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Original Research Article

Comparative assessment of plasmid DNA delivery by encapsulation within or adsorbed on poly (D, L-lactide-coglycolide) nanoparticles

Abd Almonem Doolaanea^{1,2}*, Nur 'Izzati Mansor¹, Nurul Hafizah Mohd Nor¹, Mohd Affendi Bin Mohd Shafri³, Farahidah Mohamed^{1,2,4}

¹Department of Pharmaceutical Technology, ²IKOP Sdn. Bhd., Pilot Plant Pharmaceutical Manufacturing, Faculty of Pharmacy, ³Kulliyyah of Allied Health Sciences, International Islamic University Malaysia, 25200 Kuantan, ⁴International Institute of Halal Research & Training, Kulliyyah of Engineering, International Islamic University Malaysia, PO Box 10, 50728, Kuala Lumpur, Malaysia

*For correspondence: Email: abdalmonemdoolaanea@yahoo.com; Tel: +60136238628; Fax: +609 571 6775

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Abstract

Purpose: To compare the gene delivery effectiveness of plasmid DNA (pDNA) encapsulated within poly (D,L-lactide-co-glycolide) (PLGA) nanoparticles with that adsorbed on PLGA nanoparticles.

Methods: PLGA nanoparticles were prepared using solvent-evaporation method. To encapsulate pDNA within the particles, it was first complexed with cetyltrimethylammonium bromide (CTAB) and then added to the oil phase during the synthesis. For the adsorption, PLGA nanoparticles were first modified with either CTAB or chitosan and then pDNA was adsorbed on the particle surface by electrostatic interaction.

Results: Nanoparticles encapsulating pDNA exhibited better plasmid loading and protection with significantly lower burst release (p < 0.05) compared to that of the nanoparticles with adsorbed plasmid. Cell uptake of chitosan-modified nanoparticles by murine neuroblastoma (N2a) cells was significantly (p < 0.05) higher than that of chitosan-free nanoparticles. Nanoparticles encapsulating pDNA showed higher transfection efficiency (p < 0.05) in N2a cells.

Conclusion: Encapsulation of pDNA within PLGA nanoparticles presents a potential strategy for gene delivery that is superior to pDNA adsorbed on the nanoparticle surface. In addition, encapsulation keeps the particle surface free for further modifications such as the addition of targeting ligands.

Keywords: Poly (D,L-lactide-co-glycolide), Plasmid DNA, Encapsulation, Adsorption, Cellular uptake, Gene therapy, Targeting ligands

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INTRODUCTION

Gene therapy has shown promising results in the treatment of a wide range of diseases. Poly (D,L-lactide-co-glycolide) (PLGA) is a biodegradable

and biocompatible polymer [1]. Encapsulating plasmid DNA (pDNA) in PLGA microspheres protects it from enzymatic degradation and preserves its bioactivity and its ability to exhibit sustained release [2].

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Microparticles and PLGA nanoparticles have been frequently used to deliver nucleic acids after being modified to bear positive charges. The positively-charged particles interact with the negatively-charged nucleic acids by means of electrostatic interactions. Some of the commonly used surface modifying materials are chitosan [3], cetyltrimethylammonium bromide (CTAB) [4] and poly-L-lysine (PLL) [5]. Another approach for delivering nucleic acids is by encapsulating them within PLGA microparticles or nanoparticles [6]. Every approach for encapsulation of PDNA may have its own advantages and disadvantages.

However, there are no available studies on the identification of comparative differences among the various methods of pDNA delivery. Furthermore, comparing different methods based on what is reported in the literature is arguably unreliable because different studies may use different materials obtained from different suppliers, which would vastly affect nanoparticle properties such as PLGA molecular weight and hydrophobicity, chitosan molecular weight and degree of de-acetylation, and fabrication procedure.

The present study was carried out to compare pDNA encapsulated in PLGA nanoparticles, with pDNA adsorbed onto PLGA nanoparticles, in terms of particle size, zeta potential, suspension stability, pDNA loading, release profile, cell uptake by neuron cells and the transfection efficiency.

