectively.**

## RESULTS AND DISCUSSION

### Single crystal X-ray diffraction

The crystal structure of 1 and 2 mol%  $\text{MgCl}_2$  doped LACC single crystal samples were carried out using a Bruker AXS Kappa APEXII single crystal CCD diffractometer equipped with graphite monochromated Mo K ( $\lambda = 0.07107$  nm) radiation. This analysis has revealed that the single crystals of 1 and 2 mol% of  $\text{MgCl}_2$  doped LACC samples crystallized in monoclinic crystal structure and lattice angle  $\alpha = \gamma = 90^\circ$  and  $\beta = 116.41^\circ$ . The calculated lattice parameters of 1 and 2 mol% of  $\text{MgCl}_2$  doped LACC crystals in ( $\text{\AA}$ ) were  $a = 16.282$ ,  $b = 7.261$ ,  $c = 8.008$  and  $a = 16.286$ ,  $b = 7.268$ ,  $c = 8.011$  respectively. The unit cell parameters and crystal structure for pure LACC crystal in ( $\text{\AA}$ ) were  $a = 16.298$ ,  $b = 7.259$ ,  $c = 7.981$  and monoclinic respectively as reported by other

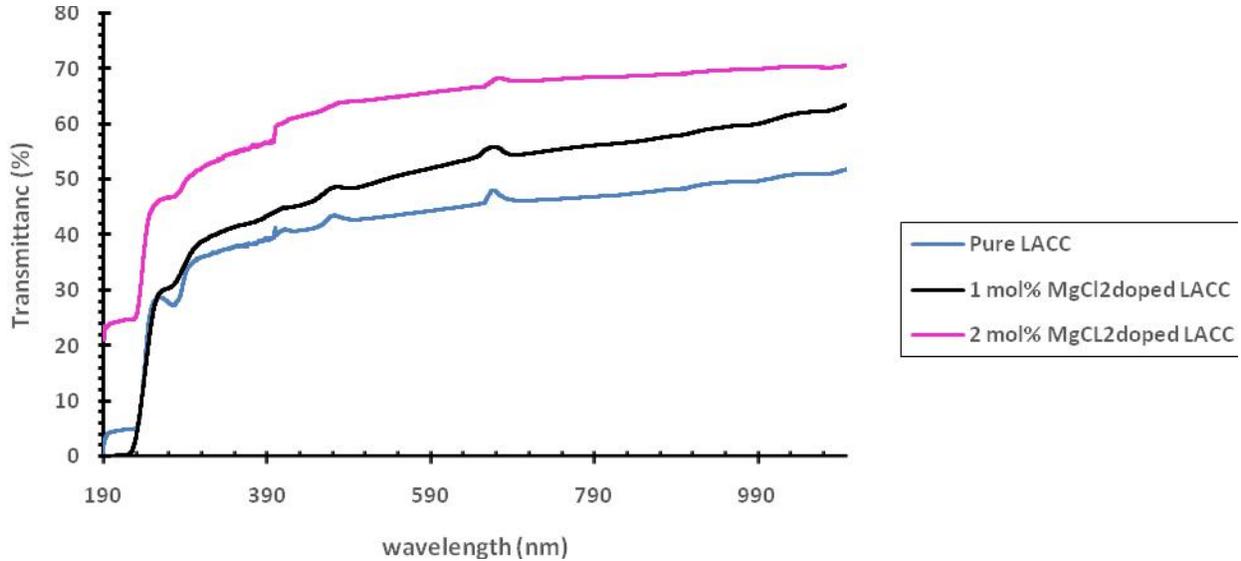
authors (Dhanuskodi *et al.*, 2007). It is observed that both undoped and  $\text{MgCl}_2$  doped LACC crystals crystallize in same structure however, slight change in the lattice parameters were observed from the doped crystal compared to the pure LACC crystal. The changes in the lattice parameter may be due to the incorporation of  $\text{MgCl}_2$  in LACC crystal lattice. The grown crystals belongs to space group  $C_2$  which is recognized as non-centrosymmetric, thus satisfying an essential material criteria for the SHG activity of the crystal (Fentaw *et al.*, 2019).

### UV-Vis NIR analysis

From the optical application point of view, it is necessary to analyze the optical parameters, such as the transmission, cut-off wavelength and band gap energy of the NLO single crystal. In this study these optical parameters have been

studied by using Perkin Elmer Lambda 35 UV-VIS-NIR spectrophotometer in the wave length range between 190-1100 nm to cover near

ultraviolet, visible and near IR regions. The recorded spectrums were shown in Fig.2..



