Tropical Journal of Pharmaceutical Research, March 2008; 7 (1): 921-927
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Research Article

A New Flavanone from *Flemingia strobilifera* (Linn) R. Br. and its Antimicrobial Activity

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Abstract

Purpose: To carry out a bioactivity guided fractionation and isolation of the antimicrobial constituent(s) of the roots of Flemingia strobilifera against some bacteria and fungi.

Methods: The root of F. strobilifera was extracted with methanol, butanol and dichloromethane. Antimicrobial activity of the extracts was determined against both bacteria and fungi while the isolation and characterization of compounds of the extracts was done by column chromatography, 2D spectroscopic studies and spectral data (U.V, I.R, NMR and MS).

Results: Flemingiaflavanone (8, 3'-diprenyl-5, 7, 4'-trihydroxy flavanone), Genistin (5, 4'-dihydroxy isoflavone 7-O-glucoside) and β - sitosterol-D glucoside were isolated from the extracts. Flemingiaflavanone showed significant antimicrobial activity against Gram-positive (S. aureus, S. epidermidis, MRSA), Gram-negative bacteria (Ps. aeruginosa, E. coli) and fungi (C. albicans). Genistin showed moderate activity against Gram-positive, Gram-negative bacteria and fungi.

Conclusions: The isolation and antimicrobial activity of presence of Flemingiaflavanone is being reported for the first time.

Key words: Flemingia strobilifera, Flemingiaflavanone (8, 3-diprenyl-5, 7, 4-trihydroxy flavanone), Genistin (5, 4-dihydroxy isoflavone 7-O-glucoside), MeOH (Methanol), BuOH (Butanol), Antimicrobial activity, MIC (Minimum inhibitory concentration).

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INTRODUCTION

Flemingia strobilifera (R.Br.), an important medicinal plant, commonly known as Kusrunt and belongs to the *Leguminosae* family ^{1, 2}. The plant is found in Sind, Rajputana, Bengal, South India and Andamans³. The roots of this plant have been indigenously used in epilepsy and hysteria and the leaves are reported to be used as vermifuge⁴. Previous phytochemical investigations reported various chalkones^{4, 5}, flavonoid glycosides⁶, aurone glycosides⁷ and epoxy chromenes⁸.

This paper reports the isolation and structural elucidation of a new flavanone from the roots of *Flemingia strobilifera* characterized as 8, 3-diprenyl-5, 7, 4-trihydroxy flavanone (1) that is being reported for the first time from the genus *Flemingia* along with two known compounds and their antimicrobial activity.

Experimental

General Melting point was determined on a Buchi Melting Point B-540 (Switzerland) apparatus. U.V was performed in U.V-Perkin Elmer double beam U.V Spectrophotometry (Germany). I.R spectra were recorded on a Jasco FT/IR 410 (U.S.A) in KBr. NMR spectra measured Bruker 400 were in MHz Ultrashield. Advance 400 (Germany) spectrometer, using TMS as internal standard. NMR experiment included the HSQC, HMBC, COSY pulse sequences. Coupling constants (J values) were given in Hz. Quatro Micro Mass; Waters (U.S) was used to record Mass experiment. Silica gel used for column chromatography was normal phase 60-120 mesh size while TLC was carried out on silica gel GF254 sheets (Merck, Germany).

Plant material

The roots of *F. strobilifera* were collected from forests of Shann Power House, Joginder Nagar, (Distt Mandi) Himachal Pradesh in October 2006. The identity of the plant material was verified by Dr. B. Naag ExBotanist Research Institute of Ayurveda, Joginder Nagar, (H.P) and Dr. H.B Singh, Head, Raw Materials Herbarium and Museum, NISCAIR, New Delhi and a voucher specimen number NISCAIR/RHMD/Consult/06/757/74

was deposited at the Herbarium of National Institute of Science Communication and Information Resources, New Delhi.

Extraction and isolation

The root (2.94 Kg) of F. strobilifera were airdried, ground and extracted with five liters of methanol for 24 h by maceration. The mark left was repeatedly extracted five times similarly, for complete extraction. The MeOH extract was evaporated in rota-vapor to yield a semisolid (1000 g), 900 g of which was suspended in five liters of water and partitioned with ten liters of DCM for five times and also partitioned with ten liters of *n*-BuOH for five times to yield 18 g and 34 g of extracted material, respectively. The DCM fraction (16 g) was column chromatographed over silica gel using petroleum ether (PE) and ethyl acetate (EtoAc), step gradient as eluents to yield compound 1 and 3. The PE and EtoAc (85:15) fractions were collected and these fractions (203.9)mg) were further chromatographed using DCM and MeOH (95:5) over silica gel to yield compound 1, (6.7) mg) and compound **3** (81.8 mg), β- sitosterol-D glucoside was eluted from PE: EtoAc (40:60) eluents. The *n*-BuOH fraction (25 g) was subjected to column chromatography on silica gel eluted with chloroform (CHCl₃): MeOH (85: 15) to yield compound 2, (58.5 mg).

