



# THE ENVIRONMENTAL DEPOSITION OF SOME ELEMENTS ENRICHMENT USING DITCH CUTTINGS SAMPLES FROM X – WELL IN OLIOGO, ONSHORE NIGER DELTA

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## ABSTRACT

*Ditch cuttings were collected at various intervals from X well in onshore Niger Delta. The study area which is Oliogo community Delta State Nigeria is located in North – North East (NNE) of Umutu. The purpose of the study is to evaluate the elemental concentration and determine the environment as well as the prevailing conditions under which the elements were deposited. The elemental concentration was determined with Atomic Absorption Spectrometer (AAS) model 969 unicam series. From the lithos stratigraphic study of the selected interval, two lithologic types were determined which are sandy shale and shale. Studies on the geochemical concentration of some elements and their variations with depth show various levels of elemental enrichment in the following order of relative abundance >Na>Ca>Fe>Mg>Mn> Ni>Co>Pb. The degrees of enrichment suggest a significant difference in depositional environment during accumulation. Further studies on the paleo redox proxies of the samples suggest a high oxygen (oxic) environment.*

**Keywords:** lithologic, elemental, enrichment, depositional environment, and ditch cuttings.

## 1. INTRODUCTION

Elements occur naturally in soils; some at a high concentration while others at low concentration. The sources of these elements may depend not only on anthropogenic and lithogenic input but also upon the textural characteristics, organic matter contents, mineralogical composition and depositional environment of sediments [1]. The occurrence of these elements in sediment is closely tied to several factors such as the organic matter content and depositional environment of the sediments (since measurable quantities of these elements do not automatically infer anthropogenic enrichment).

The magnitude of these elements enrichment in sediments varies. According to [2], the approximate order of enrichment relativities to an average shale is Mo>Pb>Zn>V>Ni>Cu>Cr>Co. However, the degree of enrichment may suggest differences in depositional environment during accumulation. The aforementioned elements can be of considerable help in elucidating probable physicochemical conditions prevailing in a given environment. Other influence and controls of these elements enrichment in sediments

include, temperature, salinity, pH, Eh, ionic potential etc. Thus inasmuch as these same factors control to a large extent the physicochemical conditions prevailing in the environment, and indirectly then the availability of both the different elements that are essential to life processes of organism in that environment and the means of incorporating these same elements into the sediments, we ought to expect that when certain elements or groups of elements are identified they will be reflective of specific environment characteristics [3-6].

Various studies have utilized either Al-normalized elemental ratios such as Mo/Al [7&8] or some element ratios such as Ni/Co, V/Cr and V/(V+Ni) [9 &10], to evaluate Paleo-redox conditions. Ni and V occur in tetrapyrrole structures that are preserved preferentially under anoxic conditions [11]. [12] Proposed that the ratio of vanadium to nickel and crude oils which is unaltered by diagenesis, indicates environmental conditions at the time of deposition. They demonstrated that V/(V+Ni) ratio for organics formed under euxinic conditions are greater than 0.5. In a comparison of several geochemical redox

indicators including the degree of pyritization. According to [10], they suggested that V/(V+Ni) ratios greater than 0.84 for euxinic, 0.54 -0.82 for anoxic, and 0.46 -0.60 for dysoxic conditions. However, the use of other element ratios has led to different interpretations. V/Cr and Ni/Co were used previously to evaluate for redox condition in different sediment studies [9]. Vanadium (V) initially incorporated into tetraply role structures under anoxic conditions, also may be absorbed unto clay minerals but most likely this occur following burial [13]. Also, Cr is believed to be associated only with the detrital fraction and is not influenced by redox conditions [14]. High V/Cr values (>4.25) are thought to indicate anoxic conditions [9]. Both Ni and Co occur in pyrite but high Ni/Co ratio are thought to be associated with anoxic conditions. Similarly [9] suggested that Ni/Co ratio <5 inferred oxic conditions, 5-7 dysoxic conditions and >7 suboxic to anoxic conditions.

Others include works of [6, 15 & 17 - 33].

## 2. STUDY AREA

The study area is located in Oliogo town at the south western part of Delta State of Nigeria. It lies within latitude 5°06' -5°50'N and longitude 6°10' -6°17'E (Fig. 1). The study area is accessible through major roads that run from Umutu to Umuaja (The source of River Ethiope) and environ. Beside the Agbor-Warri Road, other road networks include link roads to oil well heads, minor roads and footpaths connecting neighbouring villages.

