An Appraisal of Copper and Zinc as Contaminants of Potential Ecological Concern (COPEC) on onions (*Allium Cepa L*.)

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ABSTRACT

This research estimated the phyto-toxic effects of two heavy metals (copper and zinc) on onion (Allium cepa L). The effective concentration (EC₅₀) for copper and zinc following a 96 hour exposure was estimated at 1.04 mg/L \pm 0.14 mg/L and 1.94 \pm 0.17 mg/L respectively. The results indicated that at concentrations of 1.04 mg/L, copper may be considered as Contaminant of Potential Ecological Concern (COPEC) on the growth of onion (Allium cepa L) i.e. harmful effects of copper cannot be ruled out but may probably be a threat at concentrations ≤ 1.04 mg/L. However, zinc may not be considered as COPEC at concentration less or approximately $< \approx 2$ mg/L (1.94 mg/L) i.e. zinc alone is not likely to cause ecological risk at concentrations lower than 2 mg/L. The lower the EC₅₀ value, the more toxic the contaminant and vice versa. Similarly, if exposure to copper and zinc is likely to occur frequently in areas with heavy anthropogenic activities, then Allium cepa L would be able to absorb significant concentration of the toxicants and this could have multiplier effects on humans who daily consume significant quantity of the crop in their delicacies as a rich source of anti-oxidants.

Keywords: copper and zinc, contaminant of potential ecological concern (COPEC), growth inhibition, onions.

INTRODUCTION

The increasing presence of a myriad of pollutants like heavy metals, fertilizers, and pesticides in the biosphere (water, soil, and air) as a result of anthropogenic contribution to industrialization and urbanization has become an alarming phenomenon around the of world. Major sources metals in environmental media include; industrial activities, application of fertilizer, mining activities. indiscriminate dumping of hazardous waste, etc. Metals enter into the ecosystem intentionally and unintentionally and contaminate food materials, which could affect humans¹.

Although, some heavy metals are essential micronutrients for plant growth, constituents of enzymes, hormones and regulators of a variety of biochemical and physiological processes, they could become harmful if the required concentration is exceeded. Similarly, some heavy metals may destroy endocrine and chemoreceptors in organisms, cause gene alteration, affect sensitive enzymes, inhibit growth, impair reproduction and cause death^{2,3,4}.

The toxicity of heavy metals to plants could be affected by several factors, which include: plant species, specific metal, concentration, chemical form, soil composition and pH. Heavy metals are recalcitrant and do not readily biodegrade in the environment or are easily metabolized and as such bioaccumulate in organisms along the ecological food chain. Heavy metals are usually available in very large proportion in soil and aquatic ecosystems and exposure to high concentrations could induce oxidative stress in various species 5,6.

Soils used for agricultural activities in most regions of the world are contaminated with heavy metals as a result of excessive application of fertilizers, pesticides. herbicides, agrochemicals, sewage sludge (used manure) and contaminated as wastewater for irrigation practices. Consequently, phyto-toxicity results in weak plant growth and yield depression in addition to other toxic alterations at cell and molecular levels⁷. Heavy metals in environmental media render water, soils and sediment unsafe for aquatic, terrestrial and human lives.

Onion (Allium cepaL.) is cultivated in Nigeria and all over the world as a very important vegetable. Scientific investigations on onions revealed that it contains high quantity of bioactive compounds which are effective in the prevention of cancer, heart diseases and have lots of beneficial effects on human health^{8,9}. Allium cepa L was chosen for this research because it is an important halophyte, with remarkable ability to survive under stress conditions. It is readily available all the year round and a viable vegetable consumed daily humans¹⁰. This research therefore by examined the phyto-toxic effects of copper and zinc on onions (Allium Cepa. L), since these metals play a vital role in the growth

and developments of plant species and could be toxic at relatively high exposure levels.

MATERIALS AND METHODS

Materials

The test chemicals – copper (Cu) and zinc (Zn) metals of Analar grade were used for the bio-assessment. Stock solutions of 1000 mg/L Cu and Zn were prepared and serially diluted to obtain the concentrations required for the bioassay. The test species used was *A. cepa L* with mean radius of 6.91 ± 0.07 cm and mean weight of 75.70 ± 0.69 g.

Methods

The bioassay was assessed using the Organization for Economic Co-operation and Development, (OECD) protocol $#208^{11}$. Onion bulbs (*A. cepa L*) that had been airdried for 7 days were prepared for the experiment by shaving off the dried root at the base of the onions to expose the fresh meristematic tissues¹².

