



The Use of Kitchen Effluent as Alternative Nutrient Source for Bioremediation of Oil Based Drilling Muds

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ABSTRACT: The recovery of oil based drilling muds was monitored for a period of 12 weeks upon inoculation with kitchen effluent. Oil based drilling muds inoculated with varying volumes (200ml, 250ml and 300ml) of kitchen effluent constituted the experimental set-ups, while the control set-ups were made up of oil based drilling muds inoculated with varying volumes (200ml, 250ml and 300ml respectively) of sterile distilled water. In the experimental set-ups which received additional hydrocarbon utilizers ($4.15 \times 10^4 \pm 0.12$ cfu/ml) as well as phosphate (3.63 mg/l), sulphate (3.0 mg/l) and nitrate (15.60 mg/l) present in the kitchen effluent, both dissolved oxygen (DO) and residual oil concentration (ROC) decrease rapidly while the increasing trends in biochemical oxygen demand (BOD) were much more pronounced over the period when compared with the control set-ups. However, analysis of data obtained from this study revealed that the differences between the mean residual oil concentration of the experimental set-ups and that of the control set-ups for A and B are insignificant while that of C differed significantly at 5% probability level indicating that the magnitude of loss in residual oil concentration increases as the volume of kitchen effluent inoculated increases. Therefore, it may be necessary to use kitchen effluent as alternative inoculum in bioremediation of oil contaminated drilling muds. @ JASEM

Decades of offshore and inland petroleum drilling and production operations have provide the much needed energy and chemical feedstock for the nation, but they also have made an impact on the environment in the area (Fang, 1990; Nwachukwu, 2000a; Nweke and Okpokwasili, 2003; Okolo *et al.*, 2005). Various operations are needed to support the production of oil and gas. In these operations, a broad range of chemicals are used, and some of them have important environmental implications (Petrazzuolo *et al.*, 1995; Fang, 1990).

An important aspect of the environmental impact of offshore drilling operations is the discharge of oil based drilling muds. Drilling muds are chemical mixtures used for rotary drilling in the extraction of oil and gas from the earth's crust (Okpokwasili and Nnubia, 1995). Essentially, a drilling mud is a suspension of solids (e.g. clays, barite, small cuttings, etc.) in liquids (i.e. water or oil) or in liquid emulsions, with chemical additives as required to modify its properties (Ifeadi *et al.*, 1985). The chemical additives include bactericides, surface active agents, dispersants and viscosifiers (Okpokwasili and Nnubia, 1995). The general functions of drilling mud in the borehole include; lifting of formation cuttings to the surface, control of subsurface pressures, lubrication of drill stings and sides of the hole, bottom hole cleaning and cooling, provision of aid to formation evaluation, and maintenance of stability of uncased sections of the borehole (Ifeadi *et al.*, 1985; Okpokwasili and Nnubia, 1995).

Diesel oils used in oil based drilling mud formulations were found to be toxic to a wide range of microorganisms due to high concentration of potentially toxic materials (20-30% 2-, 3- and 4-ring aromatics) (Nweke and Okpokwasili, 2003). Consequently, highly refined low toxicity base oil that has low concentration of aromatic compounds (1-10%) becomes the alternative (Candler *et al.*, 1993). Nevertheless, the used drilling mud, additives and oil-laden drill cuttings still pose a serious threat to the biota of aquatic and terrestrial ecosystems (Nweke and Okpokwasili, 2003). The discharge of oil based drilling muds as well as the release of crude petroleum and petroleum products into human environments have resulted in many problems that are of much concern globally (Atlas, 1981; Nwachukwu, 2000b; Nweke and Okpokwasil, 2003). The long persistence of these pollutants in environments was attributed to low numbers of hydrocarbon utilizers and to the toxicity of crude oil on natural flora (Atlas, 1991; Nwachukwu, 2000b).

