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GEOCHEMICAL EVALUATION OF CAMPANIAN-MAASTRITCHIAN CLAY-SHALE SEDIMENTS OF PATTI FORMATION, SOUTHERN BIDA AND MAMU FORMATION, NORTHERN ANAMBRA BASINS

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ABSTRACT

Two basins (Southern Bida and Northern Anambra Basins) were investigated to deduce weathering, paleooxygenation, provenance, depositional environment and tectonic setting, as well as to establish a relationship between the two basins. The obtained high values of calculated weathering indices such as Chemical index of alteration (CIA > 90), Chemical Index of Weathering (CIW > 90), Plagioclase Index of Alteration (PIA > 90) and the Al_2O_3 -(CaO + Na_2O)-K₂O ternary relationship for the clay – shale sediments from both basins indicate intense weathering in the source area. Important geochemical ratios such as V/Cr, Cu/Zn, Ni/Co, (Cu+Mo)/Zn, revealed predominantly oxic conditions for the clay – shale sediments from both basins, although, a more reducing or an anoxic condition cannot be ruled out for the clay – shale sediments from the Southern Bida basin due to high ratios of U/Th (1.93-5.67) and Cu/Zn (0.19-5.00). In addition, the Sr/Ba ratios (0.16–3.50) for the clay-shales from the Southern Bida basin indicated an alternated marine and continental paleo-depositional settings and only continental setting (Sr/Ba ratios = 0.22 – 0.50) for the Northern Anambra basin. The Th/Sc, La/Sc, Th/Co and the LREE/HREE ratios showed a derivation of the shale and clay deposits from similar felsic-rich source rock while the log of (K₂O/Na₂O) vs SiO₂, revealed a Passive Margin tectonic setting for the two Basins. There is insignificant differences between the geochemical classifications, weathering, source rock/provenance and tectonic settings of clay-shale sediments of both sedimentary basins, however, there exist slight disparity in their salinity conditions and redox settings.

KEYWORDS: Geochemistry, Clay-shale, Provenance, Tectonic Setting, Northern Anambra and Southern Bida Basins

1. INTRODUCTION

Geochemical signatures are crucial for determination of prehistoric and depositional events in rocks. Previous researchers have addressed questions such as depositional environment, provenance, weathering conditions and tectonic settings of the clay-shale deposits in Southern Bida and Northern Anambra Basins of Nigeria (Okunlola and Idowu, 2012; Odoma et al., 2015; Bolarinwa et al., 2019). However, the relationship between the two basins on the basis of environment of deposition and variation in chemical constituents remain unresolved. Bhatia, 1983; Roser and Korsch, 1986, used distinct ratios like Th/La and Th/Sc to discriminate tectonic setting. TiO₂ with La, Y, Sc, Cr, Th, Zr, Hf and Nb trace elements in combination are powerful tool for provenance and tectonic setting determination due to their relatively low mobility and they are not significantly redistributed in the course of sedimentation, lithogenesis, and metamorphism

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(McLennan et al., 1983; Fatima and Khan, 2012; Zaid, 2012).

A tool that has been used to infer felsic and basic sources in clays and shales from different tectonic environments is the relative distribution of immobile elements such as La, Th Sc, Cr and Co differing in concentration (Armstrong-Altrin, 2009; Bakkiaraj et al., 2010). La and Th are felsic rock-enriched while Sc, Cr and Co are enriched in basic rocks (Wronkiewicz and Condie, 1990). Factors that control chemical composition of sedimentary rocks includes: depositional processes, source rocks, weathering, sorting, tectonic setting and paleoclimate (Bhatia, 1983; Wronkiewicz and Condie, 1990; McLennan and Taylor, 1991; McLennan et al., 1993; Armstrong-Altrin, 2009; Bakkiaraj et al., 2010).

Mid-Niger Basin also known as the Bida Basin or the Nupe Basin is a NW–SE trending intracratonic sedimentary basin extending from Kontagora in Niger State of Nigeria to areas slightly beyond Lokoja in the south (Adeleye, 1974; Figs.1 and 2). On the hand, Mid-Santonian deformation in the Benue Trough displaced the major depositional axis westward which led to the formation of the Anambra Basin. Post-deformational sedimentation in the Lower Benue Trough, led to the formation of the Anambra Basin (Figs. 1 and 2)



Fig. 1: Geologic Map of Nigeria showing the location of the studied Basins/Area (Modified after Obaje et al., 2004)



Fig. 2: Geological map showing the Southern Bida and Northern Anambra Basins With sample locations

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Sedimentation in the Anambra Basin thus commenced with the Campanian-Maastrichtian marine and paralic shales of the Enugu and Nkporo Formations, overlain by the coal measures of the Mamu Formation (Obaje, 2009). The Bida Basin is assumed to be a northwesterly extension of the Anambra Basin (Akande et al., 2005). Although, the sedimentary successions of these two basins are lateral equivalents, their geochemical variations, chemical weathering in the source area, provenance, geological and depositional histories are worth studying. They will help to further understand the geology of the two basins, and to compare the two basins based on the clay-shale geochemistry (Overare et al., 2020).

2. GEOLOGY AND STRATIGRAPHY

The origin of the Bida Basin is connected with possibly the Santonian orogenic movements of southeastern Nigeria and the Lower Benue Trough of Nigeria (Ojo and Ajakaiye, 1989). It is a NW–SE basin, which extends from Kontagora, Niger State, Nigeria to slightly beyond Lokoja in the south (Figs. 1, 2 and 3). This basin experienced a northeast and southwest separation by the basement complex but connected with Anambra and Sokoto basins containing post orogenic molasse facies and few thin unfolded marine sediments (Adeleye, 1974). The Bida basin can possibly be regarded as the northwestern extension of the Anambra Basin, with its deposition during the major third transgressive cycle of southern Nigeria at the Late Cretaceous times (Agyingi, 1991).



Fig. 3: Stratigraphic successions in the Bida Basin and correlations with adjacent Upper Cretaceous Succession in Anambra Basin (After Ojo and Akande, 2009) Note, NW = Northwest, SE = Southeast, S = South, + + = Crystalline basement rock

Stratigraphically, the Bida basin is divided into two sectors: the Northern Bida and the Southern Bida basin (Agyingi, 1991).

The Northern Bida basin is made up of Bida sandstone, Sakpe ironstone, Enagi siltstone and Batati ironstone (Fig. 3) while the Southern Bida basin consists of Lokoja sandstone, Patti and Agbaja ironstone Formation (Agyingi, 1991). On the other hand, the Anambra Basin is located in the southeastern part of Nigeria. It is bordered to the north by Bida Basin (Fig. 3), a NE-SW trending, folded, aborted rift basin that runs obliquely across Nigeria (Obaje, 2009; Fig. 3). Hence its origin was linked to the tectonic processes that accompanied the separation of the African and South American plates in the Early Cretaceous (Murat, 1972; Burke et al., 1971).

3. METHODOLOGY

Twelve samples (six clay and six shale) were collected from an exposed clay-shale section along Ahoko and Abaji along Lokoja/Abuja highway within the Southern Bida Basin and another twelve samples (six clay and six shale) at Ojodu along Ayingba/Itobe highway within the Mamu Formation in the Northern Anambra Basin. Representative clay and shale samples were collected from different vertical section of the exposures (Figs. 2 and 3) avoiding weathered horizons. Samples were pulverized to 0.07 mm size for XRD and XRF analyses Bundesanstalt für Geowissenschaften at und Rohstoffe (BGR) Hannover, Germany.

XRD pattern of the representative claystone- shale samples were determined using a PANalytical X'pert PRO MPD diffractometer equipped with a variable divergence slit (20 mm irradiated length), primary and secondary soller, scientific X' Celerator detector (Active

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length 0.59), and a sample changer (Sample diameter 28 mm). The samples were investigated from 2 to 85 2 theta with a step size of 0.0167 2theta and a measuring time of 10 s. per step. For the specimen preparation the top loading technique was used

For the XRF analysis, powdered samples were analyzed using a PANalytical Axios. Samples were prepared by mixing a flux material and melting into glass beads. The beads are analyzed by wavelength dispersion X-ray fluorescence spectrometry (WD-XRF) for the following oxides and elements determination; SiO₂, Al₂O₃, Fe₂O₃, CaO, MgO, TiO, P₂O₅, K₂O, MnO and LOI, Trace

4. RESULTS AND DISCUSSION

The major oxide concentrations, values of plagioclase index of alteration (PIA), chemical index of alteration (CIA) and chemical index of weathering (CIW) as well as calculated geochemical ratios for the studied Cretaceous sediments from Southern Bida and Northern Anambra Basins are presented in Tables 1 and 2. The obtained results were compared to average shales worldwide (Pettijohn, 1957), NASC (Gromet et al., 1984; Turekan and Wedephol, 1961) and shales from other elements such as As, Ba, Co, Cr, Cu, Ga, Hf, Nb, Ni, Pb, Sr, Th, U, Zr and rare earth elements such as Heavy rare earth elements (HREE) e.g. Y and Sc and Light rare earth elements (LREE) e.g. Ce, La, Nd and Sm. To determine loss on ignition (LOI) 1000 mg of sample material was heated to 1030°C for 10 min. after mixing the residue with 5.0 g lithium metaborate and 25 mg lithium bromide, it is fused at 1200°C for 20 min. the calibrations are validated by analysis of Reference materials. Monitor samples and 130 certified reference materials (CRM) are used for the correction procedures.

parts of Nigeria (Tables 3 and 4). The investigated sediments are characterized by high contents of SiO₂, moderate Al₂O₃, small variation in Fe₂O₃ (Tables 1 and 2) but low in TiO₂, CaO, Na₂O and K₂O. The low K₂O values indicated a lack of expandable clays in both sediments, such as montmorillonite (Akpokodje et al., 1991) while the low content of TiO₂ and CaO and MgO for all the samples is ascribed to strong weathering (Roy et al., 2008).