EXPERIMENTAL

Materials

PLGA5004 [lactic to glycolic acid ratio of 50:50, intrinsic viscosity (IV) of 0.2 dl/g, and acid terminated] was obtained from PURAC (Gorinchem, Netherland). Polyvinyl alcohol (PVA; MW ~115 kDa, 88 % hydrolysed) was purchased from BDH Laboratory Supplies (Poole, England). Dichloromethane (DCM), ethyl acetate (EA), cetvltrimethylammonium bromide (CTAB). low molecular weight chitosan (LCS) of 50 - 190 kDa and medium molecular weight chitosan (MCS) of 190-310 kDa were obtained from Sigma-Aldrich (Steinheim, Germany). Glacial acetic acid (GAA) was product of Merck (Hohen-brunn, Germany). All other chemicals used in this study were of analytical grade unless otherwise stated. Plasmid DNA (pDNA)-encoding green fluorescent protein (GFP) was generously provided by Dr. Michael Bradbury (Mount Sinai School of Medicine, Department of Medicine, Manhattan, USA).

Nanoparticle synthesis

Five formulations (F1-5) of nanoparticles were synthesized by a modified emulsion solventdiffusion method [3]. The components of the formulations are presented in Table 1. Formulations F1-F4 were modified with either CTAB or chitosan in order to confer positive charge on the particles. Two of the formulations were modified with CTAB pre-dissolved in either aqueous phase (F1) or oil phase (F2). Another two formulations were modified by either LCS (F3) or MCS (F4). The oil phase was mixed with the aqueous phase (Table 1) and homogenised at about 20,500 rpm for 3 min using IKA® T10 basic homogeniser (IKA Werke GmbH and Co., Germany). The resulting emulsion was added to the dispersion medium and stirred for 2 h under vacuum. For F5, pDNA was encapsulated within the nanoparticles after complexing it with CTAB. The pDNA was complexed with CTAB at a 1:1 molar ratio, and the resultant precipitate was collected by centrifugation at 14,000 g for 5 min, and rinsed with deionised water. The supernatant was discarded and the precipitated complex was dried at 50 °C for 15 min. The dried CTAB-pDNA complex was dissolved in glacial acetic acid (GAA) before being added to the oil phase. Like F1 and F2, the oil phase was homogenised with the aqueous phase and then added to the dispersion medium. The nanoparticles were collected by centrifugation, and washed and suspended in one millilitre of deionised water. Part of each formulation was kept in suspension form in ice bath whereas the other part was lyophilised. To calculate the concentration of the nanoparticle suspensions, aliquots of about 100 µL of the nanoparticle suspensions were and supernatants centrifuged the were discarded. This was followed by complete drying of the nanoparticles. Nanoparticle suspension concentration was calculated by dividing the weight of dried nanoparticles by the volume of the aliquot used (100 μ L).

Nanoparticle characterization

To determine the particle size and zeta potential, each nanoparticle suspension was diluted with deionised water and analysed by dynamic light scattering. For determination of the particle size, Malvern Zetasizer Nano Series Nano-S was used, while for the determination of zeta potential. Nano-Z was utilised (Malvern Instruments Ltd, Malvern, Worcestershire, UK). The polydispersity index (Pdl), which is a dimensionless number indicating the width of the size distribution (values range between 0 and 1), was also obtained from the particle size measurement.

 Table 1: Composition of the nanoparticle formulations

Medium	F1	F2	F3	F4	F5
Oil phase	PLGA: 1 % w/v	PLGA: 1 % w/v	PLGA: 1 % w/v	PLGA: 1 % w/v	PLGA: 1 % w/v
	DCM:EA (1:5)	CTAB: 1 % w/v	DCM:EA (1:5)	DCM:EA (1:5)	DCM:EA:GAA (1:4:1)
		DCM:EA (1:1)			pDNA ^a : 1 % w/v
Aqueous	1 % w/v PVA	1 % w/v PVA	1 % w/v PVA	1 % w/v PVA	1 %w/v PVA
phase	1 % w/v CTAB		1 % w/v LCS	1 % w/v MCS	
Dispersion	1 % w/v CTAB	Deionised water	1 % w/v LCS	1 % w/v MCS	Deionised water

^apDNA already complexed with CTAB and dissolved in glacial acetic acid

For scanning electron microscopy (SEM), dried nanoparticles were sprinkled onto aluminium stubs pre-pasted with carbon adhesive tapes. Samples were sputter-coated with gold and viewed using Carl Zeiss Evo® 50 Scanning Electron Microscope (Oberkochen, Germany).