Investigations of biodegradation of cutting piles and the microbial population associated with drilling muds have been conducted (Nnubia and Okpokwasili, 1993; Benka-Coker and Olumagin, 1995; Nweke and Okpokwasili, 2003). Both bacteria and fungi have also been developed and used to enhance rehabilitation of environments polluted with crude petroleum and its products (Nwachukwu *et al.*, 1999; Eriksson *et al.*, 2000; Nwachukwu, 2000a; Saylor and Ripp, 2000, Mishra *et al.* 2001). Poultry manure has over time been used to improve soil fertility (Okolo *et al.*, 2005). Its efficacy in promoting

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plant growth in crude oil polluted Nigerian soils, and the impact of its addition alone and in combination with surfactants or alternate carbon substrate to enhance crude oil degradation has also been reported (Amadi and Uebari, 1992; Ogboghodo *et al.*, 2004; Okolo *et al.*, 2005). Similarly, domestic effluent has over time been used to improve soil fertility (Bond, 1998; Barton *et al.*, 2005). In this study, the impacts of addition of kitchen effluent to enhance rehabilitation of oil based drilling muds were investigated.

MATERIALS AND METHODS

Study Materials: The oil based drilling muds treated in this study was obtained from an oil field in Nigeria, while the kitchen effluent used for the treatment was obtained from restaurants drainage channel at the University of Lagos, Nigeria. The kitchen effluent sample collected was analysed microbiologically for the presence of hydrocarbon utilizers (HCU), and the total heterotrophic microorganisms (THM) present in the kitchen effluent was estimated. The sulphate, nitrate and phosphate contents as well as the biochemical oxygen demand (BOD) and dissolved oxygen (DO) of the kitchen effluent collected were determined.

Bioremediation Protocols: The oil based drilling muds was mixed thoroughly to homogenize it. Thereafter, 2.0kg each of the homogenized drilling muds was weighed into six pans (internal dimensions, 40cm x 40cm x 7.5cm) and were labelled "experimental set-up A (ESUA), control set-up A (CSUA), experimental set-up B (ESUB), control set-up B (CSUB), experimental set-up C (ESUC) and control set-up C (CSUC)". The experimental set-ups (ESUA, ESUB, and ESUC) were inoculated with 200ml, 250ml and 300ml kitchen effluent respectively (containing approximately $8.14 \times 10^8 \pm 0.06$ cfu/ml total heterotrophic microorganisms (THM); $4.15 \times 10^4 \pm 0.12$ cfu/ml total hydrocarbon utilizers (THCU); nitrate, 15.60mg/l; phosphate, 3.63mg/l; sulphate, 3.0mg/l; pH, 7.6, dissolved oxygen (DO) 2.10 ± 0.02 mg/l and biochemical oxygen demand (BOD), 240.01 ± 2.01 mg/l), while the control set-ups (CSUA, CSUB and CSUC) were inoculated with 200ml, 250ml and 300ml sterile distilled water respectively for comparison of results. The progress of bioremediation of the experimental set-ups as well as that of the control set-ups was monitored by determining the mean changes in the total counts of heterotrophic microorganisms, hydrocarbon utilizers, and in the concentrations of residual oil, dissolved oxygen (DO) and biochemical oxygen demand

(BOD), at intervals of 2 weeks over a period of 12 weeks (Nwachukwu, 2000a).

Sample Collection: Before the set up of the experiments and controls, 3 random samples each of kitchen effluent (20ml each) and drilling muds (10g each) were collected. The samples from each sampling point were combined and mixed thoroughly before analysis. Immediately after the set up of the experiments and controls, 3 random samples (10g each) were collected from both the experiments and the controls. The samples from each point were also combined and mixed thoroughly before analysis. Other samples analyzed were collected from each experiment and control at 2 weeks intervals for a period of 12 weeks.