Table 1: Oxide composition (%) of clay and shale samples from Southern Bida basin

Elements	Abj1.1	Abj1.2	Abj1.3	Abj1.4	Abj1.5	Abj1.6	Abj1.7	Ahk1.1	Ahk1.2	Ahk1.3	Ahk1.4	Ahk1.5
SiO ₂	79.20	69.33	97.02	83.84	90.13	80.05	68.62	61.62	49.69	67.94	64.12	84.57
Al_2O_3	13.77	19.76	1.71	9.41	5.85	13.48	20.43	19.70	24.81	15.94	20.92	7.70
Fe ₂ O ₃	0.69	1.47	0.13	1.11	0.57	0.62	1.58	1.56	2.91	1.41	1.63	1.57
MgO	0.02	0.04	0.01	0.03	0.03	0.01	0.02	0.33	0.49	0.11	0.14	0.10
CaO	0.015	0.009	0.007	0.015	0.008	0.015	0.008	0.040	0.089	0.032	0.065	0.030
MnO	0.006	0.006	0.002	0.003	0.006	0.003	0.003	0.048	0.059	0.022	0.034	0.015
Na ₂ O	0.01	0.01	0.01	0.01	0.01	0.01	0.01	0.04	0.10	0.05	0.04	0.03
K ₂ O	0.062	0.163	0.007	0.078	0.046	0.068	0.143	0.912	0.905	1.30	1.289	0.828
TiO ₂	0.319	0.945	0.062	1.179	0.590	0.218	0.729	2.976	1.505	2.009	1.983	1.634
P_2O_5	0.021	0.048	0.115	0.087	0.021	0.017	0.046	0.241	0.571	0.075	0.136	0.082
LOI	5.72	7.92	0.80	3.84	2.59	5.42	8.18	12.00	18.15	10.69	9.19	3.07
Total	99.84	99.68	99.85	99.60	99.85	99.87	99.75	99.45	99.25	99.49	99.55	99.62
PIA	99.78	99.9	98.84	99.68	99.66	99.78	99.9	100	99.21	99.46	99.44	99.13
CIA	99.35	99.1	98.84	98.95	98.98	99.34	99.22	95.22	95.64	92.03	93.77	89.64
CIW	99.78	99.9	98.84	99.68	99.66	99.78	99.9	99.6	99.24	99.5	99.48	99.28
SiO ₂ /Al ₂ O ₃	5.75	3.51	56.74	8.91	15.41	5.94	3.56	3.13	2.00	4.26	3.07	10.98
K ₂ O/Na ₂ O	6.2	16.3	0.7	7.8	4.6	6.8	14.3	22.8	9.05	26.00	32.23	27.6
K_2O/Al_2O_3	0.09	0.01	0.004	0.01	0.01	0.01	0.01	0.05	0.04	0.08	0.06	0.11
$Al_2O_{3/}\ TiO_2$	43.17	20.91	27.58	7.98	9.92	61.83	28.02	21.6	16.49	7.93	10.55	4.71
TiO ₂ / Al ₂ O ₃	0.02	0.05	0.04	0.13	0.10	0.02	0.04	0.15	0.06	0.13	0.09	0.21
ICV	0.06	0.09	0.10	0.13	0.11	0.05	0.09	0.15	0.18	0.18	0.15	0.33
D*	0.95	0.93	0.93	0.89	0.91	0.96	0.93	0.92	0.89	0.92	0.93	0.83

D*=Al₂O₃/(Al₂O₃+MnO+Fe₂O₃; Machhour et al., 1994) *Abj1.1- Abj1.7= Claystone samples *Ahk1.1-Ahk1.5= Shale samples Table 2: Oxide composition (%) of clay and shale samples from Northern Anambra basin

	0:11.1	0.11.0	0:11.2	0:11.4	0:11.5	0.11	0.11.7	0:10.1	0:12.2	0:12.2	0.10.4	0:10.5
	Ojd1.1	Ojd1.2	Ojd1.3	Ojd1.4	Ojd1.5	Ojd1.6	Ojd1.7	Ojd2.1	Ojd2.2	Ojd2.3	Ojd2.4	Ojd2.5
SiO_2	87.31	84.39	76.99	75.47	80.39	85.83	84.35	82.42	77.86	81.51	72.99	76.03
Al_2O_3	8.07	10.08	14.77	15.88	12.70	9.02	10.09	9.65	11.64	10.16	8.96	12.31
Fe_2O_3	0.26	0.34	0.66	0.53	0.47	0.29	0.33	0.76	0.84	0.70	9.37	1.31
MgO	0.02	0.04	0.05	0.05	0.04	0.03	0.03	0.08	0.10	0.09	0.06	0.11
CaO	0.01	0.011	0.026	0.024	0.007	0.013	0.009	0.011	0.012	0.026	0.014	0.015
MnO	0.003	0.005	0.005	0.005	0.004	0.005	0.004	0.006	0.009	0.022	0.105	0.011
Na ₂ O	0.01	0.01	0.02	0.02	0.01	0.01	0.01	0.03	0.02	0.01	0.01	0.04
K_2O	0.094	0.136	0.243	0.247	0.166	0.124	0.138	0.632	0.523	0.289	0.316	0.644
TiO ₂	0.712	1.036	1.336	1.433	1.161	0.947	1.004	0.952	1.189	1.066	1.055	1.480
P_2O_5	0.026	0.037	0.047	0.040	0.030	0.035	0.034	0.040	0.049	0.053	0.054	0.058
LOI	3.32	3.68	5.63	6.07	4.83	3.49	3.81	5.21	7.49	5.78	6.80	7.71
Total	99.85	99.79	99.77	99.76	99.78	99.80	99.77	99.79	99.71	99.71	99.69	99.67
PIA	97.56	99.80	99.66	96.66	97.21	98.45	97.17	87.32	91.22	94.10	92.90	89.70
CIA	98.66	98.44	98.07	98.07	98.21	98.6	98.36	98.44	93.51	95.41	96.85	94.62
CIW	99.75	99.8	99.66	99.75	99.84	99.78	99.80	99.59	99.74	99.61	99.78	99.51
SiO ₂ /Al ₂ O ₃	10.82	8.37	5.21	4.75	6.33	9.52	8.36	8.54	6.69	8.02	8.15	6.18
K ₂ O/Na ₂ O	9.4	13.6	12.5	12.35	16.6	12.4	13.8	21.07	26.15	28.9	31.6	16.1
K ₂ O/ Al ₂ O ₃	0.01	0.01	0.02	0.02	0.01	0.01	0.01	0.07	0.04	0.03	0.04	0.05
$Al_2O_{3/}$ TiO ₂	11.33	9.73	11.06	11.08	10.94	9.52	10.05	10.14	9.79	9.53	8.49	8.31
TiO ₂ /Al ₂ O ₃	0.09	0.10	0.09	0.09	0.90	0.10	0.10	0.10	0.10	0.11	0.12	0.12
ICV	0.05	0.05	0.07	0.06	0.05	0.05	0.05	0.16	0.13	0.11	1.10	0.17
D*	0.07	0.07	0.07	0.07	0.07	0.07	0.07	0.02	0.02	0.02	0.40	0.00
D*	0.97	0.97	0.96	0.97	0.96	0.97	0.97	0.93	0.93	0.93	0.49	0.90

D*=Al₂O₃/(Al₂O₃+MnO+Fe₂O₃; Machhour et al., 1994)

*Ojd1.1-Ojd1.7= Claystone samples *

*Ojd2.1-Ojd2.5= Shale samples

Table 3: Average chemical composition of southern Bida and northern Anambra claystone and shale compared to shale from other sedimentary basins in Nigeria

Oxides	Claystone ¹		Shale	21	Asu River group (Amajor.1987)	Ezeaku shale (Amajor,1987)	Auchi shale	Ifon Shale Ajayi <i>et al</i> , 1989)
	SBB	NAB	SBB	NAB	()
SiO ₂	81.17	82.10	65.59	78.16	69.94	44.91	51.68	63.30
TiO ₂	0.58	1.09	2.02	1.15	0.52	0.65	1.95	1.02
Al_2O_3	12.06	11.52	17.81	10.54	10.00	15.71	18.76	18.47
Fe ₂ O ₃	0.88	0.41	1.82	2.60	4.04	6.24	4.67	1.26
MnO	0.004	0.004	0.04	0.03	0.04	0.06	0.06	0.01
MgO	0.02	0.04	0.23	0.09	0.87	2.58	4.39	0.82
CaO	0.01	0.01	0.05	0.02	3.38	15.42	1.90	0.09
Na ₂ O	0.01	0.01	0.05	0.02	0.40	0.42	0.93	0.42
K_2O	0.08	0.16	1.05	0.48	1.15	2.36	1.16	2.36
P_2O_5	0.05	0.04	0.22	0.05	0.17	0.46	0.25	0.46
LOI	4.92	4.40	10.62	6.60	9.21	11.1	14.05	11.6
Total	99.78	99.78	99.50	99.74	99.69	99.91	99.87	99.81

¹Present study, SBB= Southern Bida Basin, NAB= Northern Anambra Basin

Table 4: Comparing average chemical composition of the southern bida and northern anam	bra
claystone shale studied here to published average shales	

Oxides	Claystone ¹		Shale		Average shale (Pettijohn, 1957)	Turekan & Wedephol (1961)	PAAS	NASC (Gromet <i>et</i> <i>al.</i> , 1984)
	SBB	NAB	SBB	NAB				
SiO ₂	81.17	82.10	65.59	78.16	58.10	58.50	62.40	64.82
TiO ₂	0.58	1.09	2.02	1.15	0.60	0.77	0.99	0.80
Al_2O_3	12.06	11.52	17.81	10.54	1540	15.00	18.78	17.05
Fe_2O_3	0.88	0.41	1.82	2.60	6.90	4.72	7.18	5.70
MnO	0.004	0.004	0.04	0.03	Trace	-	-	-
MgO	0.02	0.04	0.23	0.09	2.40	2.50	2.19	2.83
CaO	0.01	0.01	0.05	0.02	3.10	3.10	1.29	3.51
Na ₂ O	0.01	0.01	0.05	0.02	1.30	1.30	1.19	1.13
K_2O	0.08	0.16	1.05	0.48	3.20	3.10	3.68	3.97
P_2O_5	0.05	0.04	0.22	0.05	0.20	0.16	0.16	0.15

¹Present study, SBB= Southern Bida Basin, NAB= Northern Anambra Basin

The interrelationships between major oxides, some trace and rare elements are given in Tables 5 and 6.

The abundance of silica and alumina are attributed to the clayey-silty nature of the samples as well as presence of biogenic SiO₂ as indicated by the SiO₂-Al₂O₃ association (Fig. 4A). The observed strong negative correlation between SiO₂ and Al₂O₃; r = -0.96(Table 5) and r = -0.58 (Table 6) for the sediments from Southern Bida Basin and Northern Anambra Basin respectively, suggest a terrigenous origin for the studied sediments (Moosavirada et al., 2011). This is also supported by the Fe₂O₃/TiO₂ VS. Al₂O₃/(Al₂O₃+Fe₂O₃+MnO) diagram (Fig. 4B) where the investigated clay-shale sediments plot near the terrigenous end-member around PAAS and UCC. On the other hand, Al_2O_3 and Fe_2O_3 show significant

positive correlation value (r = 0.79, Table 5) in the samples from the Southern Bida Basin, signifying a representative of detrital input into the the basin (Mishra et al., 2019). In contrast, the Northern Anambra Basin, showed negative correlation for Al_2O_3 and Fe_2O_3 (r = 0.24, Table 6), this could be attributed to lack of Feoxides in the Northern Anambra Basin. This is true as ironstone of Agbaja Formation is associated with the Southern Bida basin as documented by Adeleye and Dessauvagie (1972).

