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Measurement Of NO₂ And O₃ Ambient Concentrations In A Project Area In Niger Delta Region Of Nigeria.

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ABSTRACT

A pre-impact assessment of NO₂ and O₃ in a proposed project area for associated gas gathering in the Niger Delta region of Nigeria has been conducted. Passive samplers (diffusion tubes) which are devices capable of taking samples of gaseous pollutants from the atmosphere through diffusion to their interior and subsequently trapping them on an adsorbing material without the active movement of the gas through the sampler were utilized for this study. Monitoring was done for 8 months to include the rainy (May - August 2000) and dry (November 2000- February 2001) seasons. Results available indicate low levels of NO₂ and O₃ in the proposed project area with annual ranges of 4.10 - 11.48 μgm⁻³ NO₂ and of 2.72 - 3.78 μgm⁻³ O₃. Temporal and spatial variations were observed in the obtained results.

INTRODUCTION

The Niger Delta region of Nigeria is endowed with natural resources like crude oil, with a proven reserve of 17.8x 10⁹ barrels, and natural gas, with a proven reserve of 2.5x 10¹²m³. In this region, industries related to petroleum exploration, exploitation and production are dominant and have caused a gross pollution of the air environment. In the past, much attention has been focused on aqueous and ecological monitoring in the oil producing areas¹ but little or no results have been reported on the air quality. Air quality assessment in the Niger Delta region is therefore imperative even if the emission of air pollutants is assumed to be comparatively low since high concentrations of some air pollutants have been obtained previously^{2,3} at urban locations in Nigeria. Areas covered in this study are towns and villages in the Niger Delta region of Nigeria for which air *

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quality data do not exist although they have had their own share of environment degradation resulting from petroleum exploration activities. These towns and villages have just been surveyed for the Associated Gas Gathering Project and gaseous pollutants measured in this study are NO₂ and O₃.

In Nigeria, the following sources generally contribute to air pollution⁴: traffic with high emissions from passenger cars and utility vehicles, various types of industries - primarily oil refineries with their area sources, gas flares and electrical power stations, biomass/bush burning, biomass burning for cooking purposes, uncontrolled waste burning and harmattan dust. In Niger Delta, the major sources of air pollution include gas flaring due to the large volumes of natural gas that is produced in association with crude oil and flared away, automobiles, power plants, industrial and commercial boilers and other stationery turbines and engines all of which involve

combustion of fuels, biomass/bush burning¹.

Ozone as an ambient air pollutant has become a subject of great interest because of its effect on the respiratory system of both humans and plants. It is the most phytotoxic and widespread air pollutant affecting plants⁵. Troposphere ozone is equally a "greenhouse gas" because of its ability to absorb terrestrial thermal radiation and contributes significantly to global warming if its concentrations were increased^{6,7}. The effects of ozone on plant growth and yield has been extensively reviewed⁸. Nitrogen dioxide on the other hand, is an oxidant gas which at high concentration causes lung injury. Toxicological studies have shown that nitrogen dioxide reduces the efficacy of lung defense mechanism against infection⁹. Nitrogen dioxide contributes to stratospheric ozone depletion and tropospheric ozone production. Nitrogen dioxide can equally cause serious injury to vegetation and a level of 0.3 to 5ppm for 10 to 20 days inhibits the growth of tomato and bean seedlings¹⁰. Because of the importance of these pollutants, the Nigerian government through its Federal Environmental Protection Agency (FEPA) has prescribed Air Quality Standards for these pollutants¹¹.

The purpose of this study is to ascertain the degree of pollution of the air environment of the project area with NO₂ and O₃. It is hoped that the data generated in this study will regulate public concern and be a further basis of action for pollution remediation in the region.

EXPERIMENTAL

Sampling locations

Air samples were collected at six different locations in the area proposed for associated gas gathering project. The sampling points are described below and

the reference codes for the sampler tubes for the respective points are also given.

Delta Steel (DS): This sampling location was created in the forest along the proposed gas pipeline. It is remote and low in human presence.

Ughelli West (UW): This sampling site is about 100 meters from the proposed gas pipeline, about 300 meters from a flow station and about 350 meters from an electricity generating station.

Ughelli Town (UT): This sampling location was created in a city center with high traffic volume, strong human presence and activities.

Igbede Community (IC): This sampling location created in Igbede community is again remote and close to a flow station.

Ozoro Community (OC): This sampling location was created along the proposed gas pipeline in this community and is remote.

Eriemu Community (EC): This sampling site was created at Eriemu community about 1000 metres from the closest village.

Passive samplers for nitrogen dioxide

Nitrogen dioxide gas was collected by trapping it using passive diffusion tubes developed and used by Palmes and others^{12,13,14}. They consist of small acrylic tubing 7.4cm long, 1cm in diameter having 3 stainless grids as support for adsorbing material at its one end. Triethanolamine was used as adsorbent for nitrogen dioxide.

