

Corrosion behavior of austempered ductile iron (ADI) in iron ore slurry

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Abstract

Corrosion behavior of austempered ductile iron (ADI) in iron ore slurry was studied as a function of the microstructure developed by austempering at 380 and 300°C for different exposure time in the slurry. The corrosion rates of the ADI balls immersed in the iron ore slurry was determined using weight loss method. It was found that the pH of the iron ore slurry increased with time and the corrosion behavior is also influenced by the pH of the slurry and the austempering treatment. The ADI austempered at higher temperature showed better corrosion resistance than the ones austempered at lower temperature. The complex nature of corrosion of ADI is influenced by contents of retained austenite, ferrite and the carbon content in a collective way.

Keywords: Austempered ductile iron, iron ore slurry, weight-loss method

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1. Introduction

Austempered ductile iron (ADI) is well known for its superior wear resistance, fatigue properties and fracture toughness. Because of its wear resistance properties, the ADI has replaced many forged and cast steel components and conventional ductile iron castings. Iron ore grinding is the last and most critical stage in the process of comminution as it governs the economy and efficiency of ore processing since it attributes to 40-50% of the total cost of ore processing (Rao and Natarajan, 1991; Jain, 2001). The particle size is reduced by a combination of impact and abrasion using hard steel balls. The major problem in ore grinding is the wear of the grinding media balls and hence it becomes important to select a material which offers best wear resistance (Siddan, 2014). One of the main problems with ball mill is corrosion of the media balls (Eric et al., 2010; Afolabi, 2011). It was reported that (Prasanna *et al.*, 2009, Afolabi 2011; Arunima Banerjee *et al.*, 2013) austempering affects microstructure and corrosion properties of ductile iron. It was also reported that the percentage of retained austenite in ADI influences its corrosion behavior, higher the percentage of retained austenite lower the corrosion rate (Cheng-Hsun et al., 2010). In general single phases such as ferrite, austenite and martensite perform better in corrosive atmospheres than two phase mixtures such as pearlite, bainite and tempered martensite (Muthukumaraswamy et al., 1990). The present study is an attempt to study the corrosion resistance of ADI grinding balls with different austempering treatments in iron ore slurry using weight loss method.

2. Experimental

The chemical composition of the iron ore used for the study is given in the Table 1. A ball mill setup was used to prepare the iron ore slurry using the feed materials, iron ore and water. The slurry was prepared by wet grinding method in a sealed rubber lined ball mill operated 65 rpm for an hour. A set of 200 balls of forged EN31 steel and austempered ductile iron were used as the grinding media. The ground iron ore slurry was then used as a medium to study the corrosion behavior of the ADI balls. The pH of the ground iron ore slurry was measured as a function of time using model EQ614 Equiptronics Digital pH Meter.

Table 1 Chemical composition of iron ore

Element	Fe ₂ O ₃	SiO ₂	Alumina	Phosphorus	Sulphur
Composition	94.87	1.7	1.8	0.07	0.0007

The chemical composition of the ductile iron balls (dia. 25mm) is given in the Table 2. Ductile iron balls were coated with the Cu by dipping the balls into CuCl₂ solution to prevent it from surface oxidation during the heat treatment. Ductile iron balls were austenitised at 900 C for 60 minutes using a muffle silicon carbide electrical resistance furnace and quickly transferred to a molten salt bath consisting of a mixture of sodium nitrate and potassium nitrate for bainitic transformation. Then the samples were austempered at 380 and 300 C for time durations of 30, 60 and 90 minutes. The austempered samples were air cooled to room temperature.

Table 2 Chemical composition (wt%) of grinding balls

Material	C	Si	Mn	Cr	Mo	S	P	Mg
Ductile Iron	3.6	2.8	0.4	-----	0.3	0.01	0.01	0.04

The corrosion behaviors of the ADI samples were determined by the conventional weight loss method using grounded iron ore slurry as the corrosion medium. For weight loss method analysis, the ADI ball samples were immersed in the grounded iron ore slurry for different time periods. The corrosion rate, *CR*, is calculated using the equation,

$$CR = k \frac{W}{DxAxT} \quad (1)$$

where, *W* is the weight loss in the sample, *D* is the density of the sample, *A* is the area of the sample, *T* is the exposure time and *k* is a constant which depends on the units of the parameters used. Values of *k* for different units used in the formula are given in Table 3.