Table 5: Pearson's Correlation coefficients of major and trace elements for the investigated

 Cretaceous sediment of Southern Bida Basin

	Al_2O_3	Ba	CaO	Co	Fe ₂ O ₃	K2O	La	MgO	Na ₂ O	Ni	SiO_2	Sr	Th	TiO_2	V	Zr
Al_2O_3	1															
Ba	0.5664	1														
CaO	0.5643	0.9407	1													
Co	0.615	0.9663	0.9157	1												
Fe ₂ O ₃	0.7888	0.7994	0.7998	0.768	1											
K_2O	0.4848	0.5467	0.6913	0.6478	0.6288	1										
La	0.5112	0.1693	0.1607	0.1417	0.4964	0.2683	1									
MgO	0.596	0.9761	0.9016	0.985	0.8003	0.6119	0.1603	1								
Na ₂ O	0.5668	0.945	0.987	0.9284	0.8228	0.7355	0.1662	0.9156	1							
Ni	0.615	0.9663	0.9157	1	0.768	0.6478	0.1417	0.985	0.9284	1						
SiO_2	-0.9642	-0.7373	-0.7404	-0.7895	-0.8757	-0.6301	-0.4615	-0.767	-0.7494	-0.7895	1					
Sr	0.26	0.8842	0.756	0.7958	0.5294	0.2335	0.0732	0.808	0.7603	0.7958	-0.4286	1				
Th	0.4428	0.2499	0.3307	0.2967	0.5964	0.5852	0.7773	0.3092	0.3424	0.2967	-0.4936	-0.0231	1			
TiO_2	0.4859	0.506	0.5274	0.6539	0.609	0.8383	0.36	0.6436	0.5696	0.6539	-0.6221	0.1917	0.7203	1		
V	0.6991	0.695	0.7181	0.8109	0.8084	0.8435	0.3493	0.7811	0.7641	0.8109	-0.8331	0.3723	0.6444	0.9131	1	
Zr	0.0805	-0.1133	0.061	-0.0603	0.2932	0.5345	0.5266	-0.0489	0.0779	-0.0603	-0.1207	-0.3475	0.8432	0.5715	0.4078	1

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Table 6: Pearson's Correlation coefficients of major and trace elements for the investigated Cretaceous sediments of the Northern Anambra Basin

	Al ₂ O ₃	Ba	CaO	Co	Fe ₂ O ₃	K ₂ O	La	MgO	Na ₂ O	Ni	SiO ₂	Sr	Th	TiO ₂	v	Zr
Al ₂ O ₃	1							Ū								
Ba	0.2067	1														
CaO	0.5389	0.3388	1													
Co	-0.2368	0.1526	-0.0796	1												
Fe ₂ O ₃	-0.2413	0.0744	-0.1505	0.9884	1											
K ₂ O	0.1322	0.9257	0.1016	0.2104	0.1427	1										
La	0.6971	0.4343	0.4739	0.1575	0.1194	0.3264	1									
MgO	0.1811	0.9546	0.3442	0.1946	0.1251	0.8939	0.4521	1								
Na ₂ O	0.3694	0.8326	0.2315	- 0.0602	-0.1119	0.8575	0.342	0.6919	1							
Ni	0.1299	0.3896	0.2056	0.8948	0.8835	0.3813	0.4609	0.4526	0.1573	1						
SiO_2	-0.5872	-0.5007	-0.3553	- 0.5947	-0.5821	- 0.4837	-0.7135	- 0.5327	-0.4124	- 0.8611	1					
Sr	0.0397	0.9243	0.3401	0.4309	0.3556	0.851	0.4262	0.9502	0.6326	0.6151	-0.582	1				
Th	0.4121	-0.2783	0.1865	0.3306	0.3788	- 0.3779	0.5215	0.2133	-0.3022	0.427	- 0.4975	-0.139	1			
TiO ₂	0.8658	0.5318	0.5384	0.011	-0.0018	0.4114	0.765	0.5045	0.5787	0.3886	- 0.7696	0.4274	0.438	1		
v	0.7219	0.5925	0.7437	0.0246	-0.0374	0.4418	0.7855	0.6758	0.4001	0.4079	- 0.6854	0.5851	0.3094	0.8052	1	
Zr	0.0398	0.5781	0.2284	0.4926	0.4745	0.4349	0.4805	0.6987	0.1883	0.6573	- 0.5873	0.7588	0.3798	0.4576	0.5601	1



Fig. 4: (A). Al₂O₃ vs. SiO₂ diagram showing the presence of biogenic contribution in the

investigated clay-shale sediments (Barbera et al., 2006), (**B**). Bostrom (1973) diagram; analysed sediments are compared to argillite (T) and hydrothermal (H) end members whose mixing is modelled by the H-T curve. PAAS (Taylor McLennan, 1985) and UCC (Rudnick and Gao, 2003) data are reported for comparison, and (**C**). K₂O-Fe₂O₃—Al₂O₃ compositional space (Date after Condie, 1993) showing the major element distribution in the clay-shale sediments for both Southern Bida and Northern Anambra basins respectively. The detrital features and continental crust precursor of the analysed clay-shales are further supported by the high values (0.49-0.97; Tables 1 and 2) of the parameter D*=Al₂O₃/(Al₂O₃+MnO+Fe₂O₃; Machhour et al., 1994), which connects Al₂O₃, strongly assembled in the continental crust (D*=0.79; Taylor and McLennan, 1985). The ternary plot of K₂O-Fe₂O₃-Al₂O₃ (Fig. 4C) reveals that all of the investigated clay and shales plots close to Al₂O₃ region, suggesting enrichment in Al₂O₃, which also indicates that clay minerals in these clay-shale sediments largely controlled the abundance of elements (Wronkiewicz and Condie, 1987).

MgO content ranges between 0.01 to 0.14% (Table 1) for clay-shale sediments from Southern Bida Basin and 0.02 to 0.11% (Table 2) for sediments from Northern Anambra Basin, its lower than UCC value (2.22 %). The correlation between MgO and Fe₂O₃ is strong and positive (r = 0.59 and 0.80 respectively; Table 5) for the Southern Bida Basin sediments but for the Northern Anambra Basin Sediments, it was low and insignificant (r = 0.18 and 0.13, respectively; Table 6). This oxides are sourced from ferromagnesian silicates, such as, biotites and amphiboles, which though may be minor constituents of felsic igneous or metamorphic rocks within the basins (Mishra et al. 2019).

The composition of TiO_2 varied from 0.06 to 2.98 for the Southern Bida Basin sediments and 0.71% to 1.48% for the Northern Anambra Basin sediments (Tables 1 and 2). The high variation in value of TiO_2 in some of the investigated samples is an indication of the detrital nature of the sediments in the presence of titanium minerals such as rutile and ilmenite, transported and deposited mechanically (Liu et al., 2009).

The positive correlation between Al_2O_3 and TiO_2 (r = 0.49) for the Southern Bida Basin and Northern Anambra Basin (r = 0.87) suggest terrigenous origin (Saccà et al., 2011).

There is an observed general depletion in oxides of Na, K and Ca in the investigated samples (Tables 1 and 2) compared to the Upper Continental Crust values, (3.9 %, 3.4 % and 4.2 %). This could be due to their hydration energy resulting in high mobility during weathering process (Cullers, 1995). The observed positive correlation between Al₂O₃ and both of K₂O and Na₂O (r = 0.48 and 0.74; Table 5) for the Southern Bida Basin and Northern Anambra Basin (r = 0.13 and 0.86; Table 6) confirms that these oxides are associated and related with clay minerals like smectite and illite. Above assertion was supported with the strong positive correlation recorded between Na₂O and K₂O (r = 0.74) for the Southern Bida and (r = 0.86) for the Northern Anambra Basins respectively.

Cox et al., (1995) put forward values of K_2O/Al_2O_3 ratios for clay minerals and feldspars at 0.0 to 0.3 and 0.3 to 0.9. The range of K_2O/Al_2O_3 ratio for the investigated claystone varied from 0.01 to 0.09 and 0.04 to 0.10 for the claystone and shales from the Southern Bida basin (Table 1) while 0.01 to 0.02 and 0.03 to 0.07 for claystone and shale of the Northern Anambra Basin (Table 2). The K_2O/Al_2O_3 ratios were close to the lower clay mineral range limit in both the shale and clay samples from both basins.

Tables 7 and 8 give the concentration of the trace elements in the investigated Cretaceous samples from the Southern Bida and Northern Anambra Basins. It was generally observed that the sediments rich in quartz grain are characterized by lower composition of trace elements except Zr, which is attributed to dilution by quartz.

Transitional metals like Cr, Co, Ni and V in the studied samples (Tables 7 and 8) show relatively high values of concentration compared to the corresponding values for UCC (35, 10, 20, and 60 ppm, respectively). This suggests the presence of mafic components in the source area resulting in an increase of Ni, Co, Cr and V in the detritus (Liu et al., 2013). This is supported with the enrichment of the clayey constituents as Cr. Co. Ni and V elements are readily adsorbed onto clay minerals during weathering process. There was an observed strong positive correlation, (r = 0.69; 0.85; 0.62 and 0.62) of Al₂O₃ concentration with V, Cr, Ni and Co for sediments of the Southern Bida Basin respectively (Table 5). This strongly suggests their presence as adsorbed ions in clay minerals and strongly controlled by the nature of the source rocks (EL-Wekeil and Abou El-Anwar, 2013). On the other hand, within the Northern Anambra Basin, there was a strong positive correlation of Al_2O_3 with V and Cr (r = 0.72 and 0.91) but insignificantly correlated with Ni and Co (r = r = 0.13 and -0.24) (Table 6). This must be as a result of renewed tectonic activity within the Northern Anambra Basin during the mid Santonia when the Lower Benue trough was displaced from its major depositional axis westward, thereby disturbing the sedimentation dynamics, possibly resulting in sediment mixing and reworking and finally forming the Anambra Basin (Obaje, 2009). The strong positive correlations between Co and Ni with Fe_2O_3 (r = 0.77 and r = 0.77; Table 5) for the sediments from Southern Bida Basin and (r = 0.99 and r = 0.88; Table 6)for the Northern Anambra Basin sediments strongly suggest possible association of these two elements with Fe-oxides minerals.

Sr and Ba show a general depletion pattern (Tables 7 and 8) compared to UCC (350 ppm and 550 ppm, Fig. 5), this is probably due to their hydration energy resulting in their preferential loss during weathering and erosion (Cullers, 1995 and Liua et al., 2013). The observed positive correlation between contents of Fe₂O₃ and Ba for the Southern Bida Basin (r=0.799) and the Northern Anambra Basin (r = 0.074) indicates that Ba is mainly associated with Fe-oxides (EL-Wekeil and Abou El-Anwar, 2013). This is in good agreement with reports documented by Adeleye and Dessauvagie (1972) that Southern Bida Basin is a host to deposits of ironstone.