Diffusion tubes preparation

In preparing the tubes, uniformity was maximized, for example; the same drying time for each of the tube grids, a freshly prepared triethanolamine/acetone mixture. A mixture of two parts of acetone and one part of triethanolamine was prepared. The grids were dipped in the mixture and allowed to dry. Three of the steel grids were pushed into the diffusion tube after

drying. The prepared tubes were mounted vertically with the coloured cap uppermost at a height of approximately 2.0m above the ground surface on the supporting stands at each sampling station.

Sampling routine

Two weeks sampling period was observed to allow a reasonable quantity of nitrogen dioxide to be adsorbed. Nitrogen dioxide monitoring was done for a period of 8 months to include rainy season (May – August 2000) measurement and dry season (November 2000 – February 2001) measurement. The analytical finish was accomplished by using Saltzmann reagent and a clean visible Spectronic 21D Spectrophotometer at zero extinction to determine the extinction of both the blanks which were prepared like the exposed tubes, except that they were not exposed, and the air samples at a wavelength of 540nm. Nitrogen dioxide concentration was calculated from the amount adsorbed by using the following equation¹⁵.

$$C = \frac{22.4 \times T/273 \times F \times Z \times E}{T \times D \times A}$$

F = Calibration factor

Z = Length of the diffusion tube (cm)

Where:

C = NO₂ conc. (ppb)

E = Extinction minus Extinction of the blank

t = Exposure time (s)

D = Diffusion coefficient of the compound in air (cm²/sec)

A = Cross – sectional area of the tube (cm²)

Note: 1ppb NO₂ = 1.91µgNO₂/m³

Passive sampler for Ozone

Air samples for tropospheric ozone were collected by using passive diffusion tubes described and used by Monn and Hangatner¹⁶ and Hauser and Bradley¹⁷. The tubes used are made of polystyrene, 2.9cm long and 0.9cm in diameter. 1, 2-di (4-pyridyl) ethylene (DPE) was used as the adsorbent.

Diffusion tubes preparation

The adsorbent, a solution of DPE was prepared by dissolving it in acetic acid and ethylene glycol. “Whatman EPM 200M” glass fibre filters were used as solid sorbent because of their large surface. The filters were cut on a glass dish mould into a size of 0.8cm by 0.8cm, soaked in the adsorbent solution and hung – dried for 15minutes in a light protected box. With the aid of tweezers, the filters were then transferred into the polystyrene tubes. The prepared tubes were exposed at each site at a height of approximately 2.0m. In this method, the absorption of ozone takes place by reaction with DPE, the ozonide formed undergoes cleavage and yields an aldehyde.

Sampling routine

One-week sampling duration was observed, and as in the case of NO₂ tubes, monitoring was done for 8 months. The quantity of pyridylaldehyde (PA) formed from the reaction of the adsorbent and the ambient tropospheric ozone was determined by using Methylbenzthioazolinonehydrazone (MBTH) in acetic acid. The absorbance of the azine formed was measured at 442nm using a clean UV/Visible spectronic 21D Spectrophotometer at zero extinction. This quantity of azine is proportional to ozone collection.

Table 1: Measured NO₂ concentrations

Sampling location	Tube reference	Ambient NO ₂ concentration (µgm ⁻³)							
		May 2000	June 2000	July 2000	Aug 2000	Nov 2000	Dec 2000	Jan 2001	Feb 2001
Delta Steel	DS	3.02	2.78	3.72	3.64	4.96	5.08	5.12	7.08
Ughelli West	UW	7.84	8.20	8.08	7.22	9.66	10.11	11.04	12.00
Ughelli Town	UT	10.22	9.88	10.62	8.79	9.64	12.98	15.20	14.44
Igbede Community	IC	4.48	6.02	4.22	3.98	5.23	5.28	5.96	6.43
Ozoro Community	OC	5.62	5.40	5.44	4.87	6.08	6.89	7.46	8.02
Eriemu Community	EC	2.66	2.42	2.86	3.20	4.08	5.24	6.32	5.99

RESULTS AND DISCUSSION.

The time-weighted average concentrations of NO₂ and O₃ measured between May 2000 and February 2001 are presented in Tables 1, 2, 3 and 4. As can be seen from Table 1, low concentrations of NO₂ were recorded throughout the sampling duration. Spatial and temporal variations were equally noticed in the obtained results. The highest NO₂ level of 15.20µgm⁻³ was measured at Ughelli town and the lowest of 2.42µgm⁻³ recorded at Eriemu community. An annual range of 4.10 -

11.84µgm⁻³NO₂ was observed in this study. The FEPA¹¹ allowed daily average limit for NO₂ is presently 75 - 113µgm⁻³. The US National Ambient Air Quality Standard¹⁸ for NO₂ is presently 100µgm⁻³ annual average while the World Health Organization has recommended a 1- hourly average short-term standard of 320µgm⁻³ not to be exceeded more than once per month¹⁹. Though levels of NO₂ obtained in this study are very much lower than the national and

Table 2: Seasonal and annual mean NO₂ concentrations

Sampling season	Tube reference	Dry Annual mean
Location (µgm ⁻³)	reference mean (µgm ⁻³)	mean (µgm ⁻³)