The bio-assessment for root growth inhibition started with a range-finding test in which the exposed triplicate species were to concentrations of 1, 10 and 100 mg/L of copper and zinc for 2 days. This was done to establish the concentrations that would be used for the screening (actual) test, which had serial dilutions of 4.0, 2.0, 1.0, 0.5 and 0.25 mg/L. The base of each onion was placed on the brim of the specimen containers such that the root system made contact with the extracts of the metals and control (water) in the dark for 96 h.

At the end of the exposure period, the roots of each onion bulb were removed with forceps and the length measured in cm. The results were used to determine the percentage root growth inhibition in relation to the control and the effective concentration (EC₅₀) which amounts to 50% effects of species exposed to the toxicants. The effect of the toxicants on the morphology of growing roots was also assessed.

Estimation of growth rate and toxicological risk assessment

Root growth inhibition

The root growth inhibition assessment was estimated using the effective concentration (EC₅₀). The percentage inhibition efficiency amounted to 50% of the controls for root inhibition and this was used to determine the responsiveness of onions to the test chemicals at 96 hours¹¹. The growth rate, percentage growth rate relative to control and the percentage growth inhibition efficiency of the metals were calculated using the equations (1) -(3).

Growth rate (cm/hr) = $\frac{mean \ length}{time}$ (1)

Percent growth rate (%) = $\binom{GRs}{GRc} \times 100$ (2)

Percent growth rate = $\frac{GRc - GRs}{GRc} \times 100$ (3) relative to control (%)

Where:

GRs = growth rate for sample

GRc = growth rate for control

Statistical analysis

Data analysis was performed using one way Analysis of Variance (ANOVA) in Statistical Package for Social Science (SPSS), IBM software (version 22). Concentration-response curves (EC₅₀) from the phyto-toxicity tests were analyzed with STATA software (version 8.2). The level of significance at P = 0.05between the controls and the exposed species was also assessed. Other data were represented using different graph styles and patterns.

RESULTS AND DISCUSSIONS

The results of phyto-toxicological effects and metal analysis on onion (*A. cepa L*) exposed to different concentrations of copper and zinc are presented in Tables 1–5, Figures 1 – 2 and Plate 1.

Effective Concentration (EC_{50}) - growth retardation (inhibition)

The data from this study showed that there was root growth inhibition in the treatment tanks when compared to the control experiment which grew significantly (Tables 1-2 and Plate 1). The respective estimated effective concentrations (EC_{50}) for copper and zinc were 1.04 ± 0.14 mg/L and 1.94 ± 0.17 mg/L with a safe limit of 0.104 ± 0.05 mg/L and 0.194 ± 0.02 mg/L. The results indicated that at concentrations of 1.04 mg/L, copper may be considered as Contaminant of Potential Ecological Concern (COPEC) on the growth of onion (Allium cepa L) i.e. harmful effects of copper cannot be ruled out but may probably be a threat at concentrations \leq 1.04 mg/L. However, zinc may not be considered as COPEC at concentration less or approximately $< \approx 2 \text{ mg/L}$ (1.94 mg/L) i.e. zinc alone is not likely to cause ecological risk at concentrations lower than 2 mg/L. The lower the EC_{50} value, the more toxic the contaminant and vice versa¹³.

The mean percentage growth rate relative to the control recorded for the respective concentrations of copper and zinc (0.25, 0.5, 1.0, 2.0, 4.0 mg/L) was 80%, 67%, 57%, 42%, 9% and 100%, 92%, 67%, 49%, 30% respectively. The mean percentage growth rate inhibition efficiency relative to the control recorded for copper and zinc were 20, 33, 43, 58, 91% and 0, 8, 33, 51, 70 respectively (Figures 1 and 2). Phyto-toxicity impacted by the metals include decolouration of the test solutions and stunted growth especially in the higher concentrations of 2 mg/L and 4 mg/L. Other effects to the onions

specie include; broken roots, twists, roots bent upwards, stunted growth, bulb deformation and tissue damage.