**Fig.2. Optical transmittance of pure, 1 and 2 mol % of MgCl<sub>2</sub> doped LACC crystals.**

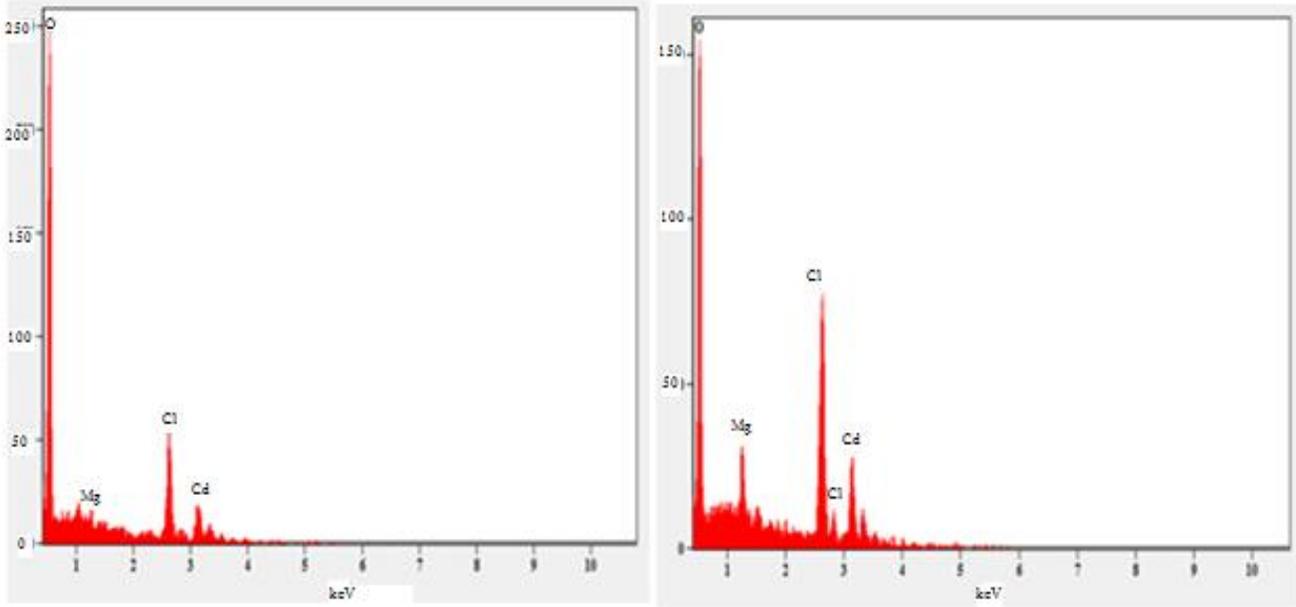
From the spectrum it is observed that the crystals were transparent in the wavelength range 230 -1100 nm. NLO materials can be of practical use only if they have a wide transparency range, thus the optical transmittance range of the grown crystals make them convenient materials for nonlinear optical applications (Ahlam *et al.*, 2013; Charoen-In *et al.*, 2010). The result confirmed that the lower cutoff wavelength of the crystals were almost the same irrespective of the dopant concentration, however, the percentage of transmittances have increased for 1mol% and 2 mol% MgCl<sub>2</sub> doped LACC crystal. The increase in optical transmittance could be due the enhancement in the crystalline quality and absence of major defects as crystalline defect affect the optical properties (Jothimani and Selvarajan, 2017). The band gap energy was calculated using the relation  $E_g = 1240/\lambda_{min}$ ,

where  $\lambda_{min}$  is the cut-off wavelength of the light which is 230 nm and it was found to be 5.4 eV for pure and doped crystals (Raguram *et al.*, 2016).

### Energy dispersive X-ray spectroscopy (EDX) analysis

Energy dispersive X-ray spectroscopy (EDX) is an instrument that used to identify the elemental composition of the samples. The incorporation of 1 and 2 mol% of MgCl<sub>2</sub> into the crystal lattice of LACC was confirmed by JEOL-6390LV scanning electron microscope attached to EDX and it is shown in Fig.3. The EDX spectrum confirmed presence of expected elements such as oxygen, chlorine, magnesium and cadmium in both 1 and 2 mol% of MgCl<sub>2</sub> doped LACC crystal. The weight and atomic percentage of identified elements in the doped NLO samples were presented in Table1. The

quantitative and qualitative EDX results clearly showed increase in the atomic percentage of Mg and Cl signifying incorporation of the  $MgCl_2$  in the LACC single crystal structure.