Preparation and characterisation of nanoparticle-pDNA (NP-pDNA) complexes

Formulations F1-F4 were used to prepare NPpDNA complexes by adsorption of pDNA on the nanoparticle surface. An amount of 10 μ g pDNA from 1 % (w/v) solution in Tris-EDTA (TE) buffer was mixed with nanoparticle suspensions. The total reaction volume was made up to 100 μ L with deionised water by gentle mixing using upand-down pipetting, and the mixture was incubated at room temperature for 1 h without shaking. The amounts of nanoparticles were varied and yielded NP:pDNA weight ratios of 5:1, 10:1, 20:1 and 50:1. The complex formation was verified by gel retardation. The NP-pDNA complexes were also subjected to zeta potential measurement.

NP-pDNA complexes and pDNA encapsulated F5 were examined for their ability to protect pDNA from serum by agarose ael electrophoresis as described by Agarwal et al [7]. Foetal bovine serum (FBS) was added to the NPpDNA complexes to a final concentration of 10 % FBS and incubated at 37 °C for 1 h or 24 h. Immediately following the incubation, 5 µL of 0.5 M EDTA was added and the samples were placed on ice to inhibit DNA nuclease. The control was naked pDNA-treated as NP-pDNA complexes. The samples and the control were loaded on agarose gel and run for 1 h, as described earlier [7].

Release profile study

NP-pDNA complexes at NP/pDNA ratio of 20:1, and F5 were subjected to release profile study in phosphate buffer (pH 7.2) at 37 °C with shaking at 120 rpm to avoid nanoparticle precipitation, using a rotary incubator shaker (Innova 4000; New Brunswick Scientific Inc., Edison, NJ). The nanoparticles were collected by centrifugation at seven pre-determined time intervals (3 h, 6 h, 12 h, 1 day, 3 days, 7 days, and 14 days). Then, the supernatant fractions were used to quantify the released pDNA by measuring the absorbance at 260 nm. Thereafter, the nanoparticles were re-suspended in equal volume of fresh buffer before being withdrawn and returned back to the incubator shaker.

Cellular uptake of nanoparticles by murine neuroblastoma (N2a) cells

Cellular uptake studies were carried out in accordance with the procedures earlier [8]. Nanoparticle formulations were prepared with inclusion of coumarin-6 as a fluorescent probe. Murine neuroblastoma (N2a) cells were seeded in a 96-well white flat-bottom microplate at a density of 2.5×10^5 cells per well and incubated for 48 h at 37 °C and 5 % CO₂. Nanoparticles added to the cells at different were concentrations: 5 µg, 10 µg and 20 µg per 100 µL. Following 1 h incubation, the cells were washed three times in ice-cold phosphate buffered saline (PBS) to remove the noninternalised nanoparticles. The cells were then visualised under an Olympus FSX100 fluorescence inverted microscope (Olympus, Tokyo, Japan). They were subsequently lysed with Promega cell culture lysis buffer (CCLR) (Promega; Madison, WI) and then lyophilised. This was followed by the solubilisation of coumarin-6 with methanol. Finally, the fluorescence intensity was measured in a fluorescence spectrophotometer (Perkin Elmer Corp., Norwalk, CT, USA) at 450 nm excitation and 505 nm emission.

Transfection efficiency

Transfection efficiency of pDNA adsorbed on the nanoparticles (F1-F4) or encapsulated within the particle (F5) was determined by monitoring the gene expression of GFP in N2a cells. Cells were seeded at a density of 25×10^5 cells per well in a 24-well flat-bottom microplate and incubated for 48 h at 37 °C and 5 % CO₂. After that, the media were replaced with fresh media containing the

nanoparticles and incubated for 48 h. The cells were visualised in a Nikon Eclipse Ti fluorescence microscope and the captured images were analysed with ImageJ (NIH, Bethesda, MD, USA) to quantify the green colour produced by GFP.

Statistical analysis

Comparisons amongst the formulations were performed using analysis of variance (ANOVA) with Tukey's post hoc test. Values of p < 0.05 were considered statistically significant. Minitab software version 16 (Minitab, State College, PA) was used to perform the statistical analysis.

RESULTS

Characteristics of nanoparticles

With water as suspending medium, the formulation with CTAB-pDNA complex (F5) exhibited smaller particle size than the chitosanmodified nanoparticles (F3 and F4) but was not significantly different from the formulation with CTAB in oil phase (F2, Figure 1). The suspending medium of 10 % FBS strongly affected the particle size and the PdI of chitosanmodified nanoparticles (but not chitosan-free nanoparticles, i.e., F1, F2 and F5). The particle size of MCS-modified nanoparticles (F4) was more affected by serum: it dropped down to resemble the non-modified nanoparticles (F5). One important feature of all formulations was that they did not exhibit any detectable aggregation in the presence of serum. This is an essential compatibility pre-requisite for *in vivo* administration.