Fig. 5: UCC normalized for trace and rare earth elements of sediments from A: Southern Bida and B: Northern Anambra Basins (UCC After Taylor and Mclennan, 1981).

The observed negative correlation between Sr and Ba with Zr (r = -0.11 and- 0.35; Table 5) for Southern Bida Basin suggest an abundance of fine-grained sediments whereas in the Northern Anambra Basin, the strong positive correlation values between Sr and Ba with Zr (r = 0.76 and 0.58; Table 6) suggest an abundance of coarse grained detrital sands. This confirms that the Northern Anambra Basin is more sandy than the Southern Bida Basin, evident from the high silica composition.

The compositional value of Th (Tables 7 and 8) is close to the corresponding value in the UCC (10.7 ppm, Fig. 5). Samples with higher Th contents suggest high sand enrichment. This is supported by the positive correlation between Th and Zr (r = 0.84; Table 5) for the Southern Bida Basin and (r = 0.34; Table 6) for the Northern Anambra Basin. The positive correlation values of Fe, Ti, Mg, V, Ni, Co and Cr (r= 0.91, 0.79, 0.75, 0.93, 0.76and 0.76; Table 5) for Southern Bida Basin and the positive correlation values Ti, Mg, V, Ni and Cr (r = 0.92, 0.52, 0.87, and 0.32 respectively, Table 6) for the Northern Anambra Basin indicate that these elements may be incooperated with heavy minerals and/or trace elements during depositional episode or sedimentation within the Southern Bida and Northern Anambra Basins.

Table 7	: Major	element	concentra	tion of the	Southern	Bida cla	ystone	and shale	samples

	Abj1.1	Abj1.2	Abj1.3	Abj1.4	Abj1.5	Abj1.6	Abj1.7	Ahk1.1	Ahk1.2	Ahk1.3	Ahk1.4	Ahk1.5
Ba	36.00	56.00	119	134	36	36	70	872	1947	384	419	242
Sr	11.00	26.00	417	142	15	14.00	23	324	914	64	65	50
Ni	2.00	3.00	2.00	2.00	2.00	2.00	2.00	32.00	46	17	13	5
Co	2.00	3.00	2.00	3.00	2.00	2.00	3.00	26	32	20	6	3
Cu	8.00	10.00	10.00	10.00	10.00	9.00	10	28	26	18	26	9
Cr	24.00	78.00	4.00	53.00	22.00	19.00	85	109	120	87	88	53
Ga	18.00	25.00	6.00	15.00	11.00	15.00	25	30	41	27	36	19
Hf	10.00	24.00	6.00	36.00	6.00	6.00	14	21	7	34	39	40
Sc	8.00	15.00	8.00	10.00	8.00	8.00	10	18	17	14	27	9
Zn	3.00	5.00	2.00	9.00	6.00	2.00	6.00	108	140	90	30	16
V	20.00	51.00	5.00	37.00	25.00	17.00	53.00	109	96	96	71	58
Zr	385	1005	96.00	1475	423	192	569	697	321	1438	1367	1554
Mo	2.00	2.00	2.00	2.00	2.00	2.00	2.00	2.00	2.00	2.00	3	3
Nb	11.00	22.00	3.00	29.00	15.00	7.00	17.00	70	33	59	57	43
Pb	12.00	11.00	10.00	27.00	11.00	10.00	28	62	47	43	51	31
Th	16.00	27.00	7.00	37.00	17.00	8.00	22.00	27.00	25	30	35	26
U	3.00	8.00	3.00	8.00	3.00	3.00	4.00	14	8	11	11	8

Abj1.1- Abj1.7= Claystone samples

Ahk1.1-Ahk1.5= Shale samples

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Table 8: Major element concentration of the Northern Anambra claystone and shale samples

	Ojd1.1	Ojd1.2	Ojd1.3	Ojd1.4	Ojd1.5	Ojd1.6	Ojd1.7	Ojd2.1	Ojd2.2	Ojd2.3	Ojd2.4	Ojd2.5
Ba	47	82	95	127	68	48	61	198	180	179	115	281
Sr	19	24	30	31	20	24	23	45	50	53	48	63
Ni	2	3	6	6	4	2	2	4	6	6	14	7
Co	2	2	3	3	2	2	3	4	3	4	15	4
Cu	10	9	15	20	11	6	10	13	10	11	18	13
Cr	31	40	62	74	53	34	44	48	57	52	43	62
Ga	13	17	22	23	19	14	16	16	16	16	13	18
Hf	9	20	17	13	13	14	20	14	21	21	19	25
Sc	8	9	10	10	8	8	8	8	8	8	11	11
Zn	5	6	11	11	8	6	7	10	16	14	56	15
V	22	41	77	75	50	32	46	44	76	80	48	70
Zr	455	768	644	651	623	640	685	546	920	906	945	953
Mo	2	2	2	3	2	2	2	2	2	2	3	3
Nb	15	21	27	31	22	18	20	20	25	22	21	28
Pb	11	20	23	16	12	18	14	17	20	18	18	25
Th	12	17	18	18	14	13	17	7	15	13	19	14
U	3	3	3	3	3	3	3	3	5	3	6	3

^{*}Ojd1.1-Ojd1.7= Claystone samples

On the other hand, the observed negative correlation values recorded for Fe, Co and Cr (r = -0.10 and -0.06 respectively, Table 6) for some samples in the Northern Anambra Basin suggest probable disturbance of the sedimentation dynamics, resulting in sediment mixing, reworking and size sorting (Moosavirada et al., 2010).

4.1. Paleo-Weathering Indices and Maturity of the Clay-Shale deposit

The average values of Chemical index of alteration (CIA), Plagioclase Index of Alteration (PIA) and Chemical Index of Weathering (CIW) for the clay samples in the Southern Bida basin are 99.11%,

99.65%, 99.65%, respectively whereas the shale samples have average values of 93.26%, 99.45%, 99.42%, respectively for these indices. On the other hand, CIA, PIA and CIW indices for clay samples from the Northern Anambra averaged 98.34%, 98.07%, 99.77% whereas shale averaged 95.77%, 91.05%, 99.65%, respectively. The high PIA indicates that nearly all of the plagioclase has been transformed into clay minerals. This is consistent with the calculated CIA and CIW indices that reveal strong or prolonged weathering in the source area (Nesbitt and Young, 1982; Fedo et al., 1995; Tables 1, 2 and 9).

Table 9: Summary of source area weathering for Southern Bida and Northern Anamb	ora
claystone and shale deposits	

Indices	Standard	¹ SBB	² SSB	¹ NAB	² NAB
CIA					
Range	Intense weathering 75–100%	98.95-99.35	89.64-95.64	98.07-98.66	93.51-98.44
Mean		99.11	93.26	98.34	95.77
Median CIW					
Range	Intense weathering	99.66-99.90	99.24-99.50	99.75-99.84	99.51-99.74
	75–100				
Mean		99.65	99.42	99.77	99.65
Median					

¹SBB and ²SSB = Southern Bida Basin Claystone and shale ¹NAB and ²NAB = Northern Anambra Basin Claystone and shale

The Al_2O_3 -(CaO + Na_2O)- K_2O (A-CN-K) diagram (Fig. 6A) of Nesbith and Young (1982) and Fedo et al. (1995) permit the segregation in compositional variations as related to chemical weathering and/or source rock composition (Madhavaraju et al., 2016). it demonstrated intense weathering history. The clay-shale sediments sourced from both Southern Bida and Northern Anambra Basins plot close to the high Al₂O₃ contents (A-Apex), thus revealing a high level of weathering to a point where there is liberation of major amounts of alkali and alkali earth elements from the sediments (Overare et al., 2020). The plot of CIA vs. SiO_2 (Fig. 6B) also proved a similar trend with the investigated samples plotting typically in the upper part between illite and kaolinite, signifying a high degree of weathering.

To identify climatic conditions which prevailed in the provenance, a bivariant plot of SiO_2 against total ($Al_2O_3+K_2O+Na_2O$) proposed by Suttner and Dutta, (1986) was used. The plot revealed semi-humid and arid conditions for the clay and shales in both basins (Fig. 6C1 and 6C2). The index of compositional variation

(ICV) can be used to categorize the original nature and maturity of the sediments with the prevailed climatic

conditions (Cox

et

al., 1995).

GIBBSITE, KAOLINITE, CHLORITE в A 120 100 LLITE 100 SMECTIT **⊘₩₩∕™™** ^ 80 AUSCOVITE . Illite 80 Muscovite Average Shale PAAS CIA Graphite eldspar CIA PLAGIOCL FELDSPAR GRANODIORITE GRANITE 40 Basalt TONALITE GRANITE CHARNOKIT 20 VPE 20 0 ò 20 80 100 SiO₂(Wt%) Northern Anambra Basin □ Ojodu clay ◊ Ojodu shale Southern Bida Basin ∆Ahoko clay thern Bida Basin ∆Ahoko clay Northern Anambra Basin 🗖 Ojodu clay 🔹 Ojodu shale **⊽** Ahoko shale v Ahoko shale C2 3 C1 10 15 54 12 ŝ SG, 50. 78-75 74 72 78 (AL2O3+K2O+Na2O) (AL₂O₃+K₂O+Na₂O) Weak weathering Intense weathering D 70.75 mmature PAAS CIA= CV 0 **PAAS ICV = 0.85 ∇⊗08** 50 40 60 70 80 90 100 CIA Southern Bida Basin Northern Anambra Basin □ Ojodu clay ◊ Ojodu shale ∆Ahoko clay **∇**Ahoko shale

Fig. 6: (A). Ternary plot of Al₂O₃-(CaO + Na₂O)-K₂O for the sediments (Nesbith and Young,

1982; Fedo et al., 1995), **(B)**. Plot of CIA vs. SiO2 (Nesbitt and Young 1982), SiO₂ Vs (AL₂O₃+K₂O+Na₂O) for claystone-shale sediments of **C1:** Southern Bida and **C2:** Northern Anambra Basin showing trend of maturity with reference to climatic conditions (Suttner and Dutta, 1986) **(D).** ICV vs. CIA plot showing the maturity and intensity of chemical weathering for the Shales (After Long et al., 2012).

 $ICV = (Fe_2O_3 + K_2O + Na_2O + CaO + MgO + MnO)/Al_2O_3$

Values of ICV < 1 are characteristic of minerals such as kaolinite, illite and muscovite whereas higher values points toward plagioclase, K-feldspar, amphiboles and pyroxenes Also, the more mature the sediment, the low the ICV values. Tables 1 and 2 illustrate that the documented ICV values for the investigated Southern Bida basin and Northern Anambra basin sediments ranged from 0.06 to 0.33 and 0.05 to 1.10, respectively. Hence, these sediments are considered to be highly matured and are derivative of intense chemical weathering. This may also imply distinctive clay rich mature sediments associated with either tectonically inactive or intracratonic setting, where sediment recycling is dynamic (Cox et al., 1995) or from intensely

weathered crystalline basements, which is consistent with Fig. 6D established by Long et al. (2012) for the estimation of sediment maturity and weathering intensity. This also conforms well to the results obtained from the association between SiO₂ and (Al₂O₃ +Na₂O+ K₂O) shown in Fig. 6C1 and 6C2.