The study area is sparsely populated. The topography of the area is quite flat and low lying with no major outcrops. On the average, the low lying area does not exceed an elevation of 30 -37m above sea level. The area falls within the broad Niger Delta described by many workers which include [34-40].

## 3. MATERIALS AND METHODS

Ditch cuttings were collected at depth intervals from X well in Oliogo and subjected to both granulometric and geochemical analysis. 100g of the dried sample were obtained from the prepared samples; 50g was used for megascopic studies while the other 50g was subjected to geochemical analysis in order to determine the element concentration of the selected elements.

Two grams (2g) of the sample were also weighed in a 250ml conical flask. 50ml of 0.05N Hcl and 50ml 0.1 N H<sub>2</sub>SO<sub>4</sub> was subjected to heat for about an hour after which the sample was allowed to cool and then 10 percent dilute Hcl added. Samples were filtered into a

120ml rubber bottle and made up to the mark with distilled water. Metal concentrations were then determined with the aid of the atomic absorption spectrophotometer (AAS) unicam serie medel 969 series, [41].

## 4. RESULTS AND DISCUSSION

A lithostratigraphic section of the studied well was established based on megascopic description of the ditch cuttings.(Table 1) and (Fig. 2).

The results of geochemical analysis of selected elements (Table 2) indicate significant environmental factors. The varying degrees of enrichment suggest differences in the depositional environment during accumulation. The degree of this enrichment is influenced and controlled by factors such as temperature, salinity, pH, Eh, ionic potential etc. The Eh represents

Table 1. Megascopic Description of X-Well Section.

| Well depth (ft) | Lithology   | Lithology Description                                  | % Shale |
|-----------------|-------------|--|---------|
| 9850            | Sandy-shale | Dark brown sandy shale                                 | 88      |
| 9880            | Sandy-shale | Light grey shady shale                                 | 70      |
| 9910            | Sandy-shale | Light grey sandy shale                                 | 75      |
| 9940            | Sandy-shale | Grey sandy shale                                       | 82      |
| 9970            | Sandy shale | Light brown sandy shale                                | 80      |
| 10000           | Sandy-shale | Light - dark grey sandy shale                          | 75      |
| 10030           | Sandy-shale | Dark brown sandy shale                                 | 60      |
| 10060           | Sandy-shale | Dark grey sandy shale                                  | 68      |
| 10090           | Sandy-shale | Dark grey sandy shale                                  | 85      |
| 10120           | Sandy-shale | Dark grey - dark brown sandy shale                     | 75      |
| 10130           | Sandy-shale | Grey - dark brown sandy shale                          | 80      |
| 10210           | Sandy-shale | Light grey sandy shale                                 | 85      |
| 10240           | Sandy-shale | Light grey sandy shale                                 | 88      |
| 10300           | Sandy-shale | Grey fairly cemented with organic material sandy shale | 75      |
| 10330           | Shale       | Light grey shale                                       | 100     |
| 10360           | Shale       | Light grey shale                                       | 100     |
| 10460           | Shale       | Brown grey shale parthy laminated                      | 100     |
| 10510           | Shale       | Light grey shale                                       | 100     |
| 10610           | Shale       | Medium grey shale                                      | 100     |
| 10640           | Shale       | Dark grey shale with termination                       | 100     |
| 10680           | Shale       | Dark grey shale with lamination                        | 98      |

| Well depth (ft) | Lithology   | Lithology Description           | % Shale | Well depth (ft) | Lithology | Lithology Description             | % Shale |
|-----------------|-------------|---------------------------------|---------|-----------------|-----------|-----------------------------------|---------|
| 10710           | Shale       | Dark grey shale with lamination | 99      | 11130           | Shale     | Dark grey laminated shale         | 100     |
| 10740           | Shale       | Dark grey shale with lamination | 100     | 11160           | Shale     | Dark grey laminated shale         | 100     |
| 10770           | Shale       | Dark grey laminated shale       | 100     | 11190           | Shale     | Dark brown laminated shale        | 98      |
| 10800           | Shale       | Dark grey laminated shale       | 100     | 11220           | Shale     | Grey laminated shale              | 100     |
| 10850           | Sandy shale | Dark grey sandy shale           | 70      | 11250           | Shale     | Dark brown laminated shale        | 100     |
| 10910           | Sandy shale | Dark grey sandy shale           | 65      | 11370           | Shale     | Light grey laminated shale        | 100     |
| 10940           | Sandy shale | Dark grey sandy shale           | 63      | 11460           | Shale     | Light browns hale                 | 97      |
| 11010           | Sandy shale | Dark grey sandy shale           | 78      | 11490           | Shale     | Light brown poorly cemented shale | 90      |
| 11070           | Shale       | Brown - grey shale              | 97      | 11550           | Shale     | Dark grey laminated shale         | 100     |
|                 |             |                                 |         | 11580           | Shale     | Dark grey laminated shale         | 100     |
|                 |             |                                 |         | 11670           | Shale     | Dark grey laminated shale         | 100     |
|                 |             |                                 |         | 11730           | Shale     | Dark grey laminated shale         | 100     |