Chemical Analysis: The pH of the liquid sample, (kitchen effluent) was determined using portable pH meter with combine glass and calomel electrodes. The pH meter was standardized with buffer solutions pH 4.0, 9.0 and 7.0, after which the samples were tested in turn. The pH of the drill cuttings samples was determined in the same manner but using a suspension of a 1:5 (Drilling muds – distilled water) ratio (APHA, 1995; Nwachukwu and Ugoji, 1995; Nwachukwu, 2000b). Phosphate, sulphate and nitrate concentrations were determined spectrophotometrically using the method of APHA (1995). The residual oil was extracted from the samples using n-hexane : dichloromethane solvent system (1:1) and quantified by gravimetric method (Yveline *et al.*, 1997). To achieve this, 5 g of homogenized drilling mud sample was weighed into a 75 ml beaker and 50 ml n-hexane : dichloromethane was added to extract the residual oil in the sample. After shaking vigorously, the mixture was allowed to stand for about 5 minutes and then filtered through whatman No 1 filter paper into a 75 ml beaker of known weight (W_1) as residual oil extract (ROE). The residual oil extract was placed in an oven at 80°C for 5-10 minutes to evaporate the solvent system (Nwachukwu, 2000b). The combined weight of the residual oil and the beaker was taken and recorded as W_2 . The residual oil content (ROC) was then obtained by difference in mass ($W_2 - W_1 = ROC$) (Yveline *et al.*, 1997; Nwachukwu, 2000b; Okolo *et al.*, 2005). The dissolved oxygen (DO) was determined using Orion portable DO meter after dilution with oxygenated water. For biochemical oxygen demand (BOD) determination, the initial DO of the oxygenated samples was determined and repeated after 5 days incubation period using Winkler method (Odiere, 1999; Singh *et al.*, 1999; Nwachukwu, 2000a). Then, BOD was calculated as outlined by APHA (1980). The difference between

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dissolved oxygen at day zero and that at the fifth day multiply by a dilution factor gives the BOD₅²⁰.

Microbiological Analysis: Heterotrophic microbial counts were determined by plating serially diluted samples on nutrient agar (for bacteria) and on potato dextrose agar (for fungi). Oil-degrading microorganisms were enumerated on minimal salt agar using crude petroleum as carbon and energy source as previously reported (Nwachukwu, 2000a).

Statistical Analysis: The data obtained for the samples from the experiments and controls were fitted to t-test statistic for significance evaluation (Esan and Okafor, 1995; Gupta, 1981).

RESULTS AND DISCUSSION

The initial drop in total heterotrophic microorganisms (THM) of the experimental set-ups (Tables 1, 3 and 5) confirms the toxic impacts of the hydrocarbon products present in the drilling muds which is in accordance with the previous reports by Atlas (1981), Nwachukwu and Ugoji (1995), and Nwachukwu (2000b), and probably prove that some of the microorganisms introduced into the experimental set-ups from the kitchen effluent can not survive in an oil polluted environment (Nwachukwu, 2000b), compared to the control set-ups (Tables 1, 3 and 5) which do not show any initial decrease in THM. Microorganisms require inorganic nutrient sources (mineral elements) for growth (Andrew and Jackson, 1996), however, If any of the required inorganic nutrients is lacking or becomes limiting particularly those needed in relatively large amounts known as macro mineral elements such as P, S, C, N₂, the oil pollutant degraders will not grow even though other factors such as carbon and energy required for growth are available (Nwachukwu *et al.*, 2000).

Therefore, the introductions of kitchen effluent which contain higher content of P, S, and N₂ into the experimental set-ups would have probably increased the quantity of these elements available for the growth of the oil pollutant degraders. Table 1, 3 and 5 also show a more rapid increase in the mean population densities of hydrocarbon utilizers present in the experimental set-ups compared to the control set-ups and this could also be attributed to the addition of kitchen effluent to the experimental set-ups. For example, the percentage total hydrocarbon utilizers (% THCU) calculated for both experimental set-ups and control set-ups relative to the total heterotrophic microorganisms, THM (Tables 1, 3 and 5) differed significantly at 5% probability level ($t = 3.833; 3.844$ and 3.840 for A, B and C respectively; t from statistical table = 2.179).