Relationship between Th/U ratio and Th concentration can be used to estimate the degree of weathering in sedimentary rocks. The observed result indicates intense weathering in the source areas or sedimentary recycling for the clay in both basins but the shale indicative of moderate to high weathering (Tables 10 and 11).

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Table 10: Calculated geochemical ratios for the investigated samples of Southern Bida Basin

-	Abil 1	Abil 2	Abil 3	Abil 4	Abi1 5	Abil 6	Abil 7	Abk1.1	Ahk1 2	Ahk1 3	Ahk14	Ahk1 5
	11051.1	11051.2	11051.5	11051.1	11051.5	11051.0	11051.7	7 11111 1	74661.2	7 нист.5	7 (11)(1)	7 нист.5
Th/Co	8.00	9.00	3.5	12.33	8.5	4.00	7.33	1.04	0.78	1.5	5.83	12
Co/Th	0.13	0.11	0.29	0.08	0.12	0.25	0.14	0.96	1.28	0.67	0.17	0.12
Co/La	0.05	0.03	0.04	0.02	0.07	0.06	0.03	0.36	0.34	0.30	0.04	0.06
Th/U	5.33	3.40	2.33	4.63	5.67	2.67	5.50	1.93	3.13	2.73	3.18	3.25
La/Sc	5.38	7.33	6.13	14.3	3.5	4.38	11.1	4.06	5.47	4.79	5.44	5.89
La/Co	21.5	63.67	24.5	47.67	14	17.5	37	2.81	2.91	3.35	24.5	17.67
Th/Cr	0.67	0.35	1.75	0.70	0.77	0.42	0.23	0.25	0.21	0.34	0.40	0.15
Th/Sc	2.00	1.80	1.75	3.7	2.13	1.00	2.20	1.50	1.47	2.14	1.30	2.89
Cu/Zn	2.67	2.00	5.00	1.11	1.67	4.50	1.67	0.26	0.19	0.20	0.87	0.56
Ni/Co	1.00	1.00	1.00	0.67	1.00	1.00	0.67	1.23	1.44	0.85	2.17	1.67
Zr/Sc	48.13	67	12.00	147.5	54	24	56.9	38.72	18.88	102.71	50.63	172.67
Cr/Th	1.5	2.89	0.57	1.43	1.29	2.38	3.86	4.04	4.80	2.90	2.51	2.04
Cr/La	0.56	0.71	0.08	0.37	0.79	0.54	0.77	1.49	1.29	1.30	0.60	1.00
U/Th	0.19	0.30	0.43	0.22	0.18	0.34	0.18	0.52	0.32	0.37	0.31	0.31
V/Cr	0.83	0.65	1.25	0.70	1.14	0.89	0.62	1.00	0.80	1.10	0.81	1.09
Ni/Co	1.00	1.00	1.00	0.67	1.00	1.00	0.67	1.23	1.44	0.85	2.17	1.67
Cr/Ni	12.00	26.00	2.00	26.5	11.00	18.50	42.50	3.41	2.61	5.12	6.77	10.60
V/Ni	10.00	17.00	2.50	18.50	12.50	8.50	26.50	3.41	2.09	5.65	5.46	11.60
Cu/Zn	2.67	2.00	5.00	1.11	1.67	4.50	1.67	0.26	0.19	0.20	0.96	0.56
Cu+Mo/Zn	3.33	2.4	6.00	1.33	6.00	5.50	6.00	0.28	0.20	0.22	0.97	0.69
Sr/Ba	0.31	0.46	3.50	1.06	0.42	0.39	0.33	0.37	0.47	0.17	0.16	0.21
D*	0.93	0.93	0.89	0.91	0.96	0.93	0.92	0.89	0.92	0.93	0.83	0.95

D*=Al₂O₃/(Al₂O₃+MnO+Fe₂O₃; Machhour et al., 1994) Abj1.1- Abj1.7= Claystone samples Ahk1.1-Ahk1.5= Shale samples

4.2. Paleo-Environmental Conditions and

Provenance of the Clay-Shale deposit Sr/Ba ratio have proven to be a very important tool employed by previous studies to estimate paleo-salinity (Chen et al., 2016). Sr/Ba ratios >1, 0.5–1, and < 0.5 propose seawater, brackish water, and freshwater conditions, correspondingly (Li et al., 2018). The Sr/Ba ratios (0.16–3.50; Table 10) for the investigated clayshales from the Southern Bida basin reveal an array of depositional paleo-environment that alternated between marine and continental settings. On the other hand, the recorded Sr/Ba ratios (0.22 - 0.50; Table 11) revealed a continental setting for the Northern Anambra basin. In addition, K₂O/Al₂O₃ and MgO/Al₂O₃ diagram of Roaldset (1978) was used to differentiate between marine and non-marine sediments.

Table 11: Calculated geochemical ratios for the investigated samples of Northern Anambra

	Dasiii											
	Ojd1.1	Ojd1.2	Ojd1.3	Ojd1.4	Ojd1.5	Ojd1.6	Ojd1.7	Ojd2.1	Ojd2.2	Ojd2.3	Ojd2.4	Ojd2.5
Th/U	4	5.67	6	6	4.67	4.33	5.67	2.33	3	4.33	3.17	4.67
Th/Co	6	8.5	6	6	7	6.5	5.67	1.75	5	3.25	1.27	3.5
La/Sc	2.75	5	5.1	7.1	3.5	4	4.5	4.5	6.88	6	4.12	4.36
La/Co	11.00	22.50	17.00	23.67	14.00	16.00	12.00	9.00	18.33	11.50	3.07	12.00
Co/Th	0.17	0.12	0.17	0.17	0.14	0.15	0.18	0.57	0.20	0.31	0.79	0.29
Co/La	0.09	0.04	0.06	0.04	0.07	0.06	0.08	0.11	0.05	0.09	0.33	0.08
Th/Cr	0.39	0.43	0.29	0.24	0.26	0.38	0.39	0.15	0.26	0.25	1.46	0.23
Th/Sc	1.50	1.89	1.80	1.80	1.75	1.63	2.13	0.88	1.88	1.63	1.73	1.27
Cu/Zn	1.25	1.50	1.36	1.82	1.38	1.00	1.43	1.30	0.63	0.79	0.32	0.89
Ni/Co	1.00	1.50	2.00	2.00	2.00	1.00	0.67	1.00	2.00	1.50	0.93	1.75
Zr/Sc	56.88	85.33	64.40	65.10	77.88	80.00	85.63	68.25	115	113.25	85.91	86.64
Cr/Th	2.58	2.35	3.44	4.11	3.79	2.62	2.59	6.86	3.80	4.00	2.26	4.43
Cr/La	1.41	0.89	1.22	1.04	1.89	1.06	1.22	1.33	1.04	1.13	0.93	1.29
U/Th	0.25	0.18	0.17	0.17	0.21	0.23	0.18	0.43	0.33	0.23	0.32	0.21
V/Cr	0.71	1.03	1.24	1.01	0.94	0.94	1.05	0.92	1.33	1.54	1.12	1.13
Ni/Co	1.00	1.50	2.00	2.00	2.00	1.00	0.67	1.00	2.00	1.50	0.93	1.75
Cr/Ni	15.50	13.33	10.33	12.33	13.25	17.00	22.00	12.00	9.50	8.67	3.07	8.86
V/Ni	11.00	13.67	12.83	12.50	12.50	16.00	23.00	11.00	12.67	13.33	3.43	10.00
Cu/Zn	2.00	1.50	1.36	1.82	1.38	1.00	1.43	1.30	0.63	0.79	0.32	0.19
Cu+Mo/Zn	2.40	1.83	1.55	2.09	1.63	1.33	1.71	1.50	0.75	0.93	0.38	0.23
Sr/Ba	0.40	0.29	0.32	0.24	0.29	0.50	0.38	0.23	0.28	0.30	0.42	0.22
D*	0.97	0.97	0.96	0.97	0.96	0.97	0.97	0.93	0.93	0.93	0.49	0.90

 $D^*=AI_2O_3/(AI_2O_3+MnO+Fe_2O_3; Machhour et al., 1994)$

*Ojd1.1-Ojd1.7= Claystone samples

*Ojd2.1-Ojd2.5= Shale samples

The Southern Bida and Northern Anambra basin's clay shales sediments plot in the non-marine to slightly near marine environment (Fig. 7A), suggesting a transitional/mixed environment of deposition, consistent with bivariant plot of V vs. Al_2O_3 (Fig. 7B; Mortazavi et al., 2014). This is supported by the chemical classification on the basis of (Al_2O_3)-(K_2O+Na_2O+CaO)- (Fe_2O_3+MgO) contents (AKF) proposed by Englung and Jorgensen (1973) were the clay – shale sediments under investigation plots in the continental zone and in an argillaceous area (Fig. 7C1-C2).

Trace element geochemical ratios like Ni/Co and V/Cr have proved to be reliable tools for redox determination in depositional environments (Jones and Manning, 1994;

Nagarajan et al., 2007; Madhavaraju et al., 2016). Jones and Manning (1994) proposed that V/Cr ratios < 2 infer oxic conditions, 2–4.25 dysoxic conditions, and > 4.25 suboxic to anoxic conditions. They also found that < 5 Ni/Co ratios assumed oxic conditions, 5-7 means dysoxic conditions while > 7 imply suboxic to anoxic conditions. The Ni/Co and V/Cr of the investigated clay – shale sediments (Tables 10 and 11; Fig. 7D) suggest oxic environments.



Fig. 7: Paleo-environmental reconstruction and redox-sensitivity of the investigated clay-shale

sediments from Southern Bida and Northern Anambra basins on **A).** Log K₂O/Al₂O₃ vs Log MgO/Al₂O₃ bivariate plot (After Roaldset, 1978), **B).** V vs. Al₂O₃ bivariate plot (After Mortazavi et al., 2014), **C1-C2).** Al₂O₃ - (K₂O + Na₂O + CaO) - (Fe₂O₃+MgO) [AKF] Ternary plots and D) Cross plot of V/Cr vs. Ni/Co (After Jones and Manning 1994).