Delta Steel	DS	5.56
		4.43±1.42
Ughelli West	UW	10.70
		9.27±1.70
Ughelli Town	UT	13.07
		11.48±2.40
Igbede Community	IC	5.73
		5.21±0.91
Ozoro Community	OC	7.11
		6.22±1.12
Eriemu Community	EC	5.41
		4.10±1.56

international limits for NO₂, higher concentrations were recorded in Ughelli Town. A NO₂ range of 8.79 - 15.20µgm⁻³ was obtained at this sampling site. Mention has already been made of the high traffic

density and intense human activities prevalent at this location. The observed NO₂ level could therefore be ascribed to the heavy traffic around the location since traffic is a contributing factor to NO₂ level

in the atmosphere. The largest single source of NO₂ emissions and, therefore, NO₂ in the atmosphere is from motor vehicles²⁰. The location with the second highest NO₂ concentration is Ughelli West which is about 300 metres from a flow station and about 350 metres from an electricity generating station. The increased level of NO₂ at this site may be due to the emissions of NO₂ gas from the combustion processes going on at the

electricity generating station and the flow station.

Other sampling locations which are remote from vehicular and human traffic gave very

low concentrations of NO₂. Temporal variations in the obtained NO₂ data were observed in all the sampling sites (Table 2). Higher NO₂ concentrations were measured in the dry

Table 3: Measured O₃ concentrations

Sampling location	Tube reference	Ambient O ₃ concentration (µgm ⁻³)							
		May 2000	June 2000	July 2000	Aug 2000	Nov 2000	Dec 2000	Jan 2001	Feb 2001
Delta Steel	DS	2.86	2.89	2.00	2.37	3.29	4.12	3.99	4.02
Ughelli West	UW	3.20	2.73	2.82	2.02	4.33	4.29	5.20	5.64
Ughelli Town	UT	2.63	2.06	2.72	2.08	2.98	4.48	4.98	5.22
Igbede Community	IC	2.02	2.12	2.92	2.08	3.22	3.01	3.24	3.08
Ozoro Community	OC	2.08	2.10	2.96	2.23	3.62	4.12	4.09	4.24
Eriemu Community	EC	2.32	2.28	2.17	2.92	3.13	3.18	3.74	3.86

season at all the locations (Tables 1 and 2). The differences in the concentrations recorded for the two seasons can be attributed to the wet deposition (rain out and washout) of NO₂ as HNO₃ in the wet season. Concentrations of O₃ recorded were equally low throughout the period of study (Table 3) with the highest concentration of 5.64µgm⁻³ measured in the month of February at Ughelli West monitoring station and the lowest concentration of 2.02 µgm⁻³ measured in the months of May and August at two different sampling locations. The ground level distribution of ozone is influenced by the two independent factors, emission rates and meteorological conditions²¹. It is believed that NO₂ is the limiting precursor

for the formation of ozone, a secondary pollutant in the free troposphere²². Low concentrations of O₃ obtained could, to a reasonable extent, be due to the low levels of NO₂ equally measured at the different sampling locations. An annual range of 3.08 - 5.64µgm⁻³ O₃ was found. Ughelli West gave the highest concentrations of O₃ for half of the sampling months. Ughelli town gave the second highest annual mean of 3.40µgm⁻³ O₃. Other remote sampling sites gave very low levels of O₃. Ozone is a respiratory irritant. It is also the most phytotoxic air pollutant affecting plants⁵. Low levels recorded in these remote sites inside rubber plantations is therefore heart warming for the region.

Table 4: Seasonal and annual mean O₃ concentrations

Sampling location	Tube reference	Dry season mean (μgm^{-3})	Wet season mean (μgm^{-3})	Annual mean (μgm^{-3})
Delta Steel	DS	3.86	2.53	3.20±0.80
Ughelli West	UW	4.87	2.69	3.78±1.28
Ughelli Town	UT	4.42	2.37	3.40±1.30
Igbede Community	IC	3.14	2.29	2.72±0.54
Ozoro Community	OC	4.02	2.23	3.18±0.95
Eriemu Community	EC	3.48	2.42	2.95±0.65

Seasonal variation in the concentration of O₃ was equally observed. Table 4 represents the annual and seasonal average ozone concentrations. For all the sampling locations, dry season gave higher ozone concentrations.

Ambient concentrations of NO₂ and O₃ obtained in this study compare favorably with levels obtained previously in studies conducted in Benin City, Nigeria^{23,24}. In consideration of NO₂ values obtained in work done in United Kingdom²⁵ and O₃ values from previous study conducted in Frankfurt, Germany²⁶, communities surveyed for this associated gas-gathering project are relatively free from NO₂ and O₃ pollution.

CONCLUSION

This pre - project assessment of O₃ and NO₂ levels in the environment set aside for an associated gas gathering project indicate that ambient concentrations of NO₂ and O₃ are within the short and long term Federal Environmental Protection Agency limits.

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