Table 3 *k* values in equation (1) for different units used

Corrosion rate unit	Units used	<i>k</i> Value
mpy	W (mg), D (g/cc), A (cm ²), T (h)	3450
mpy	W (mg), D (g/cc), A (in ²), T (h)	535
mpy	W (g), D (g/cc), A (in ²), T (days)	22,300
mpy	W(g), D(g/cc), A(cm ²), T (days)	14370
mm/year	W(g), D(g/cc), A(in ²), T (days)	566
mm/year	W(g), D(g/cc), A(cm ²), T (days)	3650
mm/year	W (mg), D (g/cc), A (in ²), T (h)	1.36
mm/year	W (mg), D (g/cc), A (cm ²), T (h)	87.6

mpy = milli-inches per year or mils per year

Metallographic examinations were carried out on the polished specimen after etching with 3% nital solution, under the scanning electron microscope (JEOL). Quantitative information on the volume fraction of retained austenite, ferrite and the carbon content of austenite was obtained through X-ray diffraction on JEOL JDX 8P diffractometer using CuK radiation [Cullity 1978].

3. Results and Discussion

In the case of ductile iron, the high carbon content causes the presence of spheroidal graphite nodules embedded in the iron matrix. Ductile iron balls were first austenitised to dissolve carbon, then quenched rapidly to the austempering temperature to avoid the formation of deleterious pearlite or martensite. When the iron balls are held at the austempering temperature nucleation and growth of acicular ferrite occurs, accompanied by rejection of carbon into the austenite. The resulting microstructure, known as "Ausferrite", consists of a mix of acicular ferrite and high carbon Austenite.

It was reported (Rao and Putatunda, 2005; Keough and Hayrynen, 2010; Guesser et al., 2012; Janji et al., 2016) that high temperature ADIs (where the temperature of isothermal transformation in the range 350 °C - 420 °C) produce a typically upper ausferritic microstructure, which consists of broad blades of isolated ferrite . The microstructure consists of spheroidal graphite and the matrix of ausferrite needles. Typical microstructure of the high temperature and low temperature ADIs are shown Fig 1a and Figure 1b, respectively. At 380°C upper bainite with a feathery morphology was stable. At this temperature, while the amount of retained austenite decreased with increase in austempering time, ferrite content increased with time. The microstructure of low temperature ADIs (where the temperature of the isothermal transformation in the range 250 °C - 330 °C) showed a change in bainitic structure. At 300°C lower bainite with acicular morphology was observed. At this temperature also the variation of amount of retained austenite and ferrite are similar to that of the high temperature ADIs with the difference that the amount retained austenite and ferrite being less compared with high temperature ADIs.

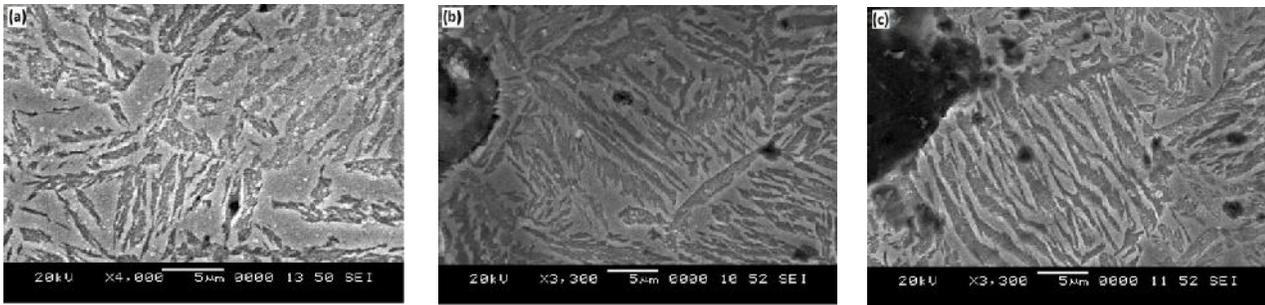


Figure 1a. Microstructure of ADI, austempered at 380 C. (a) 30 mins (b) 60 mins (c) 90 mins.

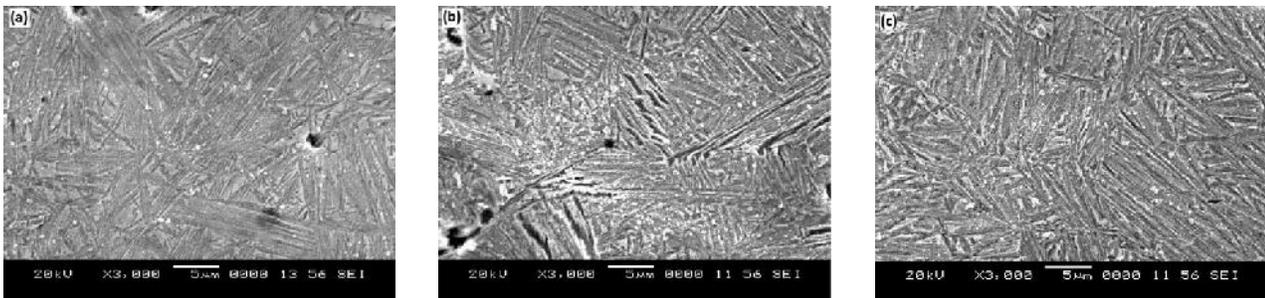


Figure 1b. Microstructure of ADI, austempered at 300 C. (a) 30 mins (b) 60 mins (c) 90 mins.