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Concentration, (mg/L)	Mean root length $(RL) + SD (cm)$	Growth rate (cm/hr)	Percent growth rate relative to control (%)	Percent growth rate inhibition efficiency (%)
Control	7.9 ± 0.06	0.082	100 ± 0.0	0 ± 0.0
0.25	6.33 ± 0.09	0.066	80 ± 1.0	20 ± 0.0
0.50	5.3 ± 0.06	0.055	67 ± 0.5	33 ± 0.5
1.0	4.5 ± 0.04	0.047	57 ± 0.0	43 ± 0.5
2.0	3.3 ± 0.06	0.034	42 ± 0.5	58 ± 1.0
4.0	$0.7 {\pm}~ 0.02$	0.007	9 ± 0.5	91 ± 1.0



Figure 1: Mean (%) root growth (RG) of onions (*Allium Cepa L*) against log concentration of copper

I able 2: Mean root length (RL) and growth rate (%) of Allium cepa L exposi-	osure to zinc
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Concentration	Mean root length	Growth rate	Percent growth rate	Percent growth rate
(mg/L)	$(RL) \pm SD (cm)$	(cm/hr)	relative to control (%)	inhibition efficiency (%)
Control	7.9 ± 0.06	0.082	100 ± 0.0	0 ± 0.0
0.25	7.9 ± 0.06	0.082	100 ± 0.0	0 ± 0.0
0.50	7.3 ± 0.12	0.076	92 ± 1.0	8 ± 0.0
1.0	5.28 ± 0.04	0.055	67 ± 0.5	33 ± 1.0
2.0	3.84 ± 0.05	0.040	49 ± 0.5	51 ± 1.0
4.0	2.4 ± 0.01	0.025	30 ± 0.0	70 ± 2.0



Figure 2: Mean (%) root growth (RG) of onions (*Allium Cepa L*) against log concentration of zinc



Plate 1: Plate showing the effects of exposure of copper to onions (*Allium cepa L*)

Ecological risk assessment of the heavy metals

On the Eco toxicological Risk Assessment Matrix (ERAM), risk levels can be classified as low, medium, or high (Table 3)^{14,15}. If an environment is contaminated with heavy metals, animals (A), plants (P), environment (E) and community (C) may be affected and classification can be done based on exposure concentration, exposure duration and potency of the toxicant^{14,15}. The risk levels are categorized in a numbered format. Each hazard is given a rating, and this was multiplied by the probability that these hazards would occur using the relationship:

Risk level = Hazard severity x likelihood of exposure (Table 4).

Hazard severity are rated as 1 (slight effect), 2 (minor effect), 3 (localized effect or damage). 4 [major effect (deaths)] and 5 [extensive effect (death of population)]. Similarly, the likelihood of occurrence or

exposure are rated as 1 (seldom - A yearly), 2 (frequent -B - quarterly), 3 (very likely - C - monthly). 4 (near certain - D weekly) and 5 (certain $-E - daily)^{15,16}$. The release of the chemicals - metals (hazard) into the environment could result in major damage or death, which could be given a rating of 4 (D). The next step is to consider how often each hazard is likely to occur due to exposure of the chemical or pollutant as indicated in Table 5. A daily uptake of the test metals by the plant species would be represented as D4 (P; E; C) or 16 (P; E; C) for copper (extremely toxic - $EC_{50} = 1.04 \pm$ 0.14 mg/L) and zinc (extremely toxic - $EC_{50} = 1.94 \pm 0.17 \text{ mg/L}$). Although Cu and Zn have the same eco toxicological risk assessment value, both were extremely toxic $(EC_{50} - 1.0-10 \text{ mg/L})^{17}$. However, the data from this assessment showed that copper was more toxic to onion (Allium *cepa L*) than zinc i.e. the lower the EC_{50} concentration. the more toxic the contaminant¹³.

 Table 3: Ecotoxicological Risk Assessment Matrix (ERAM)

	CONSEQUENCE						INCREAS	ING PROB.	ABILITY		
SEVERITY			Р	Α	Е	С	Α	В	С	D	E
							Never experienc e the chemical in the area	Had been expose d / used in the area	Had been exposed / used in the area and other location s	Had been expose d / used several times in the area	Had been exposed / used several times in the area and other location
0	Practically non-toxic	>100 0	No iniury	No effect	No effect	No impact	*Area 1				_
1	Practically non-toxic	>100 0	Slight	Slight effect	Slight effect	Slight impact					
2	Slightly toxic	100- 1000	Minor injury	Minor effect	Minor effect	Limited impact					
3	Very toxic	10- 100	Major injury	Localize d effect	Localize d effect	Considerabl e impact					
4	Extremely toxic	1.0- 10	Single fatality	Major effect (deaths)	Major effect	National impact		Area 2		Area 3	
5	Super toxic	<1.0	Multipl e fatality	Extensiv e effect (kills)	Massive effect	Internationa l impact					