**Fig. 3. EDX spectrum of 1 and 2 mol% of  $MgCl_2$  doped LACC crystals respectively.**

**Table 1: Energy dispersive X-ray (EDX) analysis**

Elements	1mol% $MgCl_2$ +LACC		2mol% $MgCl_2$ +LACC	
	Weight%	Atom%	Weight%	Atom%
<b>OK</b>	74.28	91.19	62.49	82.57
<b>MgK</b>	0.72	0.57	5.41	4.70
<b>ClK</b>	10.20	5.65	16.38	9.77
<b>CdL</b>	14.80	2.59	15.72	2.96
<b>Total</b>	100.00	100.00	100.00	100.00

**Second Harmonic Generation (SHG) test**

The second harmonic generation efficiency analysis of the samples were carried out by Kurtz-Perry powder technique by packing very fine powder of  $MgCl_2$  doped LACC sample in a microcapillary tube and a fundamental beam

(1064 nm) from Nd-YAG laser was incident on the  $MgCl_2$  doped LACC crystals (Kushwaha *et al.*, 2011). A second harmonic generated green light beam (532 nm) was emerged from the samples. Potassium dihydrogen phosphate  $KH_2PO_4$  (KDP) crystalline sample in the form of

very fine powder was used as a reference material. The beam voltage of the transmitted radiation was 12 mV, when KDP crystalline sample in the form of powder was used as a reference material. The measured values of the transmitted beam voltage through LACC doped by 1 and 2 mol%  $\text{MgCl}_2$  samples were 21 and 24 mV respectively. Thus, the SHG efficiency of 1 and 2 mol%  $\text{MgCl}_2$  doped LACC samples are 1.75 and 2 times greater than that of the standard KDP crystal respectively. The SHG efficiencies of pure LACC single crystal was 0.87 times greater than that of the standard KDP crystal reported by (Kalaiselvi *et al.*, 2013). The enhancement of SHG efficiency may be either due to the improved in crystallinity or the change in electronic structure of LACC crystals due to the doping of  $\text{MgCl}_2$  (Bhagavannarayana and Kushwaha, 2010). The current results indicate that 1mol% and 2mol%  $\text{MgCl}_2$  doped LACC crystals are the better candidates for NLO applications than the undoped LACC single crystal.

## CONCLUSION

Pure, 1 and 2 mol% of  $\text{MgCl}_2$  doped LACC single NLO crystals were grown by using slow evaporation solution growth method at room temperature. The grown single crystals were subjected to structural and optical properties, elemental composition and SHG efficiency characterizations. The single crystal XRD analysis of 1 and 2 mol% of  $\text{MgCl}_2$  doped LACC crystals confirmed monoclinic crystal system with space group  $C_2$ . The UV-Vis-NIR spectral studies revealed that Pure, 1 and 2 mol% of  $\text{MgCl}_2$  doped LACC crystals transmit in the entire UV-visible and IR region. The transmission of the LACC single crystal increased for 1mol% and 2 mol%  $\text{MgCl}_2$  doped

LACC crystal. The EDAX analysis of the doped LACC single crystal indicated the presence of  $\text{MgCl}_2$ . Second harmonic generation tests revealed that the efficiency of 1 and 2 mol% of  $\text{MgCl}_2$  doped LACC single crystals were 1.75 and 2 times greater than that of the standard KDP crystal respectively. The present results indicated that the pure and  $\text{MgCl}_2$  doped LACC single crystals were a potential candidate for nonlinear optical applications.

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## References

- Ahlam M., Hemaraju B., and Prakash A.G. 2013. Growth and characterization of pure and doped organic nonlinear optical single crystal: L-Alanine alaninium nitrate (LAAN). *Optik*. **124(23)**: 5898-5905.
- Bhagavannarayana G. and Kushwaha S.K. 2010. Enhancement of SHG efficiency by urea doping in ZTS single crystals and its correlation with crystalline perfection as revealed by Kurtz powder and high-resolution X-ray diffraction methods. *J. Appl. Crystallogr.* **43(1)**: 154-162.
- Boyd R.W. 2019. Nonlinear optics: Academic press.
- Charoen-In U., Ramasamy P. and Manyum P. 2010. Comparative study on L-alaninium maleate single crystal grown by Sankaranarayanan-Ramasamy (SR) method and conventional slow evaporation solution technique. *J. Cryst. Growth* **312(17)**: 2369-2375.
- Dhanuskodi S., Vasantha K. and Mary P.A. 2007. Structural and thermal characterization of a semiorganic NLO material: L-alanine cadmium chloride. *Spectrochim. Acta A: Mol. Biomol. Spectrosc.* **66(3)**: 637-642.
- Fentaw D.A., Peter M.E. and Abza T. 2019. Synthesis and characterization of lanthanum chloride doped L-alanine maleate single crystals. *J. Cryst. Growth* **522**: 1-4.