Complexation of nanoparticles with pDNA resulted in significant changes in the particle sizes of F1, F3 and F4 but not F2. F1 complexed with pDNA underwent significant increase in both particle size and polydispersity. In contrast, F3 and F4 had significant reduction in particle size when complexed with pDNA. Interestingly, both formulations had similar particle size after complexation. Furthermore, F1, F3 and F4 revealed similar particle size when complexed with pDNA (p > 0.05).

The two formulations (F1 and F2) which contained CTAB as a cationic modifier, did not exhibit positive zeta potential (Figure 2). On the other hand, chitosan-modified nanoparticles (F3 and F4) were positively charged. F3 (with LCS)



Figure 1: Particle size (mean \pm standard error) of the nanoparticles suspended in different conditions. Numbers over the bars are the PdI \pm standard deviation (SD). Significant differences are represented by letters. Groups that do not share the same letter are significantly different (p < 0.05)



Figure 2: Zeta potential (mean \pm standard error) of the nanoparticle formulations after suspending in different conditions. Significant differences are represented by letters; groups that do not share a common letter are significantly different (p > 0.05)

showed higher zeta potential. F5, in which CTAB was already complexed with pDNA, exhibited negative zeta potential. When the particles were suspended in phosphate buffer, pH 7.2, the zeta potential of chitosan-modified nanoparticles (F1 and F2) was reversed to slightly negative. In the presence of 10 % FBS, FBS was shown to confer a negative charge to all nanoparticle formulations with remarkably lower zeta potential in F1 and F2.

The zeta potential values before and after lyophilisation were compared. Interestingly, F1 and F2 manifested zeta potential inversion after lyophilisation with no significant difference (Figure 2). On the other hand, chitosan-modified nanoparticles remained positively charged after lyophilisation, but F3 zeta potential decreased while that of F4 increased. In contrast to CTABand chitosan-modified nanoparticles, F5 remained negatively charged after lyophilisation with slight decrease in zeta potential.

Scanning Electron Microscopy (SEM) revealed that the formulations had particles with smooth surfaces and without visible pores, as shown in Figure 3. The particle sizes of F3 and F4 under

SEM (Figure 3) appeared smaller than the size obtained by dynamic light scattering (Figure 1).





Characteristics of NP-pDNA complex

CTAB- and chitosan-modified nanoparticle formulations F1-F4 were used to adsorb pDNA



Figure 4: Characterisation of NP-pDNA complexes. (A) Gel retardation assay of the nanoparticles complexed with pDNA at different NP:pDNA ratios. Lanes: (1) 1 kb ladder, (2) naked pDNA, (3-6) F1-pDNA complexes at ratios 5:1, 10:1, 20:1, 50:1, respectively; (7-10) F2-pDNA complexes at ratios 5:1, 10:1, 20:1, 50:1, respectively; (11-14) F3-pDNA complexes at ratios 5:1, 10:1, 20:1, 50:1, respectively; (19) F5, (20) CTAB-pDNA complex suspension. (B) Protection of pDNA in NP-pDNA complexes at ratio 20:1 from degradation in 10 % FBS. Lanes: (1) 1 kb ladder, (2) naked pDNA, (3) naked pDNA incubated in 10 % FBS for 1 h, (4-7) NP-pDNA ratio 20:1 complexes incubated in 10 % FBS for 24 h, (13) F5 incubated in 10 % FBS for 24 h



Figure 5: Zeta potential of the NP-pDNA complexes at different NP/pDNA ratios: F1 (●); F2 (■); F3 (▲); F4 (▼)

on the particle surface. The adsorption was evaluated by gel retardation (Figure 4) and zeta potential measurement (Figure 5) at different NP:pDNA ratios. Although F1 revealed negative zeta potential, it could complex with pDNA but not completely, even at high NP:pDNA ratio. In addition, the pattern of pDNA migration on the gel at ratios 20:1 and 50:1 appeared similar to that when CTAB-pDNA complex was subjected to gel electrophoresis (lane 20 in Figure 4A). Zeta potential measurement revealed that F1pDNA complexes at different ratios were negatively charged. F2, on the other hand, could not adsorb pDNA efficiently because free pDNA was detected in all ratios. Moreover, zeta potential showed negative charge that was similar in all ratios. In contrast, chitosan-modified nanoparticles (F3 and F4) adsorbed pDNA even at the lowest NP:pDNA ratio. In addition, zeta potential of the NP-pDNA complexes for F3 and F4 increased gradually with increase in NP:pDNA ratio.

