Isolation and Characterization of 5-Hydoxy-4', 7- Dimethoxyisoflavone from the Root Bark of *Millettia Mannii* (Leguminoceae)**

Ngwendson J. NGUNDE⁺, Samuel N.Y. FANSO-FREE^{+*}, Joseph T. MBAFOR ⁺⁺, Z. Tanee FOMUM^{++*} and Bernard BODO⁺⁺⁺

⁺ Department of Chemistry, University of Buea, P.O. Box 63 Buea, Cameroon

+ + Department of Organic Chemistry, University of Yaounde 1, BP 812 Yaounde, Cameroon

+++ Laboratoire de Chimie URA, 401 CNRS, Museum National d'Histoire Naturelle, 63 Rue Buffon, 75005 Paris, France.

ABSTRACT

The phytochemical studies on the root bark of *Millettia Mannii*, a species belonging to the family leguminoceae and sub-family papillionaceae, led to the isolation and determination of the structure of 5-hydroxy-4',7-dimethoxyisoflavone (6). Though previously synthesized, characterized and prescreened (SEKIZAKI and YOKOSAWA, 1988), compound (6) has been isolated for the first time from a natural source.

Keywords: Leguminoceae; Papillionaceae; Root bark; *Millettia mannii*, 5-hydroxy-4', 7-dimethoxyisoflavone

RÉSUMÉ

Les études phytochimiques sur les écorces des racines de *Millettia mannii*, une espèce de la grandefamille des légumineuses (sous-famille des papillionacées), ont permi l'isolement et la détermination de la structure de 5-hydroxy-4′, 7-diméthoxyiso-flavone (6). Bien que précédemment synthétisé, caractérisé et pre-testé, le composé (6) a été isolé pour la première fois d'une source naturelle.

Mots clés: Léguminocées; Papillionacées; écorces de racines; *Millettia mannii*, 5-hydroxyl-4', 7-diméthoxyisoflavone

Authors to whom correspondences can be addressed.

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INTRODUCTION

The genus Millettia has about 200 species worldwide, with about 41% found in tropical Africa alone (GILLET et al; 1971). Interest on the genus Millettia resulted from their constant use in Africa as a source of insecticides and molluscides, vermifuges and as substances for stupefying fish. The number of Millettia species in Cameroon is about 21, a good number of which have been shown already to possess biologically active constituents (SINGHAL et al; 1982 and DEWICK, 1993). We report on the isolation and structural determination of. 5-hvdroxy-4',7dimethoxyisoflavone (6), which was first synthesized by SEKIZAKI and YOKOSAWA, (1988) and also shown in their study to have a significant attracting activity to Aphanomyces euteiches zoospore (Table 1).

EXPERIMENTAL

The melting point was determined on a Yanagimoto micro melting point apparatus and is uncorrected. The proton nuclear magnetic resonance (¹H NMR) spectrum was recorded at 300 MHz in CDC1₃ with TMS as internal standard, using the Bruker AC 300 spectrometer. The mass spectrum was determined at 70 eV using an apparatus of type R10-10.

The root bark of Millettia mannii was collected from the edges of the sub-equatorial forest around Yaounde in February 1995. Voucher material documenting the collection was authenticated by the Director of the National Herbarium, Yaounde, Cameroon, where a specimen has been deposited. The dried, ground root bark gave 2.1 kg of coarse powder which was successively and exhaustively extracted with MeOH and MeOH/CH₂Cl₂ (1:1). Concentration of both filtrates at reduced pressure yielded 205 g and 90 g respectively. On the basis of their tlc, both extracts were combined to give a total of 295 g of crude extract which was chromatographed on a silica gel column packed in hexane. Elution was then effected with a cyclohexane-CH₂Cl₂ gradient. On elution with

Table 1: Attracting Activity of Isoflavonoids to zoospores of Aphanomyces euteiches (AE-F3)^a