The relative amounts of austenite, ferrite and the carbon content of austenite are shown in Figure 2a and Figure 2b for different austempering treatment. The carbon content in the retained austenite decreases with austempering time. High temperature and low temperature ADIs showed a decrease in carbon content with ageing time. At 90 minutes ageing the carbon content in the retained austenite show a larger decrease compared to that of high temperature ADI.

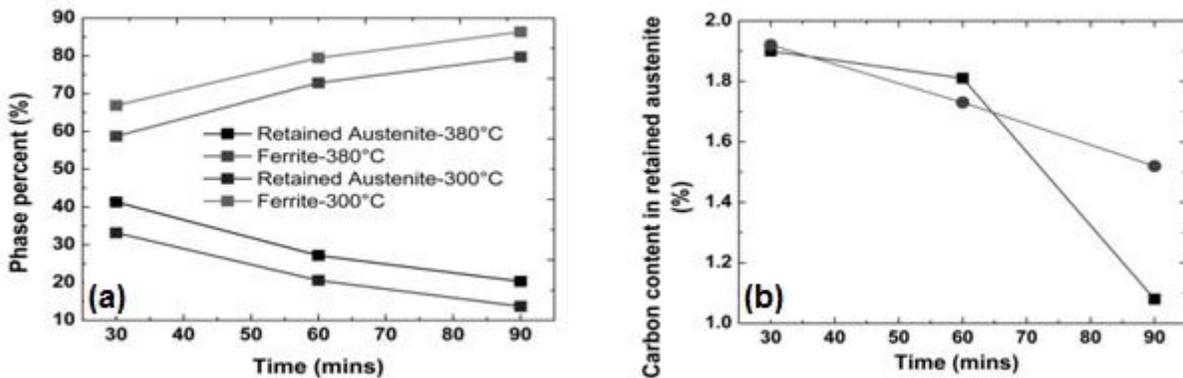


Figure 2 Volume percentage of (a) retained austenite and ferrite (b) carbon content in retained austenite by XRD analysis

Figure 3a shows the cumulative loss in weight (a measure of the degree of corrosion) occurred in different ADI sample for different time periods in the iron ore slurry. High temperature ADIs showed lower weight loss. It was also noted that the pH of the ground iron ore slurry increased with time in a more or less linear fashion as shown in Figure 3b.

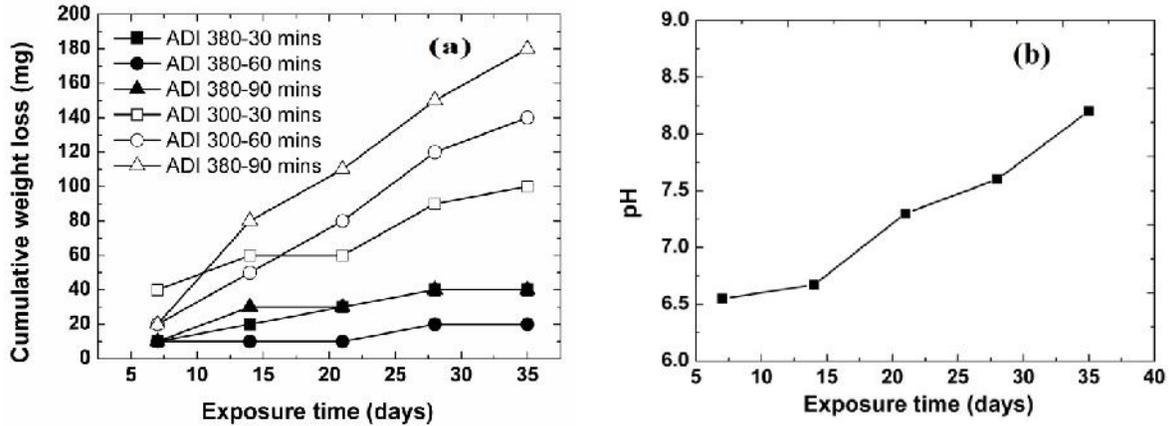


Figure 3(a) Cumulative weight losses versus exposure time for different samples, (b) Variation in pH of the iron ore slurry with time

The corrosion rates calculated from the weight loss at different pH levels of the iron ore slurry is presented in the Table 4. It was inferred from the results that the corrosion rates of the balls were strongly influenced by the pH of the iron ore slurry. The corrosion rate as a function of pH of the iron ore slurry for high temperature ADIs and low temperature ADIs are shown in Figure 4.