Antimicrobial activity method

The minimal inhibitory concentration (MIC) of and isolated compounds were determined by the broth microdilution method according to National Committee for Clinical Laboratory Standards guidelines⁹ as well as for non-filamentous fungi in 96-well microtitre plates with MHB (Muller Hinton broth) made in-house. 96-well microtitre plates contained the antimicrobial agents in serial twofold dilutions from 136 to 0.53 µg/ml, depending on the antimicrobial agent being tested. Inocula were prepared in MHB from cultures grown on tryptic soya agar. The final concentration was 1 × 10⁵ CFU/ml. All microtitre plates were prepared in duplicate and incubated at 35°C for 24 hrs. The susceptibility of the standard

drugs vancomycin, linezolid, fluconazole and itraconazole were defined as the lowest concentration of drug that resulted in total inhibition of microbial growth. The MIC was defined as the minimum inhibitory concentration of the extract or compound that resulted in total inhibition of microbial growth.

RESULTS

The DCM fraction was column chromatographed over silica gel using PE and EtoAc to yield compound 1 and 3. The n-BuOH fraction was column chromatographed over silica gel using CHCl3: MeOH to yield compound 1 showed R_f value of 0.52 in DCM: MeOH (95:5) & showed R_f value of 0.58 in CHCl₃: MeOH (95:5) solvent system. It gave orangish vellow colour with 10% methanolic sulfuric acid and pink colour with Shinoda confirmed the presence of flavanone. Compound 2 showed R_f value of 0.61 in CHCl₃: MeOH (70:30) solvent system and showed R_f value of 0.52 in PE: EtoAc (75:25) solvent system. It gave yellow colour with 10% methanolic sulfuric acid and greenbrown colour with alcoholic FeCl₃.

Compound 3 showed R_f value of 0.50 in EtoAc: MeOH: H_2O (10: 1: 0.5) and showed R_f value of 0.54 in CHCl₃: MeOH (80:20) solvent system and gave purple colour with 10% methanolic sulfuric acid and also showed positive Molisch test with formation of violet ring.

The antimicrobial activity of compound 1 has shown the most significant activity against Gram-positive, Gram-negative bacteria and fungi.

Compound **2** showed moderate activity against Gram-positive, Gram-negative bacteria and fungi.Compounds **2** and **3** are known and identified as Genistin (**2**) and β - sitosterol-D glucoside (**3**) by comparisons of their spectral data (U.V, I.R, NMR and MS) with those reported previously ^{10, 11, 12, 23, 24}. The compound **1** was isolated for the first time from the plant and showed MIC of $17\mu g/ml$ against Staphylococcus aureus,

Staphylococcus epidermidis, Methicillin resistant Staphylococcus aureus, Pseudomonas aeruginosa, Escherichia coli and Candida albicans.

DISCUSSION

Compound 1 Compound 1 designated as 8, 3-diprenyl-5, 7, 4-trihydroxy flavanone was obtained as pale yellow crystals. Its molecular formula was established as C25H28O5 due to parent ion at m/z 408 $[M]^+$ and [M + 1] at 409. The I.R spectrum showed strong absorptions at 1630cm⁻¹ (chelated C=O group) and The characteristic 3300cm⁻¹ (OH). U.V absorption bands $[\lambda_{\text{max}}^{\text{MeOH}}]$ nm = 228 sh, 293,339 + NaoMe, 248, 285, 333; + AICI₃, 221, 316, 392; + NaoAc, 287, 297,334 + H₃BO₃] suggested a flavanone structure. That was confirmed by the detection of 1H-NMR signals characteristic at δ_H 5.40 (1H, dd, J = 12.1, 2.7 Hz, C_2 - β H), 3.20 (1H, dd, J = 17.4, 6.6Hz, C_3 - α H), 2.73 (1H, dd, J = 17.0, 2.9 Hz, C₃-βH) attributed to the flavanone C-ring protons and at $\delta_{\rm C}$ 78.1 and 41.8 in its $^{13}{\rm C}$ -NMR spectrum (Table 1).

It also indicated the presence of two-prenyl unit at δ_H 1.65, 1.54 [(each 6 H) s (CH₃ × 4)], 3.07, 3.17 (each 2H, d, J = 4.6 Hz, Ar-CH₂-CH = × 2), 5.08, 5.26 (each 1H, t, J = 6.6 Hz, CH₂-CH = × 2), three hydroxyl groups [δ_H 12.00, 6.20 and 5.32 (each 1H, s) which shifted in DMSO- d_6 to δ_H 12.10, 10.74 and 9.40]. Aromatic protons at δ_H 5.95 (1H, s) were assigned to the H-6 in A ring¹³. A characteristic ABX system at δ_H 6.78, 7.16 and 7.11 indicated the presence of a C-3, 4 disubstitution on the B-ring moiety.