The results summarized in tables 2, 4 and 6 revealed that the magnitude of loss in residual oil concentration (ROC), dissolved oxygen level (DO), and the increasing trends in biochemical oxygen demand (BOD) were much more pronounced in the experimental set-ups compared with the control set-ups. However, the values obtained from t-test analysis of residual oil content calculated for the experimental and control set-ups at 5% level of significance ($t = 1.646; 1.990$; and 2.219 for A, B and C respectively, t from statistical table = 2.179) shown that the differences between the mean residual oil concentration (ROC) of the experimental set-ups and that of the control set-ups for A and B are insignificant while that of experimental and control set-ups for C differed significantly at 5% probability level, indicating that the magnitude of loss in residual oil concentrations increases as the volume of kitchen effluent introduced into the experimental set-ups increases.

Table 1: Mean population densities of different microbial groups enumerated in experiment A and Control A

Sampling Time (Weeks)	Mean Population Densities (cfu/g±SD) x 10 ⁴			
	Experimental set-up A		Control set-up A	
	Total HM	Total HCU;%	Total HM	Total HCU; %
0 (a)	271.7±0.2	1.2±0.2;0.4	20.3±0.2	0.1±0.0;0.6
(b)	20.4±0.3	0.1±0.0;0.9	20.4±0.3	0.1±0.0;0.7
2	25.4±0.3	2.1±0.1;8.4	21.2±0.5	0.2±0.0;0.7
4	33.0±0.4	3.2±0.2;9.8	23.1±0.4	0.2±0.0;0.9
6	39.5±0.2	4.8±0.3;12.1	25.0±0.5	0.4±0.0;1.4
8	45.1±0.5	7.0±0.2;15.5	26.3±0.3	0.7±0.2;2.8
10	52.1±0.4	9.3±0.5;17.9	29.8±0.5	1.1±0.2;3.8
12	61.0±0.5	12.1±0.2;19.9	35.8±0.4	1.7±0.2;4.8

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Table 2: Mean change in chemical parameters in experiment A and control A

Sampling Time (Weeks)	Mean Changes In Chemicals Parameters \pm SD					
	Experiment A			Control A		
	Dissolved oxygen (mg/kg)	Biochemical Oxygen demand (mg/kg)	Residual oil content (g/kg)	Dissolved oxygen (mg/kg)	Biochemical Oxygen demand (mg/kg)	Residual oil content (g/kg)
O (a)	4.2 \pm 0.1	10.0 \pm 2.1	48.7 \pm 1.1	4.2 \pm 0.0	10.0 \pm 1.1	48.7 \pm 1.2
(b)	4.2 \pm 0.0	10.0 \pm 1.0	49.1 \pm 1.3	4.2 \pm 0.0	10.0 \pm 1.0	49.1 \pm 1.3
2	3.1 \pm 0.1	32.1 \pm 3.0	45.8 \pm 2.1	3.7 \pm 0.0	18.0 \pm 1.8	46.9 \pm 2.6
4	3.0 \pm 0.0	45.0 \pm 2.6	42.0 \pm 1.1	3.7 \pm 0.1	15.2 \pm 3.3	45.0 \pm 2.0
6	2.5 \pm 0.0	46.1 \pm 2.6	36.8 \pm 0.9	3.6 \pm 0.1	14.6 \pm 4.1	43.0 \pm 2.0
8	2.3 \pm 0.0	42.0 \pm 3.2	31.7 \pm 1.3	3.6 \pm 0.0	16.3 \pm 1.1	41.1 \pm 3.1
10	2.2 \pm 0.0	40.2 \pm 1.1	26.9 \pm 2.1	3.8 \pm 0.0	17.0 \pm 1.6	39.1 \pm 1.1
12	2.2 \pm 0.0	43.3 \pm 2.2	22.7 \pm 1.1	4.0 \pm 0.0	16.1 \pm 2.0	37.1 \pm 1.1

SD, standard deviation; a, after application of kitchen effluent or sterile distilled water; b, before application of kitchen effluent or sterile distilled water.