Abbreviations: LC50 median lethal concentration in ppm. Data from (OECD¹⁷, GESAMP¹⁸)

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Concentration of	Frequency of	Hazard severity	Risk level	Hazard rating	
test metal	exposure (a)	(b)	(a X b)		
Control	D	0	D 0	D 0 or 0 (P,E,C)	
1.0 (Cu); 2.0 (Zn)	D	5	D 5	D 5 or 20 (P,E,C)	

Table 4:Ecotoxicological risk assessment for the study

Table 5:Consequences of the effects of heavy metals using Ecotoxicological RiskAssessment Matrix (ERAM)

Concentration	Consequences	Toxic				
of test metals	Plant (P)	Animal (A)	Environment (E)	Community (C)	consequence	
Control	No injury	No effect	No effect	No impact	Practically non- toxic	
1.0 (Cu); 2.0 (Zn)	Major effect (deaths)	Major effect	Major effect	Major effect	Extremely toxic	

Discussion

Globalization and industrial technology / advancement had introduced toxicants including agrochemicals, production chemicals and electronic waste containing significant amount of heavy metals into environmental media. A large number of chemicals are being used by industries during production processes and the waste containing toxins are discharged into recipient systems, which could runoff into soils and as a result soils used for agricultural purposes are contaminated. Industrial wastes are among the major sources of heavy metal pollution in organisms, plants and the environment. Metal contaminants are mixed in the environmental system through human activities such as discharge of sewage, smelting processes, effluents and leaching of garbage¹⁹.

Plant may accumulate significant amounts of heavy metals, decreasing the quality and biological value of products without physical symptoms of their damage. Similarly, since metals are used in the production of fertilizers, when such fertilizers are used excessively or indiscriminately, the metals rise to a toxic level in the receiving plant. The consumption of such plants could be toxic to man and other organisms^{6,20}.

Onion (Allium cepa L.) is a poor competitor and as a result of the inability to compete with weeds, herbicides and fertilizers are used on onion plantation. Some of the direct toxic effects caused by high concentration of heavy metals include inhibition of cytoplasmic enzymes and damage to cell structures due to oxidative stress. Some researches revealed that some substances could cause a decrease in the mitotic index in onion (Alliumcepa L) root tips^{21,22}. This research showed that the release of copper and zinc into the environment through the use of compounds (substances) containing these heavy metals e.g. industrial wastewater and electronic wastes could have negative environmental impact on the growth of onion (Allium cepa *L*.). Although both copper and zinc affected the growth of onion in this appraisal, copper was more toxic than zinc to onion (Allium cepa L).

Copper is an essential micronutrient involved in a number of biological

processes needed to sustain life, however, it can be toxic when present in excess. He et. al.,²³, reported that soil properties could influence absorption and accumulation of heavy metals in vegetables. Despite the role of copper as a micro nutrient, high concentrations can pose a threat to man when consumed²⁴. Similarly, zinc is essential to all organisms and has an important role in metabolism, growth, development and general well-being. It is an essential co-factor for a large number of enzymes in the body. Zinc deficiency could lead to a myriad of metabolic disorders, while excess amount could result in many health related issues²⁵. Zinc has its primary site of accumulation and may persist for many years without decomposition²⁶. The bioavailability of these metals could disrupt different species' ability to fight disease and function.

The high mobility of these metals from the solution to the vegetable (onion), which could have resulted in the root growth inhibition amongst other effects, is an indication that in times of heavy water movement such as irrigation, uptake of these metals is bound to occur through waste water, pesticides or fertilizer that will be washed (runoffs) along farm lands. Thus it is imperative that these types of vegetables should not be cultivated in farms and fields close to heavy industrialized which areas have heavy vehicular movements irrigated with or are questionable water quality which could be sources of heavy metal contamination.

CONCLUSION

This appraisal revealed that copper and zinc were extremely toxic to the exposed test specie, *Allium cepa L*, with copper being

slightly more toxic than zinc. The production, use and disposal of chemicals containing heavy metals should be properly regulated to minimize the adverse effect these metals and other harmful substances to organisms in could cause the environment and humans. It should be high concentrations noted that of Contaminants of Potential Ecological Concern (COPEC) (e.g. heavy metals) in environmental media could pose deleterious impact and ecological risk on species and this could have multiplier effects on humans, who daily consume significant quantity of the crop (Allium cepa L) in their delicacies as a rich source of anti-oxidants.

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