In addition to the paleo-environmental conditions, other trace elemental ratios like U/Th and Cu/Zn were also considered. According to Jones and Manning, (1994), ratios of U/Th < 0.75 infers oxic conditions, while U/Th ratios > 1.25 indicate anoxic conditions. Moreso, Hallberg (1976) proposed that low ratios of Cu/Zn in a basin reveals oxidizing depositional conditions, whereas high Cu/Zn ratios suggests reducing depositional conditions. The primarily high ratios of U/Th (1.93-5.67; Table 9) and Cu/Zn (0.19-5.00; Table 9) for the Southern Bida basin affirms a more reducing or an anoxic condition of deposition for the studied clay-shale sediments while the predominantly low ratios of U/Th (0.17 - 0.25; Table 10) and Cu/Zn (0.19 - 2.00; Table 10) for the Northern Anambra reveals an oxic condition of deposition for sediments studied.

Authors like Taylor and Mclennan, 1985; Condie et al, 1992; Cullers, 1995; Armstrong-Altrin et al., 2004; 2013; Madhavaraju et al., 2016; Chen et al., 2016 reports that geochemical signatures of clastic sediments can be used to ascertain provenance characteristics. Al₂O₃/TiO₂ ratios for most clastic sediments can be used to infer source rock composition. According to Hayashi et al.,

1997, Al₂O₃/TiO₂ ratios increase from 3 to 8 for mafic igneous rocks, 8 to 21 for intermediate rocks and 21 to 70 for felsic igneous rocks. The Al₂O₃/TiO₂ ranged from 9.92 to 61.83 in the clays and 4.71 to 21.70 in the shale samples of the Southern Bida Basin; hence, this study's Al₂O₃/TiO₂ ratio suggest intermediate and felsic rocks for the clay samples while mafic and intermediate rocks were suggested as being source rocks for the shale For the clays and shale samples of the samples. Northern Anambra Basin, Al₂O₃/TiO₂ ratio ranged from 9.52 to 11.33 and 8.49 to 10.14, respectively, suggesting intermediate rocks as being probable source rocks for the clay and shale (Tables 1 and 2). To also discriminate felsic, intermediate, and mafic provenance, bivariate plot of Al₂O₃/TiO₂ vs. SiO₂ (Le Bas et al., 1986) was employed. The investigated clay - shale sediments for both basins plot in both intermediate and felsic fields (Fig. 8A), indicating contributions from both sources.

Trace elements such as Y, Th, Nb, Zr, Hf, and Sc with REE in clastic sedimentary rocks are useful tool for their source rock interpreting. These elements have very low mobility during sedimentary processes, and are probably quantitatively transferred to the clastic sediments during 110

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the weathering, transportation, and diagenetic processes, thus reflecting the signatures of the precursors (e.g., Bhatia and Crook, 1986; McLennan, 1989; Armstrong Altrin et al., 2004). Different authors have also employed geochemical variations between elements such as La and Th (felsic source rocks) and Sc and Cr (mafic source rocks) to distinguish between felsic and mafic origin (Ramachandran et al. 2016). The plot of Th/Co vs. La/Sc (Fig. 8B) proved that the clay shale sediments for both basins were derivative of felsic origin. This was supported with Hf vs. La/Th diagram (Fig. 8C; Floyd and Leveridge, 1987), TiO₂ (wt. %) Vs. Ni (ppm) diagram (Fig. 7D; Floyd et al., 1989), V-Ni-Th*10 ternary relationship (Fig. 9A; Bracciali et al., 2007) and the Th/Sc vs. Zr/Sc relationship (Fig. 9B; McLennan et al. 1993; Willan, 2003). The plot of Th/Sc vs. Zr/Sc indicated an igneous differentiation trend, close to PAAS composition for the clay – shale sediments under investigation, therefore, no significant signs of heavy mineral concentration due to zircon enrichment. Thus, the recycling of older mature sediments of felsic and crustal origin probably did not play a significant role (Mikes, 2006). In addition, Co/Th vs. La/Sc diagram (Fig. 9C) demonstrated a main distribution between felsic and andesitic sources, while the La-Th-Sc relationship (Fig. 9D; Cullers, 1994a, b) revealed contributions from felsic sources, intermediate sources or possible mixing between felsic and basic source rocks.



Fig. 8: A) Scatter diagram of Al₂O₃/TiO₂ vs. SiO₂ for the investigated clay and shale samples

(Le Bas et al., 1986; Zaid and Al Gahtani, 2015), B) Bivariate plot of La/Sc vs. Th/Co (after Cullers, 2002) C) La/Th ratio vs. Hf plot (Fields after Floyd and Leveridge, 1987 and Gu et al., 2002) D) TiO2 vs. Ni plot. Fields and trends after Floyd et al. (1989).



Fig. 9: (A) V-Ni-Th*10 plot of the shales (Bracciali et al., 2007), (B) Th/Sc vs. Zr/Sc plot

showing a magmatic arc trend (McLennan et al. 1993; Willan 2003), (C) Co/Th vs. La/Sc bivariate plot (McLennan et al., 1993) and (D) La-Th-Sc Ternary plot for the shales after Cullers (1994a) compared with Post-Archean Average Shale (data is from Taylor and McLennan, 1985).

La/Sc, Th/Co and Th/Sc ratios are significantly different in felsic and basic rocks which can be used to infer provenance composition (Wronkiewicz and Condie, 1990; Cox et al., 1995; Cullers, 1995). The Th/Co, Th/Sc and La/Sc ratios for shale and clay samples from this study were compared to those of felsic and basic rockderived sediment (fine fraction) upper continental crust (UCC) and PAAS values (Tables 10, 11 and 12). These comparisons also indicated that such ratios came within the range of felsic source rocks.

Table 12: Range of claystone and shale element ratios in this study compared to ratios for similar
fractions derived from felsic rocks, mafic rocks, upper continental crust and post-archean Australia
ahala

shale

Elemental ratio	Claystone ¹		Shale ¹		Range of	sediments ²	Upper Continental Crust ³	Post- archean average shale ³
	SBB	NAB	SBB	NAB	Felsic rocks	Mafic rocks		
Th/Sc	1.00-3.70	1.50-2.13	1.30-3.70	0.88-1.88	0.84- 20.50	0.05- 0.22	0.79	0.90
Th/Co	3.50- 12.33	5.67-8.50	0.78- 12.00	1.27-5.00	0.67- 19.40	0.04- 1.00	0.63	0.63
La/Sc	3.50- 11.10	2.75-7.10	4.06-5.89	4.12-6.88	2.50- 16.30	0.43- 0.86	2.21	2.40

4.3. Tectonic Setting and Depositional Environment of the Clay-Shale deposit

In this research, major element-based discrimination diagrams of Roser and Korsch (1986) were employed. The binary diagram of $log(K_2O/Na_2O) - SiO_2$ discrimination diagram (Figs. 10A and 10B; Roser and Korsch, 1986) reveals that the clay – shale sediments for both basins plot entirely in the field of passive margin.

Passive-margin type sediments are generally enriched in SiO₂ and depleted in Na₂O, CaO and TiO₂, thereby revealing their highly recycled and matured nature (Bhatia, 1983). Major element analysis of the studied shale and clay samples confirm this, as all these samples were enriched in SiO₂ but depleted in Na₂O, CaO and TiO₂.



Fig. 10: Tectonic Discriminant Diagram for sediments of A: Southern Bida Basin and B: Northern Anambra Basin (Roser and Korsch, 1986).



Fig. 11: Al₂O₃ - (K₂O+ Na₂O + CaO) - (Fe₂O₃+MgO) [AKF; after Englund and Jorgensen, 1973).] Ternary plots for A: Southern Bida and B: Northern Anambra Basin

The compositional values for clay and shale samples' rare earth element (REE) are given in Tables 13 and 14. A slight variation in sum of REE content was observed between the claystone and shale samples; (114-353) and (267-459) for Southern Bida, (138-255) and (142-234) for the Northern Anambra Basins, respectively,

these values were compared with notable world averages (Table 15). The enrichment of LREE and moderately negative Eu anomaly reflect their relative abundance in the crust, while the depletion of the HREE is due to their ability to form soluble complexes in seawater. Table 13: Rare earth element distribution of shales and claystone samples from Southern Bida Basin

	Abi1 1	Abi1 2	Abi1 3	Abi1 4	Abi1 5	Abi1 6	Abi1 7	Ahk1 1	Ahk1 2	Ahk1.3	Ahk1 4	Ahk1 5
Light Ra	are Earth	Elements	(LREE)	7.00]1.1	7.0]1.0	7.0011.0	7.0]111	7 4 11 (11 1	/ unit 1.2	7	7	7 411(110
Ce	78	191	70	221	34	54	196	174	205	154	252	101
La	43	110	49	143	28	35	111	73	93	67	147	53
Nd	26	84	25	82	12	15	83	66	75	36	62	26
Sm	15	19	13	17	13	13	13	13	13	13	13	13
Heavy F	Heavy Rare Earth Elements (HREE)											
Sc	8	15	8	10	8	8	10	18	17	14	27	9
Y	24	59	15	62	19	17	55	73	56	82	70	65
ΣREE	185	478	180	535	114	142	468	435	459	366	571	267
LREE/H	LREE/HREE Ratios											
Ce/Sc	9.75	12.73	8.75	22.10	4.25	6.75	19.60	9.67	12.06	11.00	9.33	11.22
Ce/Y	1.79	3.24	4.67	3.54	1.79	3.18	3.56	2.38	3.66	1.88	3.60	1.55
La/Y	1.79	1.84	3.27	2.31	1.47	2.06	2.02	1.00	1.66	0.82	2.10	0.82

 Table 14: Rare earth element distribution of shales and claystone samples from Northern

 Anambra Basin

Ojd1.1 Ojd1.2 Ojd1.3 Ojd1.4 Ojd1.5 Ojd1.6 Ojd1.7 Ojd2.1 Ojd2.2 Ojd2.3 Ojd2.4 Oj	d2.5								
Light Rare Earth Elements (LREE)									
Ce 64 59 114 107 49 86 83 49 90 84 74 85									
La 22 45 51 71 28 32 36 36 55 46 46 48									
Nd 12 23 31 28 12 33 13 12 27 31 35 31									
Sm 16 13 13 22 13 13 13 13 15 13 15 13	1								
Heavy Rare Earth Elements (HREE)									
Sc 8 9 10 10 8 8 8 8 8 8 11 11									
Y 20 29 36 36 28 24 28 24 32 36 39 46	i								
ΣREE 142 178 255 267 138 196 181 142 227 218 220 23	4								
LREE/HREE Ratios									
Ce/Sc 8.00 6.56 11.40 10.70 6.13 10.75 10.38 6.13 11.25 10.50 6.72 7.	72								
Ce/Y 3.20 2.03 3.17 2.97 1.75 3.58 2.96 2.04 2.81 2.33 1.90 1.5	35								
La/Y 1.10 2.03 1.42 1.97 1.00 1.33 1.29 1.50 1.72 1.28 1.18 1.0	04								

 Table 15: Average rare earth elements of claystone and shales from Southern Bida and Northern