Positive U.V shifts after the addition of sodium acetate and aluminium chloride indicated that the three hydroxyl groups at C₅, C₇ and C₄ were free and therefore the prenyl group in the A ring must be at C-8¹⁴. Since the ¹H-NMR spectrum (B-ring) of compound showed ABX type proton signals of the aromatic ring, the prenyl group in the B-ring must be located at C-3'. These data indicated the substitution pattern of the A ring was 5-hydroxy substituted and 8-prenylated, and that of the B ring was 4- hydroxylated and 3- prenylated and the later three signals were assigned to

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Table 1: 1 H-NMR and 13 C-NMR Spectroscopic Data of Compound 1, δ (ppm) in DMSO- d_{δ}

Position	¹ H-NMR ^a	¹³ C-NMR/HSQC DEPT ^b		COSY	HMBС ^с ³J _{сн}			
2β	5.40 dd (2.7, 12.1)	78.1	СН					
3α	3.20 dd (6.6,17.4)	41.8	СН					
3 β	2.73 dd (2.9,17.0)	41.8	СН					
4		196.6						
4a		101.7	С					
5-OH	12.10 s	161	С		C-6, C-4a, C-8			
6	5.95 s	95.2	СН		C-5, C-4a, C-8			
7-OH	10.74 s	164.2	С		C-8			
8		106.9	С					
8a		159.6	С					
1		129.1	С					
2	7.16 d (2.4)	127.7	СН		C-4			
3		127.3	С					
4 ['] -OH	9.40 s	155	С		C-1			
5	6.78 d (8.2)	114.5	СН	H-5 ['] -H-6 [']	C-4 [']			
6	7.11 dd (8.2, 2.4)	125	СН		C-4 [']			
1"	3.07 d (4.6)	21.2	CH ₂	H-1 ["] -H-2 ["]	C-2 ["] , C-3 ["] , C-8			
1""	3.17 d (4.6)	28	CH ₂		C-2", C-3"			
2"	5.08 t (6.6)	122.6	СН	H-2 ["] -H-4 ["]				
2'''	5.26 t (6.4)	122.6	СН					
3"		130.1	С					
3 ^{'''} 4 ^{''} /4'''	1.65 s	131.3 17.5	C CH₃		C-5"/C-5", C-2"/C-2", C-3"/C-3"			
5"/5""	1.54 s	25.5	CH₃		C-5"/C-5", C-2"/C-2", C-3"/C-3"			

the C-5, C-2 and C-6 protons, respectively patterns. from their chemical shifts and coupling

The $^{13}\text{C-NMR}$ spectrum of **1** showed 25 carbon atoms that were classified as four methyl carbons at δ_C 17.5/17.5, 25.5/25.5 (C-4/4" and C-5/5"), three methylene carbons at δ_C 21.2, 28 (C-1"/C-1"), 41.8 (C-3), seven quaternary carbon at δ_C 106.9, 159.6, 101.7, 130.1, 129.1, 127.3, 131.3 (C-8, C-8a, C-4a, C-3", C-1, C-3', C-3"), seven methines at δ_C 78.1, 95.2, 122.6, 127.7, 125.0, 114.5, 122.6 (C-2, C-6, C-2', C-2', C-6', C-5', C-2') and three- hydroxylated carbons at δ_C 161.0, 164.2, 155.0 (C-5, C-7, C-4'), with one carbonyl carbon at 196.6 using distortion less enhancement by polarization transfer (DEPT 90° and 135°) spectral analysis.

This was confirmed by the HMBC experiment; correlations long-range were observed between the following protons and carbons: 5-OH and 6, 4a, 8-C; 7-OH and 8-C; 4-OH and 3, 1-C. In the HMBC spectrum (Fig.1), the proton at δ_H 5.95 (1H, s, H-6) was correlated with C-5, 4a, 8, (δ_C 161, 101.7, 106.9), suggesting that one prenyl unit was located at C-8 and the other prenyl unit will be attached at C-3 of the ABX system of 'B' ring. The H-5' proton of Ring-B showed correlation with H-6' in ¹H-¹H COSY spectrum and H-1" showed correlation with H-2" in ¹H-¹H COSY spectrum. The absolute configuration at C-2 was established as S by comparing the optical rotation value with literature data of euchrestaflavanone A¹⁵. This is the first report of a flavanone (1) in a Flemingia species. The related 8, 3-diprenyl-5, 7, 4-trihydroxy flavanone (1) (Euchrestaflavanone A) is found iaponica¹⁵ Euchresta Sophora moorcroftian 16, Lupinus luteus 17, Euchresta formonsa¹⁸. indica¹⁹and Azadirachta Lespedezaflavanone B is found in Lespedeza davidii²⁰ Glycyrrhiza glabra^{21 22}.