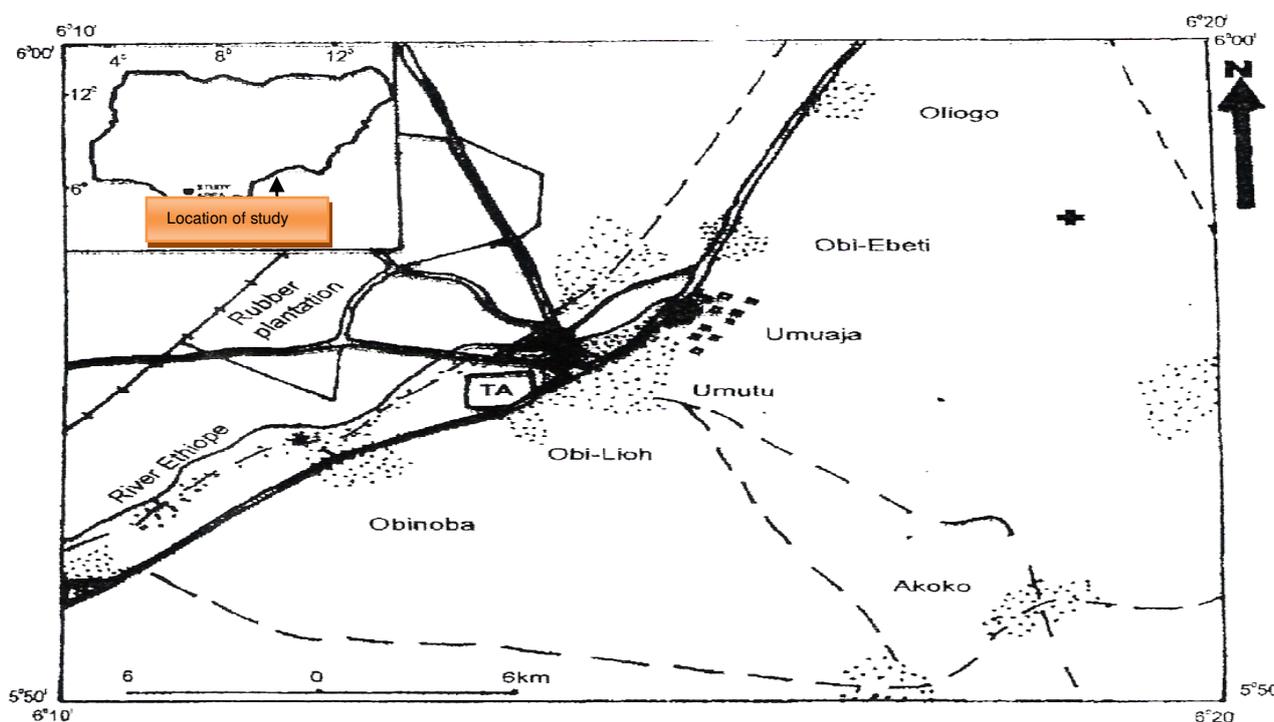


FIG. 1 : MAP OF OLIIGO AND ENVIRONS

**LEGEND**

|                          |                   |              |
|--------------------------|-------------------|--------------|
| MAJOR TOURIST ATTRACTION | X-well            | RAILWAY-LINE |
| DUAL-CARRIAGEWAY         | TOWNS             |              |
| MAJOR TARRERD ROAD       | BOREHOLE LOCATION |              |