The stabilities of the pDNA complexes were evaluated using agarose gel electrophoresis after incubating NP-pDNA complexes (ratio 20:1) in the presence of serum (10 % FBS) for 1 h and for 24 h (Figure 4B). The naked pDNA was completely degraded during the first hour of incubation. In contrast, F1-4 (lanes 4-7, Figure 4B) exhibited partial pDNA protection during 1 h incubation. F1 and F2 (lanes 4-5, Figure 4B) showed the least protection while F4 (lane 8, Figure 4B) showed the highest protection. On the other hand, F5 did not show any sign of smearing but only intense spot in the well. Overall, following 24 h incubation, only F4 and F5 conferred considerable stability on the pDNA.



Figure 6: Release profile of F5 and NP-pDNA complexes at ratio 20:1 (nanoparticles: pDNA) over 2 weeks: F1 (\diamond); F2 (\blacksquare); F3 (Δ); F4 (Δ); F5 (\Box)

Results from in vitro release study

F5 and NP-pDNA complexes (ratio 20:1) were subjected to release profile study over 2 weeks (Figure 6). Plasmid DNA adsorbed on the particle surface (F1-F4) exhibited high burst release (~50 - 80 % within 24 h). F2, which poorly adsorbed and protected the pDNA, revealed the highest burst release (79.4 % within 24 h) followed by slow release of the remaining pDNA. Interestingly, F2, F3 and F4 displayed almost similar burst release within 24 h (60.6, 62.3 and 58.1 %, respectively). On the other hand, F5 showed low burst release (only 12.9 % within 24 h) followed by sustained release, with a total of 38.9 % released over the 2 weeks.

Cellular uptake of nanoparticles by N2a cells

The results revealed that within each chitosanfree nanoparticle formulation (F1, F2 and F5), there was no significant difference in cellular uptake at all the tested concentrations (Figure 7 and Figure 8). In contrast, chitosan-modified nanoparticles (F3 and F4) showed increases in cellular uptake with increases in nanoparticle concentration.

Degree of transfection of N2a cells

Nanoparticles loaded with pDNA (encoding GFP), either by encapsulation (F5) or by adsorption (F1-F4), were used to transfect N2a cells. F2 exhibited higher transfection than F1 (Table 2); the latter almost did not show any gene expression. Chitosan-modified



Figure 7: Quantitation of the cellular uptake of coumarin-6 loaded nanoparticles by N2a cells. Significance differences are represented by letters. Groups that do not share a common letter are significantly different (p < 0.05)

nanoparticles exhibited different transfection efficiencies at different ratios; F3 and F4 showed the highest transfection at ratios 20:1 and 50:1, respectively. Amongst all formulations, F5 revealed the highest transfection. At the low NPpDNA ratio of 5:1, F2 exhibited the highest transfection but at the high ratio of 50:1, F3 showed the highest transfection.



Figure 8: Cellular uptake of coumarin 6-loaded nanoparticles by N2a as viewed by fluorescence microscopy. Arrows denote free particles present in the medium or adsorbed on the cell surface

DISCUSSION

Chitosan adsorption and the high viscosity of the aqueous phase were responsible for the increased particle size of chitosan-modified nanoparticle [9]. Serum contents could adsorb on the particle surface and may influence the particle size, polydispersity and zeta potential [11]. Schulze et al [12] reported deagglomeration while Nafee et al [13] reported agglomeration and increased particle size in the presence of serum. In contrast to the report of Nafee et al [13], it was found in this study that F3 and F4 exhibited decreased particle size in 10 % FBS.

It has been suggested that chitosan adsorption on PLGA particles followed the multi-layer pattern with different conformations [9]. Serum components (zeta potential = -16.7 ± 0.7 mV) could adsorb on the nanoparticle surface causing chitosan to desorb, and its conformation could also become more compacted. Like the effect of serum, compaction of chitosan layers due to the adsorption of pDNA was responsible for the particle size decrease of chitosan-modified nanoparticles.