Table 3: Mean Population densities of different microbial groups enumerated in experiment B and control B

Sampling Time (Weeks)	Mean Population Densities (cfu/g \pm SD) $\times 10^4$			
	Experiment B		Control B	
	Total HM	Total HCU;%	Total HM	Total HCU; %
O (a)	339.6 \pm 0.2	1.5 \pm 0.2;0.4	20.2 \pm 0.3	0.1 \pm 0.0;0.6
(b)	20.5 \pm 0.2	0.2 \pm 0.0;0.7	20.5 \pm 0.2	0.2 \pm 0.0;0.7
2	31.8 \pm 0.4	2.7 \pm 0.0;8.4	20.9 \pm 0.0	0.2 \pm 0.0;0.7
4	41.4 \pm 0.2	4.1 \pm 0.3;9.8	22.5 \pm 0.4	0.2 \pm 0.2;0.9
6	49.3 \pm 0.5	6.0 \pm 0.2;12.2	24.3 \pm 0.6	0.3 \pm 0.0;1.4
8	56.4 \pm 0.3	8.7 \pm 0.2;15.5	25.9 \pm 0.3	0.7 \pm 0.2;2.8
10	65.1 \pm 0.3	11.7 \pm 0.4;18.0	29.0 \pm 0.5	1.1 \pm 0.1;3.8
12	76.2 \pm 0.5	15.2 \pm 0.1;19.9	35.1 \pm 0.4	1.7 \pm 0.1;4.8

SD, standard deviation; a, after application of kitchen effluent or sterile distilled water; b, before application of kitchen effluent or sterile distilled water; HCU, hydrocarbon utilizers; HM, heterotrophic microorganisms; cfu, colony forming units.

Table 4: Mean change in chemical parameters in experiment B and control B

Sampling Time (Weeks)	Mean Changes In Chemicals Parameters \pm SD					
	Experiment B			Control B		
	Dissolved oxygen (mg/kg)	Biochemical Oxygen demand (mg/kg)	Residual oil content (g/kg)	Dissolved oxygen (mg/kg)	Biochemical Oxygen demand (mg/kg)	Residual oil content (g/kg)
O (a)	4.0 \pm 0.0	10.0 \pm 2.1	48.2 \pm 2.1	4.0 \pm 0.0	10.0 \pm 1.1	48.2 \pm 2.1
(b)	4.0 \pm 0.0	10.1 \pm 3.0	49.0 \pm 1.1	4.0 \pm 0.0	10.0 \pm 2.1	49.0 \pm 1.1
2	3.0 \pm 0.1	40.1 \pm 2.1	44.6 \pm 2.0	3.8 \pm 0.1	17.0 \pm 1.1	46.4 \pm 3.0
4	2.9 \pm 0.0	56.3 \pm 1.2	39.8 \pm 1.3	3.7 \pm 0.0	18.2 \pm 2.2	44.5 \pm 2.1
6	2.5 \pm 0.0	57.6 \pm 2.0	33.3 \pm 1.0	3.6 \pm 0.0	18.0 \pm 1.0	42.5 \pm 1.1
8	2.2 \pm 0.0	52.5 \pm 2.0	27.0 \pm 2.5	3.6 \pm 0.0	17.4 \pm 1.2	40.6 \pm 3.2
10	2.0 \pm 0.0	50.3 \pm 1.3	21.0 \pm 3.2	3.8 \pm 0.0	17.0 \pm 2.3	38.6 \pm 2.1
12	2.0 \pm 0.0	54.1 \pm 2.2	15.7 \pm 3.3	3.9 \pm 0.0	18.0 \pm 3.0	36.7 \pm 1.2

SD, standard deviation; a, after application of kitchen effluent or sterile distilled water; b, before application of kitchen effluent or sterile distilled water.