Table 2 Concentration of Trace Element In Mg/kg

| Depth of X-well Parameter | Ca      | Mg      | K        | Na      | Ni    | Pb    | Mn     | Fe      | Cr   | Co    |
|---------------------------|---------|---------|----------|---------|-------|-------|--------|---------|------|-------|
| X-well - 9970             | 6138.00 | 4191.60 | 11160.80 | 5573.00 | 7.92  | 0.62  | 42.83  | 4393.50 | 0.31 | 5.58  |
| 10090                     | 5940.00 | 4590.80 | 12157.30 | 5174.00 | 8.91  | 0.62  | 49.30  | 2727.00 | 0.62 | 4.96  |
| 10300                     | 2524.50 | 3792.40 | 12954.50 | 6766.00 | 11.88 | 0.62  | 104.58 | 4646.00 | 0.92 | 5.58  |
| 10360                     | 2920.50 | 4391.20 | 13751.70 | 4975.00 | 6.93  | 0.62  | 74.70  | 4797.50 | 1.23 | 8.06  |
| 10610                     | 4504.50 | 3293.40 | 10762.20 | 3283.50 | 7.92  | 3.08  | 164.43 | 1010.00 | 0.62 | 3.10  |
| 11160                     | 9702.00 | 5988.40 | 18734.20 | 6766.00 | 15.84 | 4.31  | 253.98 | 2474.50 | 1.54 | 8.06  |
| 11280                     | 3712.50 | 2794.40 | 26148.20 | 6169.00 | 9.41  | 12.92 | 263.92 | 8888.00 | 0.92 | 11.16 |

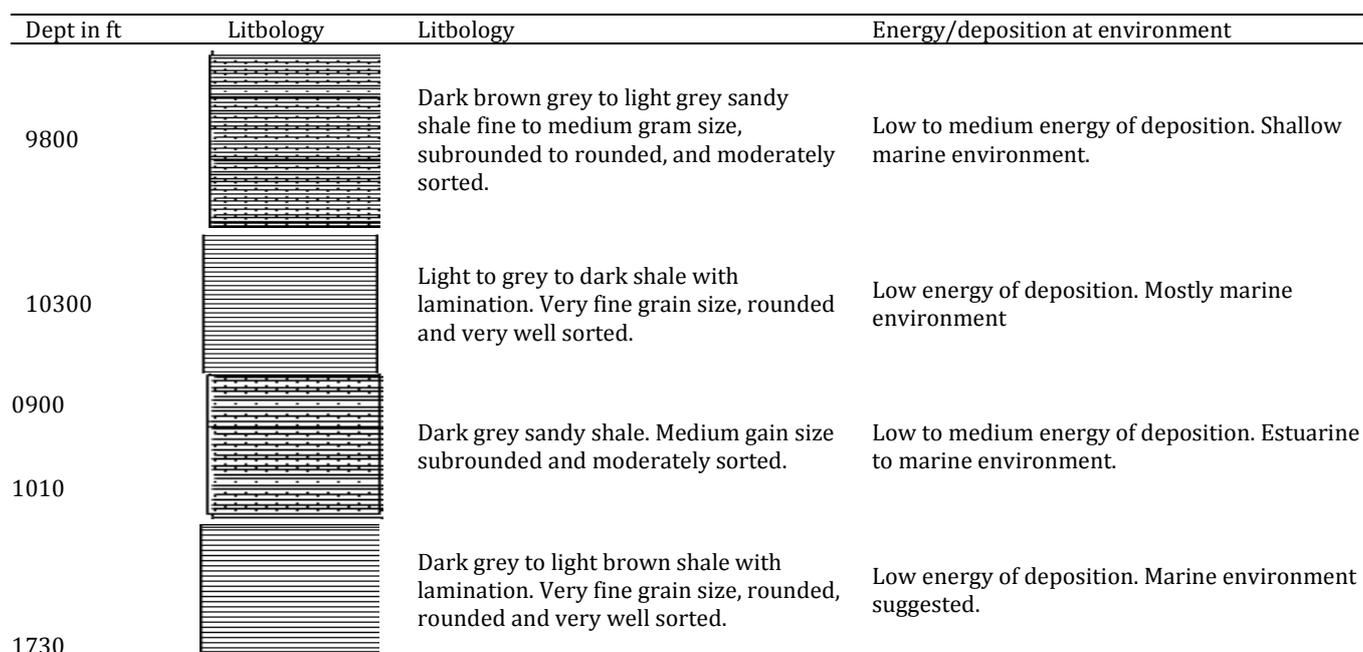


Fig. 2 Lithostratigraphy Section of the Well

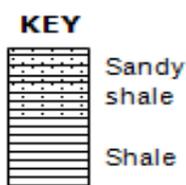


Table 3: Paleo Redox Indicators

| Depth of Borehole in ft   | Ni    | Co    | Ni / Co |
|---------------------------|-------|-------|---------|
| X <sub>-well</sub> - 9970 | 7.92  | 5.58  | 1.42    |
| 10090                     | 8.91  | 4.96  | 1.80    |
| 10300                     | 11.88 | 5.58  | 2.12    |
| 10360                     | 6.93  | 8.06  | 0.86    |
| 10610                     | 7.92  | 3.10  | 2.55    |
| 11160                     | 15.84 | 8.06  | 1.97    |
| 11280                     | 9.41  | 11.16 | 0.84    |