Table 2: Quantitation of transfection of N2a with pDNA (encoding GFP)-loaded nanoparticles

Formulation	NP:pDNA	Transfection	
	ratio	efficiency ^a	
F1	5:1	0 ± 0 (c)	
	10:1	0.541 ± 0.103 (e)	
	20:1	0.19 ± 0.06 (h)	
	50:1	0 ± 0 (j)	
F2	5:1	10.333 ±1.155 (a)	
	10:1	4.667 ± 0.577 (e)	
	20:1	100.33 ± 30.66 (f)	
	50:1	52.67 ± 7.09 (j)	
F3	5:1	0.593 ± 0.090 (c)	
	10:1	28.0 ± 4.359 (d)	
	20:1	50.0 ± 13.23 (g)	
	50:1	142.33 ± 46.32 (i)	
F4	5:1	7.333 ± 1.155 (b)	
	10:1	25.667 ± 4.041 (d)	
	20:1	139.67 ± 18.01 (f)	
	50:1	62.67 ± 11.02 (j)	
F5	-	227.33 ± 12.50	
Untreated cells	-	0.23 ± 0.2	
Negative	-	0 ± 0	
control			
Positive	-	0.86 ± 0.14	
control			

Significant differences amongst different formulations of each of the NP:pDNA ratio were denoted by letters. Groups that do not share a letter differ significantly (p < 0.05). F5 was significantly different from all other formulations with different ratios and also significantly different from positive and negative controls. (^aTransfection efficiency was calculated as relative area of the green color in the captured images and reported as relative area × 1000)

Electrolytes cause compression of the double layers around the particles, leading to decrease in the absolute value of zeta potential [9]. This clearly effect was observed where all formulations had slightly negative zeta potential in phosphate buffer. In addition, the non-specific adsorption of serum components [12,13] was responsible for the negative zeta potential of all formulations in 10 % FBS. The inversion of zeta potential values of F1 and F2 could be explained by the re-localisation and re-orientation of CTAB. During the lyophilisation, CTAB molecules may migrate to the surface of the particles orienting their polar head to the surface in a manner similar to that of micelle conformation. The change of zeta potential of chitosan-modified nanoparticles after lyophilisation may be attributed to the change in chitosan conformation. In contrast, F5 exhibited the best stability of zeta potential before and after lyophilisation.

A gene delivery vehicle should be able to protect pDNA under physiological conditions so that it can reach the target cell. The results of this study are in line with the general observation that high molecular weight chitosan produced more stable complexes with pDNA [14]. F5 offered the best protection to pDNA in the presence of serum due to encapsulation of the pDNA inside the particles. Furthermore, pre-complexation of pDNA with CTAB, followed by encapsulation within PLGA dramatically reduced the burst release of F5. This is of advantage because it would help to avoid the loss of pDNA before it reaches the target cell.

Particle surface plays important role in cellular uptake [8]. Chitosan-modified nanoparticles exhibited higher cell uptake despite the negative zeta potential in the presence of 10 % FBS. Thus, there could be other factors involved in the enhancement effect of chitosan on cell uptake. Since nanoparticle uptake is the balance between endocytosis and exocytosis, chitosan may reduce the exocytosis aspect. The observation of free nanoparticles in the medium with F2 and F5 may be attributed to exocytosis since both formulations were chitosan-free.

The unexpected high transfection observed in F2 may be attributed to the weak interaction with pDNA. Similar results had been obtained by Köping-Höggård et al [15] who reported that polyplexes with easier dissociations yielded higher gene expression. The transfection efficiency of chitosan-modified nanoparticles is in line with the zeta potential measurements of the NP-pDNA complexes. The high stability of pDNA in F5 might be responsible for its high transfection efficiency. Thus, the transfection efficiency was a consequence of synergistic effects of different factors such as the strength of interaction between pDNA and nanoparticles, pDNA stability, zeta potential, cellular uptake and ease of pDNA dissociation inside the cell.

CONCLUSION

It is evident from the results obtained in the present study that the method of loading pDNA within PLGA nanoparticles influences the characteristics of the nanoparticles. This encapsulation loading method produces better results than the adsorption method. Encapsulation of pDNA as CTAB-pDNA complex in PLGA nanoparticles is a more effective technique for delivering pDNA for gene therapy than adsorption of pDNA on the particle surface. This encapsulation method also leaves the particle surface free for further modification that may even include addition of chitosan or targeting moieties.

DECLARATIONS

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Conflict of interest

The authors declare that no conflict of interest is associated with this work.

Contribution of authors

The authors declare that this work was done by the authors named in this article and all liabilities pertaining to claims relating to the content of this article will be borne by them. A.A. Doolaaneaa, NI Mansor and NH Mohd Nor did the overall lab work. AA Doolaaneaa wrote the manuscript. MA Mohd Shafri supervised the cell culture. F Mohamed supervised the work and reviewed the manuscript.

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