Corrosive wear during ore grinding occurs when a sliding takes place in a corrosive environment. There is a possibility that additional constituents could be introduced into the slurry from the grinding media balls and the chemical nature of the slurry keeps changing continuously in the mill and the slurry becomes corrosively active (Hebbar, 2011). In the absence of sliding (idle periods), the products of the corrosion will form a film on the surface, which will slow down or even arrest the corrosion. It is seen from Figure 4 that high temperature ADIs offered better corrosion resistance compared to low temperature ADIs.

Table 4 Corrosion rates from the weight loss data

Exposure days	pH of slurry	Austempered at 380 C			Austempered at 300 C		
		30 min	60 min	90 min	30 min	60 min	90 min
7	6.55	5.41	9.98	4.15	15.9	7.25	5.45
14	6.67	5.41	4.99	6.23	11.93	9.06	10.91
21	7.3	5.3	3.33	4.15	7.95	9.66	10
28	7.6	5.2	4.99	4.15	8.94	10.87	10.23
35	8.2	5.3	1.99	3.32	7.95	10.15	9.82

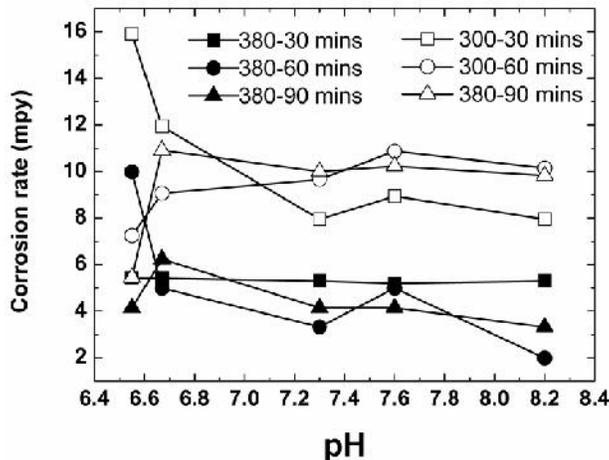


Figure 4 Variations in corrosion rate of (a) high temperature ADIs and (b) low temperature ADIs as a function of pH of the iron ore slurry

Except for large variation of corrosion rate initially at low pH value, corrosion rate is more or less independent of the pH of the slurry and the microstructure developed at the two austempering temperatures. In general single phases such as ferrite, austenite and martensite perform better in corrosive environment than the two phase mixture such as pearlite, bainite and tempered martensite. The two phases in ADI (ferrite and retained austenite) could act as electrochemical cell and aid corrosion (Ailor, 1971). Since low temperature ADIs have a fine structure, the two phases are so closely spaced that there are a number of such cells in a small area. Thus, the surface area available for corrosion is large in low temperature ADIs than the high temperature ADIs.

The corrosion of iron or carbon steels has a very complex dependence on the pH of the medium and it has been reported (Moiseeva and Rasheveskaya, 2002) that the corrosion rate increases with increase in pH. The corrosion resistance of high temperature ADIs could be due to the formation of passive layer. The high silicon content in ADI materials is responsible for the development of a passive layer (casing) after certain time period of exposure to corrosive medium and which brought down the corrosion rate of the samples (Davies, 2001). It is also reported that higher austempering temperature improves corrosion resistance by making it nobler and decreasing corrosion rate (Hemanth, 2000).

4. Conclusions

The corrosion tests were carried out for the ADI samples with the iron ore slurry as the corrosion medium. The following conclusions were drawn from the present study.

1. Isothermal transformation of austenite at high austempering temperature lead to upper bainite and retained austenite while low austempering temperature consisted of lower bainite and retained austenite.
2. The content of retained austenite decreased with increase in austempering temperature and time while the ferrite content increased with increasing austempering temperature and time
3. The corrosion behaviors of ADI were strongly dependant on the pH of the slurry which keeps changing continuously as the mill is running.
4. The ADI austempered at higher austempering temperature showed better corrosion resistance than the ones austempered at lower temperature.
5. The complex nature of corrosion of ADI is influenced by contents of retained austenite, ferrite and the carbon content in a collective way.

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