The concentrations of some elements within the X well section were observed to vary in the sample analyzed with depth. Concentration values were in the increasing order of Cr<Pb<Co<Ni<Mn<Mg<Fe<Ca<Na<K (Fig. 2). The values of K indicate a very high enrichment because its lowest and highest limits of concentration were far greater than those of other elements investigated. Absolute abundance of Potasium (K) is primarily controlled by increase in the amount of dominant clay mineral (illite). It is notable that the highest concentration of Mn, K and Fe occur at the same positions. The Mn content of the shale may be ascribed to concentration of the element by secondary oxidation

and would indicate that oxygen was present in the environment during their deposition.

The ratio of K/Na shows a fluctuating scenario for the well studied which is between 1.91 and 4.24. These differences obviously reflect in part the different compositions of the source rocks. Cross plots of concentration of these elements against depth of well (Fig. 3 & 4), show that there is great stratigraphic variability in elements concentration.

From results, the threshold value of the Ni/Co ratios from well section indicates oxic condition during accumulation of sediments (Table 3 and Fig 4).

Ni/Co as an indicator of oxygen level shows that the ditch cuttings have high oxygen values thus suggesting an oxic environment as its paleo -redox proxies.

Table 4: Trace Element Paleo-Redox Proxies indicator for Ni/Co After [9 & 10])

| Elemental Ratio | V/(V+Ni)    | V/Cr    | Ni/Co |
|-----------------|-------------|---------|-------|
| Euximic         | >0.84       |         |       |
| Suboxic/anoxic  | 0.54 - 0.82 | >4.25   | >7    |
| Dysoxic         | 0.46 - 0.60 | 2 - 4.5 | 5 - 7 |
| Oxic            |             | <2      | <5    |