 Anambra Basins

Element	Claystone ¹	Shale ¹			PAAS	Codo shale (McLennan et al., 1980)	Average shale (Levinson, 1974)
	SBB	NAB	SBB	NAB		,	
La	74.14	80.29	86.60	76.40	38.20	29.70	121.00
Ce	120.57	40.71	177.20	46.20	79.60	63.40	50.00
Pr	-	-	-	-	8.83	-	-
Nd	46.71	21.74	53.00	27.20	33.90	27.90	24.00
Sm	14.71	14.71	13.00	13.80	5.55	-	-
Eu	-	-	-	-	1.08	-	-
Gd	-	-	-	-	4.66	-	-
Tb	-	-	-	-	0.74	-	-
Dy	-	-	-	-	4.68	-	-
Ho	-	-	-	-	0.99	-	-
Er	-	-	-	-	2.85	-	-
Tm	-	-	-	-	0.41	-	-
Lu	-	-	-	-	0.43	-	-
Y	35.86	28.71	69.20	35.40	-	-	-
Sc	9.57	8.74	17.00	9.20	-	-	-

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5. CONCLUSION

The major and trace element analyses of sediments from Northern Anambra and Southern Bida Basins were carried out. The result shows that SiO₂ range from 49.69 - 97.02 wt% for Southern Bida Basin and 72.99 -87.31 wt% SiO₂ with moderate Al₂O₃, low Fe₂O₃ and K₂O values for Northern Anambra Basin, which are consistent with a quartz-rich clayey mineral assemblage. The clay-shale sediments for both basins were derived from intense chemical weathering as shown by the high CIA (>90), PIA (>90) and CIW (>90) values. This is supported with the ICV values of 0.06 - 0.33 for Southern Bida Basin and 0.05 - 1.10 for the Northern Anambra basin that indicates a highly matured and characteristic intensive chemically weathered products. Th/Sc, La/Sc, Th/Co, TiO₂ vs AI_2O_3 and REEs data indicate a dominantly felsic (granodioritic) provenance whereas the Cu/Zn, Cu+Mo/Zn and U/Th evaluations reveal an oxic depositional environment for the clay shale deposits. The log of (K₂O/Na₂O) vs SiO₂ show Passive Margin tectonic setting, in addition, Al₂O₃ - $(K_2O+ Na_2O + CaO) - (Fe_2O_3+MgO)$ [AKF] diagram indicated continental depositional environment for both basins, respectively. In summary, this study show existence of insignificant differences between the geochemical classifications, weathering, source rock/provenance and tectonic settings of clay-shale sediments of both sedimentary basins, however, there exist slight disparity in their salinity conditions and redox environment.

REFERENCES

- Akande, S. O., Ojo, O. J. and Ladipo, K. O., 2005. Upper Cretaceous Sequences in the Southern Bida Basin, Nigeria: A Field Guidebook. NAPE Geological Field Guide, No. 1. Ibadan, Mosuro Publishers. p. 60.
- Adeleye, D. R., Dessauvagie, T. F. J., 1972. Stratigraphy of the Niger embayment near Bida, Nigeria in Africa Geology. (T. F. J. Dessauvagie and A. J. Whiteman, Eds). University of Ibadan Press, page 181 – 186.
- Adeleye, D. R., 1974. Sedimentology of the fluvial Bida Sandstone (Cretaceous), Nigeria. Sedimentary Geology, 12: 1–24.
- Agyingi, C. M., 1991. Geology of Upper Cretaceous rocks in the eastern Bida, Nigeria. Unpublished Ph.D. Thesis. Department of Geology, University of Ibadan, 501pp.
- Akpokodje, E. G., Etu-Efeotor, J. O and Olorunfemi, B. N., 1991 The composition and physical properties of some ceramic and pottery clays of South Eastern Nigeria. Journal of Mining and Geology, 27, 9–15.
- Armstrong-Altrin, J. S., Lee, Y. I., Verma, S. P., Ramasamy. S., 2004. Geochemistry of sandstones from the upper Miocene Kudankulam Formation, southern India: Implications for provenance, weathering, and

tectonic setting. Journal of Sedimentary Research, 74, 285–297.

- Armstrong-Altrin, J. S., 2009. Provenance of sands from Cazones, Acapulco, and Bahia Kino beaches, Mexico. Rev. Mex. Cienc. Geol. 26 (3) 764–782.
- Armstrong-Altrin, J. S. Nagarajan, R., Madhavaraju, J., Rosalez-Hoz, L., Lee, Y., Balaram, V.,
- Cruz-Martinez, A., Avila-Ramirez, G., 2013. Geochemistry of the Jurassic and Upper Cretaceous shales from the Molango Region, Hidalgo, eastern Mexico: Implications for source-area weathering, provenance, and tectonic setting. C. R. Geoscience, 345, 185 – 202.
- Bakkiaraj, D., Nagendra, R., Nagarajan, R. and Armstrong-Altrin, J. S., 2010. Geochemistry of sandstones from the Upper Cretaceous Sillakkudi Formation, Cauvery Basin, southern India: implication for provenance. J. Geol. Soc. India 76, 453–467.
- Barbera, G., Mazzoleni, P., Critelli, S., Pappalardo, A., Lo Giudice, A., Cirrincione, R., 2006. Provenance of shales and sedimentary history of the Monte Soro Unit, Sicily. Periodico di Mineralogia, 75(2-3), 313-330.
- Bhatia, M. R. and Taylor, S. R., 1981. Trace-element geochemistry and sedimentary provinces: a study from the Tasman geosyncline, Australia. Chemical Geology, 33, 115–125.
- Bhatia, M. R. and Crook, K. A. W., 1986. Trace Element Characteristics of Graywackes and Tectonic Setting Discrimination of Sedimentary Basins. Contrib. mineral. Petrol. 92, 181 – 193.
- Bhatia, M. R., 1983. Plate Tectonics and Geochemical Composition of Sandstones. Journal of Geology, 91, pp. 611–627.
- Bolarinwa, A. T., Idakwo. S. O. and Bish, D. L. 2019. Rare-Earth, Trace Element, and Hydrogen and Oxygen Isotopic Composition of Cretaceous Kaolinitic Sediments from the Lower Benue Trough, Nigeria: Provenance, Depositional and Paleoclimatic Significance. Acta Geochim. https://doi.org/10.1007/s11631-019-00328-y.
- Bostrom, K., 1973. The origin and fate of ferromanganoan active ridge sediments. Stockholm contributions in geology, 27, 149-243.
- Bracciali, L., Marroni, M., Pandolfi, L. and Rocchi, S., 2007. Geochemistry and petrography of Western Tethys Cretaceous sedimentary covers (Corsica and Northern Apennines): From source areas to configuration of margins. Geological Society of America Special Paper, 420, 73-93.

ATABO NATHANIEL ODOMA AND SUNDAY OJOCHOGWU IDAKWO

- Burke, K. C., Dessauvagie, T. F. J., and Whiteman, A. J., 1971. The opening of the Gulf of Guinea and the Geological History of the Benue Depression and Niger Delta. Natural Physical Science, 233 (38), 51-55.
- Chen, B., Liu, G., Wu, D., and Sun, R., 2016. Comparative study on geochemical characterization of the Carboniferous aluminous argillites from the Huainan Coal Basin, China. Turkish Journal of Earth Sciences, 25, 274-287.
- Condie, K. C., Boryta, M. D., Liu, J. and Quian, X., 1992. The origin of khondalites: geochemical evidence from the Archean to Early Proterozoic granulitic belt in the North China Craton. Precambrian Research, 59, 207–223.
- Condie, K. C., 1993. Chemical Composition and Evolution of the Upper Continental Crust: Contrasting Results from Surface Samples and Shales. Chemical Geology, 104, 1-37.
- Cullers, R. L., 1994a. The chemical signature of source rocks in size fractions of Holocene stream sediment derived from metamorphic rocks in the wet mountain region, Colorado, USA. Chemical Geology, 113, 327–343.
- Cullers, R. L., 1995. The controls on the major and trace element evolution of shales, siltstones and sandstones of Ordovician to Tertiary age in the Wet Mountain region, Colorado, U.S.A. Chemical Geology, 123, 107–131.
- Cullers, R. L., 2002. Implications of elemental concentrations for provenance, redox conditions, and metamorphic studies of shales and limestones near Pueblo, CO, USA. Chemical Geology, 191(4), 305-327.
- Cox, R., Lowe, D. R. and Cullers, R. L., 1995. The influence 385 of sediment recycling and basement composition on evolution of mudrock chemistry in the southwestern United. States. Geochimica et Cosmochimica Acta, 59, 2919– 2940.
- EL-Wekeil, S. S. and Abou El-Anwar, E. A., 2013. Petrology, geochemistry and sedimentation history of Lower Carboniferous shales in Gebel Abu Durba, southwestern Sinai, Egypt. Journal of Applied Sciences Research, 9 (8): 4781-4798.
- Englund, J.O. and Jorgensen, P., 1973. A chemical classification system for argillaceous sediments and factors affecting their composition. Geol. Stockholm Forh. 95, 87-97.

- Fatima, S. and Khan, M. S., 2012. Petrographic and geochemical characteristics of Mesoproterozoic Kumbalgarh clastic rocks, NW Indian shield: implications for provenance, tectonic setting, and crustal evolution. Int. Geol. Rev. 54 (10) 1113–1144.
- Fedo, C. M., Nesbitt, H. W. and Young, G. M., 1995. Unravelling the effects of potassium metasomatism in sedimentary rocks and paleosols, with implications for paleo-weathering conditions and provenance. Geology, 23, 921– 924.
- Floyd, P. A. and Leveridge, B. E., 1987. Tectonic environment of the Devonian Gramscatho basin, Cornwall: south framework mode and evidence geochemical from turbiditic sandstones. Journal of the Geological Society London, 144, 531-542.
- Floyd, P. A., Winchester, J. A. and Park, R. G., 1989. Geochemistry and tectonic setting of Lewisian clastic metasediments from the early Proterozoic Loch Maree Group of Gairloch, NW Scotland. Precambrian Research, 45: 203–214.
- Gromet, L. P., Dymek, R. F., Haskin, L. A., and korotev, R. L., 1984. The North American shale composite. Its compilation, major and trace element characteristics. Geochimica Cosmochimica Acta 48, 2469–2482.
- Gu, X. X., Liu, J. M., Zheng, M. H., Tang, J. X. and Qt, L., 2002. Provenance and Tectonic setting of the Proterozoic turbidites in Hunan, South China: Geochemical Evidence. Journal of Sedimentary Research, 72, 393–407.
- Hayashi, K., Fujisawa, H., Holland, H., and Ohmoto, H., 1997. Geochemistry of ~1.9 Ga sedimentary rocks from northeastern Labrador, Canada. Geochimica et Cosmochimica Acta, 61, 4115– 4137.
- Hallberg, R. O., 1976. A geochemical method for investigation of paleo-redox conditions in sediments. Ambio Special Report, 4, 139-147.
- Isayeva, B.A., (1971) Relation between titanium and iron in the sediments of the Ocean. Geokhimiya, 3, 310–317.
- Jones, B. and Manning, D. A. C., 1994. Comparison of geochemical indices used for the interpretation of paleo-redox conditions in ancient mudstones. Chemical Geology, 654 111, 111-129.
- Le Bas M. J., Le Maitre R. W., Streckeisen, A. and Zanettin, B., 1986. A chemical classification of volcanic rocks based on the total alkali-silica diagram. Journal of Petrology, 27, 662 745–750.