Degree of attracting activity b and Concent							ncentrati	ration (M)
Comp No.	R ¹	\mathbb{R}^2	\mathbb{R}^3	10 ⁻⁵	10-6	10-7	10-8	10-9
1	Н	Н	Н	+++	++	+	·	<u>-</u>
2	Me	Н	Н	+++	+++	++	+	-
3	Н	Me	Н	.+	-	-	-	<u>-</u> .
4	Н	Н	Me	+++	++	+	-	· -
5	Me	Me	Н	140	-	· _	-	-
6	Me	Н	Me	+++	++	+		
7	Н	Me	Me	-			-	-
8	Me	Me	Me	-	-	· . .	-	<u>-</u> ·

AE-F3 is Aphanomyces euteiches isolate fraction 3, isolated from pea root rot soils

b-, no effect; +, attractive effect vanishes within 10 min.; ++, attractive effect persists for 30 to 60 min; ++++, attractive effect persists for more than 60 min.

hexane/CH₂Cl₂ (65:35), a series of identical fractions were obtained and assembled from which (6) crystallized out. Recrystallization in MeOH yielded 9 mg of light yellow crystals:

5-hydroxy-4',7-dimethoxyisoflavone **(6)** crystals, mp 139-140 °C (EtOH). El-MS, m/z (rel.int.): 198 ([M']+ (100), 283(10), 166(22), 138(12), 132(20), 117(12), 95(22), 89(24). HRMS m/z [M]+ 298 ($C_{17}H_{14}O_5$)).

RESULTS AND DISCUSSION

The dried and ground root bark of Millettia mannii was successively extracted with MeOH and MeOH/CH₂CI₂ (1:1). Silica gel chromatography of the combined extracts using gradient elution from cyclohexane to methylene chloride yielded 5-hydroxy-4',7-dimethoxyoisoflavone (6), which was obtained as slightly yellow crystals (mp 139-140 °C (EtOH) (SEKIZAKI and YOKOSAWA, 1988).

The EI mass spectrum of (6) showed a prominent $[M]^+$ at m/z 298 as the base peak. The molecular formula $C_{17}H_{14}O_5$ was determined from high resolution EI data. The peak at m/z 283 (10%) resulted from the loss of a CH₃ radical and those at m/z 166 (22%) and 132 (20%) arise from the RDA-fragmentation of the isoflavonoid molecular ion.

The ¹H NMR spectrum was fascinatingly simple, showing a signal for the characteristic H-2 proton of isoflavonoids (DODDRELL *et al*; 1982) at **ä** 7.84 ppm (1H, s); a chelated hydroxyl proton (offset) at **ä** 12.84 ppm, exchangeable with D₂O; two signals at **ä** 3.82 (3H, s) and 3.85 (3H, s) indicating two groups of methoxyl protons on rings B and C respectively; doublet signals at **ä** 6.95 ppm

(2H, dd) and 7.44 (2H, dd) indicating the AA'BB' signals for the four protons of the para-substituted ring C; signals at ä 6.36 ppm (1H, d) and ä 6.39 ppm (1H, d) indicating the two *meta* protons of ring B.

These data make it possible to conclude that this compound, isolated from a natural source for the first time, has the same structure as compound (6) synthesized and pre-screened by SEKIZAKI and YOKOSAWA, (1988). It has been reported that naturally occurring genistein (1), prunetin (2) and biochamin A (4) have estrogenic properties (SEKIZAKI and YOKOSAWA, 1988). Since the synthetic 5-hydroxy-4',7-dimethylisoflavone (6) exhibited significant activity to Aphanomyces enteiches zoospores (Table 1), the isolation and characterisation of natural compound (6) is strong evidence that this group of isoflavonoids, possess both estrogenic and attracting activities.

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Table 2: ¹H NMR data for 5-hydroxy-4', 7- dimethoxyisoflavone.

Proton	• (ppm)
H-2	7.84 (s)
H-6	6.36 (d)
H-8	6.39 (d)
H-2'	7.44 (dd)
H-3'	6.95 (dd)
H-5'	6.95 (dd)
H-6'	7.44 (dd)
4'-OMe	3.85 (s)
5-OH	12.84 (s)
7-OMe	3.82 (s)

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