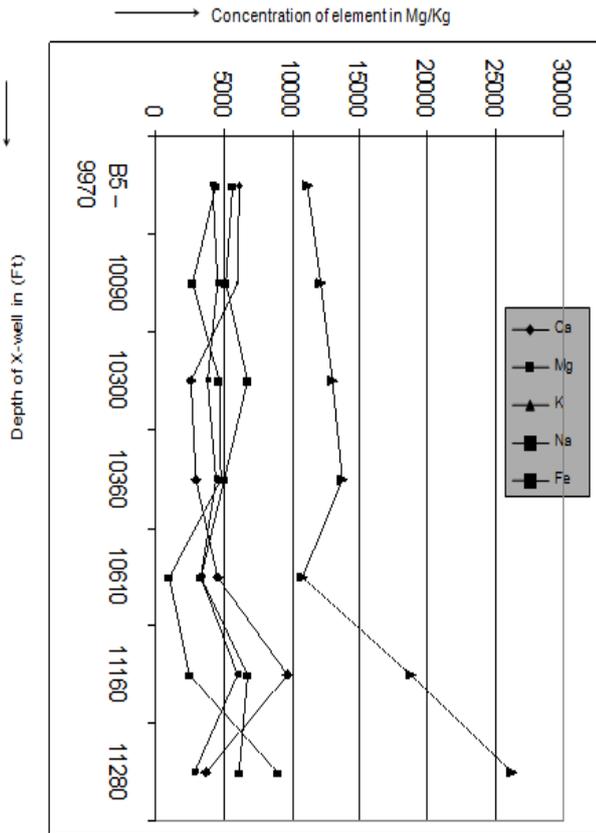


Fig. 3: Cross plot of Conc. Ca, Mg, K, Ni and Fe against Depth

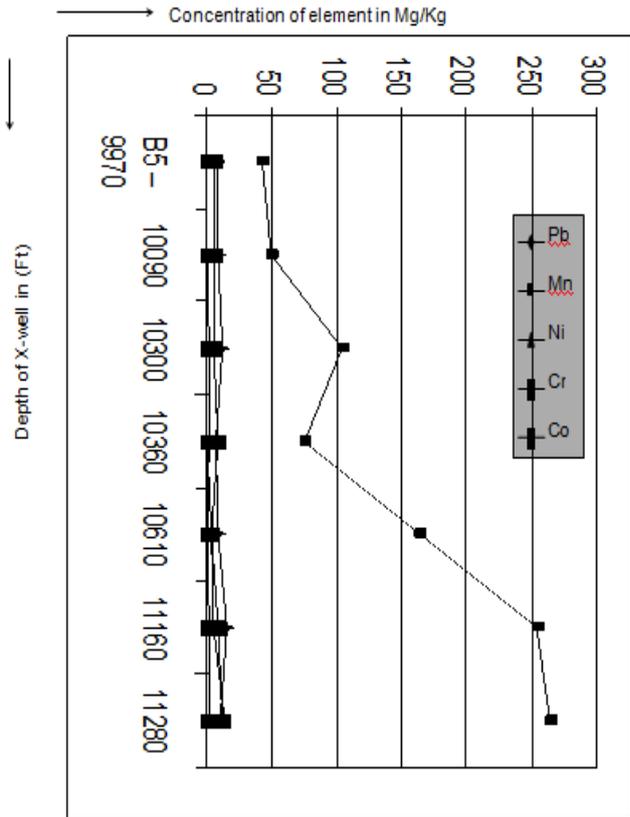


Fig. 4: Cross plot of Conc Pb, Mn, Ni, Cr and Co against Depth

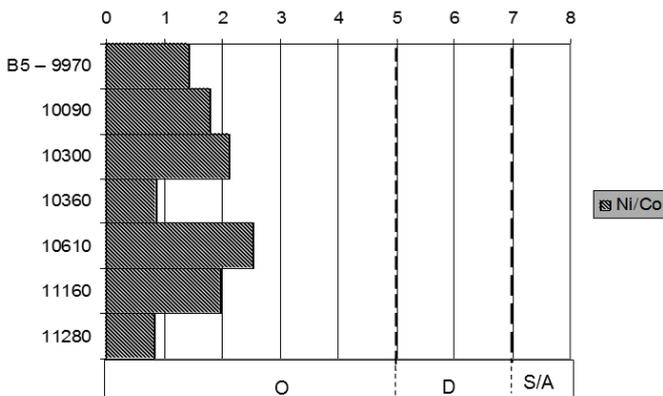


Fig. 5: Trace Element Paleo-redox proxies for Ni/Co, O is the Oxidic, D is the Dysoxic and S/A is the Suboxic/Anoxic.

5. CONCLUSION

The analyses of element concentration within the X-well section vary greatly with depth interval in relation to the different geochemical analysis result of the element. The scattered nature of these elements all over the well section is as a result of differences in depositional environment during accumulation, the physicochemical condition of that environment,

synergetic mineralization, and relative mobility of element. All of these distinctive features of the elements are influenced and controlled by different factors such as temperature, salinity, pH, Eh, ionic potential which also help in assessing an environment to be reducing or oxidizing from paleo redox indices of some trace element ratios. Some element redox indices especially *Ni/Co* used here suggest a high oxygen environment for the samples. However, there is no complete agreement between different redox trace-element ratios. Therefore, it is suggested that absolute thresholds established in other studies should be applied carefully, but that ratios may provide information on relative differences in redox conditions. Relationships of some element ratios such as *Ni/Co* may also provide clues as to whether redox conditions were the primary control on organic matter accumulation. Given the variation in the-element concentration values within the X-well section, it is recommended that investigation to determine these element concentrations in ditch cuttings should be based on the average value obtained from different wells where such previous works have successfully

been carried out. This will yield more accurate results in comparison with such standard average value from previous work.

For further investigation to determine the clastic influx which control the organic carbon concentration in such sample through dilution, by influencing burial rate, or by providing site for organic matter absorption in and on aluminosilicates of such a sample. It is recommended that elements such as Ti and aluminium in such samples should also be determined since the ratio of Ti/Al has been used as all indicator of siliclastic grain size. Hence sedimentation rate and this ratio also give an indication of the paleo-wind strength. K/Al ratio gives an idea of uniform detrital clay influx and increase in K/Al could also suggest a more micaceous clay input or possibly an increase in fine -grained K -feldspar.

All these information when tied together will give a detail understanding of these elemental enrichment in ditch cuttings and the degree of variability within the stratigraphic framework and to help decipher the paleo-redox proxies of these elements precisely and accurately in that environment with conditions that had prevailed.

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