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- Levinson, A. A., 1974. Introduction to Exploration Geochemistry, Applied Publishing, Illnois.
- Li, D., Li, R., Zhu, Z. and Xu, F., 2018. Elemental characteristics of lacustrine oil shale and its controlling factors of paleo-sedimentary environment on oil yield: a case from Chang 7 oil layer of Triassic Yanchang Formation in southern Ordos Basin. Acta Geochimica, 37(2), 228–243.
- Liu, B., Y. Wangb, X. Sua and H. Zheng., 2013. Elemental geochemistry of northern slope sediments from the South China Sea: Implications for provenance and source area weathering since Early Miocene. Chemie der Erde, 73, 61–74.
- Long, X., Yuan, C., Sun, M., Xiao, W., Wang, Y., Cai, K. and Jiang, Y. 2012. Geochemistry and Nd isotopic composition of the Early Palaeozoic flysch sequence in the Chinese Altai, Central Asia: Evidence for a northward-derived mafic source and insight into Nd model ages in accretionary orogen. Gondwana Research, 22, 554-566.
- Machhour L., Philip J. and Oudin J. L., 1994: Formation of Laminite deposits in anaerobic dysaerobic marine environments. Marine Geology, 117, 287-302.
- Madhavaraju, J., Ramirez-Montoya, E., Monreal, R., Gonzalez–Leon, C. M., Pi-Puig, T., Espinoza-Maldonado, I. G. and Grijalva–Noriega, F. J., 2016. Paleoclimate, paleo-weathering and paleo-redox conditions of Lower Cretaceous shales from the Mural Limestone, Tuape section, northern Sonora, Mexico: Constraints from clay mineralogy and geochemistry. Revista Mexicana de Ciencias Geologicas, 33(1), 34-691 48.
- McLennan, S. M., 1989. Rare earth elements in sedimentary rocks: influence of provenance and sedimentary processes. Rev. Mineral. 21, 170–199.
- Mclennan, S. M. and S. R. Taylor, 1991. Sedimentary rocks and crustal evolution: Tectonic setting and secular trends. Journal of Geology, 99, 1–21.
- Mclennan, S. M., Hemming, S., McDaniel, D. K., and Hanson, G. N., 1993. Geochemical approaches to sedimentation, provenance, and tectonics. Processes controlling the composition of clastic sediments (Johnson, M.J. and Basu, A., eds.), 21–40, Geological Society of America special paper 284.
- Mikes, T., Dunkl, I., Eynatten, H. and Frisch, W., 2006. Geochemistry of Eocene flysch sandstones in the NW External Dinarides. Acta Geologica Hungarica, 49(2), 103–124.

- Mishraa, P. K., AnkitbY., Gautamc, P. K., Lakshmidevi C. G., Singh, P. and Anoop, A., 2019. Inverse relationship between south-west and north-east monsoon during the late Holocene: Geochemical and sedimentological record from Ennamangalam Lake, southern India. Catena, 182, 104 - 117.
- Moosavirada, S. M., M. R. Janardhanab, M. S. Sethumadhava, M. R. Moghadamc and M. Shankaraa 2010. Geochemistry of lower Jurassic shales of the Shemshak Formation, Kerman Province, Central Iran: Provenance, source weathering and tectonic setting, Chemie der Erde.
- Moosavirada, S. M., Janardhanab, M. R., Sethumadhava, M. S., Moghadamc, M. R. and Shankara, M., 2011. Geochemistry of lower Jurassic shales of the Shemshak Formation, Kerman Province, Central Iran: Provenance, source weathering and tetonic setting. Chem. Erde 71: 279 – 288.
- Mortazavi, M., Moussavi-Harami, R., Mahboubi, A. and Nadjafi, M., 2014. Geochemistry of the Late Jurassic–Early Cretaceous shales (Shurijeh Formation) in the intracontinental Kopet-Dagh Basin, northeastern Iran: implication for provenance, source weathering and paleoenvironments. Arabian Journal of Geosciences, 7, 5353–5366.
- Murat, R. C., 1972. Stratigraphy and paleogeography of the Cretaceous and Lower Tertiary in Southern Nigeria. In Dessauvagie, T. T. J. and Whiteman, A. J. (eds.) African Geology. University of Ibadan Press, Ibadan, Nigeria, 251-266.
- Nagarajan, R., Madhavaraju, J., Nagendra, R., Armstrong-Altrin, J. S. and Moutte, J., 2007. Geochemistry of Neoproterozoic shales of the Rabanpalli Formation, Bhima Basin, Northern Karnataka, southern India: implications for provenance and paleoredox conditions. Revista Mexicana de Ciencias Geologicas, 24, 150-160.
- Nesbitt, H. W. and Young, G. M., 1982. Early Proterozoic Climates and Plate Motions Inferred from Major Element Chemistry of Lutites. Nature, 299, 715-717.
- Obaje, N. G., 2009. Geology and mineral resources of Nigeria. Springer-Verlag Berlin Heidelberg. 221p.
- Obaje, N. G., Wehner, H. and Scheeder, G., 2004. Hydrocarbon prospectivity of Nigeria's inland basins from the viewpoint of organic geochemistry and organic petrology. American Association of Petroleum Geologists Bulletin, 88, 325–353.

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- Odoma, A. N., Obaje, N. G., Omada, J. I., Idakwo, S. O. and Erbacher, J., 2015. Mineralogical, chemical composition and distribution of rare earth elements in clay-rich sediments from Southeasthern Nigeria. Journal of African Earth Sciences, **102**, 50-60.
- Ojo, O. J and Akande, S. O., 2009. Sedimentology and depositional environments of the Maastrichtian Patti Formation, southeastern Bida Basin, Nigeria. Cretaceous Research, 30, 1415–1425.
- Ojo, S. B. and Ajakaiye, D. E., 1989. Preliminary interpretation of gravity measurements in the Mid-Niger Basin area, Nigeria. In Kogbe, C.A. (Ed.), Geology of Nigeria, 2nd edn, Lagos, Elizabethan Publishers. pp. 347–358.
- Okunlola, O. A. and Idowu O., 2012. The geochemistry of claystone-shale deposits from the Maastritchian Patti formation, Southern Bida basin, Nigeria. Earth Sciences Research Journal 16.2: 57-67.
- Overare, B., Osokpor, J., Ekeh, P.C., Azmy, K., 2020. Demystifying provenance signatures and paleodepositional environment of mudrocks in parts of south-eastern Nigeria: Constraints from geochemistry. Journal of African Earth Sciences, doi: <u>https://doi.org/10.1016/</u> j.jafrearsci.2020.103954.
- Pettijohn, F. J., 1957 Sedimentary Rocks, 2nd edition. Harper and Row, New York. 628p.
- Ramachandran, A., Madhavaraju, J., Ramasamy, S., IL Lee, Y., Rao, S., Chawngthu, D. L. and Velmurugan, K. 2016. Geochemistry of Proterozoic clastic rocks of the Kerur Formation of Kaladgi-Badami Basin, North Karnataka, South India: implications for provenance. Turkish Journal of Earth Sciences, 25, 126-144.
- Roaldset, E., 1978. Mineralogical and chemical changes during weathering, transportation, and sedimentation in different environments with particular references to the distribution of Yttrium and lanthanide elements, Ph.D. Thesis, Geol. Inst., Univ. of Oslo, Norway.
- Roser, B. P., and Korsch, R. J., 1986. Determination of tectonic setting of Sandstone and Mudstone suites using SiO₂ content and K₂O/Na₂O ratio. Journal of Geology, 94, 635–650.
- Roy, P.D., Caballero, M., Lozano, R. and Smykatz-Kloss, W., 2008. Geochemistry of late quaternary sediments from Tecocomulco lake, central Mexico: implication to chemical weathering and provenance. Chem. Erde Geochem. 68: 383–393.

- Rudnick, R. L. and Gao, S., 2003. The Composition of the Continental Crust, In Treatise on Geochemistry, Elsevier – Pergamon, Oxford– London, 3:1-64.
- Saccà, C. D., Saccà, P., Nucera and Fazio, A. D., 2011. Composition and geochemistry of clay sediments offshore the northeastern Sicilian coast (Southeastern Tyrrhenian Sea, Italy). Estuarine, Coastal and Shelf Science, 92, 564– 572.
- Suttner, L. J. and Dutta, P. K., 1986. Alluvial sandstone composition and paleoclimate, Framework mineralogy: Journal of Sedimentary Petrology, 56, 329–345.
- Taylor S. R, McLennan S. M. 1981. The composition and evolution of the continental crust: rare earth element evidence from sedimentary rocks. Philosophical Transactions of the Royal Society london A 301:381–399.
- Taylor, S. R. and S. M. McLennan, 1985. The Continental Crust: Its Composition and Vietnam in the western South China Sea: Implications for source analysis and East Asian monsoon evolution. Science in China Series D: Earth Sciences, 50, 1674–1684.
- Turekian, K. K. and K.H. Wedepohl 1961. Distribution of elements in some major units of Earth, Crust. Geol, Soc. Amer. Bull., 77(2): 175-195.
- Willan, R. C. R., 2003. Provenance of Triassic– Cretaceous sandstones in the Antarctic Peninsula; implications for terrane models during Gondwana breakup. Journal of Sedimentary Research, 73, 1062–1077.
- Wronkiewicz D. J. and Condie K. C., 1987. Geochemistry of Archean shales from the Witwatersrand Super group, South Africa: Source area weathering and provenance. Geochimica et Cosmochimica Acta, 51, 2401-2416.
- Wronkiewicz, D. J. and Condie, K. C., 1990. Geochemistry and mineralogy of sediments from the Ventersdorp and Transvaal Super groups, South Africa: cratonic evolution during the early Proterozoic. Geochimica et Cosmochimica Acta, 54, 343–354.
- Zaid, S. M., 2012. Provenance, diagenesis, tectonic setting and geochemistry of Rudies sandstone (Lower Miocene), Warda Field, Gulf of Suez, Egypt. J. Afr. Earth Sci. 66–7, 56–71.
- Zaid, S.M. and Al Gahtani, F., 2015. Provenance, diagenesis, tectonic setting, and geochemistry of Hawkesbury Sandstone (Middle Triassic), southern Sydney Basin, Australia. Turkish Journal of Earth Sciences, 24, 72-98.

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