- Fleck M. and Petrosyan A. 2009. Comments on papers reporting IR spectra and other data of alleged L alanine alaninium nitrate and L alanine sodium nitrate crystals. *Crystal Research and Technology: Journal of Experimental and Industrial Crystallography* **44(7)**: 769-772.
- Jothimani R. and Selvarajan P. 2017. Effect of Nickel sulphate on Growth, Structural, Optical, Mechanical and thermal properties of L-alanine Single Crystals (LANS). Paper presented at the IOP Conference Series. Materials Science and Engineering (Online).
- Kalaiselvi P., Raj S.A.C. and Vijayan N. 2013. Linear and nonlinear optical properties of semiorganic single crystal: L-Alanine cadmium chloride (LACC). *Optik*. **124(24)**: 6978-6982.
- Kaliammal R., Sudhahar S., Parvathy G., Velsankar K., and Sankaranarayanan K. 2020. Physicochemical and DFT studies on new organic Bis-(2-amino-6-methylpyridinium) succinate monohydrate good quality single crystal for nonlinear optical applications. *J. Mol. Struct.*, 128069.
- Karunanithi U., Arulmozhi S. and Madhavan J. 2012. Structural, optical and dielectric studies on pure and doped *Lalaninium maleate* single crystals. *J. Appl. Phys.* **1**: 19-23.
- Kumar R.M., Babu D.R., Jayaraman D., Jayavel R. and Kitamura K. 2005. Studies on the growth aspects of semi-organic L-alanine acetate: a promising NLO crystal. *J. Cryst. Growth*. **275**:1935 - 1939.
- Kushwaha S., Vijayan N., Maurya K., Kumar A., Kumar B., Somayajulu K. and Bhagavannarayana G. 2011. Enhancement in crystalline perfection and optical properties of benzophenone single crystals: the remarkable effect of a liquid crystal. *J. Appl. Crystallogr.* **44(4)**: 839-845.
- Marudhu G., Krishnan S., Thilak T., Samuel P., Vinitha G. and Pasupathi G. 2013. Optical, thermal and mechanical studies on nonlinear optical material diglycine barium chloride monohydrate (DGBCM) single crystal. *J. Nonlinear Opt. Phys. Mater* **22(4)**: 1350043.
- Natarajan S., Britto S.M. and Ramachandran E. 2006. Growth, thermal, spectroscopic, and optical studies of L-alaninium maleate, a new organic nonlinear optical material. *Cryst. Growth Des.* **6(1)**: 137-140.
- Prabha D. and Palaniswamy S. 2010. Growth and characterization of NLO material: L-alanine potassium chloride. *Rasayan J. Chem.* **3**: 517-524.
- Radhika P., Jayakumari K. and Mahadevan C. 2013. Growth and characterization of pure and oxalic acid doped L arginine acetate single crystals. *Int J Eng Res Appl.* **3**: 1841-1849.
- Raguram T., Rajni K., Shanmugam P. and Meiyazhagan S. 2016. Synthesis and characterisation of undoped and methyl orange (dye) doped L-alanine acetate single crystal. *Int. J. ChemTech Res.* **9(7)**: 678-687.
- Ramajothi J., Muraleedharan R. and Vinitha G. 2019. Investigation of Mechanical, Thermal and Third Order Nonlinear Optical Properties of L-Alanine Cadmium Chloride Single Crystal. *Int. J. Chem. Tech. Res.* **12**: 141-154.
- Rose A.L., Selvarajan P. and Perumal S. 2011. Studies on growth and characterization of an NLO crystal: L-alanine hydrogen chloride (LAHC). *Mater. Chem. Phys.* **130(3)**: 950-955.
- Shkir M., Riscob B., Khan M.A., AlFaify S., Dieguez E. and Bhagavannarayana G. 2014. Effect of organic ligands (l-Proline and l-Methionine) on growth, structural, vibrational, crystalline perfection, SHG efficiency, microscopic and optical properties of KDP single crystals. *Spectrochim. Acta A: Mol. Biomol. Spectrosc.* **124**: 571-578.