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Table 5: Mean Population densities of different microbial groups enumerated in experiment C and control C

Sampling	10 ⁴	Mean Population Densities (cfu/g±SD) x			
		Experiment C		Control C	
Time		Total HM	Total HCU;%	Total HM	Total HCU; %
O (a)		407.5±0.2	1.8±0.0;0.5	20.2±0.0	0.1±0.0;0.7
(b)		407.5±0.2	0.2±0.0;0.8	20.5±0.4	0.2±0.0;0.8
2		38.1±0.1	3.2±0.0;8.4	20.8±0.0	0.2±0.0;0.7
4		49.7±0.4	4.9±0.0;9.8	22.5±0.2	0.2±0.1;0.9
6		59.3±0.5	7.2±0.2;12.1	24.0±0.3	0.3±0.0;1.4
8		67.6±0.0	10.5±0.1;15.5	25.8±0.5	0.7±0.0;2.8
10		78.1±0.2	14.0±0.0;17.9	28.8±0.3	1.1±0.0;3.8
12		91.5±0.4	18.2±0.0;19.9	35.0±0.0	1.7±0.0;4.8

SD, standard deviation; a, after application of kitchen effluent or sterile distilled water; b, before application of kitchen effluent or sterile distilled water; HCU, hydrocarbon utilizers; HM, heterotrophic microorganisms; cfu, colony forming units.

Table 6: Mean change in chemical parameters in experiment C and control C

Sampling	Time (Weeks)	Mean Changes In Chemicals Parameters ±SD					
		Experiment C			Control C		
		Dissolved oxygen (mg/kg)	Biochemical Oxygen demand (mg/kg)	Residual oil content (g/kg)	Dissolved oxygen (mg/kg)	Biochemical Oxygen demand (mg/kg)	Residual oil content (g/kg)
O (a)		4.0±0.0	10.1±2.7	48.0±1.2	4.0±0.0	10.1±1.1	47.9±1.1
(b)		4.0±0.0	10.1±3.0	48.7±1.3	4.0±0.0	10.1±3.1	48.7±1.3
2		2.9±0.0	48.2±2.2	43.6±2.2	3.6±0.0	18.1±2.1	46.3±1.2
4		2.6±0.0	67.5±1.3	37.9±1.1	3.6±0.0	18.1±3.0	44.2±1.0
6		2.2±0.0	69.2±2.1	30.1±3.0	3.5±0.0	18.0±0.1	42.3±1.3
8		2.0±0.0	63.0±1.1	22.5±1.2	3.5±0.0	18.0±2.2	40.3±2.2
10		1.9±0.0	60.3±2.3	15.3±4.0	3.4±0.0	18.0±1.2	38.4±1.2
12		1.9±0.0	65.0±2.2	9.0±2.2	3.1±0.0	18.0±2.0	36.4±2.2

SD, standard deviation; a, after application of kitchen effluent or sterile distilled water; b, before application of kitchen effluent or sterile distilled water.

Therefore, the decrease in ROC which was much more rapid in the experimental set-ups could be attributed to the additional activities of the hydrocarbon utilizers introduced into the experimental set-ups by the addition of kitchen effluent. Also the decrease in dissolved oxygen (DO) and the increase in biochemical oxygen demand (BOD) which was much more remarkable in the experimental set-ups (Table 2, 4 and 6) were probably due to the greater oxygen consumption in breaking down the oil pollutants resulting in higher values for BOD when compared with the control set-ups which is in line with previous reports (Singh *et al.*, 1999; Nwachukwu, 2000a). Thus oxygen is a necessary requirement in crude oil degradation (Atlas, 1981).

Conclusion: It was clear that immediately after the introduction of the kitchen effluent into the

experimental set-ups, the environment changed due to the toxicity of the oil present in the drilling muds resulting in the death of many microorganisms present in the introduced kitchen effluent, hence the initial drop in the total heterotrophic microorganisms (THM). However, some of the microorganisms that can utilize hydrocarbons soon adapted to the oiled environment and probably utilized it as a substrate for growth, hence the reduction in the ROC. Therefore, kitchen effluent could probably be used as an alternative inoculum to enhance rehabilitation of oil contaminated drilling muds before it is discharged to avoid further pollution of human environment with oil and to also ameliorate the cost of developing a particular strains or species of a particular genus of microorganisms before introducing them into the oil polluted environment.

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