From the above discussion the structure of **1** was concluded to be (S) - 8, 3-diprenyl-5, 7, 4-trihydroxy flavanone.

8, 3-diprenyl 5, 7, 4-trihydroxy flavanone (1): Green-brown with FeCl₃, Pink colour with Shinoda (Mg-HCl), Pale yellow crystal; m.p 155°C; U.V λ_{max} [MeOH, nm (log ϵ)]: 228 sh, 293, 339; + NaOMe, 248, 285, 333; + AlCl₃, 221,316, 392; + NaOAC, 287, 297, 334 (+ H_3BO_3); IR (KBr) cm⁻¹: 3300(OH), 1630

(C=O), 1000, 1500 (arom C=C), 1390, 1370 (CH₃). 1 H-NMR (400 MHz, DMSO- d_{6}) δ_{H} and 13 C-NMR (400 MHz, DMSO- d_{6}) δ_{C} given in **Table 1**. MS m/z (rel.int): 409 [M+H] $^{+}$ C₂₅H₂₈O₅ (100), 408 [M] $^{+}$ (20).

Genistin (2): Green-brown with FeCl₃, No Pink colour with Shinoda (Mg-HCl), Yellow colour after spray with 10% MeOH-H₂SO₄. Properties and spectra were identical to those reported earlier^{10, 11}.

β-sitosterol-D glucoside (3): Violet ring formation with Molisch reagent. Purple colour after spray with 10% MeOH-H₂SO₄, obtained as white crystals. Properties and spectra were identical to those reported earlier¹².

Compound 1 and 2 were tested for its in vitro antimicrobial activity by measuring their MIC 9 against selected test organisms (Table 2). 8, 3-diprenyl 5, 7, 4-trihydroxy flavanone (1) showed the higher activity against Grampositive, Gram-negative bacteria and fungi. Genistin (2) showed moderate activity against Gram-positive and Gram-negative bacteria, and fungi. The DCM and n-BuOH fractions of the plant also showed potent activity against these selected test organisms. Compound (3), already known compound and its activity have literature²³, been reported in the Vancomycin and Linezolid showed potent activity against Gram-positive bacteria in Gram-negative comparison to bacteria. Fluconazole and Itraconazole were active at MIC of>64 and >16 µg/ml respectively against fungi.

CONCLUSION

The present study has identified the isolation and characterization of a new flavanone for the first time from the *Flemingia* species. The antimicrobial activity of compound **(1)** has shown the most significant activity against Gram-positive, Gram-negative bacteria and fungi. Compound **(2)** showed moderate activity against Gram-positive, Gram-negative bacteria and fungi. The DCM and *n*-BuOH extracts of the plant also showed potent activity against these selected test organisms. The antimicrobial activity of these compounds and extracts has not been reported earlier.

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Table 2: Antimicrobial activities of Compound (1), (2), DCM extract and BuOH extract on selected test organism

	MIC μg/ml									
Test organism ^a	Test Compounds and Extracts ^b						Standards ^c			
	Cpd. (1)	Cpd. (2)	DCM ext.	BuOH ext.	٧	L	F	ı		
Gram-positive										
S. aureus ATCC 25923	17	34	2.1	34	1	2				
S. epidermidis ATCC 12228	17	34	2.1	34	2	1				
MRSA 562	17	34	2.1	34	1	2				
Gram-negative										
P. aeruginosa ATCC 7853	17	136	34	136	16	16				
E. coli ATCC 25922	17	146	17	136	2	2				
Non-filamentous fungi										
C. albicans ATCC 1122	17	136	17	68			>64	>16		

Test organisms^a: S. aureus; Staphylococcus aureus, S. epidermidis; Staphylococcus epidermidis MRSA; Methicillin resistant Staphylococcus aureus, Ps. aeruginosa; Pseudomonas aeruginosa, E. coli; Escherichia coli, C. albicans; Candida albicans

Test Compounds and Extracts^b: Cpd. (1): Compound (1); Cpd. (2): Compound (2); DCM ext.: Dichloromethane extract; BuOH ext.: n-Butanol extract.

Standards^c: V: Vancomycin; L: Linezolid; F: Fluconazole, I: Itraconazole

Fig 1: Compound 1: (8, 3 -diprenyl-5, 7, 4 -trihydroxy flavanone).

ACKNOWLEDGEMENTS

The authors are thankful to Dr. C. K. Katiyar, Dr. Anil Kanaujia, Dr. Steve Thomas, Mr. Rajeev Duggar, Dr. C. P. Gupta and Herbal Drug Research Division, Ranbaxy Research Laboratories, for providing the required facilities for the completion of this work.

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